

# ATLAS OF ZEOLITE FRAMEWORK TYPES

Ch. Baerlocher  
W.M. Meier  
D.H. Olson

Fifth Revised Edition  
2001

Published on behalf of the  
Structure Commission of the International Zeolite Association

ELSEVIER



# **ATLAS OF ZEOLITE FRAMEWORK TYPES**

**Fifth Revised Edition**

**Ch. Baerlocher  
Laboratory of Crystallography  
ETH-Zentrum  
CH-8092 Zurich, Switzerland**

**W.M. Meier  
ETH-Zentrum HUT  
CH-8092 Zurich, Switzerland**

**D.H. Olson  
Department of Chemical Engineering  
University of Pennsylvania  
Philadelphia, PA 19104, USA**

Formerly: *Atlas of Zeolite Structure Types*

2001

Published on behalf of the  
Structure Commission of the International Zeolite Association by



Amsterdam - London - New York - Oxford - Paris - Shannon - Tokyo



## TABLE OF CONTENTS

Preface			1
Introduction and Explanatory Notes			3
Framework Type Data Sheets (arranged by 3-letter code in alphabetical order)			19
<b>ABW</b>	Li-A (Barrer and White)	<b>DOH</b>	Dodecasil 1H
<b>ACO</b>	ACP-1	<b>DON</b>	UTD-1F
<b>AEI</b>	AlPO-18	<b>EAB</b>	TMA-E (Aiello and Barrer)
<b>AEL</b>	AlPO-11	<b>EDI</b>	Edingtonite
<b>AEN</b>	AlPO-EN3	<b>EMT</b>	EMC-2
<b>AET</b>	AlPO-8	<b>EPI</b>	Epistilbite
<b>AFG</b>	Afghanite	<b>ERI</b>	Erionite
<b>AFI</b>	AlPO-5	<b>ESV</b>	ERS-7
<b>AFN</b>	AlPO-14	<b>EUO</b>	EU-1
<b>AFO</b>	AlPO-41	<b>FAU</b>	Faujasite
<b>AFR</b>	SAPO-40	<b>FER</b>	Ferrierite
<b>AFS</b>	MAPSO-46	<b>FRA</b>	Franzinite
<b>AFT</b>	AlPO-52	<b>GIS</b>	Gismondine
<b>AFX</b>	SAPO-56	<b>GME</b>	Gmelinite
<b>AFY</b>	CoAPO-50	<b>GON</b>	GUS-1
<b>AHT</b>	AlPO-H2	<b>GOO</b>	Goosecreekite
<b>ANA</b>	Analcime	<b>HEU</b>	Heulandite
<b>APC</b>	AlPO-C	<b>IFR</b>	ITQ-4
<b>APD</b>	AlPO-D	<b>ISV</b>	ITQ-7
<b>AST</b>	AlPO-16	<b>ITE</b>	ITQ-3
<b>ASV</b>	ASU-7	<b>JBW</b>	Na-J (Barrer and White)
<b>ATN</b>	MAPO-39	<b>KFI</b>	ZK-5
<b>ATO</b>	AlPO-31	<b>LAU</b>	Laumontite
<b>ATS</b>	MAPO-36	<b>LEV</b>	Levyne
<b>ATT</b>	AlPO-12-TAMU	<b>LIO</b>	Liottite
<b>ATV</b>	AlPO-25	<b>LOS</b>	Losod
<b>AWO</b>	AlPO-21	<b>LOV</b>	Lovdarite
<b>AWW</b>	AlPO-22	<b>LTA</b>	Linde Type A
<b>*BEA</b>	Beta	<b>LTL</b>	Linde Type L
<b>BIK</b>	Bikitaite	<b>LTN</b>	Linde Type N
<b>BOG</b>	Boggsite	<b>MAZ</b>	Mazzite
<b>BPH</b>	Beryllophosphate-H	<b>MEI</b>	ZSM-18
<b>BRE</b>	Brewsterite	<b>MEL</b>	ZSM-11
<b>CAN</b>	Cancrinite	<b>MEP</b>	Melanophlogite
<b>CAS</b>	Cesium Aluminosilicate	<b>MER</b>	Merlinoite
<b>CFI</b>	CIT-5	<b>MFI</b>	ZSM-5
<b>CGF</b>	Co-Ga-Phosphate-5	<b>MFS</b>	ZSM-57
<b>CGS</b>	Co-Ga-Phosphate-6	<b>MON</b>	Montesommaite
<b>CHA</b>	Chabazite	<b>MOR</b>	Mordenite
<b>-CHI</b>	Chiavennite	<b>MSO</b>	MCM-61
<b>-CLO</b>	Cloverite	<b>MTF</b>	MCM-35
<b>CON</b>	CIT-1	<b>MTN</b>	ZSM-39
<b>CZP</b>	Chiral Zincophosphate	<b>MTT</b>	ZSM-23
<b>DAC</b>	Dachiardite	<b>MTW</b>	ZSM-12
<b>DDR</b>	Deca-dodecasil 3R	<b>MWW</b>	MCM-22
<b>DFO</b>	DAF-1	<b>NAT</b>	Natrolite
<b>DFT</b>	DAF-2	<b>NES</b>	NU-87

<b>NON</b>	Nonasil	<b>SFE</b>	SSZ-48
<b>OFF</b>	Offretite	<b>SFF</b>	SSZ-44
<b>OSI</b>	UiO-6	<b>SGT</b>	Sigma-2
<b>OSO</b>	OSB-1	<b>SOD</b>	Sodalite
<b>-PAR</b>	Partheite	<b>STF</b>	SSZ-35
<b>PAU</b>	Paulingite	<b>STI</b>	Stilbite
<b>PHI</b>	Phillipsite	<b>STT</b>	SSZ-23
<b>RHO</b>	Rho	<b>TER</b>	Terranovaite
<b>-RON</b>	Roggianite	<b>THO</b>	Thomsonite
<b>RSN</b>	RUB-17	<b>TON</b>	Theta-1
<b>RTE</b>	RUB-3	<b>TSC</b>	Tschörtnerite
<b>RTH</b>	RUB-13	<b>VET</b>	VPI-8
<b>RUT</b>	RUB-10	<b>VFI</b>	VPI-5
<b>SAO</b>	STA-1	<b>VNI</b>	VPI-9
<b>SAS</b>	STA-6	<b>VSV</b>	VPI-7
<b>SAT</b>	STA-2	<b>WEI</b>	Weinebeneite
<b>SAV</b>	Mg-STA-7	<b>-WEN</b>	Wenkite
<b>SBE</b>	UCSB-8Co	<b>YUG</b>	Yugawaralite
<b>SBS</b>	UCSB-6GaCo	<b>ZON</b>	ZAPO-M1
<b>SBT</b>	UCSB-10GaZn		

## Appendices

A. Additional information for the data sheets of	
<b>CGF, CON, DFO, GIS, MER, MFI, MOR, RSN, SBS, SBT, VSV</b>	287
B. Rules for Framework Type Assignment	293
C. Topological Densities	295
D. Origin of 3-Letter Codes and Type Material Names	297
Isotypic Material Index	299

## PREFACE

A forerunner of the ATLAS was first published in 1970<sup>(1)</sup>. This early survey comprised 27 zeolite structures known at the time. Then the "Atlas of Zeolite Structure Types" by W.M. Meier and D.H. Olson, with 38 entries, was published by the Structure Commission of the IZA in 1978. This was followed by the much expanded 2<sup>nd</sup> edition of the ATLAS in 1988 comprising 64 entries, the updated 3<sup>rd</sup> edition in 1992 with 85 entries, and the fully revised 4<sup>th</sup> edition in 1996 with 98 entries. This 5<sup>th</sup> edition is again an updated version of the previous compilation, and the number of entries has risen significantly to 133.

The ATLAS contains an entry for each unique zeolite framework type. The term zeolite framework refers to a corner-sharing network of tetrahedrally coordinated atoms. In compliance with the changes in zeolite nomenclature recommended by IUPAC in 2001<sup>(2)</sup> the title of the Atlas had to be changed from 'Atlas of Zeolite Structure Types' to 'Atlas of Zeolite Framework Types'. This is because the term 'structure' implies both the framework and the extra-framework constituents of a zeolite and the latter are excluded in the framework description.

As a frequently quoted work of reference, the ATLAS must be updated on a regular basis to be of full use. Not only must new framework types be added, but corrections and new information on existing entries must also be disseminated. This compilation is based on information that was available to the authors by the end of 2000. We have been very grateful for preprints and unpublished data in a number of instances and this is acknowledged on the respective pages. New framework types (formerly called structure types) will be published on the world wide web (<http://www.iza-structure.org/databases/>) as they are approved.

To make it easier for the reader we have rearranged some of the data and prepared new stereo drawings. We have separated the information pertinent to the Framework Type from that corresponding to the Type Material only. In addition, we have included idealized cell parameters for the framework types, added the vertex symbols for each T-atom in a framework and have re-introduced the secondary building units (SBU's) which had been dropped in the 4<sup>th</sup> edition. The data sheets have all been generated from a zeolite database built by Ch. Baerlocher and L.B. McCusker using the contents of the 4<sup>th</sup> edition of the ATLAS as a basis. This zeolite database is also used to publish the data on the world wide web under <http://www.iza-structure.org/databases/>.

We wish to acknowledge the assistance and collaboration of many fellow scientists in our field. We are indebted to the members of the IZA Structure Commission for their extensive proof reading and for providing additional information. In particular we wish to express our appreciation to Dr. Lynne McCusker for maintaining the reference database, for her help in preparing the stereo drawings and last but not least for her highly valued advice throughout this work. It does not seem possible to assemble such a compilation absolutely free of errors, so the authors will be grateful for any additions and/or corrections for future updates.

February 2001

Ch. Baerlocher

W.M. Meier

D.H. Olson

**Structure Commission of the International Zeolite Association**

D. Akporiaye	R.F. Lobo
G. Artioli	L.B. McCusker*
Ch. Baerlocher	W. M. Meier***
W. H. Baur	W.J. Mortier
A. Chippindale	D.H. Olson***
H. Gies	J. V. Smith
R.W. Grosse-Kunstleve	M.M.J. Treacy
J. B. Higgins**	H. van Koningsveld
R.M. Kirchner	P.A. Wright

\* Chairperson

\*\* Co-Chairperson

\*\*\* Honorary Life Member

**Previous IZA Special Publications:**

- W.J. Mortier, "Compilation of Extra Framework Sites in Zeolites" (1982)
- W.M. Meier and D.H. Olson, "Atlas of Zeolite Structure Types", 2nd edition (1987)
- W.M. Meier and D.H. Olson, "Atlas of Zeolite Structure Types", 3rd edition (1992)
- W.M. Meier , D.H. Olson and Ch. Baerlocher, "Atlas of Zeolite Structure Types", 4<sup>th</sup> edition (1996)
- R. von Ballmoos and J.B. Higgins, "Collection of Simulated XRD Powder Patterns for Zeolites", 2nd edition (1990)
- M.M.J. Treacy, J.B. Higgins and R. von Ballmoos, "Collection of Simulated XRD Powder Patterns for Zeolites", 3rd edition (1996)
- H. Robson and K.P. Lillerud, "Verified Synthesis of Zeolitic Materials" (1998)

## INTRODUCTION AND EXPLANATORY NOTES

Zeolites and zeolite-like materials do not comprise an easily definable family of crystalline solids. A simple criterion for distinguishing zeolites and zeolite-like materials from denser tectosilicates is based on the framework density (FD), the number of tetrahedrally coordinated atoms (T-atoms) per 1000 Å<sup>3</sup>. Figure 1 shows the distribution of these values for porous and dense frameworks whose structures are well established<sup>(3)</sup>. A gap is clearly recognizable between zeolite-type and dense tetrahedral framework structures. The lower boundary ranges from 19 to over 21 T-atoms per 1000 Å<sup>3</sup>, depending on the type of smallest rings present. Strictly speaking the boundaries defined in Figure 1 for the framework densities apply to fully crosslinked frameworks only. Therefore, Figure 1 does not include interrupted frameworks.

For each framework type code (see below), two pages of data are included in the ATLAS. The left hand page lists the information that characterizes the framework type. This includes crystallographic data (highest possible space group, cell constants of the idealized framework), coordination sequences, vertex symbols and loop configurations. Taken together the last three pieces of information define the framework type. On the second page, data for the type *material* (i.e. a real material) can be found. Although the channel dimensionality is a property of the framework type, the channel description also includes the observed ring dimensions, and must therefore refer to the type material. For all framework types, a list of isotopic materials and their references are also given. The different entries in the data sheets are described in more detail below in the order in which they appear on these pages.

### Framework Type Page

#### **Framework type codes** (previously called structure type codes)

Following the rules set up by an IUPAC Commission on Zeolite Nomenclature in 1978<sup>(4)</sup>, designations consisting of three capital letters (in bold face type) have been used throughout. The codes are generally derived from the names of the type materials (see Appendix D) and do not include numbers and characters other than capital Roman letters. The assignment of Framework Type codes is subject to review and clearance by the IZA Structure Commission according to a decision of the IZA Council (taken at the time of the 7th IZC in Tokyo, 1986). Codes are only assigned to established structures that satisfy the rules of the IZA Structure Commission (see Appendix B for a listing of these rules). For interrupted frameworks, the 3-letter code is preceded by a hyphen. These mnemonic codes should not be confused or equated with actual materials. They only describe and define the network of the corner sharing tetrahedrally coordinated framework atoms. Thus, designations such as NaFAU are untenable. However, a material can be described using the IUPAC crystal chemical formula<sup>(2)</sup>, as |Na<sub>58</sub>| [Al<sub>58</sub>Si<sub>134</sub> O<sub>384</sub>]-FAU or |Na-| [Al-Si-O]-FAU (Note that the chemical elements must be enclosed within the appropriate brackets, i.e. | | for guest species and [ ] for the framework host). Framework types do not depend on composition, distribution of the T-atoms (Si, Al, P, Ga, Ge, B, Be, etc.), cell dimensions or symmetry.

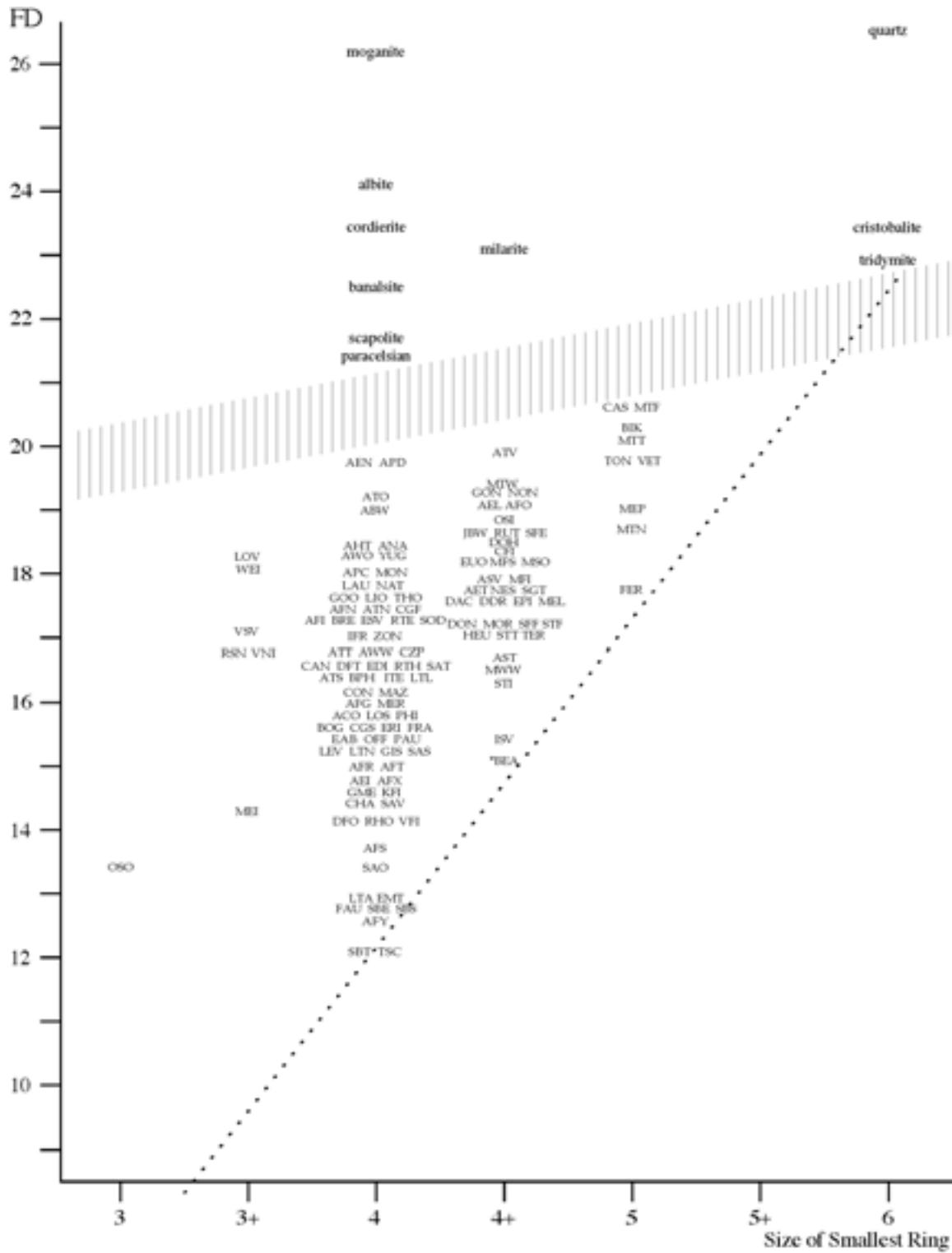


Fig. 1. Framework density vs. smallest ring in loop configuration. The + sign indicates that there are some T-positions associated with only larger rings (See Loop Configurations).

The framework types have been arranged in alphabetical order according to the framework type code, because structural criteria alone do not provide an unambiguous classification scheme. This also facilitates later insertion of new codes and allows simple indexing. The framework type code is given at the top of each page. On the left hand page this is supplemented with the maximum space group symmetry for the framework, and on the right hand page with the full name of the type material.

### Idealized cell parameters

The idealized cell parameters are obtained from a DLS-refinement<sup>(5)</sup> in the given (highest possible) symmetry for the framework type. The refinement was carried out assuming a (sometimes hypothetical) SiO<sub>2</sub> composition and with the following prescribed interatomic distances: d<sub>Si-O</sub> = 1.61Å, d<sub>O-O</sub> = 2.629Å and d<sub>Si-Si</sub> = 3.07Å using the weights of 2.0, 0.61 and 0.23, respectively.

### Coordination sequences (CS) and vertex symbols

The concept of coordination sequences was originally introduced by Brunner and Laves<sup>(6)</sup> and first applied to zeolite frameworks by Meier and Moeck<sup>(7)</sup>. In a typical zeolite framework, each T-atom is connected to N<sub>1</sub> = 4 neighboring T-atoms through oxygen bridges. These neighboring T-atoms are then linked in the same manner to N<sub>2</sub> T-atoms in the next shell. The latter are connected with N<sub>3</sub> T-atoms etc. Each T-atom is counted only once. In this way, a coordination sequence can be determined for each T-atom of the 4-connected net of T-atoms. It follows that

$$N_0 = 1 \quad N_1 \leq 4 \quad N_2 \leq 12 \quad N_3 \leq 36\dots \quad N_k \leq 4 \cdot 3^{k-1}$$

CS's are listed from N<sub>1</sub> up to N<sub>10</sub> for each topologically distinct T-atom in the framework structure along with the site multiplicity and the site symmetry (both in parenthesis).

The vertex symbol was first used in connection with zeolite-type networks by M. O'Keeffe and S.T. Hyde<sup>(8)</sup>. This symbol indicates the size of the smallest ring associated with each of the 6 angles of a tetrahedron (T-atom). The symbols for opposite pairs of angles are grouped together. For **FAU** the vertex symbol reads 4.4.4.6.6.12, indicating that one pair of opposite angles contains 4-rings, a second pair contains a 4-ring and a 6-ring, and the final pair of opposite angles contains a 6-ring and a 12-ring. It is useful for determining the smallest rings in a framework. In the case of **DOH**, for example, these are 4- and 5-rings, i.e. 4+. Sometimes more than one ring of the same size is found at a vertex.. This is indicated by a subscript like 6<sub>2</sub> or 8<sub>2</sub>.

The coordination sequence and the vertex symbol together appear to be unique for a particular framework topology, i.e. they can be used to distinguish different zeolite framework types unambiguously. In this way, isotypic frameworks can be easily recognized.

### Secondary building units (SBU's)

Zeolite frameworks can be thought to consist of finite or infinite (i.e. chain- or layer-like) component units. The finite units which have been found to occur in tetrahedral frameworks are shown in Figure 2. These secondary building units<sup>a</sup>, which contain up to 16 T-atoms, are derived

---

<sup>a</sup> The primary building units are single TO<sub>4</sub> tetrahedra.

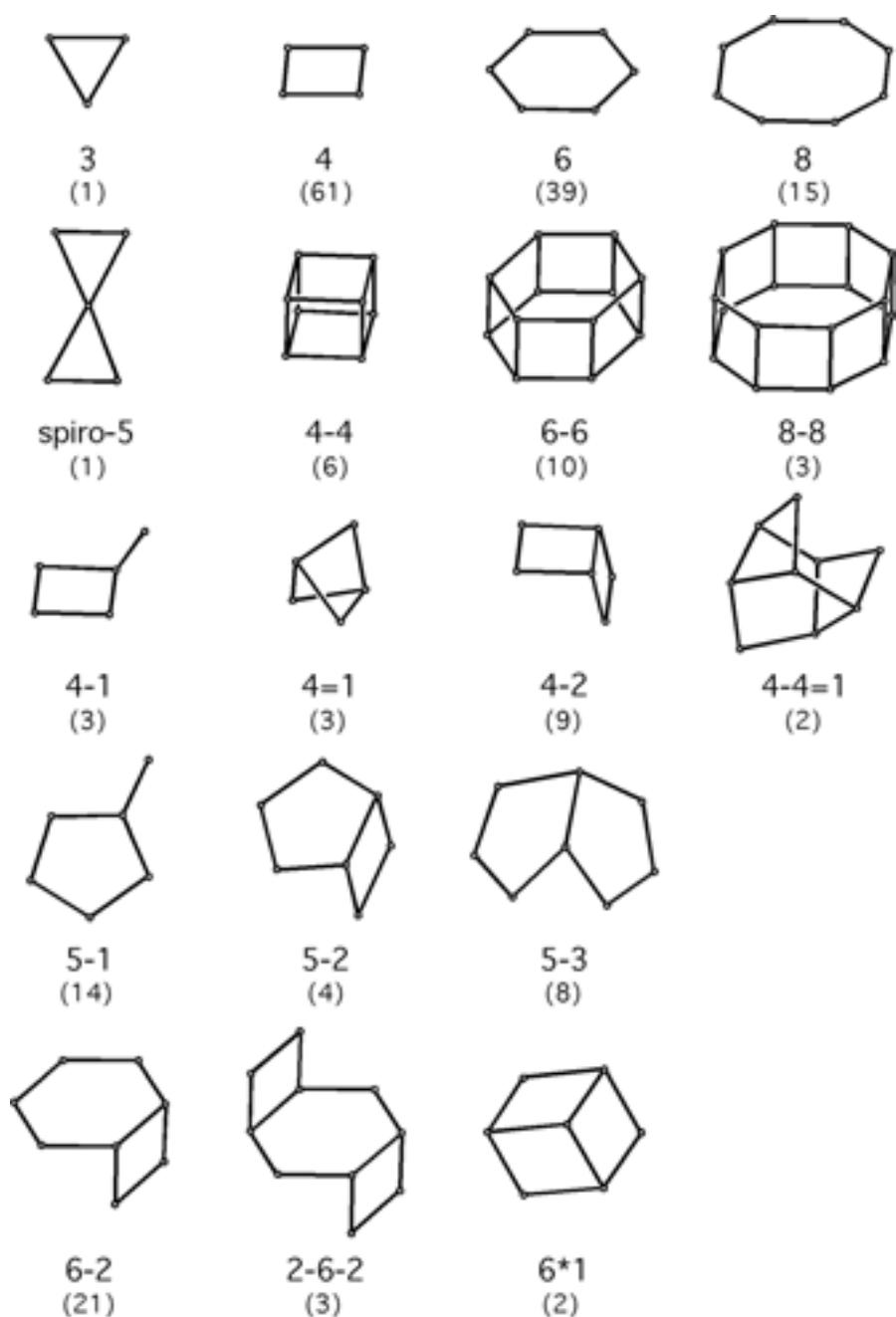


Fig. 2. Secondary building units and their symbols. Number in parenthesis = frequency of occurrence.

assuming that the entire framework is made up of one type of SBU only. It should be noted that SBU's are invariably non-chiral<sup>a</sup>. A unit cell always contains an integral number of SBU's. As far as practicable, all possible SBU's have been listed. The number of observed SBU's has increased from 16 in 1992 to 20 at present. In some instances, combinations of SBU's have been encountered. These

<sup>a</sup> This means that SBU's in the isolated state of highest symmetry are neither left- nor right-handed.

have not been listed *in extenso* because this can arbitrary. The symbols given below the drawings in Figure 2 are used in the data sheets to describe the SBU's. If more than one SBU is possible for a given framework type, all are listed. The number given in parenthesis in Figure 2 indicates the frequency of the occurrence of that SBU.

The SBU's are only theoretical topological building units and should not be considered to be or equated with species that may be in the solution/gel during the crystallization of a zeolitic material.

### **Loop configuration of T-atoms**

The loop configuration is a simple graph showing how many 3- or 4-membered rings a given T-atom is involved in. Solid lines represent T–O–T linkages whereas dotted lines indicate non-connected T–O bonds found in interrupted frameworks. Sato<sup>(9)</sup> used the term "second coordination networks". Loop configurations are likely to be of interest to spectroscopists. These data can also be used for classification purposes and for deducing rules relating to these structures which might be of predictive value<sup>(10)</sup>. Figure 3 shows all observed loop configurations and their frequency of occurrence. The information given in the loop configuration is a subset of the vertex symbol.

### **Framework description**

For all 15 framework types belonging to the so-called ABC-6-family the ABC stacking sequence is listed here. Listed are also some other structural relationships which are thought to be helpful.

### **Isotypic framework structures**

Under this heading as-synthesized materials that have the same framework type but different chemical composition or have a different laboratory code are listed. Materials obtained by post synthesis treatment (e.g. ion exchange, dealumination, etc.) are generally not included. The type material (defined on the right hand page) is given first and marked with an asterisk. Isotypic species, which have sometimes been termed "homeotypic"<sup>(11)</sup>, are very frequent and are also listed in the isotypic material index.

Zeolite-type silicates and phosphates apparently constitute two distinctive categories of microporous materials. Table 1 (which is based on the isotypes listed in the ATLAS) shows, however, that there are three, rather than two distinct groups of framework types. Apart from those associated with silicates and phosphates there is a sizable group of structure types which have been found to occur both in silicates and phosphates.

### **References**

The list of references cited is far from complete. As a general rule, references to the type materials are to the work first establishing that framework type and to subsequent work adding significant information regarding the framework topology. Thus papers on non-framework species have not been included. References to isotypes are limited to the work in which sufficient data are provided to establish the identity.

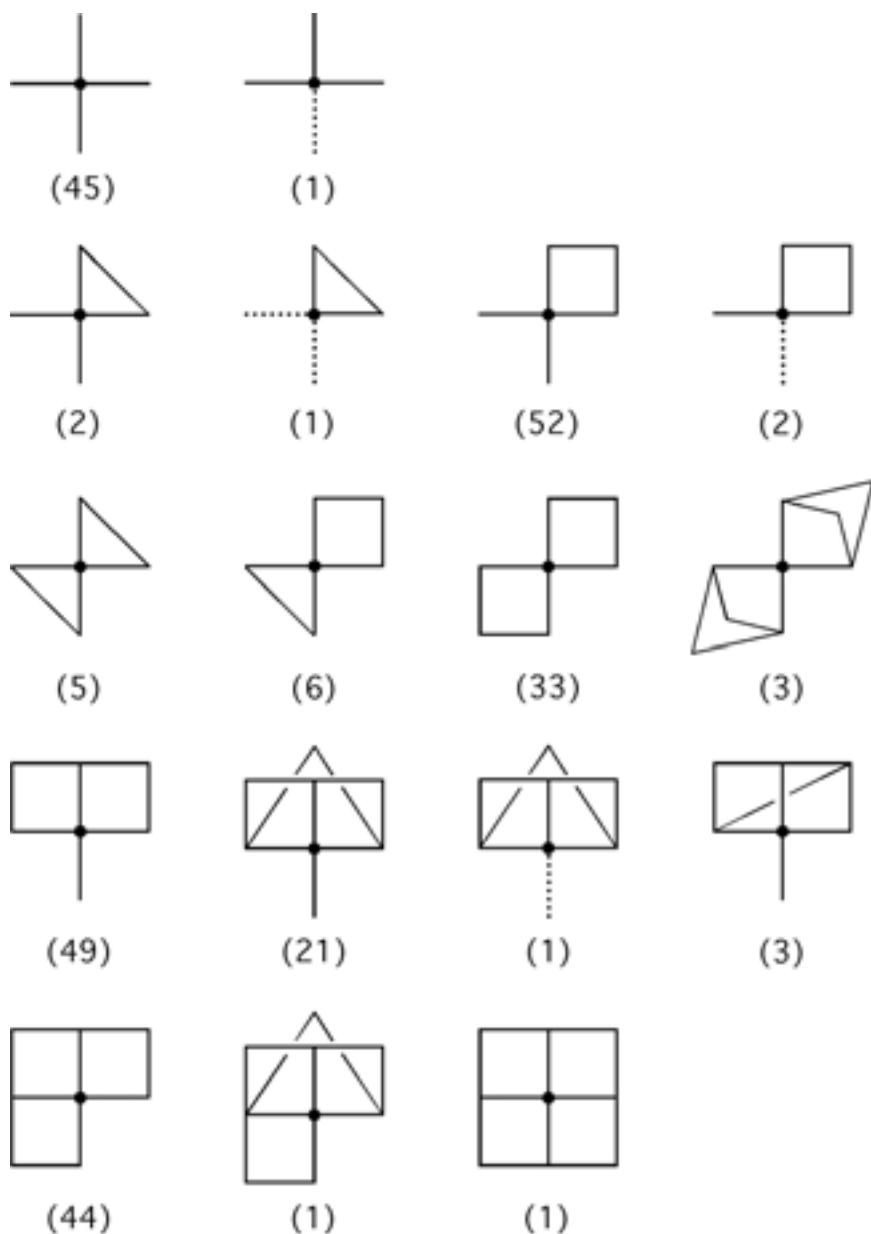


Fig. 3. Loop configurations. Number in parenthesis = frequency of occurrence.

For the 42 codes from **ABW** to **CZP**, complete references, cell constant data, space groups, site symmetries, symmetry relationships, structural diagrams, positional coordinates for all types and chemical compositions for all crystal structure determinations published up to April 2000 are to be found in: W.H. Baur and R.X. Fischer: Zeolite-Type Crystal Structures and their Chemistry<sup>(12)</sup>.

**Table 1:** Microporous zeolite-type materials

Silicates <sup>a</sup>			Both Silicates and Phosphates	Phosphates <sup>b</sup>	
AFG	IFR	OFF	ABW	ACO	SAO
ASV	ISV	OSO	AET	AEI	SAS
*BEA	ITE	-PAR	AFI	AEL	SAT
BIK	JBW	PAU	AFX	AEN	SAV
BOG	KFI	-RON	ANA	AFN	SBE
BRE	LIO	RSN	AST	AFO	SBS
CAS	LOV	RTE	BPH	AFR	SBT
CFI	LTN	RTH	CAN	AFS	VFI
-CHI	MAZ	RUT	CGS	AFT	WEI
CON	MEI	SFE	CHA	AFY	ZON
DAC	MEL	SFF	DFT	AHT	
DDR	MEP	SGT	EDI	APC	
DOH	MFI	STF	ERI	APD	
DON	MFS	STI	FAU	ATN	
EAB	MON	STT	GIS	ATO	
EMT	MOR	TER	LAU	ATS	
EPI	MSO	TON	LEV	ATT	
ESV	MTF	TSC	LOS	ATV	
EUO	MTN	VET	LTA	AWO	
FER	MTT	VNI	LTL	AWW	
FRA	MTW	VSV	MER	CGF	
GME	MWW	-WEN	PHI	-CLO	
GON	NAT	YUG	RHO	CZP	
GOO	NES		SOD	DFO	
HEU	NON		THO	OSI	

<sup>a</sup> including germanates<sup>b</sup> including arsenates

## Type Material Page

The type material is the species first used to establish the framework type. Detailed information about the material is given on this page.

### Crystal chemical data

The composition, expressed in terms of cell contents, has been idealized where necessary for simplicity. The chemical formula is given according to the new IUPAC rules<sup>(2)</sup>. The space group and cell parameters listed for each type material are those taken from the reference cited. In many instances, further refinement of the structure taking into account ordering etc. would yield a lower symmetry. It should also be noted that the space group and other crystallographic data related to the type material structure do not necessarily apply to isotypes.

In some cases, the space group setting of the type material differs from that of the framework type. In these cases, the relationship of the unit cell orientation with respect to the framework type is listed. This relationship is important when comparing the orientation of the channel direction and the viewing direction of ring drawings (which are both given for the axis orientation of the type material) with that of the framework drawing.

### Framework density (FD)

The framework density is defined as the number of T-atoms per 1000 Å<sup>3</sup>. The figures given refer to the type materials. For non-zeolitic framework structures, values of at least 20 to 21 T/1000 Å<sup>3</sup> are generally obtained, while for zeolites with fully crosslinked frameworks the observed values range from about 12.1 for structures with the largest pore volume to around 20.6. To date, FD's of less than 12 have only been encountered for the interrupted framework cloverite<sup>(13)</sup>, and for hypothetical networks<sup>(14)</sup>. The FD is obviously related to the pore volume but does not reflect the size of the pore openings. For some more flexible zeolite structure types, the FD values can vary appreciably. In these cases (e.g., gismondine) values are given for the type material and for the framework in its most expanded state. The flexibility of a framework structure is, to some extent, revealed by the possible variation in the FD. FD values may also depend on chemical composition.

### Channels

A shorthand notation has been adopted for the description of the channels in the various frameworks. Each system of equivalent channels has been characterized by

- the channel direction (relative to the axes of the type material structure),
- the number of T-atoms (in bold type) forming the rings controlling diffusion through the channels, and
- the crystallographic free diameters of the channels in Angstrom units.

The number of asterisks in the notation indicates whether the channel system is one-, two- or three-dimensional. In most cases, the smaller openings simply form windows (rather than channels)

connecting larger cavities. Interconnecting channel systems are separated by a double arrow (‘). A vertical bar (|) means that there is no direct access from one channel system to the other. The examples from Table 2 have been selected to illustrate the use of these notations. Cancrinite is characterized by a 1-dimensional system of channels parallel to [001] or *c* with circular 12-ring apertures. In offretite, the main channels are similar but they are interconnected at right angles by a 2-dimensional system of 8-ring channels, and thus form a 3-dimensional channel system. The channel system in mordenite is essentially 2-dimensional with somewhat elliptical 12-ring apertures. The 8-ring limiting diffusion in the [001] direction is an example of a highly puckered aperture. Zeolite rho is an example of a framework type containing two non-interconnecting 3-dimensional channel systems which are displaced with respect to one another (<100> means there are channels parallel to all crystallographically equivalent axes of the cubic structure, i.e., along *x*, *y* and *z*). In gismondine, the channels parallel to [100] together with those parallel to [010] give rise to a 3-dimensional channel system which can be pictured as an array of partially overlapping tubes.

**Table 2: Examples illustrating the notation for the crystallographic characterization of the Channels**

Cancrinite	[001] <b>12</b> 5.9 x 5.9*
Offretite	[001] <b>12</b> 6.7 x 6.8* ‘ ^ [001] <b>8</b> 3.6 x 4.9**
Mordenite	[001] <b>12</b> 6.5 x 7.0* ‘ {[010]} <b>8</b> 3.4 x 4.8 ‘ [001] <b>8</b> 2.6 x 5.7}*{
Zeolite Rho	<100> <b>8</b> 3.6 x 3.6***   <100> <b>8</b> 3.6 x 3.6***
Gismondine	{[100]} <b>8</b> 3.1 x 4.5 ‘ {[010]} <b>8</b> 2.8 x 4.8}***

Please note: The channel direction is given for the axis orientation of the *type material*. This orientation may be different from the orientation given in the framework drawing (see the cell relationship given under "crystal chemical data" for these cases).

A summary of the channel systems, ordered by decreasing number of T-atoms in the largest rings, is given in Table 3. The free diameter values (effective pore width) given in the channel description and on the ring drawings are based upon the atomic coordinates of the *type material* and an oxygen radius of 1.35Å. Both minimum and maximum values are given for noncircular apertures. In some instances, the corresponding interatomic distance vectors are only approximately coplanar, in other cases the plane of the ring is not normal to the direction of the channel. Close inspection of the framework and ring drawings should provide qualitative evidence of these factors. Some ring openings are defined by a very complex arrangement of oxygen atoms, so in these cases other short interatomic distances that are not listed may also be observed. It should be noted that crystallographic free diameters may depend upon the hydration state of the zeolite, particularly for the more flexible frameworks. It should also be borne in mind that effective free diameters can be affected by non-framework cations and may also be temperature dependent.

**Table 3:** Channel dimensions

<b>20-, 18- &amp;14-Ring Structures</b>		
<b>-CLO</b>	Cloverite	<100> <b>20</b> 4.0 x 13.2***   <100> <b>8</b> 3.8 x 3.8***
<b>VFI</b>	VPI-5	[001] <b>18</b> 12.7 x 12.7*
<b>AET</b>	AlPO-8	[001] <b>14</b> 7.9 x 8.7*
<b>CFI</b>	CIT-5	[010] <b>14</b> 7.2 x 7.5*
<b>DON</b>	UTD-1F	[010] <b>14</b> 8.1 x 8.2*
<b>OSO</b>	OSB-1	[001] <b>14</b> 5.4 x 7.3* ^ [001] <b>8</b> 2.8 x 3.3**
<b>12-Ring Structures</b>		
<b>AFI</b>	AlPO-5	[001] <b>12</b> 7.3 x 7.3*
<b>AFR</b>	SAPO-40	[001] <b>12</b> 6.7 x 6.9* ^ [010] <b>8</b> 3.7 x 3.7*
<b>AFS</b>	MAPSO-46	[001] <b>12</b> 7.0 x 7.0* ^ [001] <b>8</b> 4.0 x 4.0**
<b>AFY</b>	CoAPO-50	[001] <b>12</b> 6.1 x 6.1* ^ [001] <b>8</b> 4.0 x 4.3**
<b>ASV</b>	ASU-7	[001] <b>12</b> 4.1 x 4.1*
<b>ATO</b>	AlPO-31	[001] <b>12</b> 5.4 x 5.4*
<b>ATS</b>	MAPO-36	[001] <b>12</b> 6.5 x 7.5*
<b>*BEA</b>	Beta	<100> <b>12</b> 6.6 x 6.7** ^ [001] <b>12</b> 5.6 x 5.6*
<b>BOG</b>	Boggsite	[100] <b>12</b> 7.0 x 7.0* ^ [010] <b>10</b> 5.5 x 5.8*
<b>BPH</b>	Beryllophosphate-H	[001] <b>12</b> 6.3 x 6.3* ^ [001] <b>8</b> 2.7 x 3.5**
<b>CAN</b>	Cancrinite	[001] <b>12</b> 5.9 x 5.9*
<b>CON</b>	CIT-1	[001] <b>12</b> 6.4 x 7.0* ^ [100] <b>12</b> 7.0 x 5.9* ^ [010] <b>10</b> 5.1 x 4.5*
<b>CZP</b>	Chiral Zincophosphate	[001] <b>12</b> 3.8 x 7.2*
<b>DFO</b>	DAF-1	{[001] <b>12</b> 7.3 x 7.3* ^ [001] <b>8</b> 3.4 x 5.6}*** ^ {[001] <b>12</b> 6.2 x 6.2* ^ [001] <b>10</b> 5.4 x 6.4}***
<b>EMT</b>	EMC-2	[001] <b>12</b> 7.3 x 7.3* ^ [001] <b>12</b> 6.5 x 7.5**
<b>FAU</b>	Faujasite	<111> <b>12</b> 7.4 x 7.4***
<b>GME</b>	Gmelinite	[001] <b>12</b> 7.0 x 7.0* ^ [001] <b>8</b> 3.6 x 3.9**
<b>GON</b>	GUS-1	[001] <b>12</b> 5.4 x 6.8*
<b>IFR</b>	ITQ-4	[001] <b>12</b> 6.2 x 7.2*
<b>ISV</b>	ITQ-7	<100> <b>12</b> 6.1 x 6.5** ^ [001] <b>12</b> 5.9 x 6.6*
<b>LTL</b>	Linde Type L	[001] <b>12</b> 7.1 x 7.1*
<b>MAZ</b>	Mazzite	[001] <b>12</b> 7.4 x 7.4*   [001] <b>8</b> 3.1 x 3.1***

### 12-Ring Structures (cont.)

<b>MEI</b>	ZSM-18	[001] <b>12</b> 6.9 x 6.9* ↗ ^ [001] <b>7</b> 3.2 x 3.5**
<b>MOR</b>	Mordenite	[001] <b>12</b> 6.5 x 7.0* ↗ { [010] <b>8</b> 3.4 x 4.8 ↗ [001] <b>8</b> 2.6 x 5.7 } *
<b>MTW</b>	ZSM-12	[010] <b>12</b> 5.6 x 6.0*
<b>OFF</b>	Offretite	[001] <b>12</b> 6.7 x 6.8* ↗ ^ [001] <b>8</b> 3.6 x 4.9**
<b>OSI</b>	UiO-6	[001] <b>12</b> 5.2 x 6.0*
<b>-RON</b>	Roggianite	[001] <b>12</b> 4.3 x 4.3*
<b>SAO</b>	STA-1	<100> <b>12</b> 6.5 x 7.2** ↗ [001] <b>12</b> 7.0 x 7.0*
<b>SBE</b>	UCSB-8Co	<100> <b>12</b> 7.2 x 7.4** ↗ [001] <b>8</b> 4.0 x 4.0*
<b>SBS</b>	UCSB-6GaCo	[001] <b>12</b> 6.8 x 6.8* ↗ ^ [001] <b>12</b> 6.9 x 7.0**
<b>SBT</b>	UCSB-10GaZn	[001] <b>12</b> 6.4 x 7.4 * ↗ ^ [001] <b>12</b> 7.3 x 7.8**
<b>SFE</b>	SSZ-48	[010] <b>12</b> 5.4 x 7.6*
<b>VET</b>	VPI-8	[001] <b>12</b> 5.9 x 5.9*

### 10-Ring Structures

<b>AEL</b>	AlPO-11	[001] <b>10</b> 4.0 x 6.5*
<b>AFO</b>	AlPO-41	[001] <b>10</b> 4.3 x 7.0*
<b>AHT</b>	AlPO-H2	[001] <b>10</b> 3.3 x 6.8*
<b>CGF</b>	Co-Ga-Phosphate-5	{ [100] <b>10</b> 2.5 x 9.2* + <b>8</b> 2.1 x 6.7* } ↗ [001] <b>8</b> 2.4 x 4.8*
<b>CGS</b>	Co-Ga-Phosphate-6	{ [001] <b>10</b> 3.5 x 8.1 ↗ [100] <b>8</b> 2.5 x 4.6 } ***
<b>DAC</b>	Dachiardite	[010] <b>10</b> 3.4 x 5.3* ↗ [001] <b>8</b> 3.7 x 4.8*
<b>EPI</b>	Epistilbite	[100] <b>10</b> 3.4 x 5.6* ↗ [001] <b>8</b> 3.7 x 4.5*
<b>EUO</b>	EU-1	[100] <b>10</b> 4.1 x 5.4*
<b>FER</b>	Ferrierite	[001] <b>10</b> 4.2 x 5.4* ↗ [010] <b>8</b> 3.5 x 4.8*
<b>HEU</b>	Heulandite	{ [001] <b>10</b> 3.1 x 7.5* + <b>8</b> 3.6 x 4.6* } ↗ [100] <b>8</b> 2.8 x 4.7*
<b>LAU</b>	Laumontite	[100] <b>10</b> 4.0 x 5.3*
<b>MEL</b>	ZSM-11	<100> <b>10</b> 5.3 x 5.4***
<b>MFI</b>	ZSM-5	{ [100] <b>10</b> 5.1 x 5.5 ↗ [010] <b>10</b> 5.3 x 5.6 } ***
<b>MFS</b>	ZSM-57	[100] <b>10</b> 5.1 x 5.4* ↗ [010] <b>8</b> 3.3 x 4.8*
<b>MTT</b>	ZSM-23	[001] <b>10</b> 4.5 x 5.2*
<b>MWW</b>	MCM-22	^ [001] <b>10</b> 4.0 x 5.5**   ^ [001] <b>10</b> 4.1 x 5.1**
<b>NES</b>	NU-87	[100] <b>10</b> 4.8 x 5.7**
<b>-PAR</b>	Partheite	[001] <b>10</b> 3.5 x 6.9*
<b>SFF</b>	SSZ-44	[001] <b>10</b> 5.4 x 5.7*
<b>STF</b>	SSZ-35	[001] <b>10</b> 5.4 x 5.7*

### 10-Ring Structures (cont.)

<b>STI</b>	Stilbite	[100] <b>10</b> 4.7 x 5.0* ↗ [001] <b>8</b> 2.7 x 5.6*
<b>TER</b>	Terranovaite	[100] <b>10</b> 5.0 x 5.0* ↗ [001] <b>10</b> 4.1 x 7.0*
<b>TON</b>	Theta-1	[001] <b>10</b> 4.6 x 5.7*
<b>WEI</b>	Weinebeneite	[001] <b>10</b> 3.1 x 5.4* ↗ [100] <b>8</b> 3.3 x 5.0*
<b>-WEN</b>	Wenkite	<100> <b>10</b> 2.5 x 4.8** ↗ [001] <b>8</b> 2.3 x 2.7*

### 9-Ring Structures

<b>-CHI</b>	Chiavennite	[001] <b>9</b> 3.9 x 4.3*
<b>LOV</b>	Lovdarite	[010] <b>9</b> 3.2 x 4.5* ↗ [001] <b>9</b> 3.0 x 4.2* ↗ [100] <b>8</b> 3.6 x 3.7*
<b>NAT</b>	Natrolite	<100> <b>8</b> 2.6 x 3.9** ↗ [001] <b>9</b> 2.5 x 4.1*
<b>RSN</b>	RUB-17	[100] <b>9</b> 3.3 x 4.4* ↗ [001] <b>9</b> 3.1 x 4.3* ↗ [010] <b>8</b> 3.4 x 4.1*
<b>STT</b>	SSZ-23	[101] <b>9</b> 3.7 x 5.3* ↗ [001] <b>7</b> 2.4 x 3.5*
<b>VSV</b>	VPI-7	[01-1] <b>9</b> 3.3 x 4.3* ↗ [011] <b>9</b> 2.9 x 4.2* ↗ [011] <b>8</b> 2.1 x 2.7*

### 8-Ring Structures

<b>ABW</b>	Li-A	[001] <b>8</b> 3.4 x 3.8*
<b>ACO</b>	ACP-1	<100> <b>8</b> 2.8 x 3.5** ↗ [001] <b>8</b> 3.5 x 3.5*
<b>AEI</b>	AlPO-18	{[100] <b>8</b> 3.8 x 3.8 ↗ [110] <b>8</b> 3.8 x 3.8 ↗ [001] <b>8</b> 3.8 x 3.8}***
<b>AEN</b>	AlPO-EN3	[100] <b>8</b> 3.1 x 4.3* ↗ [010] <b>8</b> 2.7 x 5.0*
<b>AFN</b>	AlPO-14	[100] <b>8</b> 1.9 x 4.6* ↗ [010] <b>8</b> 2.1 x 4.9* ↗ [001] <b>8</b> 3.3 x 4.0*
<b>AFT</b>	AlPO-52	^ [001] <b>8</b> 3.2 x 3.8***
<b>AFX</b>	SAPO-56	^ [001] <b>8</b> 3.4 x 3.6***
<b>ANA</b>	Analcime	irregular distorted 8-rings
<b>APC</b>	AlPO-C	[001] <b>8</b> 3.4 x 3.7* ↗ [100] <b>8</b> 2.0 x 4.7*
<b>APD</b>	AlPO-D	[010] <b>8</b> 2.3 x 6.0* ↗ [201] <b>8</b> 1.3 x 5.8*
<b>ATN</b>	MAPO-39	[001] <b>8</b> 4.0 x 4.0*
<b>ATT</b>	AlPO-12-TAMU	[100] <b>8</b> 4.2 x 4.6* ↗ [010] <b>8</b> 3.8 x 3.8*
<b>ATV</b>	AlPO-25	[001] <b>8</b> 3.0 x 4.9*
<b>AWO</b>	AlPO-21	[100] <b>8</b> 2.7 x 5.5*
<b>AWW</b>	AlPO-22	[001] <b>8</b> 3.9 x 3.9*
<b>BIK</b>	Bikitaite	[010] <b>8</b> 2.8 x 3.7*
<b>BRE</b>	Brewsterite	[100] <b>8</b> 2.3 x 5.0* ↗ [001] <b>8</b> 2.8 x 4.1*

### 8-Ring Structures (cont.)

<b>CAS</b>	Cesium Aluminosilicate	[001] <b>8</b> 2.4 x 4.7*
<b>CHA</b>	Chabazite	^ [001] <b>8</b> 3.8 x 3.8***
<b>DDR</b>	Deca-dodecasil 3R	^ [001] <b>8</b> 3.6 x 4.4**
<b>DFT</b>	DAF-2	[001] <b>8</b> 4.1 x 4.1* ^ [100] <b>8</b> 1.8 x 4.7* ^ [010] <b>8</b> 1.8 x 4.7*
<b>EAB</b>	TMA-E	^ [001] <b>8</b> 3.7 x 5.1**
<b>EDI</b>	Edingtonite	<110> <b>8</b> 2.8 x 3.8** ^ [001] <b>8</b> 2.0 x 3.1*
<b>ERI</b>	Erionite	^ [001] <b>8</b> 3.6 x 5.1***
<b>ESV</b>	ERS-7	[010] <b>8</b> 3.5 x 4.7*
<b>GIS</b>	Gismondine	{[100] <b>8</b> 3.1 x 4.5 ^ [010] <b>8</b> 2.8 x 4.8}***
<b>GOO</b>	Goosecreekite	[100] <b>8</b> 2.8 x 4.0* ^ [010] <b>8</b> 2.7 x 4.1* ^ [001] <b>8</b> 2.9 x 4.7*
<b>ITE</b>	ITQ-3	[010] <b>8</b> 3.8 x 4.3* ^ [001] <b>8</b> 2.7 x 5.8*
<b>JBW</b>	NaJ	[001] <b>8</b> 3.7 x 4.8*
<b>KFI</b>	ZK-5	<100> <b>8</b> 3.9 x 3.9***   <100> <b>8</b> 3.9 x 3.9***
<b>LEV</b>	Levyne	^ [001] <b>8</b> 3.6 x 4.8**
<b>LTA</b>	Linde Type A	<100> <b>8</b> 4.1 x 4.1***
<b>MER</b>	Merlinoite	[100] <b>8</b> 3.1 x 3.5* ^ [010] <b>8</b> 2.7 x 3.6*
		^ [001] { <b>8</b> 3.4 x 5.1* + <b>8</b> 3.3 x 3.3*}
<b>MON</b>	Montesommaite	[100] <b>8</b> 3.2 x 4.4* ^ [001] <b>8</b> 3.6 x 3.6*
<b>MTF</b>	MCM-35	[001] <b>8</b> 3.6 x 3.9*
<b>PAU</b>	Paulingite	<100> <b>8</b> 3.6x3.6***   <100> <b>8</b> 3.6 x 3.6***
<b>PHI</b>	Phillipsite	[100] <b>8</b> 3.8 x 3.8* ^ [010] <b>8</b> 3.0 x 4.3* ^ [001] <b>8</b> 3.2 x 3.3*
<b>RHO</b>	Rho	<100> <b>8</b> 3.6 x 3.6***   <100> <b>8</b> 3.6 x 3.6***
<b>RTE</b>	RUB-3	[001] <b>8</b> 3.7 x 4.4*
<b>RTH</b>	RUB-13	[100] <b>8</b> 3.8 x 4.1* ^ [001] <b>8</b> 2.5 x 5.6*
<b>SAS</b>	STA-6	[001] <b>8</b> 4.2 x 4.2*
<b>SAT</b>	STA-2	^ [001] 3.0 x 5.5***
<b>SAV</b>	Mg-STA-7	<100> <b>8</b> 3.8 x 3.8** ^ [001] <b>8</b> 3.9 x 3.9*
<b>THO</b>	Thomsonite	[100] <b>8</b> 2.3 x 3.9* ^ [010] <b>8</b> 2.2 x 4.0* ^ [001] <b>8</b> 2.2 x 3.0*
<b>TSC</b>	Tschörtnerite	<100> <b>8</b> 4.2 x 4.2*** ^ <110> <b>8</b> 3.1 x 5.6***
<b>VNI</b>	VPI-9	{<110> <b>8</b> 3.1 x 4.0 ^ [001] <b>8</b> 3.5 x .3.6}***
<b>YUG</b>	Yugawaralite	[100] <b>8</b> 2.8 x 3.6* ^ [001] <b>8</b> 3.1 x 5.0*
<b>ZON</b>	ZAPO-M1	[100] <b>8</b> 2.5 x 5.1* ^ [010] <b>8</b> 3.7 x 4.4*

## Stability

In some cases, the type material is not stable to heating and/or removal of the template. This has been indicated where the information was available.

## Stereographic figures

Stereographic drawings of the framework and of the limiting channel windows are presented for all framework types. These drawings have been generated using the program CrystalMaker<sup>(15)</sup>. Although the depth fading helps in viewing the drawings, the use of a stereo viewer is recommended (these can be obtained from any electron microscopy supply house).

For the framework drawings, the coordinates of the idealized, highest symmetry structures were used. Only the positions of the T-atoms are shown and the T-O-T bridges are represented by straight lines. This idealization makes it easier to visualize the topology and the basic features of zeolite-like framework structures, which are often relatively complex. The unit cell has been outlined wherever possible.

In the ring drawings, all atoms are shown. Their positions are based on the crystal structure of the type material, and therefore the ring dimensions and the viewing direction are also those of the type material. As explained in the crystal chemical data section, for a few type materials, the orientation of the crystallographic axes is different from that given for the framework type. In these cases, the relationship given in the "crystal chemical data" section must be applied when comparing the viewing direction of the ring drawings with that of the framework drawing.

## Supplementary Information

### Topological densities

The coordination sequences (CS) can be used to calculate a topological density (TD). As might be expected, the CS is a periodic function. This has been established for all observed framework topologies by Grosse-Kunstleve, Brunner and Sloane<sup>(16)</sup>. They showed that the CS of any T-atom can be described exactly by a set of p quadratic equations

$$N_k = a_i k^2 + b_i k + c_i \quad \text{for } k = i + np, \quad n = 0, 1, 2, \dots \text{and} \quad i = 1, 2, 3, \dots p$$

For example, the CS of **ABW** is exactly described by a set of three quadratic equations (p=3), namely

$$N_k = 19/9 k^2 + 1/9 k + 16/9 \quad \text{for } k = 1 + 3n, \quad n=0,1,2,\dots$$

$$N_k = 19/9 k^2 - 1/9 k + 16/9 \quad \text{for } k = 2 + 3n, \quad n=0,1,2,\dots$$

$$N_k = 19/9 k^2 - 0 k + 2 \quad \text{for } k = 3 + 3n, \quad n=0,1,2,\dots$$

The number of equations p necessary to calculate all members of a particular coordination sequence varies from p=1 for **SOD** and p=42 for **FAU** to p=140,900,760 for **EUO**.

With growing index k (the shell number of the CS), the linear and constant coefficients,  $b_i$  and  $c_i$ , respectively, become less and less important. Therefore we can define the exact topological density TD as the mean of all  $a_i$  divided by the dimensionality of the topology (i.e. 3 for zeolites)

$$TD = \frac{\langle a_i \rangle}{3} = \frac{1}{3p} \sum_{i=1}^p$$

This TD is the same for all T atoms in a given structure. The values are listed for all structure types together with the  $TD_{10}$ , which was listed in the previous editions of the ATLAS, in Appendix C. There is a simple relationship between TD and  $TD_{10}$ :  $TD_{10} \sim TD * 1155$ . Since  $TD_{10}$  is an approximation, i.e. it is 'arbitrarily' terminated at  $N_{10}$ , the values obtained by this formula deviate by 11% for **-CLO** and 5% for **FAU** but the differences are generally below 3%. It seems that for very open structures, 10 steps are not sufficient for a satisfactory convergence. The correlation factor between the exact topological density TD and the framework density FD is 0.82 .

### Origin of 3-letter codes and type material names

The derivation of the 3-letter codes for the zeolite minerals is fairly obvious, because the code generally consists of the first 3 letters of the mineral name. For the synthetic materials this is sometimes more obscure. One reason for this is that numbers are frequently included to distinguish different products from a particular lab, and these numbers cannot be transferred directly to the framework code. To help the reader better understand the origin of the codes, a table that includes all framework type codes derived from synthetic type materials is given in Appendix D. In this table, the letters taken for the code are written in bold. Also, an attempt has been made to decipher the origin of the mnemonic sometimes used in the designations of these materials.

## Isotypic material index

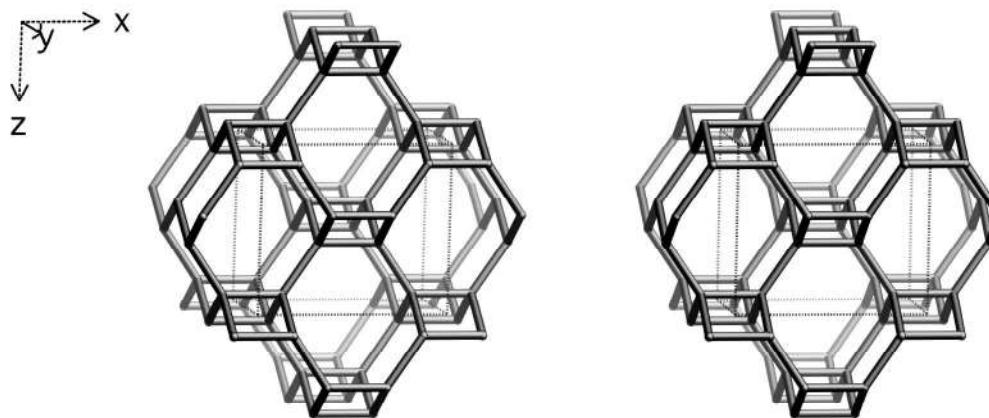
All materials are listed in alphabetical order in this index. To make the index as informative as possible, all reported materials and designations have been included in this section, provided the framework type assignment appears to be reasonably well established. Even a number of occasionally used, but discredited, names of mineral species have been included for the same reason. A full list of obsolete and discredited zeolite mineral names can be found in a report of the subcommittee on zeolites of the International Mineralogical Association<sup>(17)</sup>. Moreover, the inclusion of a synthetic material's designation in this index must not be interpreted to mean that the designation has been formally recognized or generally accepted. References are to be found on the respective framework type data sheets.

## References

- (1) W.M. Meier and D.H. Olson, *Adv. Chem. Ser.* **101**, 155 (1970).
- (2) L.B. McCusker, F. Liebau and G. Engelhardt, *Pure Appl. Chem.* **73** (2001), in press
- (3) G.O. Brunner and W.M. Meier, *Nature* **337**, 146 (1989)
- (4) R.M. Barrer, "Chemical Nomenclature and Formulation of Compositions of Synthetic and Natural Zeolites", *Pure Appl. Chem.* **51**, 1091 (1979).
- (5) Ch. Baerlocher, A. Hepp. and W.M Meier, "DLS-76, a program for the simulation of crystal structures by geometric refinement". (1978). Lab. f. Kristallographie, ETH, Zürich.
- (6) G.O. Brunner and F. Laves, *Wiss. Z. Techn. Univers. Dresden* **20**, 387 (1971) H.2.
- (7) W.M.. Meier and H.J. Moeck, *J. Solid State Chem.* **27**, 349 (1979).
- (8) M. O'Keeffe and S.T. Hyde, *Zeolites* **19**, 370 (1997)
- (9) M. Sato, Proc. 6th IZC, Reno (Butterworth, 1984) p. 851.
- (10) G.O. Brunner, *Zeolites* **13**, 88 (1993)
- (11) J.V. Smith: Geometrical and Structural Crystallography (Wiley, 1982) p. 328.
- (12) Zeolite Structure Codes ABW to CZP. Subvolume B in Landolt-Börnstein, Numerical Data and Functional Relationships in Science and Technology, New Series, Group IV: Physical Chemistry, Volume 14, Microporous and other Framework Materials with Zeolite-Type Structures, eds. W.H. Baur, R.X. Fischer, VIII & 459 pp., Springer, Berlin, 2000.
- (13) M. Estermann, L.B. McCusker, Ch. Baerlocher, A. Merrouche and H. Kessler, *Nature* **352**, 320 (1991)
- (14) W.M. Meier, Proc. 7th IZC Tokyo (Kodansha-Elsevier, 1986) p. 13.
- (15) CrystalMaker, a Crystal Structure Program for MacOS Computers. CrystalMaker Software, P.O. Box 183, Bicester, Oxfordshire, OX6 7BS, UK (<http://www.crystalmaker.co.uk>)
- (16) R.W. Grosse-Kunstleve, G.O. Brunner and N.J.A. Sloane, *Acta Crystallogr.* **A52**, 879-889 (1996)
- (17) D.S. Coombs, A. Alberti, T. Armbruster, G. Artioli, C. Colella, E. Galli, J.D. Grice, F. Liebau, J.A. Mandarino, H. Minato, E.H. Nickel, E. Passaglia, D.R. Peacor, S. Quartieri, R. Rinaldi, M. Ross, R.A. Sheppard, E. Tillmanns, G. Vezzalini, "Recommended nomenclature for zeolite minerals: Report of the subcommittee on zeolites of the international mineralogical association, commission on new minerals and mineral names", *Can. Mineral.* **35**, 1571 (1997), or, *Mineral. Mag.* **64**, 533 (1998), or, *Eur. J. Mineral.* **10**, 1037 (1998).

## **FRAMEWORK TYPE DATA SHEETS**

(arranged by 3-letter code in alphabetical order)



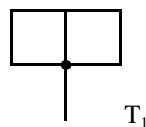
framework viewed along [010]

**Idealized cell constants:** orthorhombic, Imma,  $a = 9.9\text{\AA}$ ,  $b = 5.3\text{\AA}$ ,  $c = 8.8\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(8, m) \quad 4 \ 10 \ 21 \ 36 \ 54 \ 78 \ 106 \ 136 \ 173 \ 214$        $4\cdot6\cdot4\cdot6\cdot6\cdot8_2$

**Secondary building units:** 8 or 6 or 4

**Loop configuration of T-Atoms:**



**Isotypic framework structures:**

\*Li-A (Barrer and White)<sup>(1-3)</sup>  
 $[\text{Be-As-O}]\text{-ABW}^{(4,5)}$   
 $[\text{Be-P-O}]\text{-ABW}^{(4,6)}$   
 $[\text{Ga-Si-O}]\text{-ABW}^{(7)}$   
 $[\text{Zn-As-O}]\text{-ABW}^{(4)}$   
 $[\text{Zn-P-O}]\text{-ABW}^{(4)}$   
 $[\text{Cs-}][\text{Mg-P-O}]\text{-ABW}^{(8)}$   
 $[\text{Cs-}][\text{Al-Si-O}]\text{-ABW}^{(9,10)}$

$[\text{Cs-}][\text{Al-Ti-O}]\text{-ABW}^{(11)}$   
 $[\text{Li-}][\text{Al-Si-O}]\text{-ABW}^{(12)}$   
 $[\text{Li-}][\text{Zn-P-O}]\text{-ABW}^{(13)}$   
 $[\text{Li-}][\text{Al-Ge-O}]\text{-ABW}^{(14)}$   
 $[\text{Na-}][\text{Co-P-O}]\text{-ABW}^{(15)}$   
 $[\text{Rb-}][\text{Co-P-O}]\text{-ABW}^{(8)}$   
 $[\text{Rb-}][\text{Al-Si-O}]\text{-ABW}^{(9,10)}$   
 $[\text{Tl-}][\text{Al-Si-O}]\text{-ABW}^{(16)}$

### References:

- (1) Barrer, R.M. and White, E.A.D. *J. Chem. Soc.*, 1267-1278 (1951)
- (2) Kerr, I.S. *Z. Kristallogr.*, **139**, 186-195 (1974)
- (3) Krogh Andersen, E. and Ploug-Sørensen, G. Z. *Kristallogr.*, **176**, 67-73 (1986)
- (4) Gier, T.E. and Stucky, G.D. *Nature*, **349**, 508-510 (1991)
- (5) Harrison, W.T.A., Gier, T.E. and Stucky, G.D. *Acta Crystallogr.*, **C51**, 181-183 (1995)
- (6) Robl, C. and Gobner, V. *J. Chem. Soc., Dalton Trans.*, 1911-1912 (1993)
- (7) Newsam, J.M. *J. Phys. Chem.*, **92**, 445-452 (1988)
- (8) Rakotomanina Ralaisoa, E.L. *Ph.D. Thesis, Univ. Grenoble*, (1972)
- (9) Klaska, R. and Jarchow, O. *Naturwiss.*, **60**, 299 (1973)

**Crystal chemical data:**

$[\text{Li}^+ \cdot (\text{H}_2\text{O})_4] \cdot [\text{Al}_4\text{Si}_4\text{O}_{16}]$ -ABW  
 orthorhombic, Pna2<sub>1</sub>,  $a = 10.31\text{\AA}$ ,  $b = 8.18\text{\AA}$ ,  $c = 5.00\text{\AA}$ <sup>(2)</sup>  
 (Relationship to unit cell of Framework Type:  $a' = a$ ,  $b' = c$ ,  $c' = b$ )

**Framework density:**19 T/1000 $\text{\AA}^3$ **Channels:**

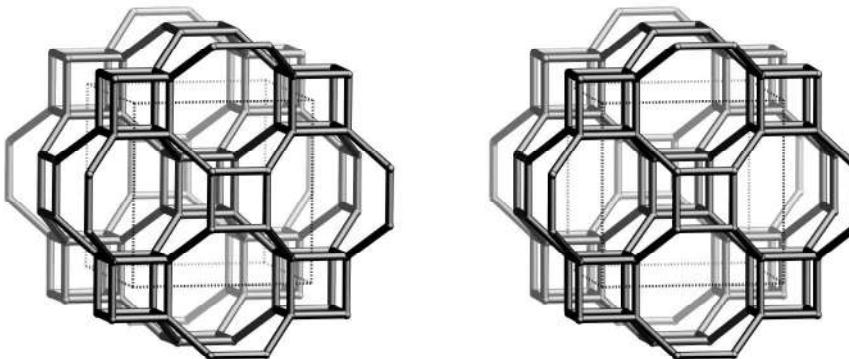
[001] 8 3.4 x 3.8\*



8-ring viewed along [001]

**References (cont.):**

- (10) Klaska, R. and Jarchow, O. Z. *Kristallogr.*, **142**, 225-238 (1975)
- (11) Gatehouse, B.M. *Acta Crystallogr.*, **C45**, 1674-1677 (1989)
- (12) Ghobarkar, H. *Cryst. Res. Technol.*, **27**, 1071-1075 (1992)
- (13) Harrison, W.T.A., Gier, T.E., Nicol, J.M. and Stucky, G.D. *J. Solid State Chem.*, **114**, 249-257 (1995)
- (14) Tripathi, A., Kim, S.J., Johnson, G.M. and Parise, J.B. *Microporous and Mesoporous Materials*, **34**, 273-279 (2000)
- (15) Chippindale, A.M., Cowley, A.R., Chen, J.S., Gao, Q. and Xu, R. *Acta Crystallogr.*, **C55**, 845-847 (1999)
- (16) Krogh Andersen, I.G., Krogh Andersen, E., Norby, P., Colella, C. and Degennaro, M. *Zeolites*, **11**, 149-154 (1991)



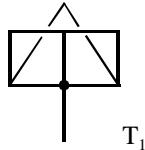
framework viewed along [001]

**Idealized cell constants:** cubic, Im $\bar{3}$ m,  $a = 9.9\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(16, 3m) \quad 4 \quad 9 \quad 19 \quad 35 \quad 52 \quad 72 \quad 100 \quad 131 \quad 163 \quad 201 \quad \dots$        $4 \cdot 8_2 \cdot 4 \cdot 8_2 \cdot 4 \cdot 8_2$

**Secondary building units:** 4-4 or 4

**Loop configuration of T-Atoms:**



**Isotypic framework structures:** \*ACP-1<sup>(1)</sup>

### References:

- (1) Feng, P.Y., Bu, X.H. and Stucky, G.D. *Nature*, **388**, 735-741 (1997)

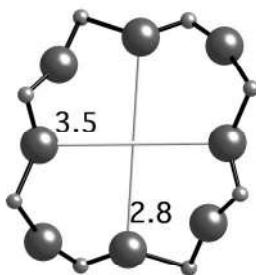
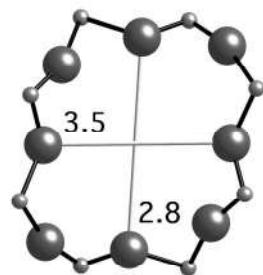
**Crystal chemical data:**  $[(C_2H_{10}N_2^{2+})_4(H_2O)_2][Al_{0.88}Co_{7.12}P_8O_{32}]$ -ACO

$C_2H_8N_2$  = ethylenediamine

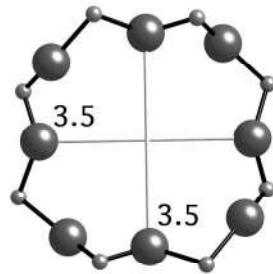
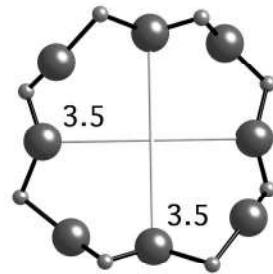
tetragonal,  $I\bar{4}2m$ ,  $a = 10.240\text{\AA}$ ,  $c = 9.652\text{\AA}$ <sup>(1)</sup>

**Framework density:** 15.8 T/1000 $\text{\AA}^3$

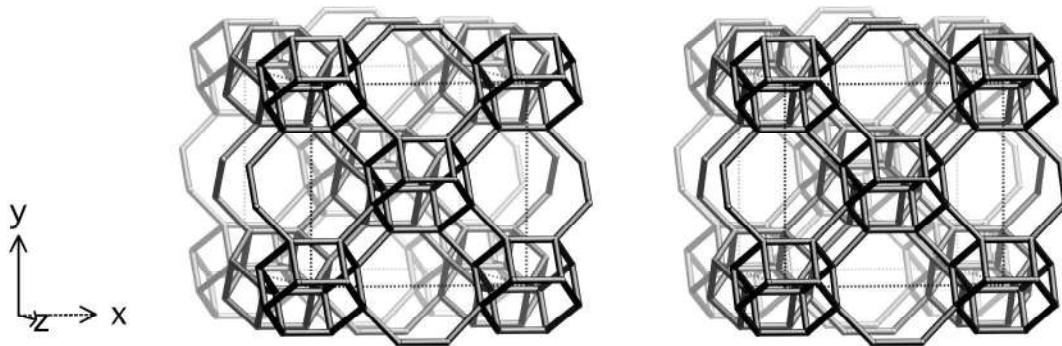
**Channels:**  $<100> \text{ 8 } 2.8 \times 3.5^{**} \leftrightarrow [001] \text{ 8 } 3.5 \times 3.5^*$



8-ring viewed along <100>



8-ring viewed along [001]



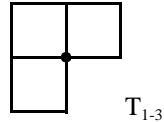
framework viewed along [001]

**Idealized cell constants:** orthorhombic, Cmcm,  $a = 13.7\text{\AA}$ ,  $b = 12.6\text{\AA}$ ,  $c = 18.5\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (16, 1)    4    9    17    29    45    64    85    111    143    177	4·4·4·8·6·8
	T <sub>2</sub> (16, 1)    4    9    17    29    45    65    88    113    143    178	4·4·4·8·6·8
	T <sub>3</sub> (16, 1)    4    9    17    29    45    65    87    113    143    176	4·4·4·8·6·8

**Secondary building units:** 6-6 or 4-2 or 6 or 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*AlPO-18<sup>(1)</sup>

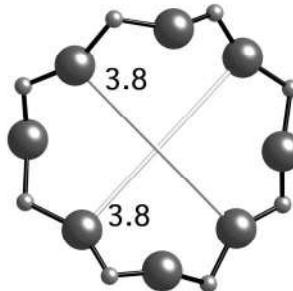
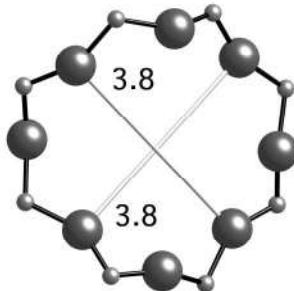
### References:

- (1) Simmen, A., McCusker, L.B., Baerlocher, Ch. and Meier, W.M. *Zeolites*, **11**, 654-661 (1991)

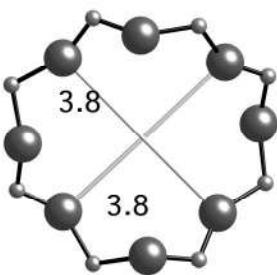
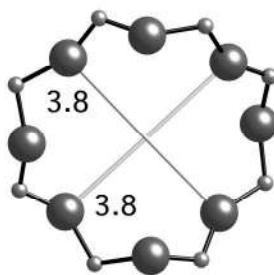
**Crystal chemical data:**  $[Al_{24}P_{24}O_{96}]$ -AEI  
monoclinic, C12/c1  
 $a = 13.711\text{\AA}$ ,  $b = 12.732\text{\AA}$ ,  $c = 18.571\text{\AA}$ ,  $\beta = 90.01^\circ$  <sup>(1)</sup>

**Framework density:** 14.8 T/1000 $\text{\AA}^3$

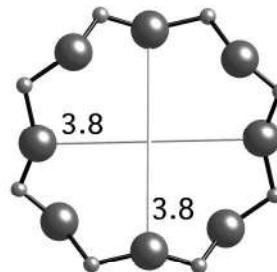
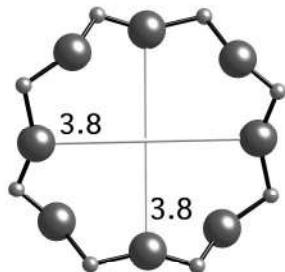
**Channels:** {[100] 8 3.8 x 3.8 ↔ [110] 8 3.8 x 3.8 ↔ [001] 8 3.8 x 3.8}\*\*\*



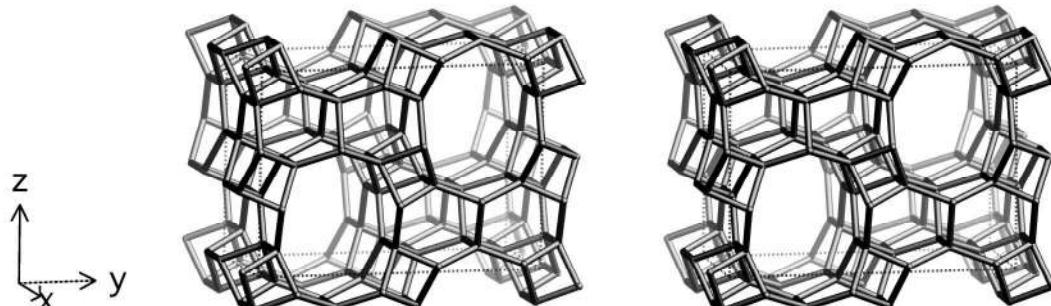
8-ring viewed along [100]



8-ring viewed along [110]



8-ring viewed along [001]



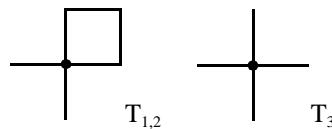
framework viewed along [100]

**Idealized cell constants:** orthorhombic, Imma,  $a = 8.3\text{\AA}$ ,  $b = 18.7\text{\AA}$ ,  $c = 13.4\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(16, 1)$	4 11 21 37 59 85 114 150 189 232	$4 \cdot 6_2 \cdot 6 \cdot 6_3 \cdot 6_2 \cdot 6_3$
	$T_2(16, 1)$	4 11 22 38 58 85 115 148 188 234	$4 \cdot 6_2 \cdot 6 \cdot 6_3 \cdot 6_2 \cdot 6_3$
	$T_3(8, m)$	4 12 24 40 59 84 115 150 186 230	$6 \cdot 6_2 \cdot 6_2 \cdot 6_2 \cdot 6_2$

**Secondary building units:** 6-2

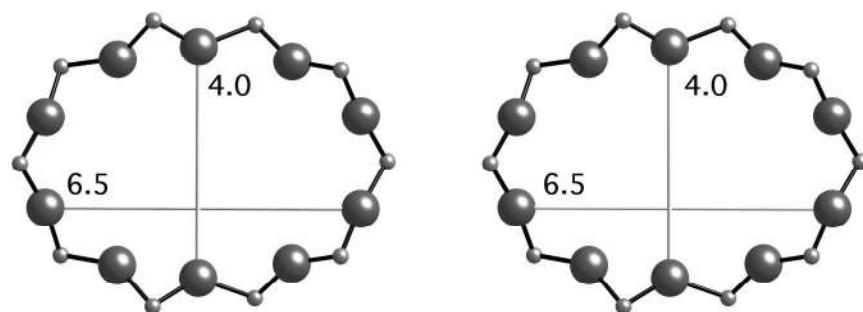
**Loop configuration of  
T-Atoms:**

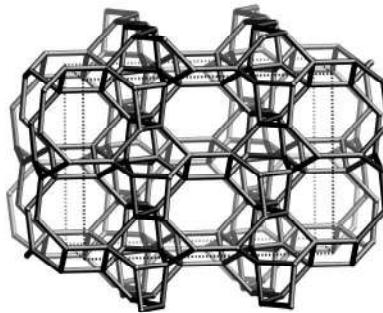
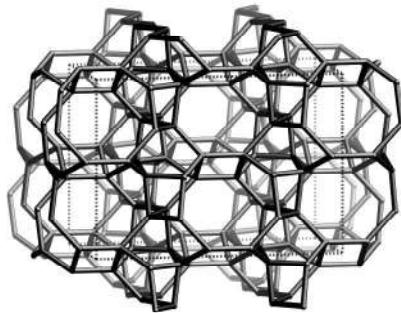
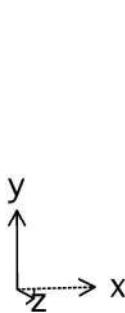


**Isotypic framework  
structures:** \*AlPO-11<sup>(1,2)</sup>  
MnAPO-11<sup>(3)</sup>  
SAPO-11 plus numerous compositional variants<sup>(4,5)</sup>

### References:

- (1) Bennett, J.M., Richardson Jr., J.W., Pluth, J.J. and Smith, J.V. *Zeolites*, **7**, 160-162 (1987)
- (2) Richardson Jr., J.W., Pluth, J.J. and Smith, J.V. *Acta Crystallogr.*, **B44**, 367-373 (1988)
- (3) Pluth, J.J., Smith, J.V. and Richardson Jr., J.W. *J. Phys. Chem.*, **92**, 2734-2738 (1988)
- (4) Flanigen, E.M., Lok, B.M., Patton, R.L. and Wilson, S.T. *Pure Appl. Chem.*, **58**, 1351-1358 (1986)
- (5) Flanigen, E.M., Lok, B.M., Patton, R.L. and Wilson, S.T. In *Proc. 7th Int. Zeolite Conf.*, (eds. Y. Murakami, A. Iijima and J.W. Ward), pp. 103-112 (1986), Kodansha, Tokyo

**Crystal chemical data:** $[Al_{20}P_{20}O_{80}]$ -AELorthorhombic, Ibm2,  $a = 13.534\text{\AA}$ ,  $b = 18.482\text{\AA}$ ,  $c = 8.370\text{\AA}$  <sup>(2)</sup>(Relationship to unit cell of Framework Type:  $a' = c$ ,  $b' = b$ ,  $c' = a$ )**Framework density:**19.1 T/1000 $\text{\AA}^3$ **Channels:**[001] **10** 4.0 x 6.5\**10-ring viewed along [001]*



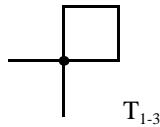
*framework viewed along [001]*

**Idealized cell constants:** orthorhombic, Cmca,  $a = 18.5\text{\AA}$ ,  $b = 13.4\text{\AA}$ ,  $c = 9.6\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(16, 1)$	4 11 24 41 60 86 123 162 199 248	$4 \cdot 6_3 \cdot 6_2 \cdot 8_2 \cdot 6_3 \cdot 8$
	$T_2(16, 1)$	4 11 22 39 64 90 119 155 201 250	$4 \cdot 6 \cdot 6 \cdot 6_2 \cdot 6_2 \cdot 6_4$
	$T_3(16, 1)$	4 11 22 38 63 90 116 155 204 250	$4 \cdot 6_3 \cdot 6_2 \cdot 6_3 \cdot 6_2 \cdot 8$

**Secondary building units:** 8 or 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*AlPO-EN3<sup>(1)</sup>

JDF-2<sup>(5)</sup>

[Ga-P-O]-AEN<sup>(2)</sup>

MSC-1<sup>(6)</sup>

AlPO-53(A)<sup>(3)</sup>

UiO-12-500<sup>(7)</sup>

AlPO-53(B)<sup>(3)</sup>

UiO-12-as<sup>(7)</sup>

CFSAPO-1A<sup>(4)</sup>

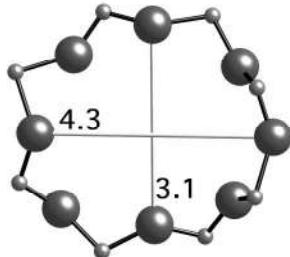
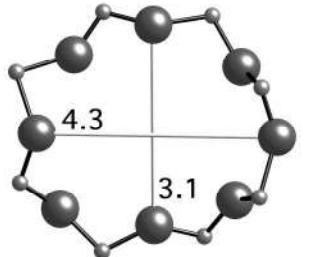
## References:

- (1) Parise, J.B. *Stud. Surf. Sci. Catal.*, **24**, 271-278 (1985)
- (2) Glasser, F.P., Howie, R.A. and Kan, Q.B. *Acta Crystallogr.*, **C50**, 848-850 (1994)
- (3) Kirchner, R.M., Grosse-Kunstleve, R.W., Pluth, J.J., Wilson, S.T., Broach, R.W. and Smith, J.V. *Microporous and Mesoporous Materials*, **39**, 319-332 (2000)
- (4) He, H. and Long, Y. *J. Incl. Phenom.*, **5**, 591-599 (1987)
- (5) Chippindale, A.M., Powell, A.V., Jones, R.H., Thomas, J.M., Cheetham, A.K., Huo, Q.S. and Xu, R.R. *Acta Crystallogr.*, **C50**, 1537-1540 (1994)
- (6) Simmen, A. *Ph.D. Thesis, ETH, Zürich, Switzerland*, (1992)
- (7) Kongshaug, K.O., Fjellvåg, H., Klewe, B. and Lillerud, K.P. *Microporous and Mesoporous Materials*, **39**, 333-339 (2000)

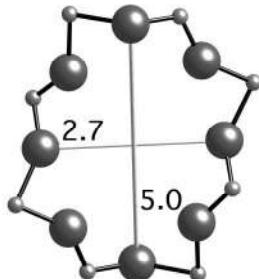
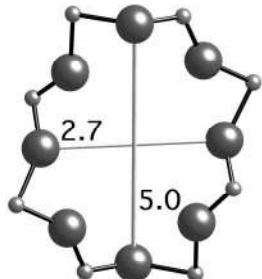
**Crystal chemical data:**  $[(C_2H_8N_2)_4(H_2O)_{16}] [Al_{24}P_{24}O_{96}]$ -AEN  
 $C_2H_8N_2$  = ethylenediamine  
orthorhombic,  $P2_12_12_1$ ,  $a = 10.292 \text{ \AA}$ ,  $b = 13.636 \text{ \AA}$ ,  $c = 17.344 \text{ \AA}$ <sup>(1)</sup>  
(Relationship to unit cell of Framework Type:  $a' = c$ ,  $b' = b$ ,  $c' = a$ )

**Framework density:**  $19.7 \text{ T}/1000\text{\AA}^3$

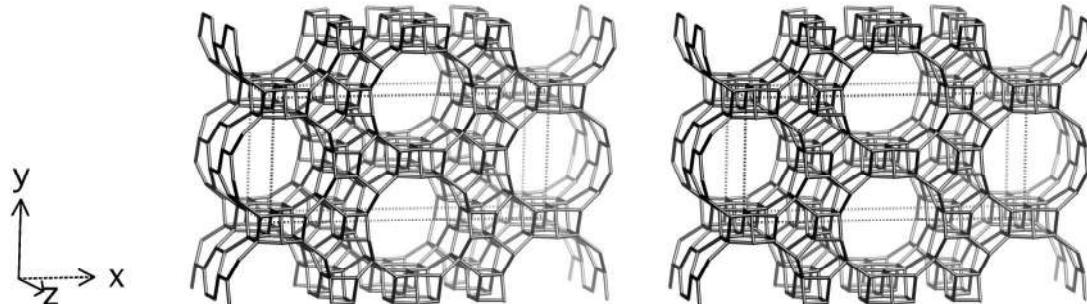
**Channels:**  $[100] \textbf{8} 3.1 \times 4.3^* \leftrightarrow [010] \textbf{8} 2.7 \times 5.0^*$



8-ring viewed along [100]



8-ring viewed along [010]



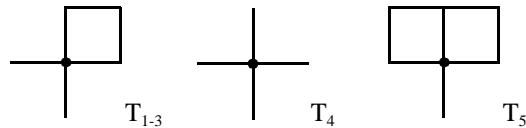
framework viewed along [001]

**Idealized cell constants:** orthorhombic, Cmcm,  $a = 32.8\text{\AA}$ ,  $b = 14.4\text{\AA}$ ,  $c = 8.4\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (16, 1)	4 11 21 35 53 78 108 140 172 208	4·6 <sub>2</sub> ·6·6 <sub>3</sub> ·6 <sub>2</sub> ·6 <sub>3</sub>
	T <sub>2</sub> (16, 1)	4 11 21 35 52 74 102 136 172 212	4·6 <sub>2</sub> ·6 <sub>2</sub> ·6 <sub>3</sub> ·6 <sub>2</sub> ·6 <sub>3</sub>
	T <sub>3</sub> (16, 1)	4 11 22 38 55 74 98 132 173 216	4·6 <sub>2</sub> ·6·6 <sub>3</sub> ·6 <sub>2</sub> ·6 <sub>3</sub>
	T <sub>4</sub> (16, 1)	4 12 23 36 52 75 103 135 172 215	6·6 <sub>2</sub> ·6 <sub>2</sub> ·6 <sub>2</sub> ·6 <sub>2</sub> ·6 <sub>2</sub>
	T <sub>5</sub> (8, m)	4 10 18 32 52 76 105 140 171 202	4·6 <sub>3</sub> ·4·6 <sub>3</sub> ·6·6 <sub>4</sub>

**Secondary building units:** 6-2

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*AlPO-8<sup>(1,2)</sup>  
MCM-37<sup>(3)</sup>

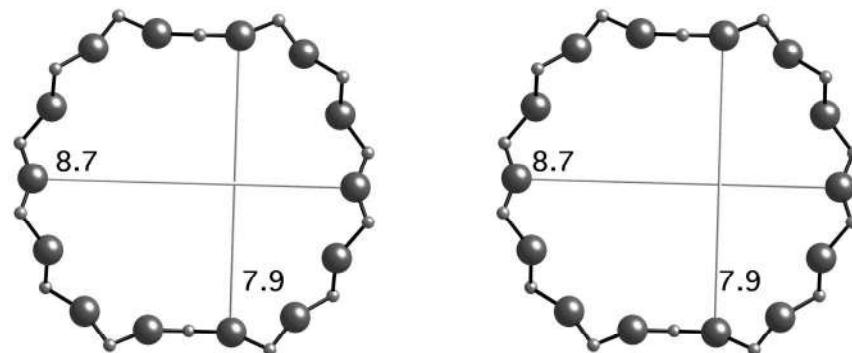
### References:

- (1) Dessau, R.M., Schlenker, J.L. and Higgins, J.B. *Zeolites*, **10**, 522-524 (1990)
- (2) Richardson Jr., J.W. and Vogt, E.T.C. *Zeolites*, **12**, 13-19 (1992)
- (3) Chu, C.T.W., Schlenker, J.L., Lutner, J.D. and Chang, C.D. U.S. Patent 5,091,073 (1992)

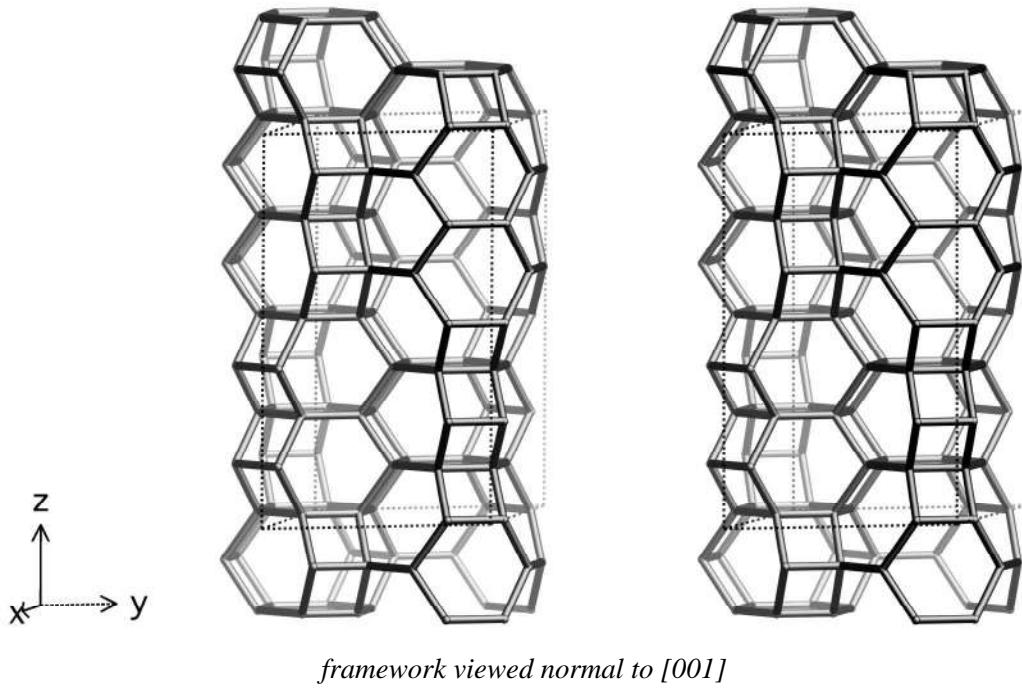
**Crystal chemical data:**  $[Al_{36}P_{36}O_{144}]$ -AET  
orthorhombic, Cmc<sub>2</sub><sub>1</sub>,  $a = 33.29\text{\AA}$ ,  $b = 14.76\text{\AA}$ ,  $c = 8.257\text{\AA}$ <sup>(1)</sup>

**Framework density:** 17.7 T/1000 $\text{\AA}^3$

**Channels:** [001] 14 7.9 x 8.7\*



14-ring viewed along [001]

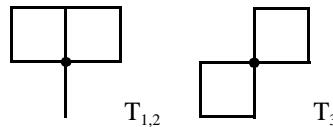


**Idealized cell constants:** hexagonal, P<sub>6</sub><sub>3</sub>/mmc,  $a = 12.5\text{\AA}$ ,  $c = 20.8\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (24, 1)    4    10    20    34    53    76    103    135    170    208	4·6·4·6·6·6
	T <sub>2</sub> (12, m)    4    10    20    34    54    78    104    134    168    210	4·6·4·6·6·6
	T <sub>3</sub> (12, 2)    4    10    20    34    54    78    104    134    168    210	4·4·6·6·6·6

**Secondary building units:** 6 or 4

**Loop configuration of  
T-Atoms:**



**Framework description:** ABABACAC sequence of 6-rings

**Isotypic framework  
structures:** \*Afghanite<sup>(1-4)</sup>

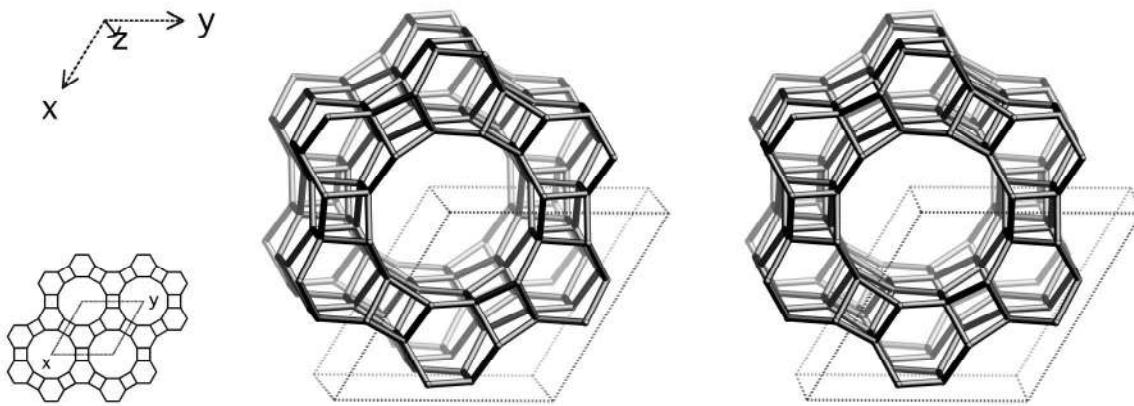
### References:

- (1) Bariand, P., Cesbron, F. and Giraud, R. *Bull. Soc. fr. Minéral. Cristallogr.*, **91**, 34-42 (1968)
- (2) Merlini, S. and Mellini, M. *Zeolite 1976, Program and Abstracts, Tucson*, (1976)
- (3) Pobedimskaya, E.A., Rastsvetaeva, R.k., Terenteva, L.E. and Saposhnikov, A.N. *Dokl. Akad. Nauk SSSR*, **320**, 882-886 (1991)
- (4) Ballirano, P., Merlini, S., Bonaccorsi, E. and Maras, A. *Eur. J. Mineral.*, **9**, 21-31 (1997)

**Crystal chemical data:**  $[\text{Ca}^{2+}_{9.8}\text{Na}^+_{22}\text{Cl}^-_2\text{SO}_4^{2-}_{5.3}\text{CO}_3^{2-}(\text{H}_2\text{O})_4][\text{Al}_{24}\text{Si}_{24}\text{O}_{96}]$ -AFG  
hexagonal , P6<sub>3</sub>mc, a = 12.761Å, c = 21.416Å<sup>(3)</sup>

**Framework density:** 15.9 T/1000Å<sup>3</sup>

**Channels:** apertures formed by 6-rings only



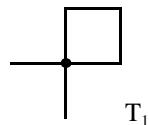
framework viewed along [001] (bottom left: projection down [001])

**Idealized cell constants:** hexagonal, P6/mcc,  $a = 13.8\text{\AA}$ ,  $c = 8.6\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(24, 1)$  4 11 21 35 53 77 105 137 172 212  $4 \cdot 6_2 \cdot 6 \cdot 6_3 \cdot 6_2 \cdot 6_3$

**Secondary building units:** 6 or 4

**Loop configuration of T-Atoms:**



**Isotypic framework structures:** \*AlPO-5<sup>(1)</sup>  
CoAPO-5<sup>(2)</sup>  
CrAPO-5<sup>(3)</sup>  
SAPO-5 and numerous compositional variants<sup>(4,5)</sup>  
SSZ-24<sup>(6)</sup>  
TPAF AlPO-5<sup>(7)</sup>

## References:

- (1) Bennett, J.M., Cohen, J.P., Flanigen, E.M., Pluth, J.J. and Smith, J.V. *ACS Sym. Ser.*, **218**, 109-118 (1983)
- (2) Chao, K.J., Sheu, S.P. and Sheu, H.S. *J. Chem. Soc., Faraday Trans.*, **88**, 2949-2954 (1992)
- (3) Radaev, S., Joswig, W. and Baur, W.H. *J. Mater. Chem.*, **6**, 1413-1418 (1996)
- (4) Flanigen, E.M., Lok, B.M., Patton, R.L. and Wilson, S.T. *Pure Appl. Chem.*, **58**, 1351-1358 (1986)
- (5) Flanigen, E.M., Lok, B.M., Patton, R.L. and Wilson, S.T. In *Proc. 7th Int. Zeolite Conf.*, (eds. Y. Murakami, A. Iijima and J.W. Ward), pp. 103-112 (1986), Kodansha, Tokyo
- (6) Bialek, R., Meier, W.M., Davis, M. and Annen, M.J. *Zeolites*, **11**, 438-442 (1991)
- (7) Qiu, S., Pang, W., Kessler, H. and Guth, J.L. *Zeolites*, **9**, 440-444 (1989)

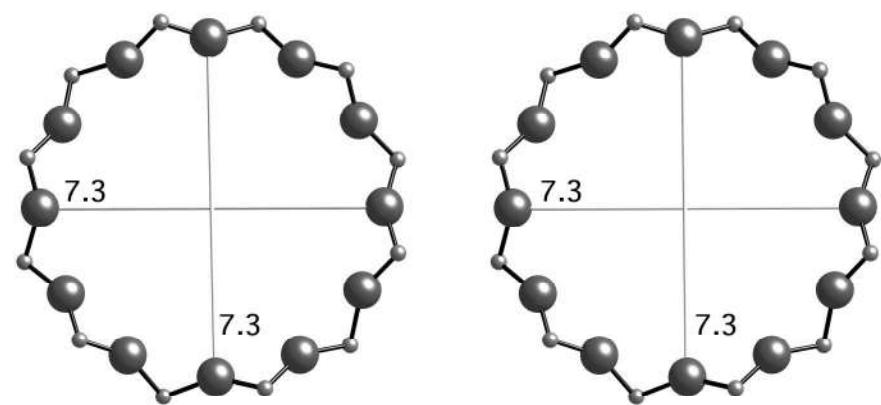
**Crystal chemical data:**  $[(C_{12}H_{28}N^+)(OH)(H_2O)_x][Al_{12}P_{12}O_{48}]$ -AFI

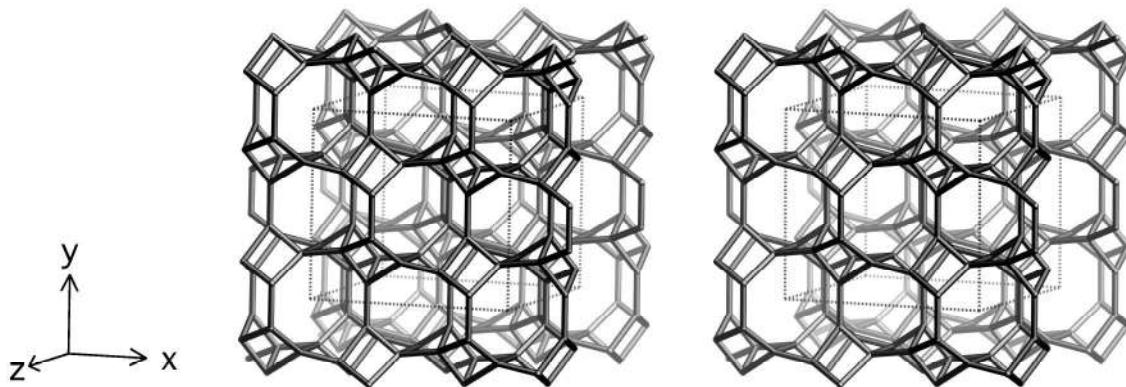
$C_{12}H_{28}N^+$  = tetrapropylammonium

hexagonal, P6cc,  $a = 13.726\text{\AA}$ ,  $c = 8.484\text{\AA}$  <sup>(1)</sup>

**Framework density:** 17.3 T/1000 $\text{\AA}^3$

**Channels:** [001] 12 7.3 x 7.3\*





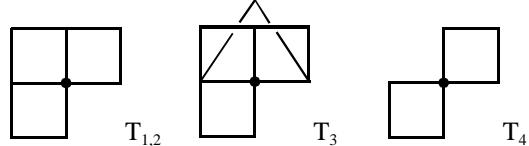
framework viewed normal to [010]

**Idealized cell constants:** monoclinic, C12/m1,  $a = 14.0\text{\AA}$ ,  $b = 13.5\text{\AA}$ ,  $c = 10.2\text{\AA}$ ,  $\beta = 107.2^\circ$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (8, 1)	4    9    19    33    51    76    98    123    162    203	4·4·4·8 <sub>2</sub> ·6·8 <sub>4</sub>
	T <sub>2</sub> (8, 1)	4    9    18    31    49    72    99    130    160    198	4·4·4·8 <sub>2</sub> ·6·6 <sub>2</sub>
	T <sub>3</sub> (8, 1)	4    9    17    30    49    75    102    125    157    202	4·6·4·8·4·8 <sub>7</sub>
	T <sub>4</sub> (8, 1)	4    10    21    35    50    71    100    132    164    198	4·4·6·8·8·8 <sub>2</sub>

**Secondary building units:** 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*AlPO-14<sup>(1)</sup>  
GaPO-14<sup>(2)</sup>

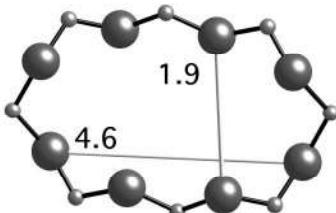
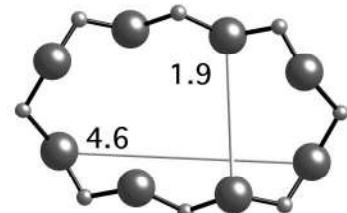
### References:

- (1) Broach, R.W., Wilson, S.T. and Kirchner, R.M. In *Proc. 12th Int. Zeolite Conf.*, (eds. M.M.J. Treacy, B.K. Marcus, M.E. Bisher and J.B. Higgins), pp. 1715-1722 (1999), MRS, Warrendale, PA
- (2) Parise, J.B. *Acta Crystallogr.*, **C42**, 670-673 (1986)

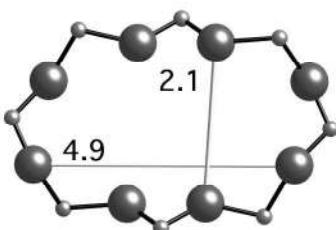
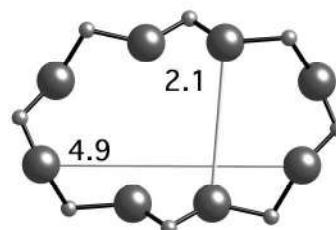
**Crystal chemical data:**  $[\text{Al}_8\text{P}_8\text{O}_{32}]\text{-AFN}$   
triclinic,  $\text{P}\bar{1}$ ,  $a = 9.704\text{\AA}$ ,  $b = 9.736\text{\AA}$ ,  $c = 10.202\text{\AA}$   
 $\alpha = 77.81^\circ$ ,  $\beta = 77.50^\circ$ ,  $\gamma = 87.69^\circ$ <sup>(1)</sup>

**Framework density:**  $17.4 \text{ T}/1000\text{\AA}^3$

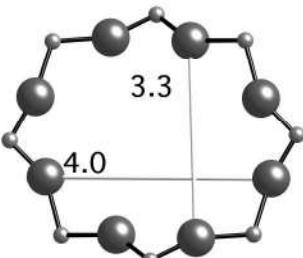
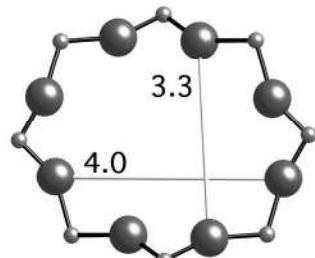
**Channels:**  $[100] \textbf{8} 1.9 \times 4.6^* \leftrightarrow [010] \textbf{8} 2.1 \times 4.9^* \leftrightarrow [001] \textbf{8} 3.3 \times 4.0^*$



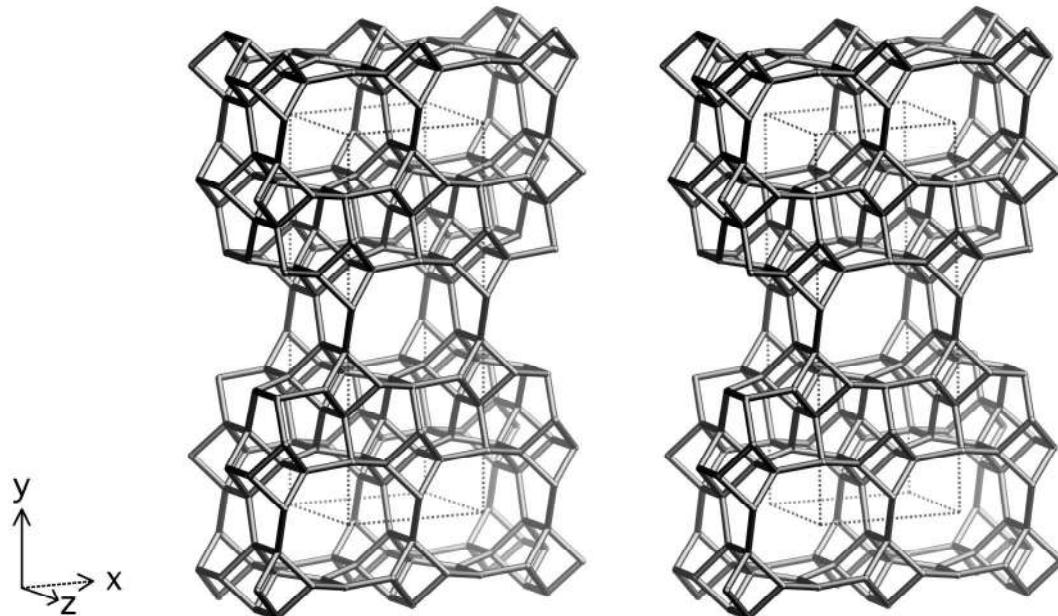
*8-ring viewed along [100]*



*8-ring viewed along [010]*



*8-ring viewed along [001]*



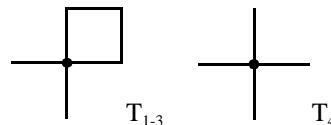
framework viewed along [001]

**Idealized cell constants:** orthorhombic, Cmcm,  $a = 9.8\text{\AA}$ ,  $b = 25.6\text{\AA}$ ,  $c = 8.3\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (16, 1)	4 11 22 38 58 85 115 149 190 235	4·6 <sub>2</sub> ·6·6 <sub>3</sub> ·6 <sub>2</sub> ·6 <sub>3</sub>
	T <sub>2</sub> (8, m)	4 11 22 41 65 88 111 145 186 231	4·6 <sub>2</sub> ·6·6 <sub>3</sub> ·6·6 <sub>3</sub>
	T <sub>3</sub> (8, m)	4 11 21 36 56 82 115 156 195 231	4·6 <sub>2</sub> ·6 <sub>2</sub> ·6 <sub>3</sub> ·6 <sub>2</sub> ·6 <sub>3</sub>
	T <sub>4</sub> (8, m)	4 12 23 37 55 82 118 155 189 232	6·6 <sub>2</sub> ·6 <sub>2</sub> ·6 <sub>2</sub> ·6 <sub>2</sub>

**Secondary building units:** 2-6-2 or 4-1

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*AlPO-41<sup>(1)</sup>

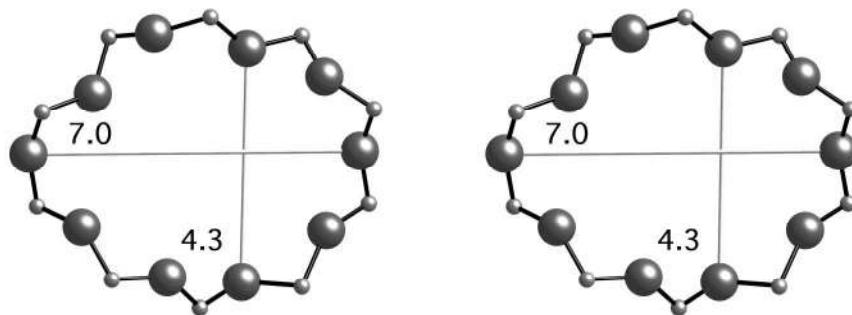
### References:

- (1) Kirchner, R.M. and Bennett, J.M. *Zeolites*, **14**, 523-528 (1994)

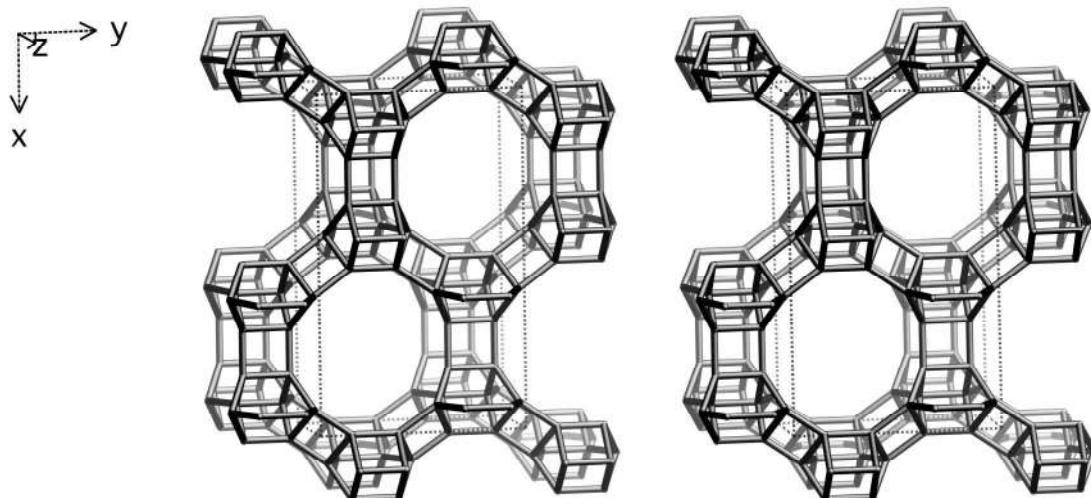
**Crystal chemical data:**  $[Al_{10}P_{10}O_{40}]$ -AFO  
monoclinic, P112<sub>1</sub>,  $a = 9.718\text{\AA}$ ,  $b = 13.792\text{\AA}$ ,  $c = 8.359\text{\AA}$ ,  $\gamma = 110.6^\circ$  <sup>(1)</sup>  
(Relationship to unit cell of Framework Type:  
 $a' = a$ ,  $b' = b \cdot \sin(\gamma)/2$ ,  $c' = c$   
or, as vectors,  $\mathbf{a}' = \mathbf{a}$ ,  $\mathbf{b}' = (\mathbf{b} - \mathbf{a})/2$ ,  $\mathbf{c}' = \mathbf{c}$ )

**Framework density:** 19.1 T/1000 $\text{\AA}^3$

**Channels:** [001] **10** 4.3 x 7.0\*



10-ring viewed along [001]



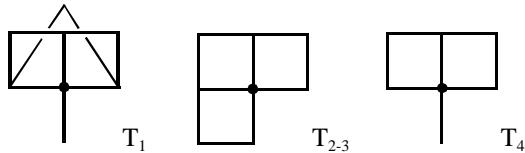
framework viewed along [001]

**Idealized cell constants:** orthorhombic, Pmmn (origin choice 2),  $a = 22.3\text{\AA}$ ,  $b = 13.6\text{\AA}$ ,  $c = 7.0\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (8, 1)    4    9    16    27    43    63    88    115    141    171	4·6·4·6 <sub>2</sub> ·4·8
	T <sub>2</sub> (8, 1)    4    9    18    30    43    64    90    111    140    181	4·4·4·8·6 <sub>3</sub> ·8
	T <sub>3</sub> (8, 1)    4    9    18    29    42    66    93    112    139    177	4·4·4·12·6·6 <sub>3</sub>
	T <sub>4</sub> (8, 1)    4    10    17    28    47    65    86    117    144    169	4·6·4·6·6·12

**Secondary building units:** 6-2 or 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**  
 \*SAPO-40<sup>(1-3)</sup>  
 AlPO-40<sup>(4)</sup>  
 CoAPSO-40<sup>(5)</sup>  
 ZnAPSO-40<sup>(5)</sup>

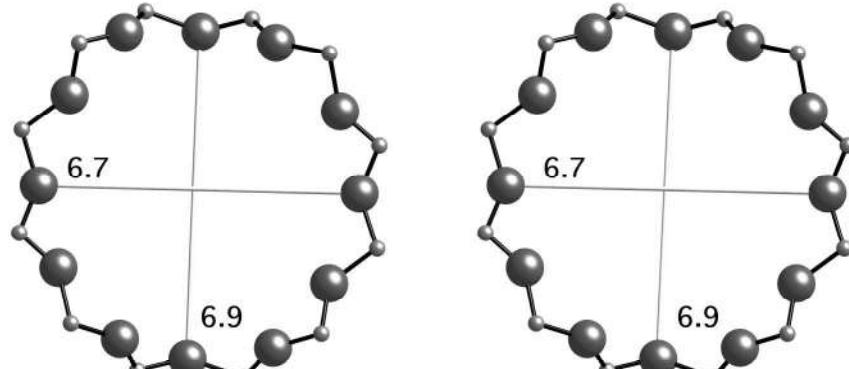
### References:

- (1) Estermann, M.A., McCusker, L.B. and Baerlocher, Ch. *J. Appl. Crystallogr.*, **25**, 539-543 (1992)
- (2) Dumont, N., Gabelica, Z., Derouane, E.G. and McCusker, L.B. *Microporous Materials*, **1**, 149-160 (1993)
- (3) McCusker, L.B. and Baerlocher, Ch. *Microporous Materials*, **6**, 51-54 (1996)
- (4) Ramaswamy, V., McCusker, L.B. and Baerlocher, Ch. *Microporous and Mesoporous Materials*, **31**, 1-8 (1999)
- (5) Lourenco, J.P., Ribeiro, M.F., Borges, C., Rocha, J., Onida, B., Garrone, E. and Gabelica, Z. , **38**, 267-278 (2000)

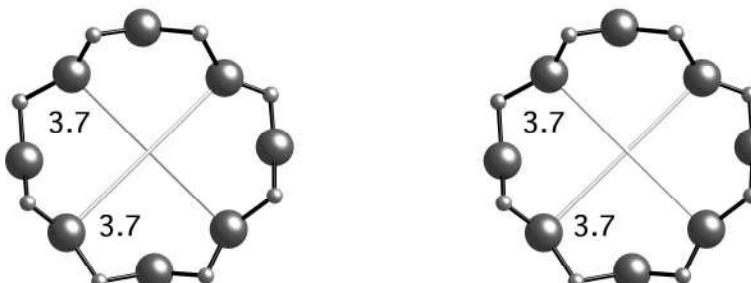
**Crystal chemical data:**  $[(C_{12}H_{28}N^+)_4(OH)^-_4][Si_8Al_{28}P_{28}O_{128}]$ -AFR  
 $C_{12}H_{28}N^+$  = tetrapropylammonium  
orthorhombic, Pccn,  $a = 21.944\text{\AA}$ ,  $b = 13.691\text{\AA}$ ,  $c = 14.249\text{\AA}$ <sup>(3)</sup>  
(Relationship to unit cell of Framework Type:  $a' = a$ ,  $b' = b$ ,  $c' = 2c$ )

**Framework density:** 15.0 T/1000 $\text{\AA}^3$

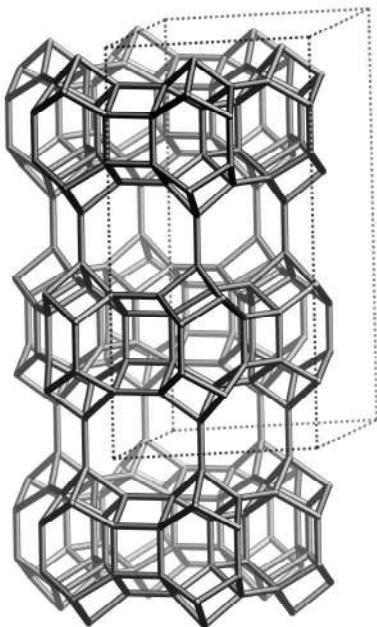
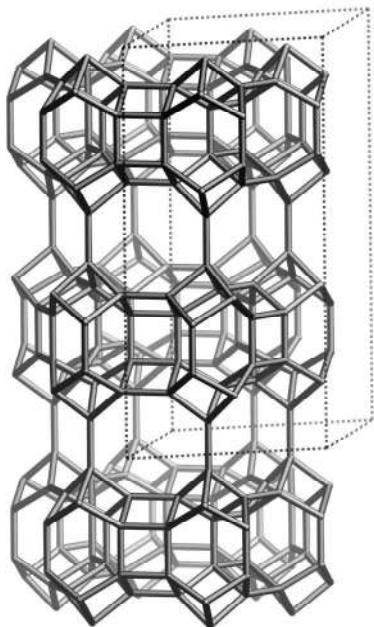
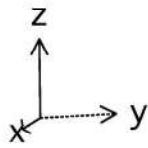
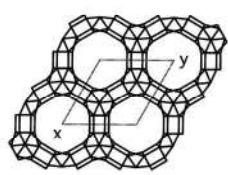
**Channels:** [001] **12** 6.7 x 6.9\*  $\leftrightarrow$  [010] **8** 3.7 x 3.7\*



12-ring viewed along [001]



8-ring viewed along [010]



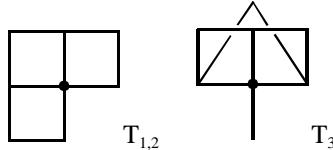
framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** hexagonal, P<sub>6</sub><sub>3</sub>/mcm,  $a = 13.1\text{\AA}$ ,  $c = 25.9\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (24, 1)    4    9    17    28    42    60    83    111    138    166	4·4·4·8 <sub>2</sub> ·6 <sub>2</sub> ·8
	T <sub>2</sub> (24, 1)    4    9    16    25    39    61    86    109    134    163	4·4·4·6·6·12
	T <sub>3</sub> (8, 3)    4    9    18    30    43    62    85    105    135    180	4·8·4·8·4·8

**Secondary building units:** 6\*1

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*MAPSO-46<sup>(1)</sup>

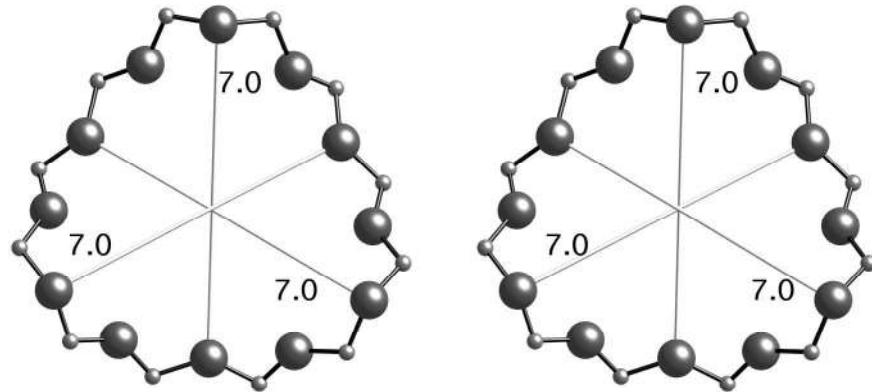
## References:

- (1) Bennett, J.M. and Marcus, B.K. *Stud. Surf. Sci. Catal.*, **37**, 269-279 (1988)

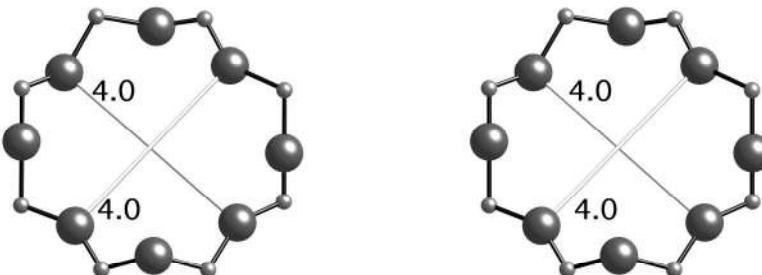
**Crystal chemical data:**  $[(C_6H_{16}N^+)_8(H_2O)_{14}] [Mg_6Al_{22}P_{26}Si_2O_{112}]$ -AFS  
 $C_6H_{15}N$  = dipropylamine  
trigonal, P3c1,  $a = 13.225\text{\AA}$ ,  $c = 26.892\text{\AA}$  <sup>(1)</sup>

**Framework density:** 13.7 T/1000 $\text{\AA}^3$

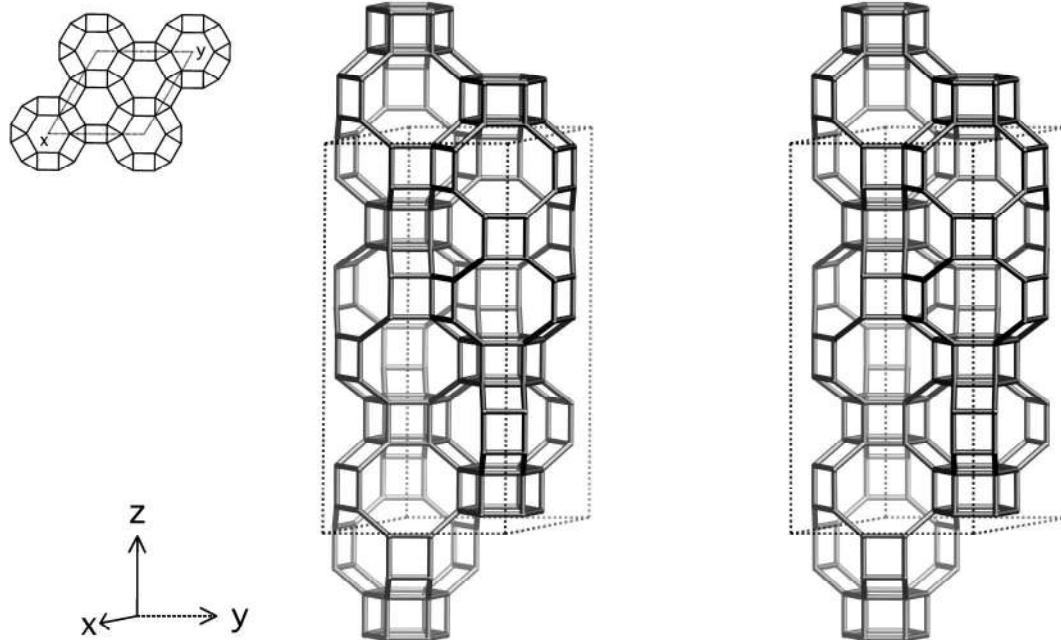
**Channels:** [001] **12** 7.0 x 7.0\*  $\leftrightarrow \perp$  [001] **8** 4.0 x 4.0\*\*



12-ring viewed along [001]



8-ring viewed normal to [001]



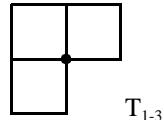
framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** hexagonal, P<sub>6</sub><sub>3</sub>/mmc,  $a = 13.7\text{\AA}$ ,  $c = 29.4\text{\AA}$

<b>Coordination sequences</b>	T <sub>1</sub> (24, 1)	4	9	17	29	45	64	85	110	140	173	4·4·4·8·6·8
<b>and vertex symbols:</b>	T <sub>2</sub> (24, 1)	4	9	17	29	45	64	86	113	144	178	4·4·4·8·6·8
	T <sub>3</sub> (24, 1)	4	9	17	29	45	65	88	113	141	175	4·4·4·8·6·8

**Secondary building units:** 6-6 or 6 or 4

**Loop configuration of T-Atoms:**



**Framework description:** AABBCCAACCBB sequence of 6-rings

**Isotypic framework structures:** \*AlPO-52<sup>(1,2)</sup>

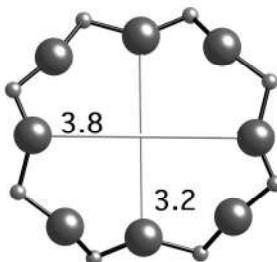
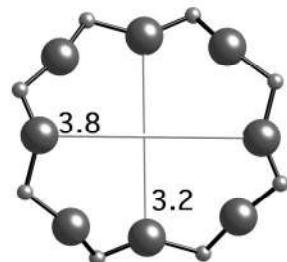
#### References:

- (1) Bennett, J.M., Kirchner, R.M. and Wilson, S.T. *Stud. Surf. Sci. Catal.*, **49**, 731-739 (1989)
- (2) McGuire, N.K., Bateman, C.A., Blackwell, C.S., Wilson, S.T. and Kirchner, R.M. *Zeolites*, **15**, 460-469 (1995)

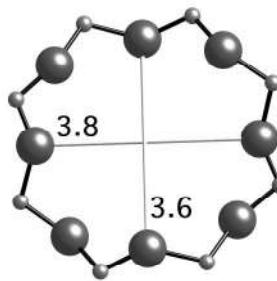
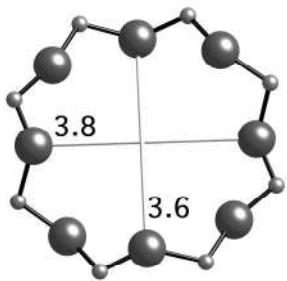
**Crystal chemical data:**  $[Al_{36}P_{36}O_{144}]$ -AFT  
trigonal,  $P\bar{3}1c$ ,  $a = 13.715\text{\AA}$ ,  $c = 29.676\text{\AA}$ <sup>(2)</sup>

**Framework density:** 14.9 T/1000 $\text{\AA}^3$

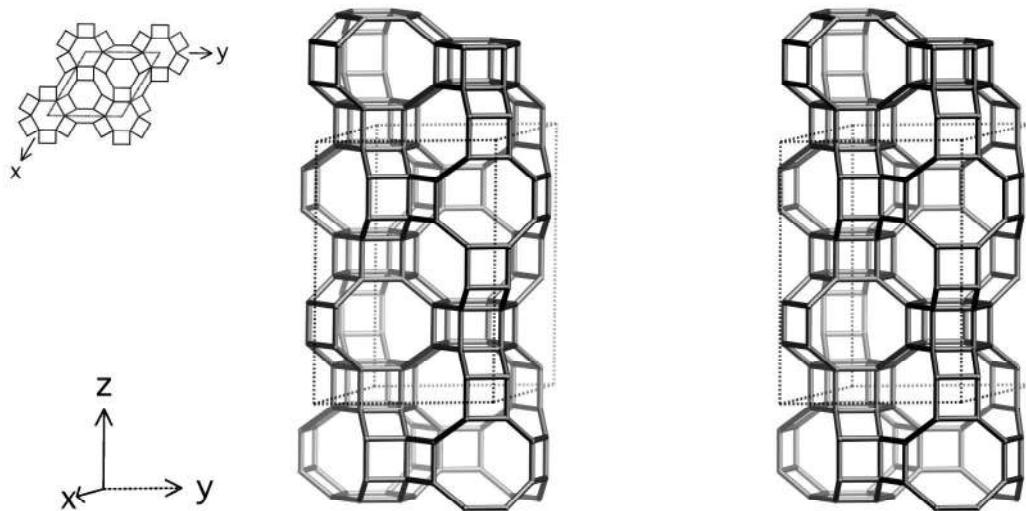
**Channels:**  $\perp [001]$  **8** 3.2 x 3.8\*\*\*



*GME cage 8-ring viewed normal to [001]*



*CHA cage 8-ring viewed normal to [001]*



framework viewed normal to [001] (top left: projection down [001])

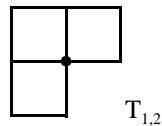
**Idealized cell constants:** hexagonal, P<sub>6</sub><sub>3</sub>/mmc,  $a = 13.7\text{\AA}$ ,  $c = 19.7\text{\AA}$

**Coordination sequences and vertex symbols:**

T <sub>1</sub> (24, 1)	4	9	17	29	45	65	89	116	144	175	4·4·4·8·6·8
T <sub>2</sub> (24, 1)	4	9	17	29	45	64	85	110	141	178	4·4·4·8·6·8

**Secondary building units:** 6-6 or 6 or 4

**Loop configuration of T-Atoms:**



**Framework description:** AABBCCBB sequence of 6-rings

**Isotypic framework structures:** \*SAPO-56<sup>(1)</sup>  
SSZ-16<sup>(2)</sup>

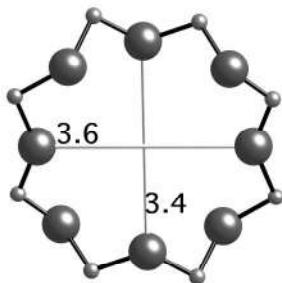
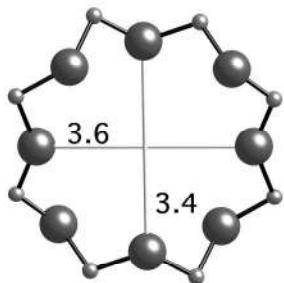
#### References:

- (1) Wilson, S.T., Broach, R.W., Blackwell, C.S., Bateman, C.A., McGuire, N.K. and Kirchner, R.M. *Microporous and Mesoporous Materials*, **28**, 125-137 (1999)
- (2) Lobo, R.F., Zones, S.I. and Medrud, R.C. *Chem. Mater.*, **8**, 2409-2411 (1996)

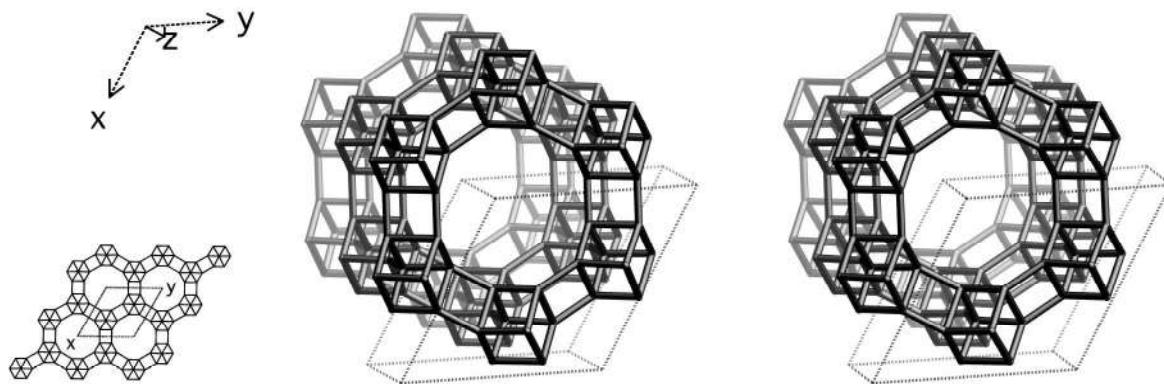
**Crystal chemical data:**  $[\text{H}_3^+][\text{Al}_{23}\text{Si}_5\text{P}_{20}\text{O}_{96}]$ -AFX  
trigonal,  $\text{P}\bar{3}1\text{c}$ ,  $a = 13.762\text{\AA}$ ,  $c = 19.949\text{\AA}$  <sup>(2)</sup>

**Framework density:** 14.7 T/1000 $\text{\AA}^3$

**Channels:**  $\perp[001]$  8 3.4 x 3.6\*\*\*



8-ring viewed normal to [001]



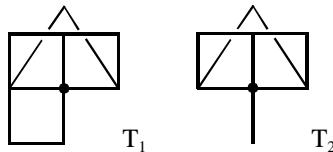
framework viewed along [001] (bottom left: projection down [001])

**Idealized cell constants:** trigonal, P $\bar{3}$ 1m,  $a = 12.3\text{\AA}$ ,  $c = 8.6\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(12, 1)$ 4    8    14    25    39    53    71    96    124    152	4.4.4.8.4.12
	$T_2(4, 3)$ 4    9    16    23    34    57    82    98    115    141	4.8.4.8.4.8

**Secondary building units:** 4-4 or 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*CoAPO-50<sup>(1,2)</sup>  
MgAPO-50<sup>(3)</sup>

### References:

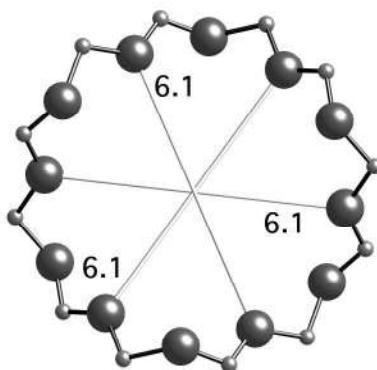
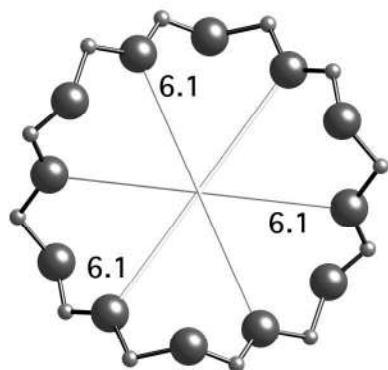
- (1) Wilson, S.T. private communication
- (2) Bennett, J.M. and Marcus, B.K. Stud. Surf. Sci. Catal., **37**, 269-279 (1988)
- (3) Akolekar, D.B. Zeolites, **15**, 583-590 (1995)

**Crystal chemical data:**  $[(\text{C}_6\text{H}_{16}\text{N}^+)_3 (\text{H}_2\text{O})_7] [\text{Co}_3\text{Al}_5\text{P}_8 \text{O}_{32}]$ -AFY  
 $\text{C}_6\text{H}_{15}\text{N}$  = dipropylamine  
trigonal,  $\text{P}\bar{3}$ ,  $a = 12.747\text{\AA}$ ,  $c = 9.015\text{\AA}$  <sup>(2)</sup>

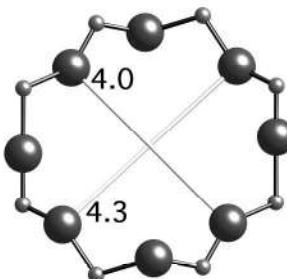
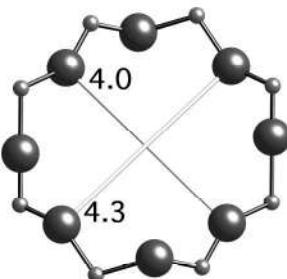
**Framework density:** 12.6 T/1000 $\text{\AA}^3$

**Channels:** [001] **12** 6.1 x 6.1\*  $\leftrightarrow$   $\perp$  [001] **8** 4.0 x 4.3\*\*

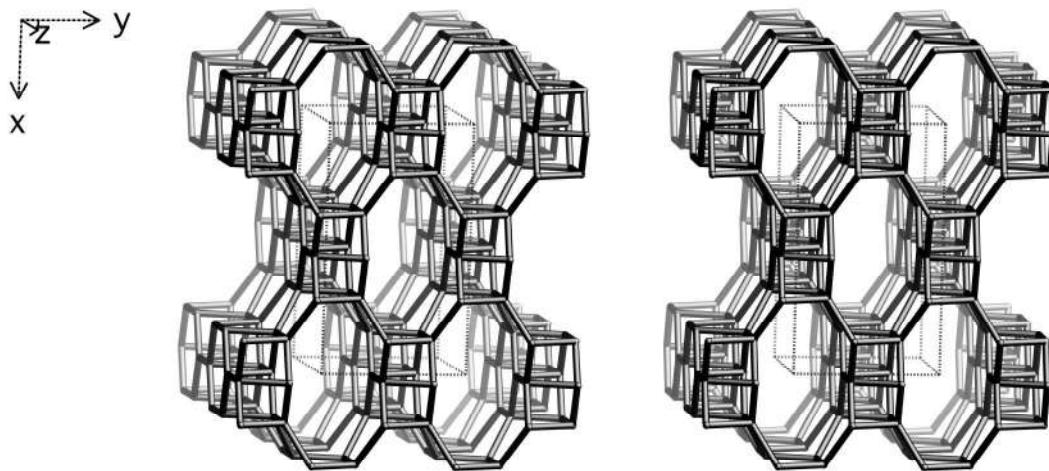
**Stability:** Unstable to removal of template <sup>(1)</sup>



12-ring viewed along [001]



8-ring viewed normal to [001]



framework viewed along [001]

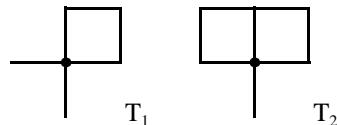
**Idealized cell constants:** orthorhombic, Cmcm,  $a = 15.8\text{\AA}$ ,  $b = 9.2\text{\AA}$ ,  $c = 8.6\text{\AA}$

**Coordination sequences and vertex symbols:**

$T_1(16, 1)$	4 11 21 36 56 81 109 142 179 221	$4 \cdot 6_2 \cdot 6 \cdot 6_3 \cdot 6 \cdot 6_3$
$T_2(8, m)$	4 10 18 32 53 78 105 140 179 218	$4 \cdot 6_3 \cdot 4 \cdot 6_3 \cdot 6 \cdot 6_4$

**Secondary building units:** 4-2

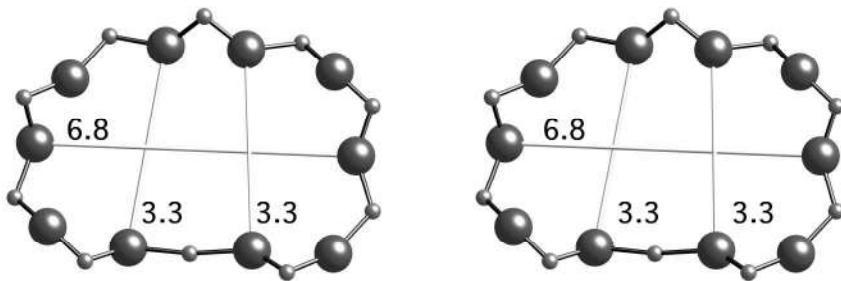
**Loop configuration of T-Atoms:**



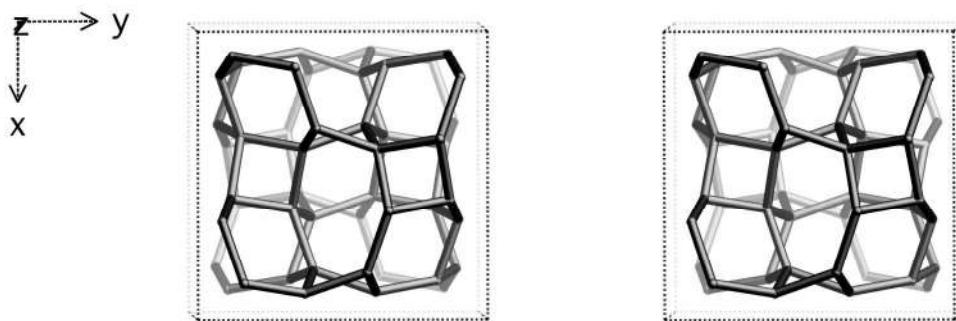
**Isotypic framework structures:** \*AlPO-H2<sup>(1-2)</sup>

### References:

- (1) Li, H.X., Davis, M.E., Higgins, J.B. and Dessau, R.M. *Chem. Commun.*, 403-405 (1993)
- (2) Kennedy, G.J., Higgins, J.B., Ridenour, C.F., Li, H.X. and Davis, M.E. *Solid State Nucl. Mag. Res.*, **4**, 173-178 (1995)

**Crystal chemical data:** $[(\text{H}_2\text{O})_8 \text{ [Al}_6\text{P}_6 \text{ O}_{24}]\text{-AHT}$ monoclinic, P112<sub>1</sub>,  $a = 9.486\text{\AA}$ ,  $b = 9.914\text{\AA}$ ,  $c = 8.126\text{\AA}$ ,  $\gamma = 121.49^\circ$ <sup>(1)</sup>(Relationship to unit cell of Framework Type:  $\mathbf{a}' = \mathbf{a}/2 \sin(\gamma)$ ,  $\mathbf{b}' = \mathbf{b}$ ,  $\mathbf{c}' = \mathbf{c}$ or, as vectors,  $\mathbf{a}' = (\mathbf{a} - \mathbf{b})/2$ ,  $\mathbf{b}' = \mathbf{b}$ ,  $\mathbf{c}' = \mathbf{c}$ )**Framework density:**18.4 T/1000 $\text{\AA}^3$ **Channels:**[001] **10** 3.3 x 6.8\***Stability:**Transforms to AlPO<sub>4</sub>-tridymite on heating<sup>(2)</sup>

10-ring viewed along [001]



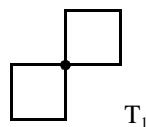
framework viewed along [001]

**Idealized cell constants:** cubic, Ia $\bar{3}$ d,  $a = 13.6\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(48, 2) \quad 4 \quad 10 \quad 22 \quad 39 \quad 60 \quad 87 \quad 118 \quad 154 \quad 196 \quad 242$        $4\cdot4\cdot6\cdot8_4\cdot8_4$

**Secondary building units:** 6-2 or 6 or 4

**Loop configuration of T-Atoms:**



**Isotypic framework structures:**

\*Analcime<sup>(1-3)</sup>  
**[Al-Co-P-O]-ANA**<sup>(4)</sup>  
**[Al-Si-P-O]-ANA**<sup>(5)</sup>  
**[Ga-Ge-O]-ANA**<sup>(6)</sup>  
 $|\text{Cs-Na-(H}_2\text{O)}|$  **[Ga-Si-O]-ANA**<sup>(7)</sup>  
 $|\text{Cs}_{16}|\text{[Cu}_8\text{Si}_{40}\text{O}_{96}]$ -ANA<sup>(8)</sup>  
**|K-|[B-Si-O]-ANA**<sup>(9)</sup>  
 AlPO-24<sup>(10)</sup>  
 AlPO<sub>4</sub>-pollucite<sup>(11)</sup>  
 Ammonoleucite<sup>(12)</sup>  
 Ca-D<sup>(13)</sup>  
 Cs berylliosilicate pollucite<sup>(14)</sup>

Cs,Fe silicate pollucite<sup>(15)</sup>  
 Hsianghualite<sup>(16)</sup>  
 Kehoeite<sup>(17)</sup>  
 Leucite<sup>(18)</sup>  
 Na-B<sup>(19)</sup>  
 Pollucite<sup>(20)</sup>  
 Synthetic analcime<sup>(21)</sup>  
 Synthetic hsinghualite<sup>(22)</sup>  
 Synthetic wairakite<sup>(23)</sup>  
 Wairakite and additional  
 compositional variants<sup>(24)</sup>

**Alternate designation:** Analcite

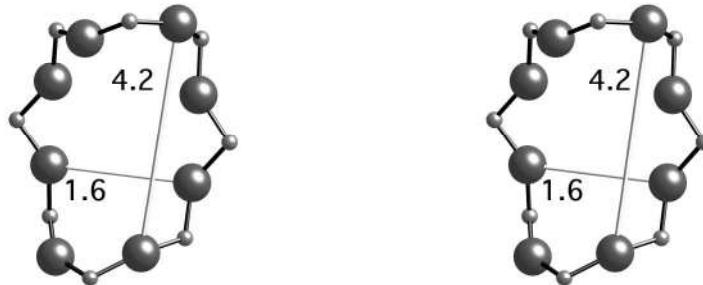
### References:

- (1) Taylor, W.H. *Z. Kristallogr.*, **74**, 1-19 (1930)
- (2) Knowles, C.R., Rinaldi, F.F. and Smith, J.V. *Indian Mineral.*, **6**, 127- (1965)
- (3) Ferraris, G., Jones, D.W. and Yerkess, J. *Z. Kristallogr.*, **135**, 240-252 (1972)
- (4) Feng, P.Y., Bu, X.H. and Stucky, G.D. *Nature*, **388**, 735-741 (1997)
- (5) Artioli, G., Pluth, J.J. and Smith, J.V. *Acta Crystallogr.*, **C40**, 214-217 (1984)
- (6) Bu, X., Feng, P., Gier, T.E., Zhao, D. and Stucky, G.D. *J. Am. Chem. Soc.*, **120**, 13389-13397 (1998)

**Crystal chemical data:**  $[\text{Na}^+_{16} (\text{H}_2\text{O})_{16}] [\text{Al}_{16}\text{Si}_{32} \text{O}_{96}]$ -ANA  
cubic, Ia $\bar{3}$ d,  $a = 13.73\text{\AA}$ <sup>(3)</sup>

**Framework density:** 18.5 T/1000 $\text{\AA}^3$

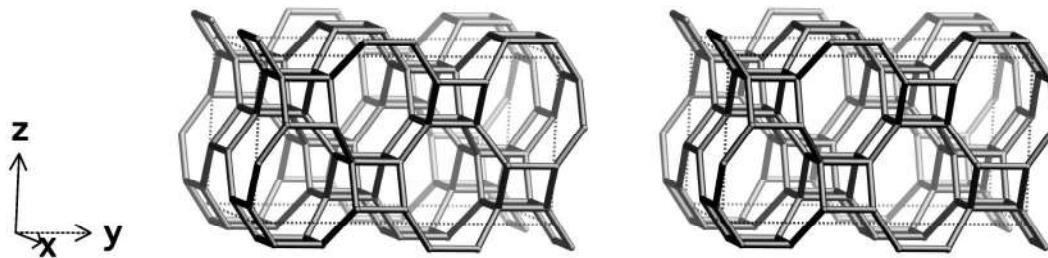
**Channels:** irregular channels formed by highly distorted 8-rings



*distorted 8-ring viewed along [110]*

#### References (cont.):

- (7) Yelon, W.B., Xie, D., Newsam, J.M. and Dunn, J. *Zeolites*, **10**, 553-558 (1990)
- (8) Heinrich, A.R. and Baerlocher, Ch. *Acta Crystallogr.*, **C47**, 237-241 (1991)
- (9) Millini, R., Montanari, L. and Bellussi, G. *Microporous Materials*, **1**, 9-15 (1993)
- (10) Wilson, S.T., Lok, B.M., Messina, C.A., Cannan, T.R. and Flanigen, E.M. *J. Am. Chem. Soc.*, **104**, 1146-1147 (1982)
- (11) Keller, E.B. *Ph.D. Thesis, ETH, Zürich, Switzerland*, (1987)
- (12) Hori, H., Nagashima, K., Yamada, M., Miyawaki, R. and Marubashi, T. *Am. Mineral.*, **71**, 1022-1027 (1986)
- (13) Ames, L.L. and Sand, L.B. *Am. Mineral.*, **43**, 476-480 (1958)
- (14) Torres-Martinez, L.M., Gard, J.A., Howie, R.A. and West, A.R. *J. Solid State Chem.*, **51**, 100-103 (1984)
- (15) Kopp, O.C., Harris, L.A., Clark, G.W. and Yakel, H.L. *Am. Mineral.*, **48**, 100-109 (1963)
- (16) Wen-Hui, H., Saho-Hua, T., Kung-Hai, W., Chun-Lin, C. and Cheng Chi, Y. *Am. Mineral.*, **44**, 1327-1328 (1959)
- (17) McConnell, D. and Foreman Jr., D.W. *Can. Mines*, **12**, 352- (1974)
- (18) Peacor, D.R. *Z. Kristallogr.*, **127**, 213-224 (1968)
- (19) Barrer, R.M. and White, E.A.D. *J. Chem. Soc.*, 1561-1571 (1952)
- (20) Nel, H.J. *Am. Mineral.*, **29**, 443-451 (1944)
- (21) Ghobarkar, H. and Franke, W. *Cryst. Res. Technol.*, 1071-1075 (1986)
- (22) Ghobarkar, H., Schaeff, O. and Knauth, P. *Annal. Chimie, Science Matériaux*, **24**, 209-215 (1999)
- (23) Ghobarkar, H. *Cryst. Res. Technol.*, K90-92 (1985)
- (24) Takeuchi, Y., Mazzi, F., Haga, N. and Galli, E. *Am. Mineral.*, **64**, 993-1001 (1979)



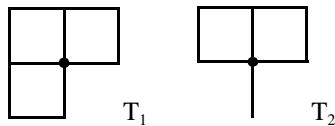
framework viewed along [100]

**Idealized cell constants:** orthorhombic, Cmca,  $a = 9.0\text{\AA}$ ,  $b = 19.4\text{\AA}$ ,  $c = 10.4\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(16, 1) \quad 4 \quad 9 \quad 19 \quad 35 \quad 53 \quad 75 \quad 102 \quad 132 \quad 168 \quad 208$        $4 \cdot 4 \cdot 4 \cdot 8_2 \cdot 8 \cdot 8_2$   
 $T_2(16, 1) \quad 4 \quad 10 \quad 20 \quad 35 \quad 54 \quad 76 \quad 104 \quad 136 \quad 171 \quad 211$        $4 \cdot 6 \cdot 4 \cdot 6 \cdot 6 \cdot 8_2$

**Secondary building units:** 8 or 4

**Loop configuration of T-Atoms:**

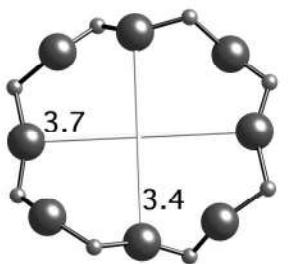
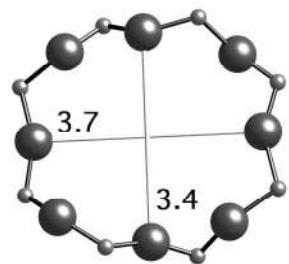


**Isotypic framework structures:** \*AlPO-C<sup>(1,2)</sup>  
AlPO-H3<sup>(3)</sup>

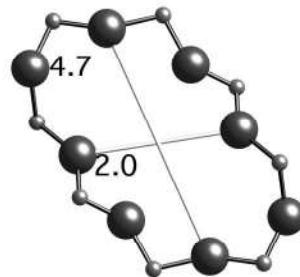
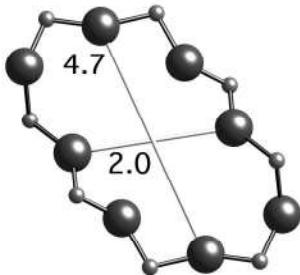
### References:

- (1) Bennett, J.M., Dytrych, W.J., Pluth, J.J., Richardson Jr., J.W. and Smith, J.V. *Zeolites*, **6**, 349-359 (1986)
- (2) Keller, E.B., Meier, W.M. and Kirchner, R.M. *Solid State Ionics*, **43**, 93-102 (1990)
- (3) Pluth, J.J. and Smith, J.V. *Acta Crystallogr.*, **C42**, 1118-1120 (1986)

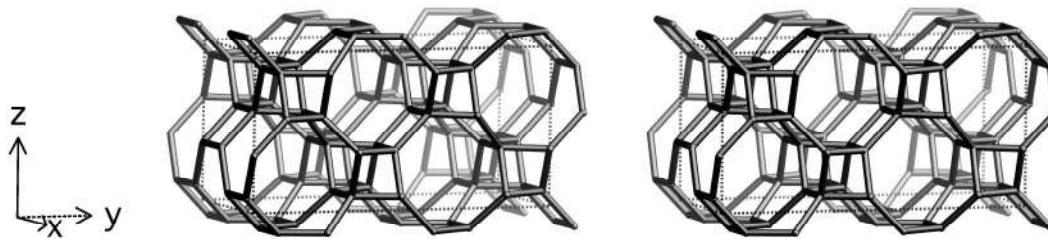
<b>Crystal chemical data:</b>	[Al <sub>16</sub> P <sub>16</sub> O <sub>64</sub> ]-APC orthorhombic, Pbca, a = 19.821 Å, b = 10.028 Å, c = 8.936 Å <sup>(2)</sup> (Relationship to unit cell of Framework Type: a' = b, b' = c, c' = a)
<b>Framework density:</b>	18.0 T/1000 Å <sup>3</sup>
<b>Channels:</b>	[001] 8 3.4 x 3.7* ↔ [100] 8 2.0 x 4.7*
<b>Stability:</b>	Transforms to AlPO <sub>4</sub> -D at ca 250°C <sup>(2)</sup>



8-ring viewed along [001]



distorted 8-ring viewed along [100]



framework viewed along [100]

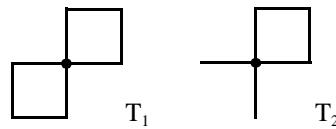
**Idealized cell constants:** orthorhombic, Cmca,  $a = 8.7\text{\AA}$ ,  $b = 20.1\text{\AA}$ ,  $c = 10.2\text{\AA}$

**Coordination sequences and vertex symbols:**

$T_1(16, 1)$	4	10	21	37	57	82	112	145	184	228	4·4·6 <sub>2</sub> ·8 <sub>3</sub> ·6 <sub>3</sub> ·8 <sub>3</sub>
$T_2(16, 1)$	4	11	22	38	59	83	113	147	186	230	4·6 <sub>2</sub> ·6·6 <sub>2</sub> ·6·6 <sub>3</sub>

**Secondary building units:** 8 or 6-2 or 4

**Loop configuration of T-Atoms:**



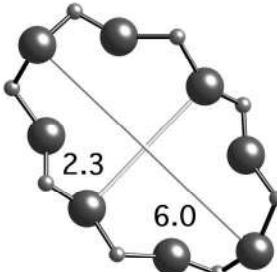
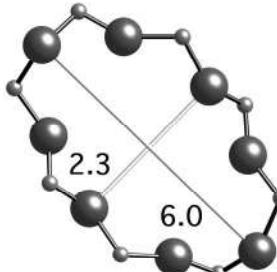
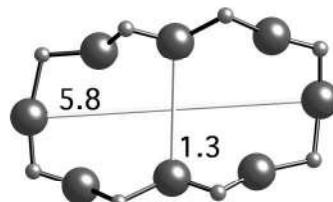
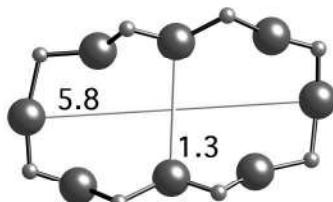
**Isotypic framework structures:** \*AlPO-D<sup>(1)</sup>

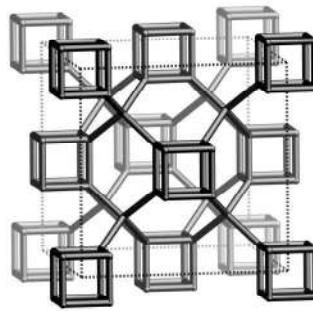
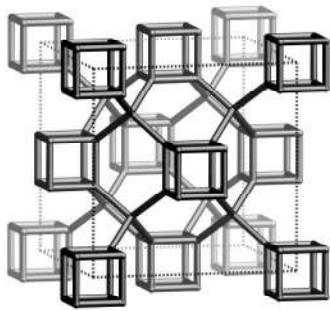
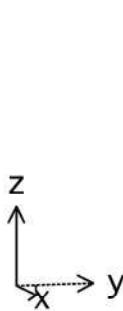
## References:

- (1) Keller, E.B., Meier, W.M. and Kirchner, R.M. *Solid State Ionics*, **43**, 93-102 (1990)

**Crystal chemical data:****[Al<sub>16</sub>P<sub>16</sub>O<sub>64</sub>]-APD**(forms irreversibly from AlPO<sub>4</sub>-C at around 200°C)orthorhombic, Pca2<sub>1</sub>,  $a = 19.187\text{\AA}$ ,  $b = 8.576\text{\AA}$ ,  $c = 9.804\text{\AA}$ <sup>(1)</sup>(Relationship to unit cell of Framework Type:  $a' = b$ ,  $b' = a$ ,  $c' = c$ )**Framework density:**19.8 T/1000 $\text{\AA}^3$ **Channels:**

[010] 8 2.3 x 6.0\* ↔ [201] 8 1.3 x 5.8\*

*distorted 8-ring viewed along [010]**distorted 8-ring along [201]*



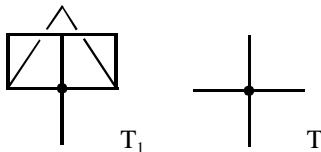
*framework viewed along [100]*

**Idealized cell constants:** cubic, Fm $\bar{3}$ m,  $a = 13.6\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(32, 3m) \quad 4 \quad 9 \quad 19 \quad 34 \quad 48 \quad 66 \quad 96 \quad 127 \quad 151 \quad 183$       4·6·4·6·4·6  
 $T_2(8, \bar{4}3m) \quad 4 \quad 12 \quad 18 \quad 28 \quad 52 \quad 78 \quad 88 \quad 112 \quad 162 \quad 204$       6·6·6·6·6·6

**Secondary building units:** 4-1

**Loop configuration of T-Atoms:**



**Framework description:** structural derivative of fluorite

**Isotypic framework structures:** \*AlPO-16<sup>(1)</sup>  
Octadecasil<sup>(2)</sup>

#### References:

- (1) Bennett, J.M. and Kirchner, R.M. *Zeolites*, **11**, 502-506 (1991)
- (2) Caulliet, P., Guth, J.L., Hazm, J., Lamblin, J.M. and Gies, H. *Eur. J. Solid State Inorg. Chem.*, **28**, 345-361 (1991)

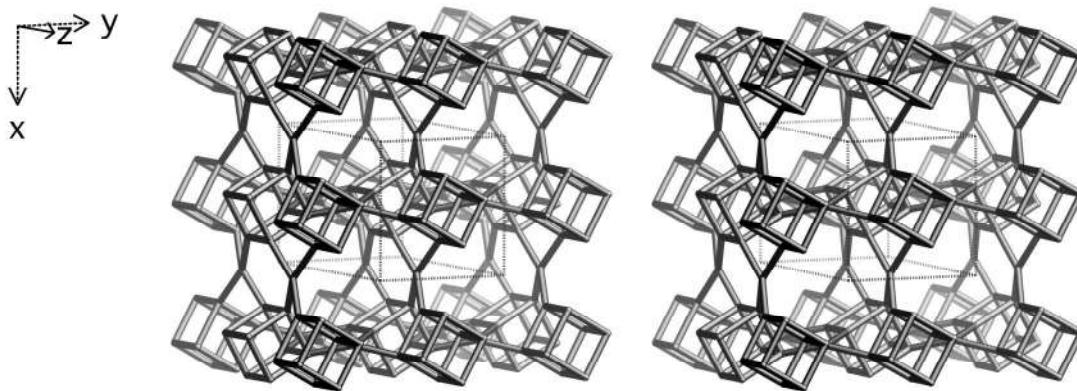
**Crystal chemical data:**  $[(C_7H_{13}N)_4(H_2O)_{16}] [Al_{20}P_{20}O_{80}]$ -AST

$C_7H_{13}N$  = quinuclidine

cubic, F23,  $a = 13.383\text{\AA}$  <sup>(1)</sup>

**Framework density:** 16.7 T/1000 $\text{\AA}^3$

**Channels:** apertures formed by 6-rings only



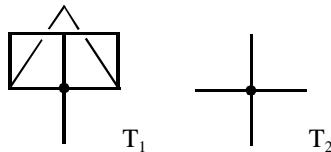
framework viewed along [001]

**Idealized cell constants:** tetragonal, P4/mcc,  $a = 8.7\text{\AA}$ ,  $c = 13.9\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(16, 1) \quad 4 \quad 9 \quad 19 \quad 35 \quad 52 \quad 72 \quad 100 \quad 131 \quad 163 \quad 201$       4·6·4·6·4·6  
 $T_2(4, 222) \quad 4 \quad 12 \quad 18 \quad 26 \quad 52 \quad 84 \quad 100 \quad 118 \quad 162 \quad 210$       6·6·6<sub>2</sub>·6<sub>2</sub>·12<sub>8</sub>·12<sub>8</sub>

**Secondary building units:** 4-1

**Loop configuration of T-Atoms:**



**Isotypic framework structures:** \*ASU-7<sup>(1)</sup>

### References:

- (1) Li, H. and Yaghi, O.M. *J. Am. Chem. Soc.*, **120**, 10569-10570 (1998)

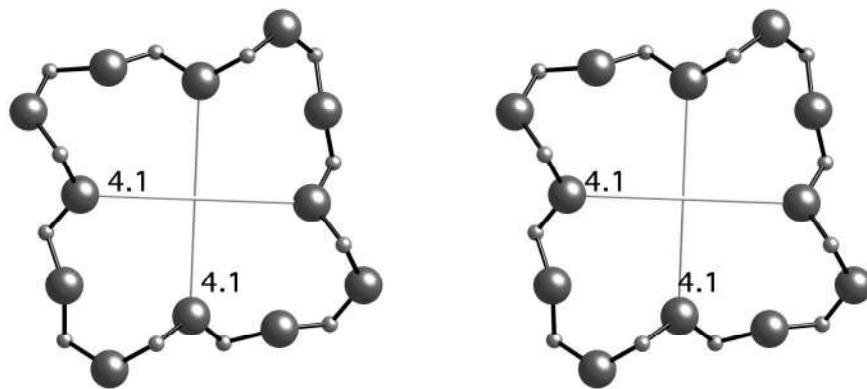
**Crystal chemical data:**  $[(C_2H_7N)_2(H_2O)_2][Ge_{20}O_{40}]$ -ASV

$C_2H_7N$  = dimethylamine

tetragonal, P4/mcc,  $a = 8.780\text{\AA}$ ,  $c = 14.470\text{\AA}$  <sup>(1)</sup>

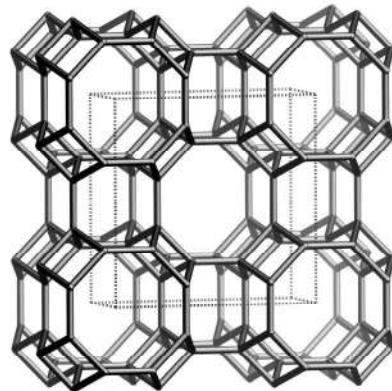
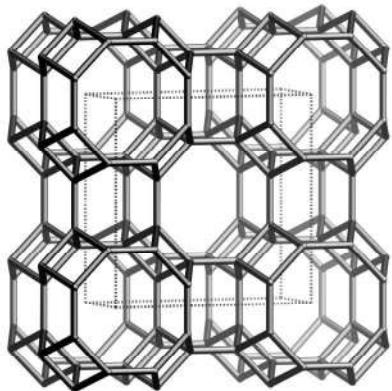
**Framework density:**  $17.9 \text{ T}/1000\text{\AA}^3$

**Channels:** [001] **12** 4.1x 4.1\*



*12-ring along [001]*

$\rightarrow z \rightarrow y$   
↓  
X



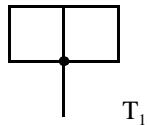
framework viewed along [001]

**Idealized cell constants:** tetragonal, I4/mmm,  $a = 13.1\text{\AA}$ ,  $c = 5.3\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(16, m) \quad 4 \ 10 \ 21 \ 36 \ 54 \ 78 \ 106 \ 136 \ 173 \ 214$       4·6·4·6·6·8

**Secondary building units:** 8 or 4

**Loop configuration of T-Atoms:**



**Isotypic framework structures:** \*MAPO-39<sup>(1,2)</sup>

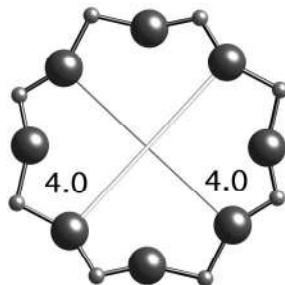
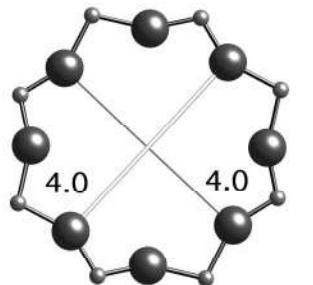
#### References:

- (1) McCusker, L.B., Brunner, G.O. and Ojo, A.F. *Acta Crystallogr.*, **A46**, C59 (1990)
- (2) Baur, W.H., Joswig, W., Kassner, D., Bieniok, A., Finger, G. and Kornatowski, J. Z. *Kristallogr.*, **214**, 154-159 (1999)

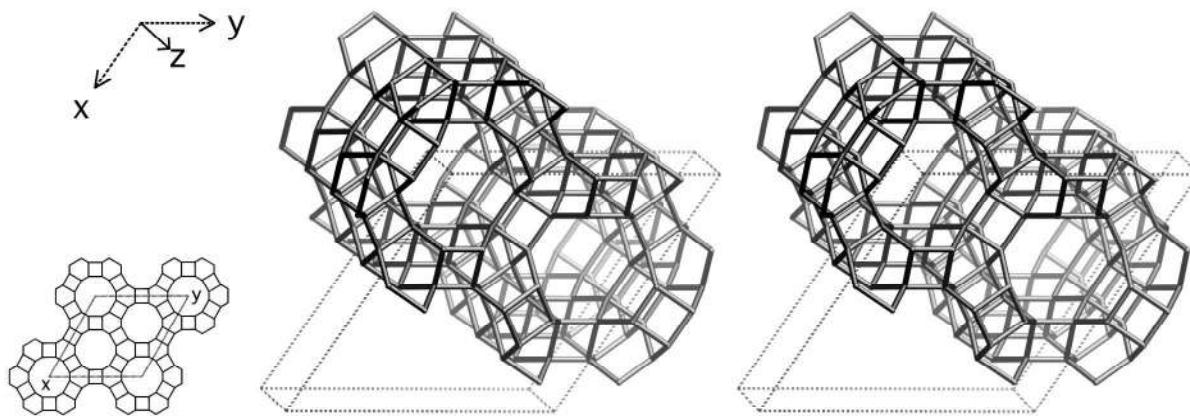
**Crystal chemical data:**  $|\text{H}^+| \text{ [Mg}_n\text{Al}_{8-n}\text{P}_8\text{O}_{32}]\text{-ATN}$   
tetragonal, I4/m,  $a = 13.209\text{\AA}$ ,  $c = 5.277\text{\AA}$  <sup>(2)</sup>

**Framework density:** 18.0 T/1000 $\text{\AA}^3$

**Channels:** [001] 8 4.0 x 4.0\*



8-ring viewed along [001]



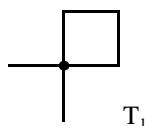
framework viewed along [001]

**Idealized cell constants:** trigonal, R $\bar{3}m$ ,  $a = 20.9\text{\AA}$ ,  $c = 5.1\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(36, 1)$  4 11 22 37 59 85 114 147 184 230  $4\cdot6_2\cdot6\cdot6_2\cdot6\cdot6_3$

**Secondary building units:** 6 or 4

**Loop configuration of T-Atoms:**



**Isotypic framework structures:** \*AlPO-31<sup>(1,2)</sup>  
SAPO-31<sup>(3-5)</sup>

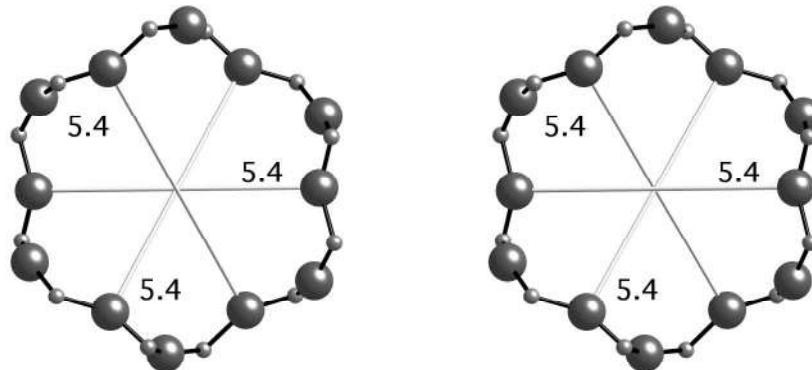
### References:

- (1) Bennett, J.M. and Kirchner, R.M. *Zeolites*, **12**, 338-342 (1992)
- (2) Baur, W.H., Joswig, W., Kassner, D. and Kornatowski, J. *Acta Crystallogr.*, **B50**, 290-294 (1994)
- (3) Flanigen, E.M., Lok, B.M., Patton, R.L. and Wilson, S.T. *Pure Appl. Chem.*, **58**, 1351-1358 (1986)
- (4) Flanigen, E.M., Lok, B.M., Patton, R.L. and Wilson, S.T. In *Proc. 7th Int. Zeolite Conf.*, (ed. eds. Y. Murakami, A. Iijima and J.W. Ward), pp. 103-112 (1986), Kodansha, Tokyo
- (5) Baur, W.H., Joswig, W., Kassner, D. and Kornatowski, J. *Acta Crystallogr.*, **B50**, 290-294 (1994)

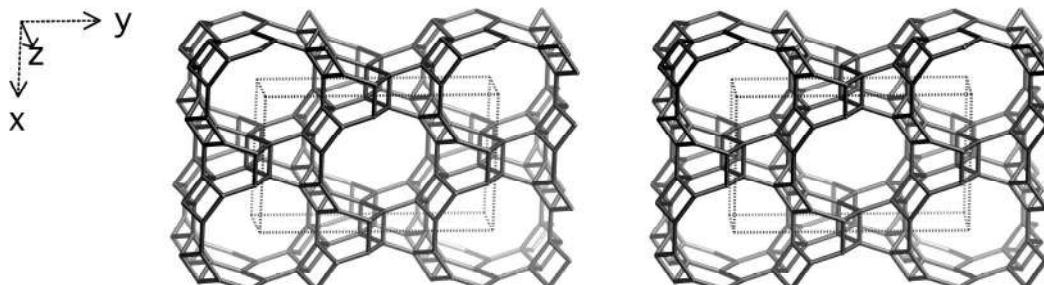
**Crystal chemical data:**  $[\text{Al}_{18}\text{P}_{18}\text{O}_{72}]\text{-ATO}$   
trigonal,  $R\bar{3}$ ,  $a = 20.827\text{\AA}$ ,  $c = 5.003\text{\AA}$ <sup>(1)</sup>

**Framework density:**  $19.2 \text{ T}/1000\text{\AA}^3$

**Channels:** [001] **12**  $5.4 \times 5.4^*$



*12-ring viewed along [001]*



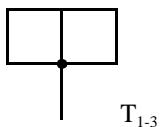
framework viewed along [001]

**Idealized cell constants:** orthorhombic, Cmcm,  $a = 13.2\text{\AA}$ ,  $b = 21.6\text{\AA}$ ,  $c = 5.3\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(8, m)$	4 10 19 30 46 67 93 124 154 189	4·6·4·6·6·6 <sub>2</sub>
	$T_2(8, m)$	4 10 20 32 49 73 97 124 157 193	4·6 <sub>2</sub> ·4·6 <sub>2</sub> ·6·12 <sub>2</sub>
	$T_3(8, m)$	4 10 19 32 51 72 96 124 155 196	4·6 <sub>2</sub> ·4·6 <sub>2</sub> ·6·12 <sub>2</sub>

**Secondary building units:** 6-2 or 4-2 or 6 or 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*MAPO-36<sup>(1)</sup>

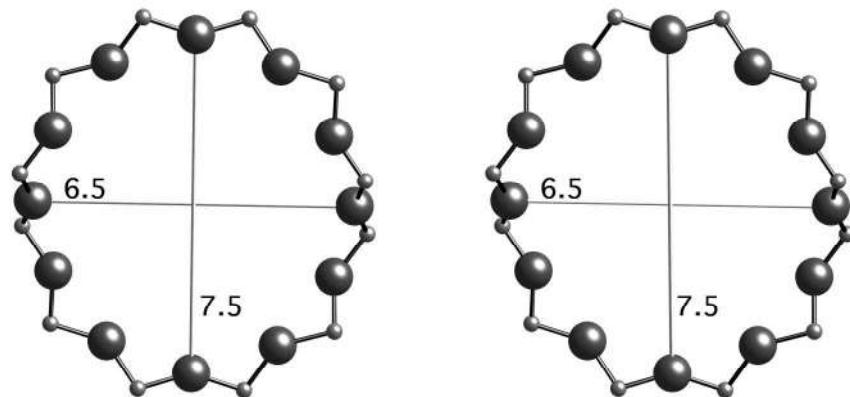
### References:

- (1) Smith, J.V., Pluth, J.J. and Andries, K.J. *Zeolites*, **13**, 166-169 (1993)

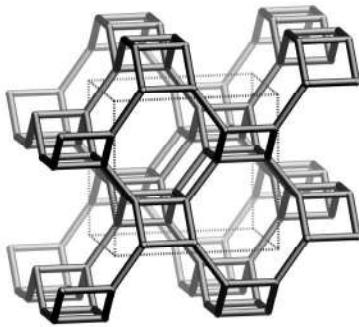
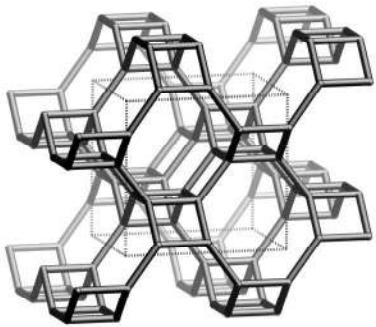
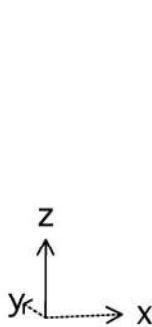
**Crystal chemical data:**  $[\text{H}^+ \text{MgAl}_{11}\text{P}_{12}\text{O}_{48}]$ -ATS  
monoclinic, C12/c1  
 $a = 13.148\text{\AA}$ ,  $b = 21.577\text{\AA}$ ,  $c = 5.164\text{\AA}$ ,  $\beta = 91.84^\circ$  <sup>(1)</sup>

**Framework density:** 16.4 T/1000 $\text{\AA}^3$

**Channels:** [001] 12 6.5 x 7.5\*



12-ring viewed along [001]



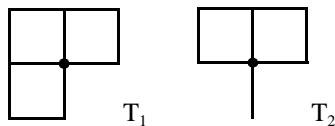
*framework viewed along [010]*

**Idealized cell constants:** orthorhombic, Pmma,  $a = 10.0\text{\AA}$ ,  $b = 7.5\text{\AA}$ ,  $c = 9.4\text{\AA}$

**Coordination sequences  
and vertex symbols:**  $T_1(8, 1) \quad 4 \quad 9 \quad 18 \quad 33 \quad 52 \quad 73 \quad 96 \quad 123 \quad 158 \quad 199$       4·4·4·6·8·8  
 $T_2(4, m) \quad 4 \quad 10 \quad 21 \quad 34 \quad 48 \quad 70 \quad 100 \quad 130 \quad 159 \quad 194$       4·8<sub>2</sub>·4·8<sub>2</sub>·6·8<sub>2</sub>

**Secondary building units:** 6-2 or 4-2 or 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*AlPO-12-TAMU<sup>(1)</sup>  
AlPO-33<sup>(2,3)</sup>

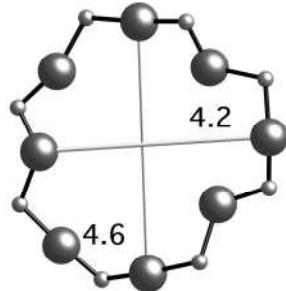
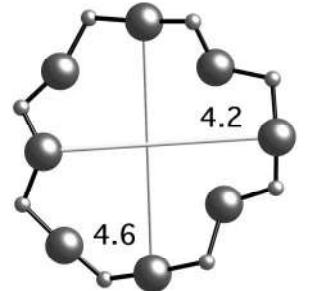
### References:

- (1) Rudolf, P.R., Saldarriaga-Molina, C. and Clearfield, A. *J. Phys. Chem.*, **90**, 6122-6125 (1986)
- (2) Smith, J.V., Pluth, J.J. and Bennett, J.M. *private communication*
- (3) Patton, R.L. and Gajek, R.T. *U.S. Patent 4,473,663* (1984)

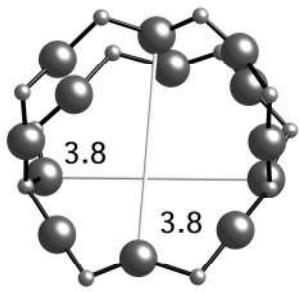
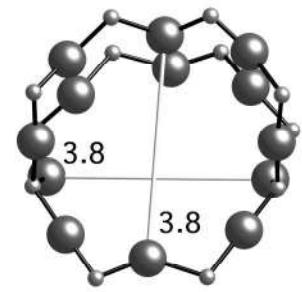
**Crystal chemical data:**  $[(C_4H_{12}N^+)_4(OH)^-]_4 [Al_{12}P_{12}O_{48}]$ -ATT  
 $C_4H_{12}N^+$  = tetramethylammonium  
orthorhombic,  $P2_12_12$ ,  $a = 10.332\text{\AA}$ ,  $b = 14.640\text{\AA}$ ,  $c = 9.511\text{\AA}$ <sup>(1)</sup>  
(Relationship to unit cell of Framework Type:  $a' = a$ ,  $b' = 2b$ ,  $c' = c$ )

**Framework density:**  $16.7 \text{ T}/1000\text{\AA}^3$

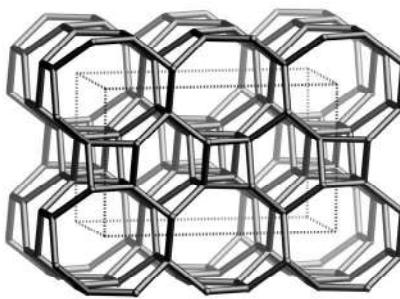
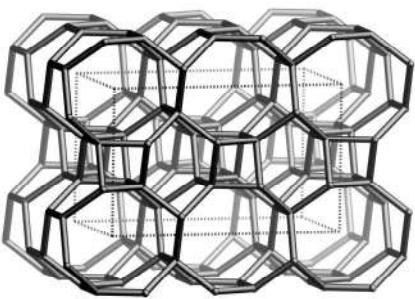
**Channels:**  $[100] \textbf{8} 4.2 \times 4.6^* \leftrightarrow [010] \textbf{8} 3.8 \times 3.8^*$



8-ring viewed along [100]



complex 8-ring viewed along [010]

**ATV****Framework Type****Cmma**

framework viewed along [100]

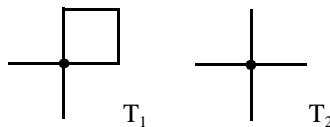
**Idealized cell constants:** orthorhombic, Cmma,  $a = 8.6\text{\AA}$ ,  $b = 15.3\text{\AA}$ ,  $c = 9.7\text{\AA}$

**Coordination sequences  
and vertex symbols:**

$T_1(16, 1)$	4	11	22	40	64	92	121	157	200	248	$4 \cdot 6_2 \cdot 6 \cdot 6_3 \cdot 6_2 \cdot 6_3$
$T_2(8, m)$	4	12	25	42	61	88	122	160	200	246	$6 \cdot 6_2 \cdot 6 \cdot 6_2 \cdot 6_2 \cdot 6_2$

**Secondary building units:** 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*AlPO-25<sup>(1)</sup>  
[Ga-P-O]-ATV<sup>(2)</sup>

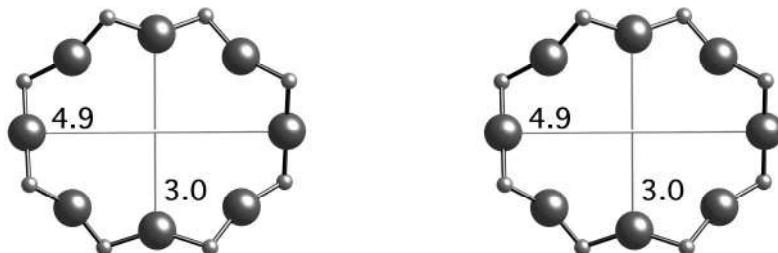
**References:**

- (1) Richardson Jr., J.W., Smith, J.V. and Pluth, J.J. *J. Phys. Chem.*, **94**, 3365-3367 (1990)
- (2) Parise, J.B. *Chem. Commun.*, 606-607 (1985)

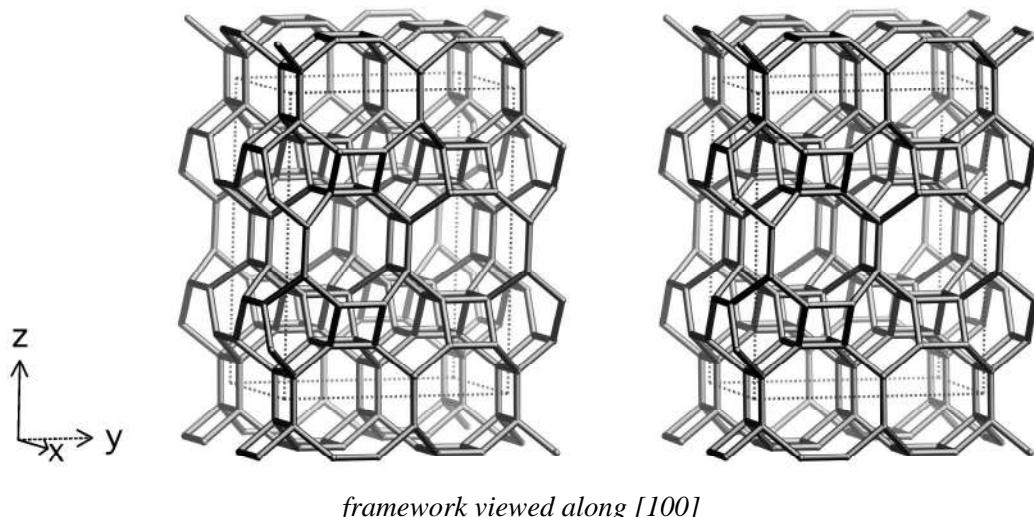
**Crystal chemical data:**  $[Al_{12}P_{12}O_{48}]$ -ATV  
orthorhombic, Acmm,  $a = 9.449\text{\AA}$ ,  $b = 15.203\text{\AA}$ ,  $c = 8.408\text{\AA}$ <sup>(1)</sup>  
(Relationship to unit cell of Framework Type:  $a' = c$ ,  $b' = b$ ,  $c' = a$ )

**Framework density:**  $19.9 \text{ T}/1000\text{\AA}^3$

**Channels:** [001] 8 3.0 x 4.9\*



8-ring viewed along [001]



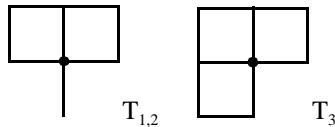
**Idealized cell constants:** orthorhombic, Cmca,  $a = 9.1\text{\AA}$ ,  $b = 15.0\text{\AA}$ ,  $c = 19.2\text{\AA}$

**Coordination sequences and vertex symbols:**

$T_1(16, 1)$	4	10	20	35	55	78	103	133	173	217	$4 \cdot 6 \cdot 4 \cdot 8_3 \cdot 6 \cdot 8_2$
$T_2(16, 1)$	4	10	21	36	53	76	108	142	173	210	$4 \cdot 6 \cdot 4 \cdot 8_2 \cdot 6 \cdot 8$
$T_3(16, 1)$	4	9	19	35	54	76	102	134	172	214	$4 \cdot 4 \cdot 4 \cdot 6 \cdot 8 \cdot 8_3$

**Secondary building units:** 6

**Loop configuration of T-Atoms:**



**Isotypic framework structures:** \*AlPO-21<sup>(1,2)</sup>  
[Ga-P-O]-AWO<sup>(3)</sup>

#### References:

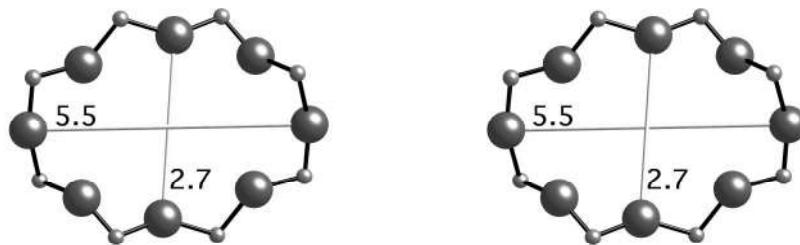
- (1) Bennett, J.M., Cohen, J.M., Artioli, G., Pluth, J.J. and Smith, J.V. *Inorg. Chem.*, **24**, 188-193 (1985)
- (2) Parise, J.B. and Day, C.S. *Acta Crystallogr.*, **C41**, 515-520 (1985)
- (3) Parise, J.B. *Chem. Commun.*, 606-607 (1985)

**Crystal chemical data:**  $[\text{H}^+ \cdot (\text{C}_2\text{H}_7\text{N})_{10.66} (\text{C}_3\text{H}_8)_{5.33} (\text{OH})_4] [\text{Al}_{12}\text{P}_{12} \text{O}_{48}]$ -AWO  
 $\text{C}_2\text{H}_7\text{N}$  = dimethylamine ,  $\text{C}_3\text{H}_8$  = propane  
monoclinic,  $P12_1/a1$   
 $a = 10.330\text{\AA}$ ,  $b = 17.524\text{\AA}$ ,  $c = 8.676\text{\AA}$ ,  $\beta = 123.37^\circ$  <sup>(1)</sup>  
(Relationship to unit cell of Framework Type:  
 $\mathbf{a}' = \mathbf{a}$ ,  $\mathbf{b}' = \mathbf{c}$ ,  $\mathbf{c}' = \mathbf{b}/2\sin(\beta)$   
or, as vectors,  $\mathbf{a}' = \mathbf{a}$ ,  $\mathbf{b}' = \mathbf{c}$ ,  $\mathbf{c}' = (\mathbf{b} - \mathbf{a})/2$ )

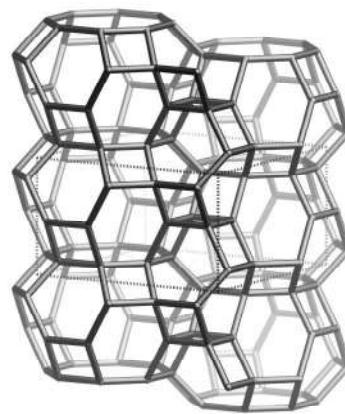
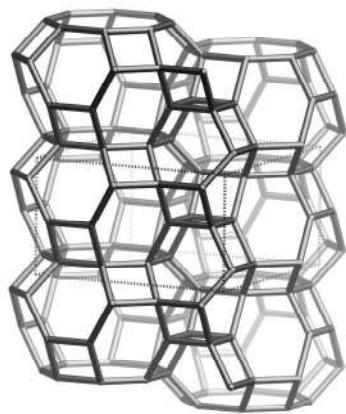
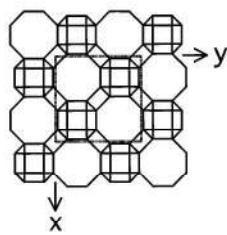
**Framework density:**  $18.3 \text{ T}/1000\text{\AA}^3$

**Channels:** [100] 8  $2.7 \times 5.5^*$

**Stability:** Transforms to AlPO-25 (ATV) upon calcination<sup>(2)</sup>



8-ring viewed along [100]



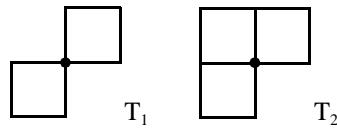
*framework viewed normal to [001] (top left: projection down [001])*

**Idealized cell constants:** tetragonal, P4/nmm (origin choice 2),  $a = 13.6\text{\AA}$ ,  $c = 7.6\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(16, 1)$	4 10 20 33 50 72 98 128 162 200	4·4·6·6·6·8
	$T_2(8, 2)$	4 9 17 30 50 74 97 123 158 198	4·4·4·6·6·6

**Secondary building units:** 6 or 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*AlPO-22<sup>(1)</sup>

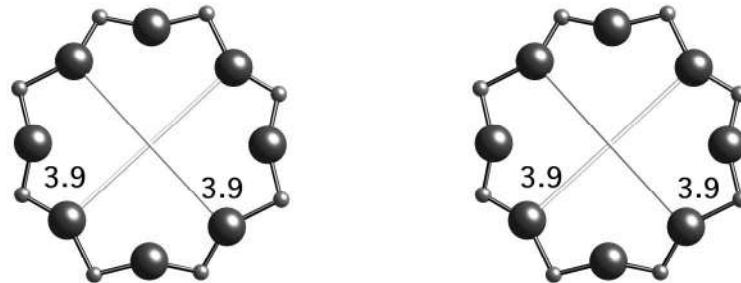
## References:

- (1) Richardson Jr., J.W., Pluth, J.J. and Smith, J.V. *Naturwiss.*, **76**, 467-469 (1989)

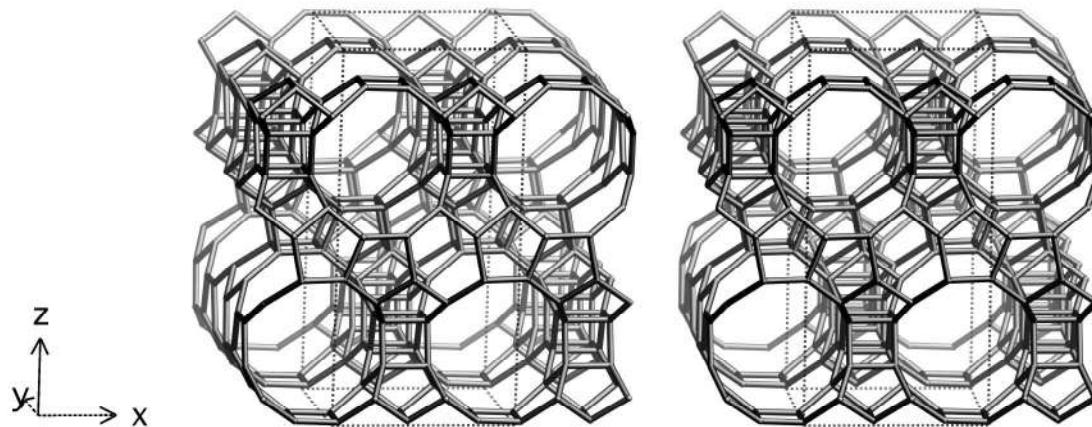
**Crystal chemical data:**  $[(C_7H_{14}N^+)_4(HPO_4^{2-})_2][Al_{24}P_{24}O_{96}]$ -AWW  
 $C_7H_{13}N$  = quinuclidine  
tetragonal, P4/ncc,  $a = 13.628\text{\AA}$ ,  $c = 15.463\text{\AA}$ <sup>(1)</sup>  
(Relationship to unit cell of Framework Type:  $a' = a$ ,  $b' = b$ ,  $c' = 2c$ )

**Framework density:** 16.7 T/1000 $\text{\AA}^3$

**Channels:** [001] 8 3.9 x 3.9\*



8-ring viewed along [001]

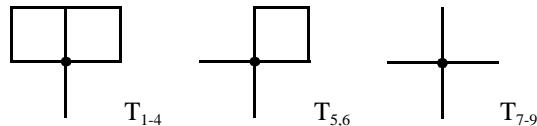


**Idealized cell constants:** tetragonal, P4<sub>1</sub>22,  $a = 12.6\text{\AA}$ ,  $c = 26.2\text{\AA}$

T <sub>1</sub> (8, 1)	4	10	19	32	51	77	105	133	167	207	4·5·4·12 <sub>3</sub> ·5·5
T <sub>2</sub> (8, 1)	4	10	19	32	51	75	102	133	170	208	4·5·4·12 <sub>6</sub> ·5·5
T <sub>3</sub> (8, 1)	4	10	21	32	49	76	109	137	170	207	4·6·4·12 <sub>3</sub> ·5·5
T <sub>4</sub> (8, 1)	4	10	21	32	49	74	105	139	173	204	4·6·4·12 <sub>6</sub> ·5·5
T <sub>5</sub> (8, 1)	4	11	18	29	48	80	107	133	160	203	4·5 <sub>2</sub> ·5·5·5·6
T <sub>6</sub> (8, 1)	4	11	18	29	48	77	106	134	160	204	4·5 <sub>2</sub> ·5·5·5·6
T <sub>7</sub> (8, 1)	4	12	18	31	51	76	109	133	164	210	5·5·5 <sub>2</sub> ·12 <sub>5</sub>
T <sub>8</sub> (4, 2)	4	12	19	32	48	75	112	134	164	206	5·5·5 <sub>2</sub> ·12 <sub>7</sub> ·6·6
T <sub>9</sub> (4, 2)	4	12	17	30	54	77	106	134	160	212	5·5·5·5 <sub>2</sub> ·12 <sub>3</sub>

**Secondary building units:** combinations only

**Loop configuration of T-Atoms:**



**Isotypic framework structures:**

\*Beta<sup>(1,2)</sup>  
[B-Si-O]-\*BEA<sup>(3,4)</sup>  
[Ga-Si-O]-\*BEA<sup>(4)</sup>

CIT-6<sup>(5)</sup>  
Tschernichite<sup>(6)</sup>

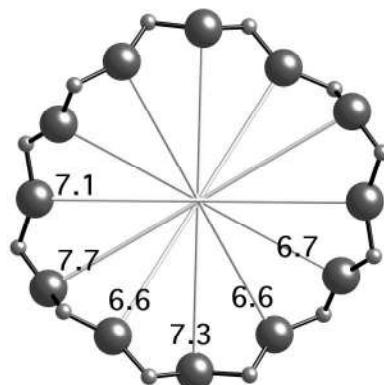
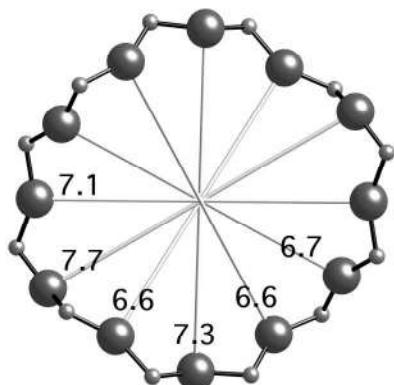
### References:

- (1) Higgins, J.B., LaPierre, R.B., Schlenker, J.L., Rohrman, A.C., Wood, J.D., Kerr, G.T. and Rohrbaugh, W.J. *Zeolites*, **8**, 446-452 (1988)
- (2) Newsam, J.M., Treacy, M.M.J., Koetsier, W.T. and de Gruyter, C.B. *Proc. R. Soc. Lond. A*, **420**, 375-405 (1988)
- (3) Marler, B., Böhme, R. and Gies, H. In *Proc. 9th Int. Zeolite Conf.*, (eds. R. von Ballmoos, J.B. Higgins and M.M.J. Treacy), pp. 425-432 (1993), Butterworth-Heinemann, Boston

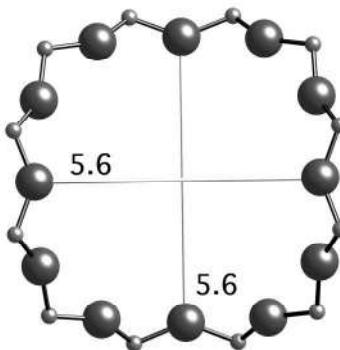
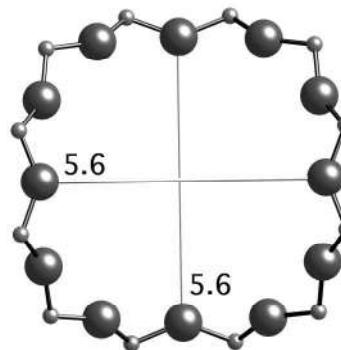
**Crystal chemical data:**  $[\text{Na}^+]_7 [\text{Al}_7\text{Si}_{57}\text{O}_{128}]$ -\***BEA**  
tetragonal,  $\text{P}4_122$ ,  $a = 12.661\text{\AA}$ ,  $c = 26.406\text{\AA}$  <sup>(2)</sup>

**Framework density:** 15.1 T/1000 $\text{\AA}^3$

**Channels:**  $<100> \textbf{12} \ 6.6 \times 6.7^{**} \leftrightarrow [001] \textbf{12} \ 5.6 \times 5.6^*$



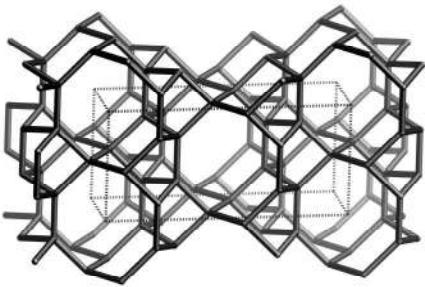
12-ring viewed along <100>

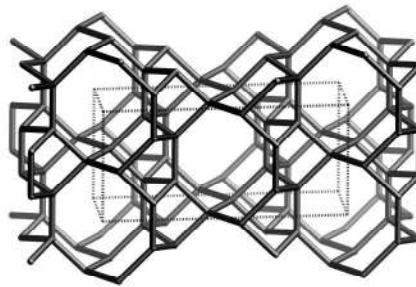


12-ring viewed along [001]

#### References (cont.):

- (4) Reddy, K.S.N., Eapen, M.J., Joshi, P.N., Mirajkar, S.P. and Shiralkar, V.P. *J. Incl. Phenom. Mol. Recogn. Chem.*, **20**, 197-210 (1994)
- (5) Takewaki, T., Beck, L.W. and Davis, M.E. *Topics in Catalysis*, **9**, 35-42 (1999)
- (6) Boggs, R.C., Howard, D.G., Smith, J.V. and Klein, G.L. *Am. Mineral.*, **78**, 822-826 (1993)





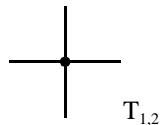
framework viewed along [001]

**Idealized cell constants:** orthorhombic, Cmcm,  $a = 7.5\text{\AA}$ ,  $b = 16.2\text{\AA}$ ,  $c = 5.3\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(8, m) \quad 4 \ 12 \ 23 \ 43 \ 71 \ 97 \ 128 \ 179 \ 226 \ 264$        $5 \cdot 5 \cdot 5 \cdot 6 \cdot 8_2$   
 $T_2(4, m2m) \quad 4 \ 12 \ 26 \ 42 \ 66 \ 102 \ 140 \ 164 \ 216 \ 288$        $5_2 \cdot 6_2 \cdot 6 \cdot 6 \cdot 6$

**Secondary building units:** 5-1

**Loop configuration of T-Atoms:**



**Isotypic framework structures:**

\*Bikitaite<sup>(1,2)</sup>  
|Cs-[Al-Si-O]-BIK<sup>(3)</sup>  
Triclinic bikitaite<sup>(4)</sup>

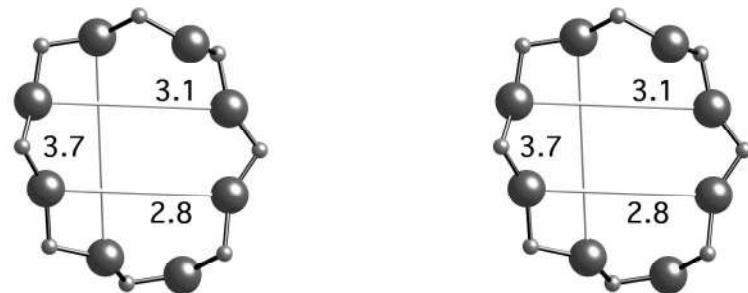
### References:

- (1) Kocman, V., Gait, R.I. and Rucklidge, J. Am. Mineral., **59**, 71-78 (1974)
- (2) Ståhl, K., Kvick, Å. and Ghose, S. Zeolites, **9**, 303-311 (1989)
- (3) Annehed, H. and Fälth, L. Z. Kristallogr., **166**, 301-306 (1984)
- (4) Bissert, G. and Liebau, L. N. Jb. Miner. Mh., 241-252 (1986)

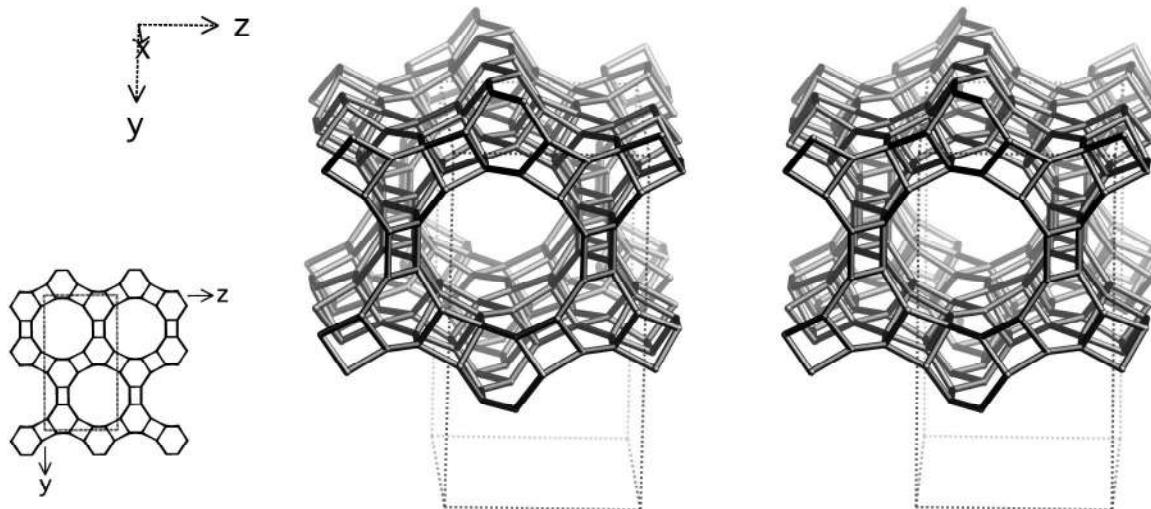
**Crystal chemical data:**  $[\text{Li}^+ \cdot (\text{H}_2\text{O})_2] \cdot [\text{Al}_2\text{Si}_4\text{O}_{12}]$ -BIK  
triclinic, P1,  $a = 8.607\text{\AA}$ ,  $b = 4.954\text{\AA}$ ,  $c = 7.597\text{\AA}$ ,  
 $\alpha = 89.90^\circ$ ,  $\beta = 114.44^\circ$ ,  $\gamma = 89.99^\circ$  <sup>(2)</sup>  
(Relationship to unit cell of Framework Type:  
 $a' = b/2 \sin(\beta)$ ,  $b' = c$ ,  $c' = a$   
or, as vectors,  $\mathbf{a}' = (\mathbf{b} - \mathbf{a})/2$ ,  $\mathbf{b}' = \mathbf{c}$ ,  $\mathbf{c}' = \mathbf{a}$ )

**Framework density:** 20.3 T/1000 $\text{\AA}^3$

**Channels:** [010] 8 2.8 x 3.7\*



8-ring viewed along [010]



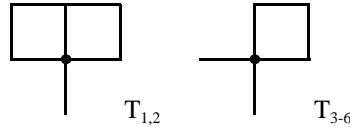
framework viewed along [100] (bottom left: projection down [100])

**Idealized cell constants:** orthorhombic, Imma,  $a = 20.0\text{\AA}$ ,  $b = 23.6\text{\AA}$ ,  $c = 12.7\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (16, 1)	4 10 19 32 51 74 101 129 158 199	4·5·4·6·5·12 <sub>2</sub>
	T <sub>2</sub> (16, 1)	4 10 20 32 48 74 104 131 159 195	4·5·4·6 <sub>2</sub> ·10·12
	T <sub>3</sub> (16, 1)	4 11 19 34 50 71 98 133 162 195	4·5 <sub>2</sub> ·5·6·5·10 <sub>3</sub>
	T <sub>4</sub> (16, 1)	4 11 21 32 49 74 101 128 162 200	4·10 <sub>5</sub> ·5·6 <sub>3</sub> ·5·6 <sub>3</sub>
	T <sub>5</sub> (16, 1)	4 11 20 31 53 76 97 126 168 199	4·5·5·6 <sub>2</sub> ·5·10
	T <sub>6</sub> (16, 1)	4 11 18 31 52 75 100 126 158 206	4·5·5·6·5·6 <sub>2</sub>

**Secondary building units:** 6 or 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Boggsite<sup>(1)</sup>

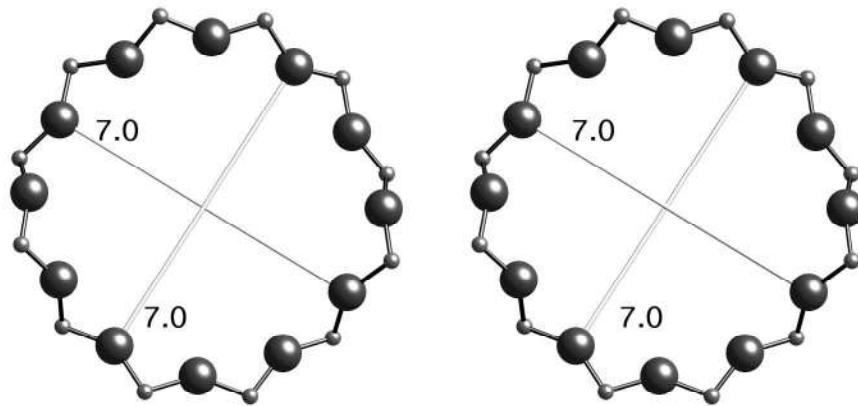
### References:

- (1) Pluth, J.J. and Smith, J.V. *Am. Mineral.*, **75**, 501-507 (1990)

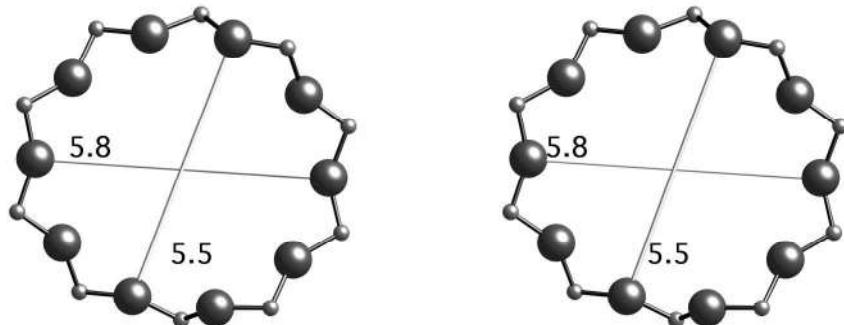
**Crystal chemical data:**  $[\text{Ca}^{2+}_7\text{Na}^+_4(\text{H}_2\text{O})_{74}] [\text{Al}_{18}\text{Si}_{78}\text{O}_{192}]$ -BOG  
orthorhombic, Imma,  $a = 20.236\text{\AA}$ ,  $b = 23.798\text{\AA}$ ,  $c = 12.798\text{\AA}$  <sup>(1)</sup>

**Framework density:** 15.6 T/1000 $\text{\AA}^3$

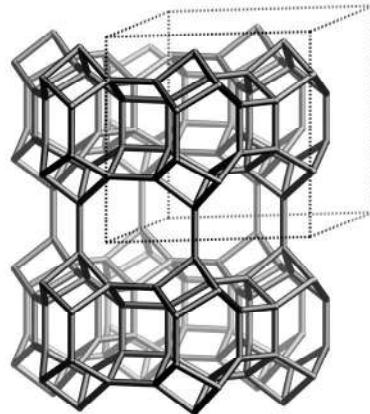
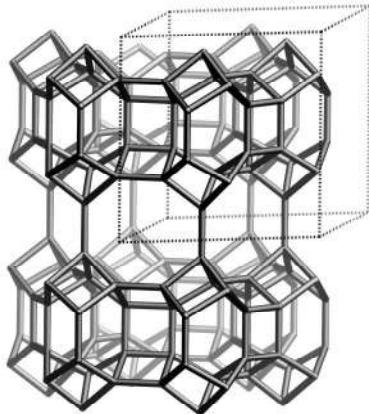
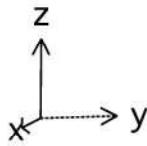
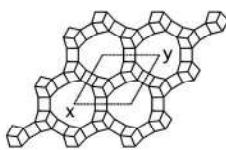
**Channels:** [100] **12** 7.0 x 7.0\*  $\leftrightarrow$  [010] **10** 5.5 x 5.8\*



12-ring viewed along [100]



10-ring viewed along [010]



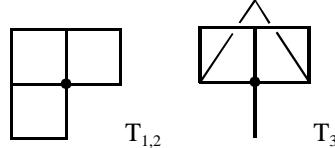
framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** hexagonal, P $\bar{6}2m$ ,  $a = 13.1\text{\AA}$ ,  $c = 13.0\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (12, 1)    4    9    17    28    42    60    84    113    140    169	4·4·4·8 <sub>2</sub> ·6 <sub>2</sub> ·8
	T <sub>2</sub> (12, 1)    4    9    16    25    39    61    86    111    141    173	4·4·4·6·6·12
	T <sub>3</sub> (4, 3)    4    9    18    30    43    62    85    105    135    180	4·8·4·8·4·8

**Secondary building units:** 6\*1

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*Beryllophosphate-H<sup>(1,2)</sup>  
Linde Q<sup>(3)</sup>  
STA-5<sup>(4)</sup>

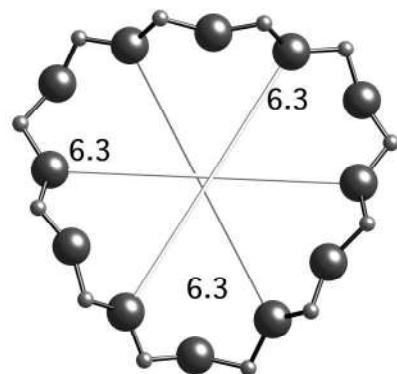
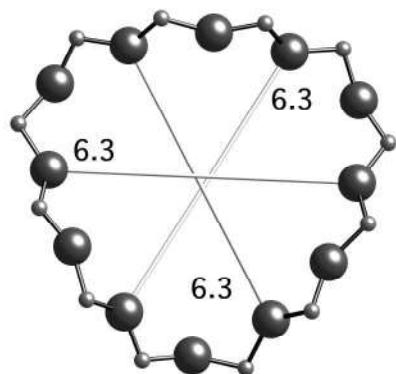
## References:

- (1) Harvey, G. Z. *Kristallogr.*, **182**, 123-124 (1988)
- (2) Harvey, G., Baerlocher, Ch. and Wroblewski, T. Z. *Kristallogr.*, **201**, 113-123 (1992)
- (3) Andries, K.J., Bosmans, H.J. and Grobet, P.J. *Zeolites*, **11**, 124-131 (1991)
- (4) Patinec, V., Wright, P.A., Aitken, R.A., Lightfoot, P., Purdie, S.D.J., Cox, P.A., Kvick, A. and Vaughan, G. *Chem. Mater.*, **11**, 2456-2462 (1999)

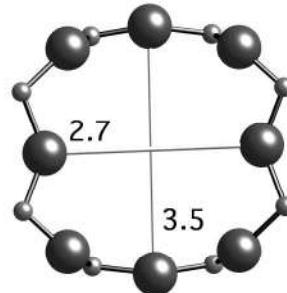
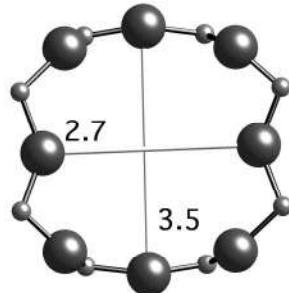
**Crystal chemical data:**  $|\text{K}^+ \text{Na}^+ \text{H}_2\text{O}_{20}| [\text{Be}_{14}\text{P}_{14}\text{O}_{56}]$ -BPH  
trigonal, P321,  $a = 12.582\text{\AA}$ ,  $c = 12.451\text{\AA}$ <sup>(2)</sup>

**Framework density:** 16.4 T/1000 $\text{\AA}^3$

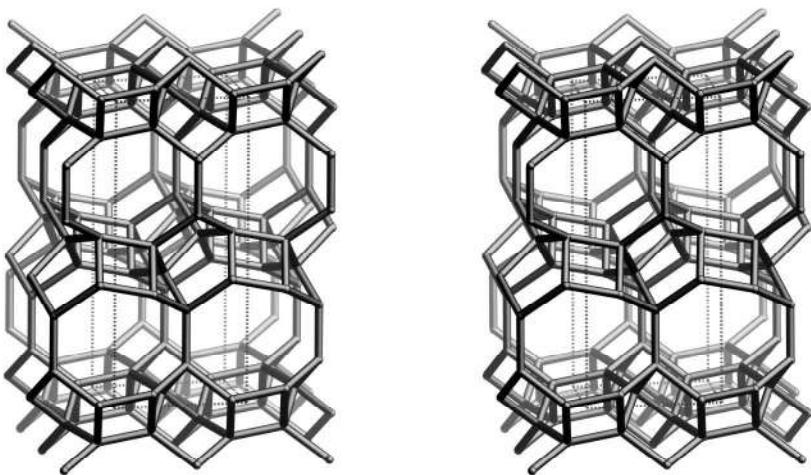
**Channels:** [001] **12** 6.3 x 6.3\*  $\leftrightarrow$   $\perp$  [001] **8** 2.7 x 3.5\*\*



12-ring viewed along [001]



8-ring viewed normal to [001]



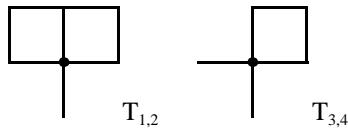
framework viewed along [100]

**Idealized cell constants:** monoclinic, P2<sub>1</sub>/m, a = 6.8 Å, b = 17.1 Å, c = 7.6 Å, β = 95.8°

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (4, 1)    4 10 20 37 61 83 110 144 192 238 T <sub>2</sub> (4, 1)    4 10 20 36 61 85 107 147 191 234 T <sub>3</sub> (4, 1)    4 11 23 37 54 82 119 152 184 233 T <sub>4</sub> (4, 1)    4 11 18 37 62 85 110 147 195 236	4·5·4·6·5·8 4·5·4·8·5·6 4·8·5·8·5·8 <sub>2</sub> 4·5 <sub>2</sub> ·5·6·5·8
---	--	---

**Secondary building units:** 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*Brewsterite<sup>(1,2)</sup>  
Ba-dominant brewsterite<sup>(3)</sup>  
CIT-4<sup>(4)</sup>  
Synthetic brewsterite<sup>(5)</sup>

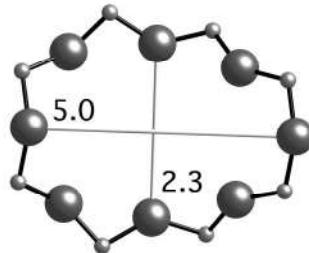
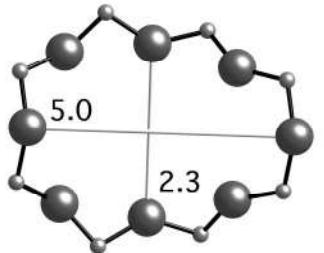
## References:

- (1) Perrotta, A.J. and Smith, J.V. *Acta Crystallogr.*, **17**, 857-862 (1964)
- (2) Schlenker, J.L., Pluth, J.J. and Smith, J.V. *Acta Crystallogr.*, **B33**, 2907-2910 (1977)
- (3) Cabella, R., Lucchetti, G., Palenzona, A., Quartieri, S. and Vezzalini, G. *Eur. J. Mineral.*, **5**, 353-360 (1993)
- (4) Khodabandeh, S., Lee, G. and Davis, M.E. *Microporous and Mesoporous Materials*, **11**, 87-95 (1997)
- (5) Ghobarkar, H. and Schaef, O. *German Patent AZ 198 24 184.4-41* (1999)

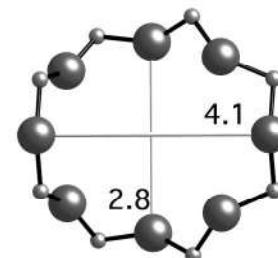
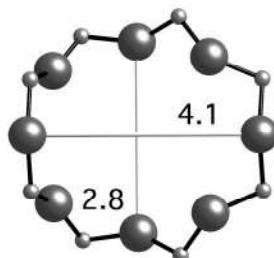
**Crystal chemical data:**  $[(\text{Ba}^{2+}, \text{Sr}^{2+})_2 (\text{H}_2\text{O})_{10}] [\text{Al}_4\text{Si}_{12} \text{O}_{32}]$ -BRE  
monoclinic, P12<sub>1</sub>/m1  
 $a = 6.793\text{\AA}$ ,  $b = 17.573\text{\AA}$ ,  $c = 7.759\text{\AA}$ ,  $\beta = 94.54^\circ$  <sup>(2)</sup>

**Framework density:** 17.3 T/1000 $\text{\AA}^3$

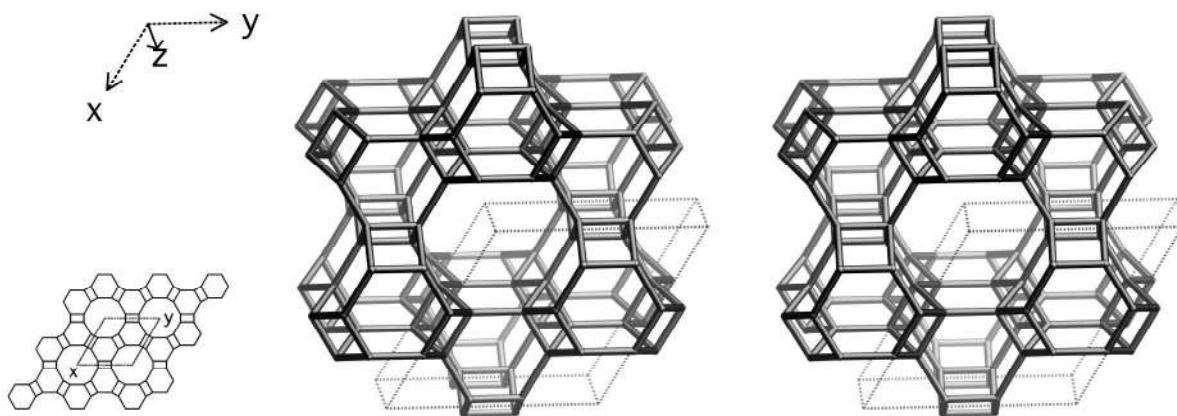
**Channels:** [100] 8 2.3 x 5.0\*  $\leftrightarrow$  [001] 8 2.8 x 4.1\*



8-ring viewed along [100]



8-ring viewed along [001]



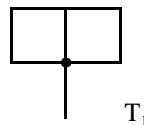
framework viewed along [001] (bottom left: projection down [001])

**Idealized cell constants:** hexagonal, P<sub>6</sub><sub>3</sub>/mmc,  $a = 12.5\text{\AA}$ ,  $c = 5.3\text{\AA}$

**Coordination sequences and vertex symbols:** T<sub>1</sub> (12, m) 4 10 20 34 54 78 104 134 168 210 4·6·4·6·6·6

**Secondary building units:** 4-2 or 6 or 4

**Loop configuration of T-Atoms:**



**Framework description:** AB sequence of 6-rings

**Isotypic framework structures:**

\*Cancrinite<sup>(1,2)</sup>  
[Al-Ge-O]-CAN<sup>(3)</sup>  
[Ga-Si-O]-CAN<sup>(4)</sup>  
[Zn-P-O]-CAN<sup>(5)</sup>  
Basic cancrinite<sup>(6,7)</sup>  
Cancrinite hydrate<sup>(8)</sup>

Davyne<sup>(9)</sup>  
ECR-5<sup>(10)</sup>  
Microsommite<sup>(11)</sup>  
Synthetic cancrinite<sup>(12)</sup>  
Tiptopite<sup>(13)</sup>  
Vishnevite<sup>(14)</sup>

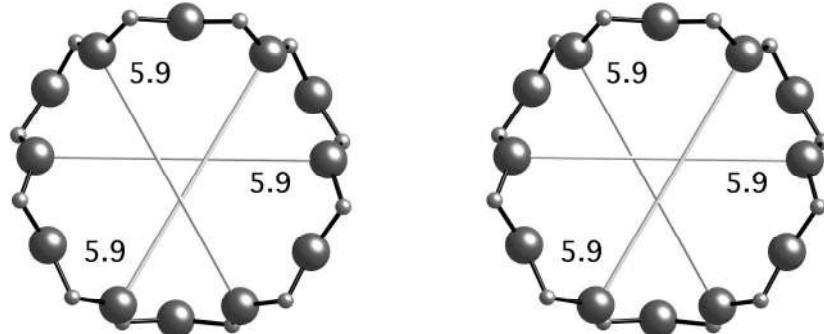
### References:

- (1) Pauling, L. *Proc. Natl. Acad. Sci.*, **16**, 453-459 (1930)
- (2) Jarchow, O. Z. *Kristallogr.*, **122**, 407-422 (1965)
- (3) Belokoneva, E.L., Uvarova, T.G. and Dem'yanets, L.N. *Sov. Phys. Crystallogr.*, **31**, 516-519 (1986)
- (4) Newsam, J.M. and Jorgensen, J.D. *Zeolites*, **7**, 569-573 (1987)
- (5) Yakubovich, O.V., Karimova, O.V. and Mel'nikov, O.K. *Crystallogr. Reports*, **39**, 564-568 (1994)
- (6) Barrer, R.M. and White, E.A.D. *J. Chem. Soc.*, 1561-1571 (1952)
- (7) Bresciana Pahor, N., Calligaris, M., Nardin, G. and Randaccio, L. *Acta Crystallogr.*, **B38**, 893-895 (1982)
- (8) Wyart, J. and Michel-Levy, M. *Compt. Rend.*, **229**, 131- (1949)
- (9) Hassan, I. and Grundy, H.D. *Can. Mineral.*, **28**, 341-349 (1990)
- (10) Vaughan, D.E.W. *E. Patent A-190,90* (1986)

**Crystal chemical data:**  $[\text{Na}^+ \text{Ca}^{2+} \text{CO}_3^{2-} (\text{H}_2\text{O})_2] [\text{Al}_6\text{Si}_6 \text{O}_{24}]$ -CAN  
hexagonal,  $P\bar{6}_3$ ,  $a = 12.75\text{\AA}$ ,  $c = 5.14\text{\AA}$ <sup>(2)</sup>

**Framework density:** 16.6 T/1000 $\text{\AA}^3$

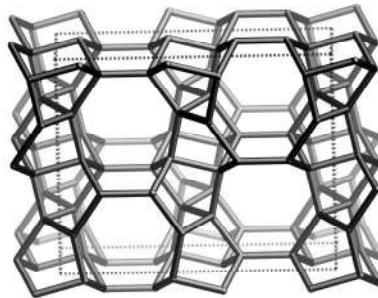
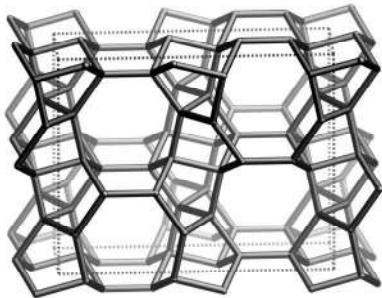
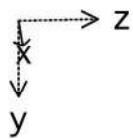
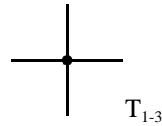
**Channels:** [001] 12 5.9 x 5.9\*



12-ring viewed along [001]

#### References (cont.):

- (11) Bonaccorsi, E., Comodi, P. and Merlini, S. *Phys. Chem. Mineral.*, **22**, 367-374 (1995)
- (12) Smolin, Y.I., Shepelev, Y.F., Butikova, I.K. and Kobyakov, I.B. *Kristallografiya*, **26**, 63-66 (1981)
- (13) Peacor, D.R., Rouse, R.C. and Ahn, J.-H. *Am. Mineral.*, **72**, 816-820 (1987)
- (14) Hassan, I. and Grundy, H.D. *Can. Mineral.*, **22**, 333-340 (1984)

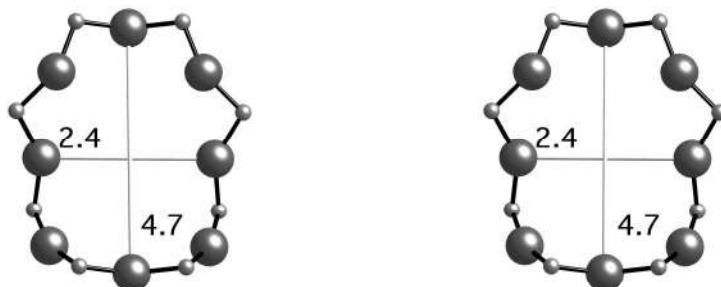
**CAS****Framework Type****Cmcm***framework viewed along [100]***Idealized cell constants:** orthorhombic, Cmcm,  $a = 5.3\text{\AA}$ ,  $b = 14.1\text{\AA}$ ,  $c = 17.2\text{\AA}$ **Coordination sequences  
and vertex symbols:**T<sub>1</sub> (8, m) 4 12 23 41 70 97 125 174 224 264 5·5·5·6·8<sub>2</sub>T<sub>2</sub> (8, m) 4 12 26 43 64 101 138 165 215 284 5·6·5·6·6<sub>2</sub>·8<sub>2</sub>T<sub>3</sub> (8, m) 4 12 23 43 72 95 128 177 225 259 5·6·5·6·5<sub>2</sub>·6**Secondary building units:** 5-1**Loop configuration of  
T-Atoms:****Isotypic framework  
structures:**\*Cesium Aluminosilicate<sup>(1)</sup>**References:**

- (1) Araki, T. Z. Kristallogr.,
- 152**
- , 207-213 (1980)

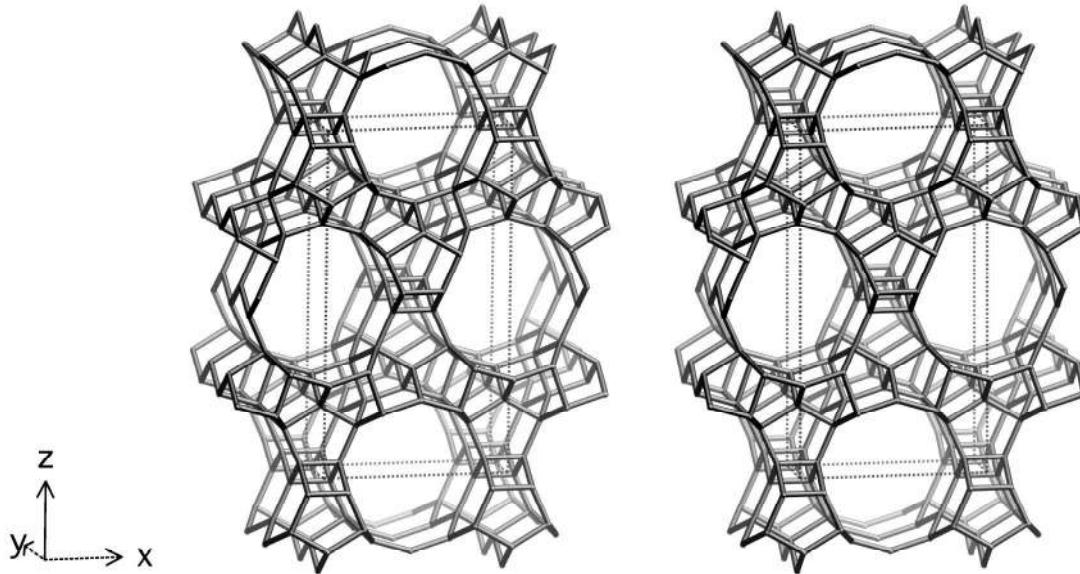
**Crystal chemical data:**  $|\text{Cs}_4^+| [\text{Al}_4\text{Si}_{20}\text{O}_{48}]\text{-CAS}$   
orthorhombic, Ama2,  $a = 16.776\text{\AA}$ ,  $b = 13.828\text{\AA}$ ,  $c = 5.021\text{\AA}$ <sup>(1)</sup>  
(Relationship to unit cell of Framework Type:  $\mathbf{a}' = -\mathbf{c}$ ,  $\mathbf{b}' = \mathbf{b}$ ,  $\mathbf{c}' = \mathbf{a}$ )

**Framework density:** 20.6 T/1000 $\text{\AA}^3$

**Channels:** [001] 8 2.4 x 4.7\*



8-ring viewed along [001]

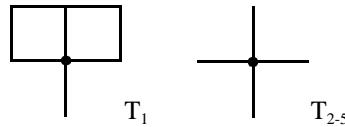


**Idealized cell constants:** orthorhombic, Imma,  $a = 14.0\text{\AA}$ ,  $b = 5.3\text{\AA}$ ,  $c = 26.0\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(8, m)$	4 10 21 36 56 84 114 143 182 231	4·6·4·6·5·6
	$T_2(8, m)$	4 12 23 37 55 83 114 153 195 222	5·6·5·6·5 <sub>2</sub> ·6
	$T_3(8, m)$	4 12 22 37 57 84 114 156 184 222	5·6·5·6·5·6 <sub>2</sub>
	$T_4(4, mm2)$	4 12 24 36 54 79 118 153 190 234	5·5·5·5·14 <sub>20</sub> *
	$T_5(4, mm2)$	4 12 20 34 56 81 116 151 186 220	5·5·5·5·6 <sub>2</sub>

**Secondary building units:** 5-3

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*CIT-5<sup>(1,2)</sup>

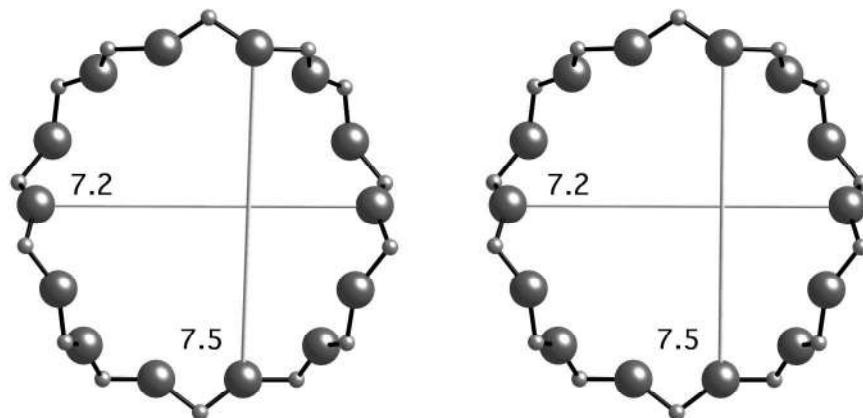
### References:

- (1) Wagner, P., Yoshikawa, M., Lovallo, M., Tsuji, K., Tsapatsis, M. and Davis, M.E. *Chem. Commun.*, 2179-2180 (1997)
- (2) Yoshikawa, M., Wagner, P., Lovallo, M., Tsuji, K., Takewaki, T., Chen, C.Y., Beck, L.W., Jones, C., Tsapatsis, M., Zones, S.I. and Davis, M.E. *J. Phys. Chem. B*, **102**, 7139-7147 (1998)

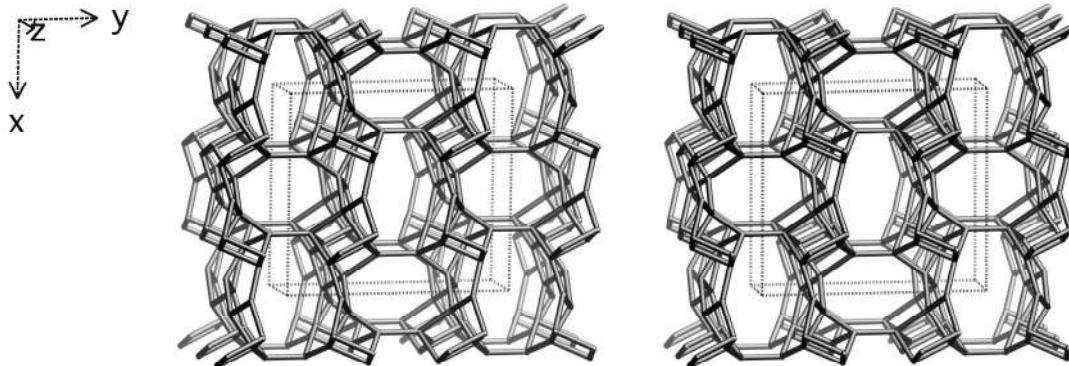
**Crystal chemical data:**  $[\text{Si}_{32}\text{O}_{64}]\text{-CFI}$   
orthorhombic,  $\text{Pmn}2_1$ ,  $a = 13.674\text{\AA}$ ,  $b = 5.022\text{\AA}$ ,  $c = 25.488\text{\AA}$ <sup>(2)</sup>

**Framework density:**  $18.3 \text{ T}/1000\text{\AA}^3$

**Channels:** [010] **14**  $7.2 \times 7.5^*$



*14-ring viewed along [010]*

**CGF****Framework Type****C12/m1**

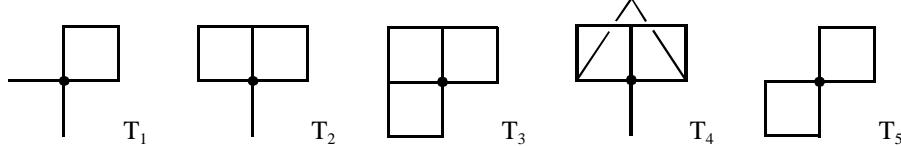
framework viewed along [001]

**Idealized cell constants:** monoclinic, C12/m1,  $a = 15.5\text{\AA}$ ,  $b = 16.9\text{\AA}$ ,  $c = 7.3\text{\AA}$ ,  $\beta = 96.1^\circ$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (8, 1)    4 11 22 34 50 76 111 142 165 199 T <sub>2</sub> (8, 1)    4 10 19 33 55 79 100 129 172 216 T <sub>3</sub> (8, 1)    4 9 18 34 55 76 97 131 177 217 T <sub>4</sub> (8, 1)    4 9 18 34 55 75 98 133 177 216 T <sub>5</sub> (4, 2)    4 10 18 32 58 80 96 124 176 228	4·8·6 <sub>3</sub> ·8·6 <sub>3</sub> ·8 4·6·4·6 <sub>2</sub> ·6·6 4·4·4·6·6·6 <sub>2</sub> 4·6 <sub>3</sub> ·4·6 <sub>3</sub> ·4·8 4·4·6 <sub>2</sub> ·6 <sub>2</sub> ·10·10
---	--	--

**Secondary building units:** 4

**Loop configuration of  
T-Atoms:**

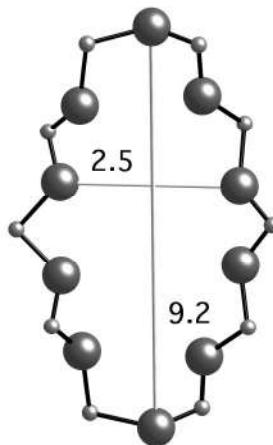
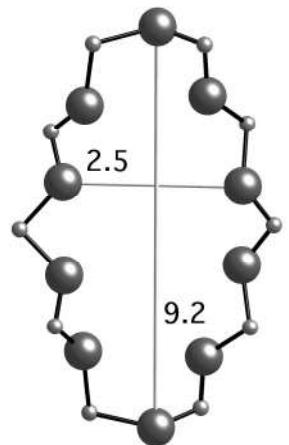


**Isotypic framework  
structures:** \*Co-Ga-Phosphate-5<sup>(1)</sup>

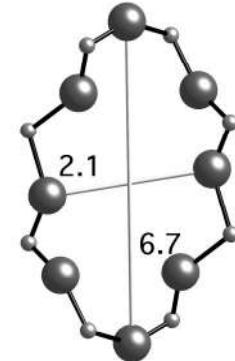
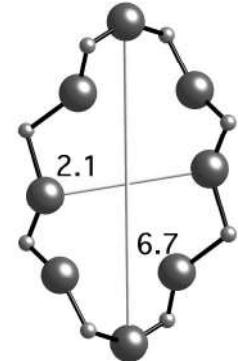
### References:

- (1) Chippindale, A.M. and Cowley, A.R. *Zeolites*, **18**, 176-181 (1997)

<b>Crystal chemical data:</b>	$[(C_6H_{14}N_2^{2+})_2][Co_4Ga_5P_9O_{36}]$ -CGF $C_6H_{12}N_2$ = DABCO monoclinic, I12/a1 $a = 15.002\text{\AA}$ , $b = 17.688\text{\AA}$ , $c = 15.751\text{\AA}$ , $\beta = 97.24^\circ$ <sup>(1)</sup> (Relationship to unit cell of Framework Type: $a' = 2c$ , $b' = b$ , $c' = a$ )
<b>Framework density:</b>	17.4 T/1000 $\text{\AA}^3$
<b>Channels:</b>	{[100] <b>10</b> 2.5 x 9.2* + <b>8</b> 2.1 x 6.7*} $\leftrightarrow$ [001] <b>8</b> 2.4 x 4.8*

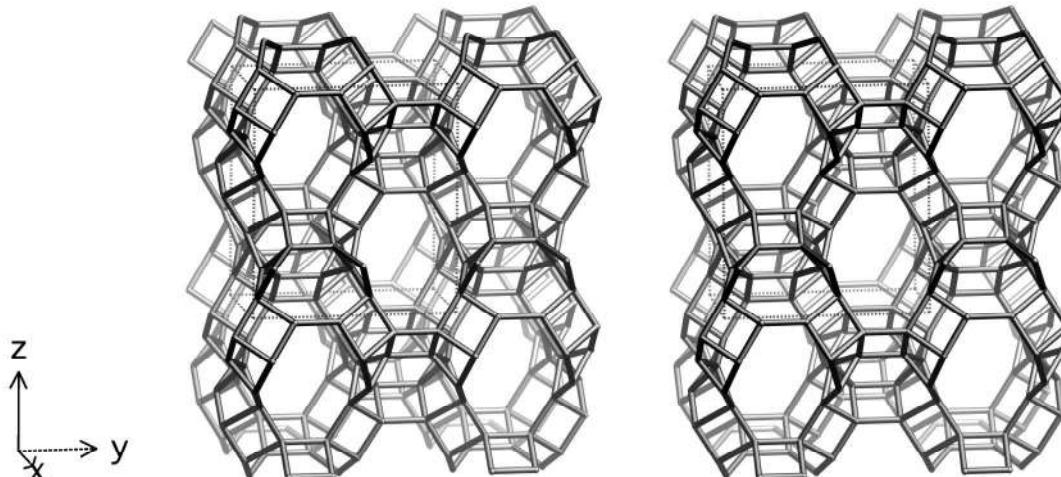


10-ring viewed along [100]



8-ring viewed along [100]

See Appendix A for 8-ring viewed along [001]



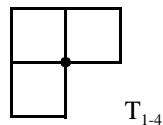
framework viewed along [100]

**Idealized cell constants:** orthorhombic, Pnma,  $a = 8.4\text{\AA}$ ,  $b = 14.1\text{\AA}$ ,  $c = 15.9\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(8, 1)$	4 9 16 26 43 67 91 116 148 188	4·4·4·6·6·8
	$T_2(8, 1)$	4 9 18 32 48 66 91 121 150 184	4·4·4·10 <sub>2</sub> ·8·8 <sub>6</sub>
	$T_3(8, 1)$	4 9 17 28 45 66 91 119 148 186	4·4·4·8 <sub>2</sub> ·6·10
	$T_4(8, 1)$	4 9 18 32 48 67 91 119 151 185	4·4·4·8·8 <sub>4</sub> ·10

**Secondary building units:** 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*Co-Ga-Phosphate-6<sup>(1)</sup>  
[Zn-Ga-P-O]-CGS<sup>(1)</sup>  
TNU-1, [Ga-Si-O]-CGS<sup>(2)</sup>  
TsG-1, [Ga-Si-O]-CGS<sup>(3)</sup>

### References:

- (1) Cowley, A.R. and Chippindale, A.M. *Microporous and Mesoporous Materials*, **28**, 163-172 (1999)
- (2) Hong, S.B., Kim, S.H., Kim, Y.G., Kim, Y.C., Barrett, P.A. and Camblor, M.A. *J. Mater. Chem.*, **9**, 2287-2289 (1999)
- (3) Lee, Y.J., Kim, S.J., Wu, G. and Parise, J.B. *Chem. Mater.*, **11**, 879-880 (1999)

**Crystal chemical data:**  $[(C_7H_{14}N^+)_4][Co_4Ga_{12}P_{16}O_{64}]$ -CGS

$C_7H_{13}N$  = quinuclidine

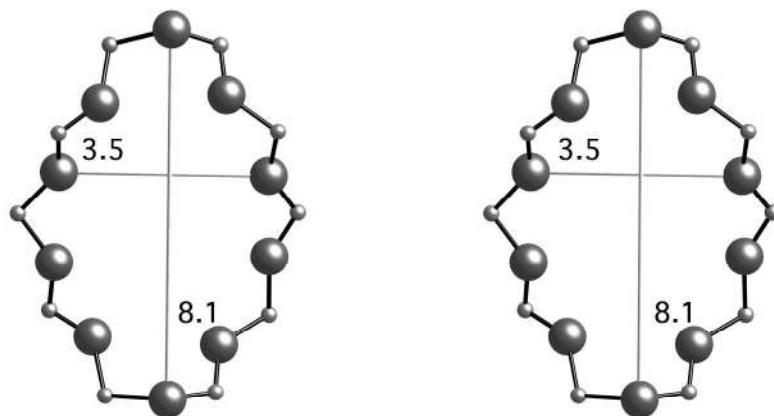
monoclinic,  $P12_1/c1$

$a = 14.365\text{\AA}$ ,  $b = 16.305\text{\AA}$ ,  $c = 8.734\text{\AA}$ ,  $\beta = 90.24^\circ$  <sup>(1)</sup>

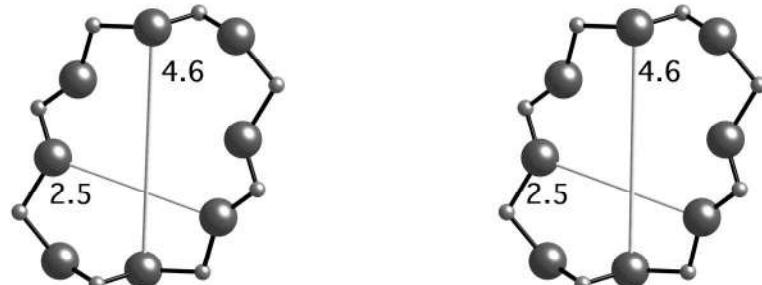
(Relationship to unit cell of Framework Type:  $a' = b$ ,  $b' = c$ ,  $c' = a$ )

**Framework density:**  $15.6 \text{ T}/1000\text{\AA}^3$

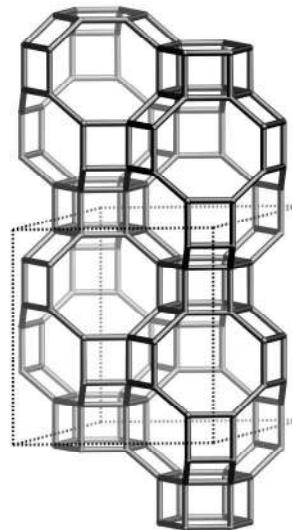
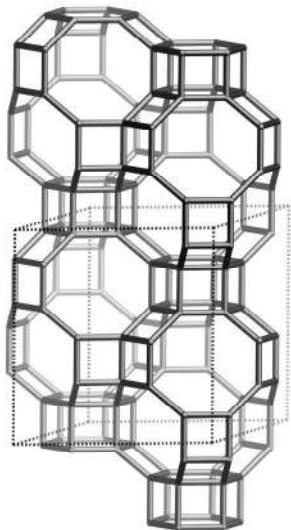
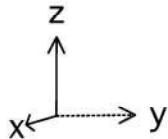
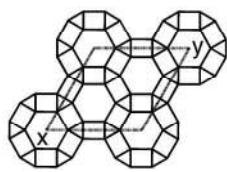
**Channels:**  $\{[001] \textbf{10} 3.5 \times 8.1 \leftrightarrow [100] \textbf{8} 2.5 \times 4.6\}^{***}$



10-ring viewed along [001]



8-ring viewed along [100]



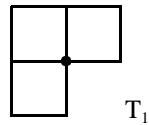
*framework viewed normal to [001] (top left: projection down [001])*

**Idealized cell constants:** trigonal, R $\bar{3}$ m,  $a = 13.7\text{\AA}$ ,  $c = 14.8\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(36, 1) \quad 4 \quad 9 \quad 17 \quad 29 \quad 45 \quad 64 \quad 85 \quad 110 \quad 140 \quad 173$       4·4·4·8·6·8

**Secondary building units:** 6-6 or 6 or 4

**Loop configuration of T-Atoms:**



**Framework description:** AABC sequence of 6-rings

**Isotypic framework structures:**

\*Chabazite<sup>(1,2)</sup>  
 $[\text{Al-Co-P-O}]\text{-CHA}^{(3)}$   
 $[\text{Co-Al-P-O}]\text{-CHA}^{(3,4)}$   
 $[\text{Mg-Al-P-O}]\text{-CHA}^{(4)}$   
 AlPO-34<sup>(5)</sup>  
 CoAPO-44<sup>(6)</sup>  
 CoAPO-47<sup>(6)</sup>  
 Dehydrated Na-Chabazite<sup>(7)</sup>  
 GaPO-34<sup>(8)</sup>  
 LZ-218<sup>(9)</sup>  
 Linde D<sup>(10,11)</sup>

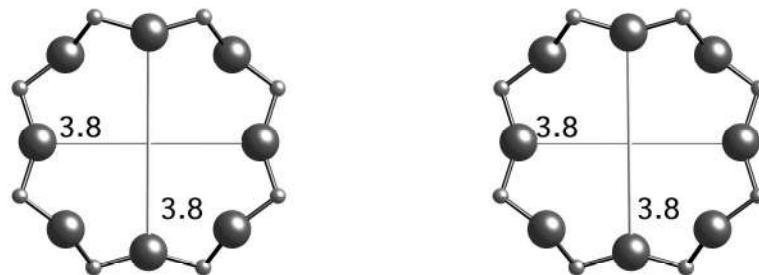
Linde R<sup>(12)</sup>  
 MeAPO-47<sup>(6,13,14)</sup>  
 MeAPSO-47<sup>(6,13,14)</sup>  
 Phi<sup>(11,15)</sup>  
 SAPO-34<sup>(16)</sup>  
 SAPO-47<sup>(17)</sup>  
 Si-CHA<sup>(18)</sup>  
 Willhendersonite<sup>(19)</sup>  
 ZK-14<sup>(20,21)</sup>  
 ZYT-6<sup>(22)</sup>

**Alternate designation:** Herschelite (discredited)

**Crystal chemical data:**  $[\text{Ca}^{2+}_6 (\text{H}_2\text{O})_{40}] [\text{Al}_{12}\text{Si}_{24} \text{O}_{72}]$ -CHA  
rhombohedral,  $\text{R}\bar{3}\text{m}$ ,  $a = 9.42\text{\AA}$ ,  $\alpha = 94.47^\circ$  <sup>(2)</sup>

**Framework density:** 14.5 T/1000 $\text{\AA}^3$

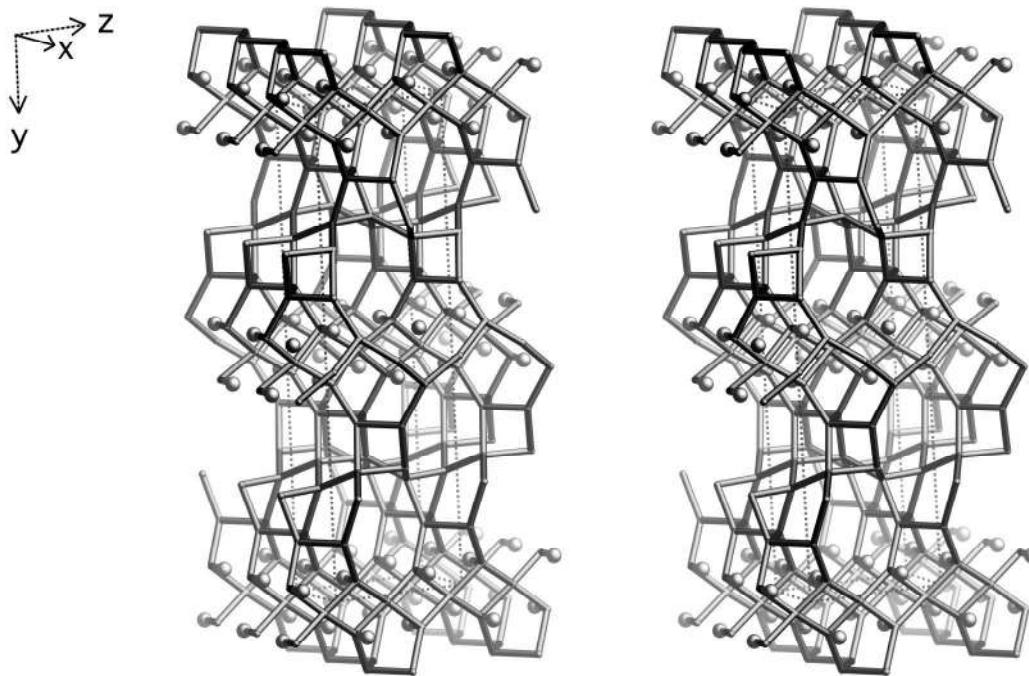
**Channels:**  $\perp [001]$  8 3.8 x 3.8\*\*\* (variable due to considerable flexibility of framework)



8-ring viewed normal to [001]

### References:

- (1) Dent, L.S. and Smith, J.V. *Nature*, **181**, 1794-1796 (1958)
- (2) Smith, J.V., Rinaldi, R. and Dent Glasser, L.S. *Acta Crystallogr.*, **16**, 45-53 (1963)
- (3) Feng, P.Y., Bu, X.H. and Stucky, G.D. *Nature*, **388**, 735-741 (1997)
- (4) Feng, P., Bu, X., Gier, T.E. and Stucky, G.D. *Microporous and Mesoporous Materials*, **23**, 221-229 (1998)
- (5) Harding, M.M. and Kariuki, B.M. *Acta Crystallogr.*, **C50**, 852-854 (1994)
- (6) Bennett, J.M. and Marcus, B.K. *Stud. Surf. Sci. Catal.*, **37**, 269-279 (1988)
- (7) Mortier, W.J., Pluth, J.J. and Smith, J.V. *Mater. Res. Bull.*, **12**, 241-250 (1977)
- (8) Schott-Darie, C., Kessler, H., Soulard, M., Gramlich, V. and Benazzi, E. *Stud. Surf. Sci. Catal.*, **84**, 101-108 (1994)
- (9) Breck, D.W. and Skeels, G.W. *U.S. Patent* 4,333,859 (1982)
- (10) Breck, D.W. and Acara, N.A. *U.S. Patent* 2,950,952 (1960)
- (11) Lillerud, K.P., Szostak, R. and Long, A. *J. Chem. Soc., Faraday Trans.*, **90**, 1547-1551 (1994)
- (12) Milton, R.M. *Brit. Patent* 841,812 (1960)
- (13) Flanigen, E.M., Lok, B.M., Patton, R.L. and Wilson, S.T. *Pure Appl. Chem.*, **58**, 1351-1358 (1986)
- (14) Flanigen, E.M., Lok, B.M., Patton, R.L. and Wilson, S.T. In *Proc. 7th Int. Zeolite Conf.*, (eds. Y. Murakami, A. Iijima and J.W. Ward), pp. 103-112 (1986), Kodansha, Tokyo
- (15) Grose, R.W. and Flanigen, E.M. *U.S. Patent* 4,124,686 (1978)
- (16) Lok, B.M., Messina, C.A., Patton, R.L., Gajek, R.T., Cannan, T.R. and Flanigen, E.M. *J. Am. Chem. Soc.*, **106**, 6092-6093 (1984)
- (17) Pluth, J.J. and Smith, J.V. *J. Phys. Chem.*, **93**, 6516-6520 (1989)
- (18) Díaz-Cabañas, M.J., Barrett, P.A. and Camblor, M.A. *Chem. Commun.*, 1881-1882 (1998)
- (19) Tillmanns, E., Fischer, R.X. and Baur, W.H. *N. Jb. Miner. Mh.*, 547-558 (1984)
- (20) Kuehl, G.H. *private communication*
- (21) Kuehl, G.H. In *Molecular Sieves*, (ed. R.M. Barrer), pp. 85-91 (1968), Soc. Chem. Indus., London
- (22) Ito, M., Shimoyama, Y., Saito, Y., Tsurita, Y. and Otake, M. *Acta Crystallogr.*, **C41**, 1698-1700 (1985)



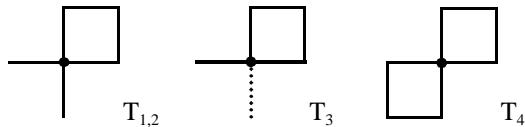
framework viewed along [100]

**Idealized cell constants:** orthorhombic, Pbcn,  $a = 5.0\text{\AA}$ ,  $b = 31.2\text{\AA}$ ,  $c = 9.0\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(8, 1)$	4 11 22 40 64 89 120 160 203 248	$4 \cdot 6_3 \cdot 6 \cdot 6_2 \cdot 6_2$
	$T_2(8, 1)$	4 10 20 36 60 86 115 157 196 238	$4 \cdot 5 \cdot 6 \cdot 9 \cdot 6_3 \cdot 10_2$
	$T_3(8, 1)$	3 8 13 29 53 80 113 147 193 231	4-5-9
	$T_4(4, 2)$	4 6 14 28 56 80 114 152 190 236	4-4-5-9-10-10

**Secondary building units:** 5-2

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*Chiavennite<sup>(1)</sup>

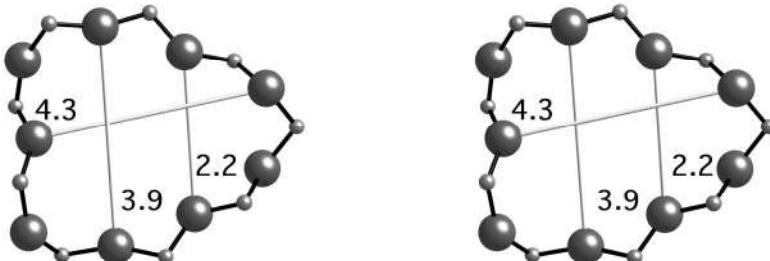
### References:

- (1) Tazzoli, V., Domeneghetti, M.C., Mazzi, F. and Cannillo, E. *Eur. J. Mineral.*, **7**, 1339-1344 (1995)

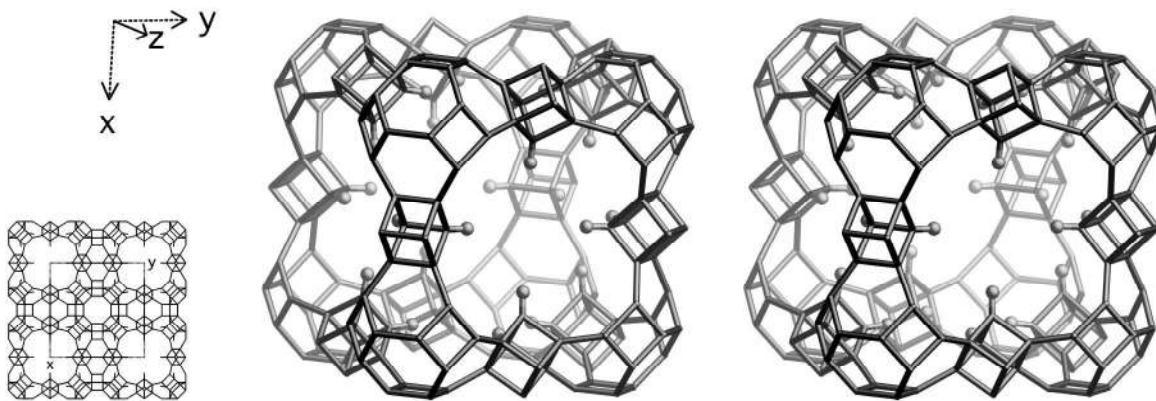
**Crystal chemical data:**  $[\text{Ca}^{2+}_4 \text{Mn}^{2+}_4 (\text{H}_2\text{O})_8] [\text{Be}_8\text{Si}_{20} \text{O}_{52}(\text{OH})_8]$ -**-CHI**  
orthorhombic, Pnab,  $a = 8.729\text{\AA}$ ,  $b = 31.326\text{\AA}$ ,  $c = 4.903\text{\AA}$ <sup>(1)</sup>  
(Relationship to unit cell of Framework Type:  $\mathbf{a}' = -\mathbf{c}$ ,  $\mathbf{b}' = \mathbf{b}$ ,  $\mathbf{c}' = \mathbf{a}$ )

**Framework density:** 20.9 T/1000 $\text{\AA}^3$

**Channels:** [001] **9** 3.9 x 4.3\*



*9-ring viewed along [001]*



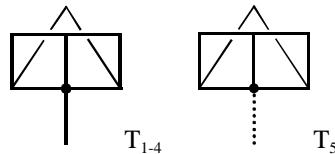
framework viewed along [001] (bottom left: projection down [001])

**Idealized cell constants:** cubic, Pm $\bar{3}$ m,  $a = 25.8\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (48, 1) 4 9 16 23 31 44 59 74 91 109	4·6·4·6·4·12
	T <sub>2</sub> (48, 1) 4 9 17 27 37 47 56 66 80 99	4·6·4·6·4·8
	T <sub>3</sub> (48, 1) 4 8 13 22 34 44 55 72 94 117	4·6·4·6·4·20 <sub>8</sub>
	T <sub>4</sub> (24, m) 4 9 16 23 32 45 58 76 98 118	4·6·4·6·4·8
	T <sub>5</sub> (24, m) 3 5 10 18 29 45 56 65 86 110	4·4·4..

**Secondary building units:** 4-4 or 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Cloverite<sup>(1)</sup>

### References:

- (1) Estermann, M., McCusker, L.B., Baerlocher, Ch., Merrouche, A. and Kessler, H. *Nature*, **352**, 320-323 (1991)

**Crystal chemical data:**  $[(\text{C}_7\text{H}_{14}\text{N}^+)_2]_8 [\text{F}_{24} \text{Ga}_{96}\text{P}_{96} \text{O}_{372}(\text{OH})_{24}]_8^-$  -CLO

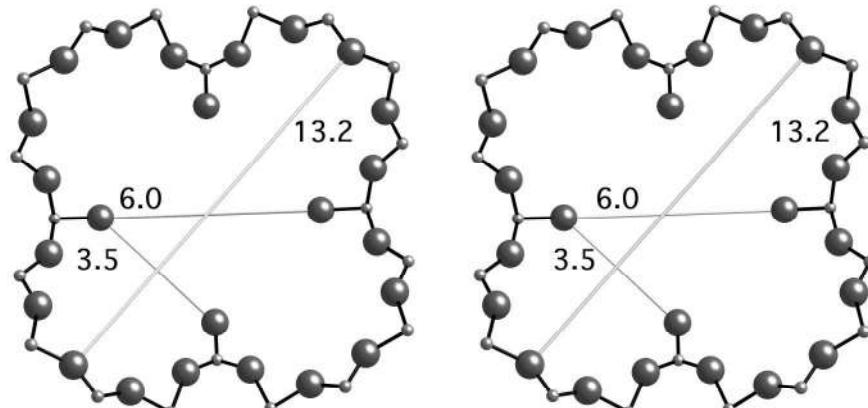
$\text{C}_7\text{H}_{13}\text{N}$  = quinuclidine

cubic,  $\text{Fm}\bar{3}\text{c}$ ,  $a = 51.712\text{\AA}$  <sup>(1)</sup>

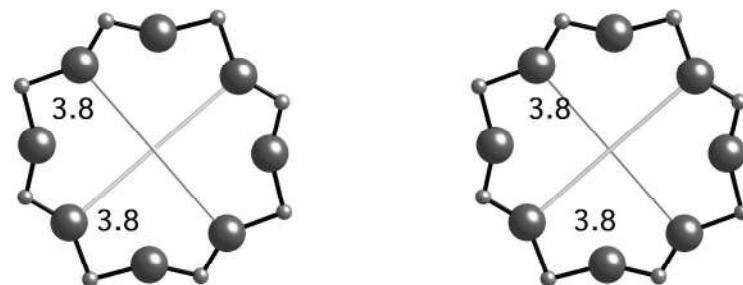
(Relationship to unit cell of Framework Type:  $a' = 2a$ )

**Framework density:**  $11.1 \text{ T}/1000\text{\AA}^3$

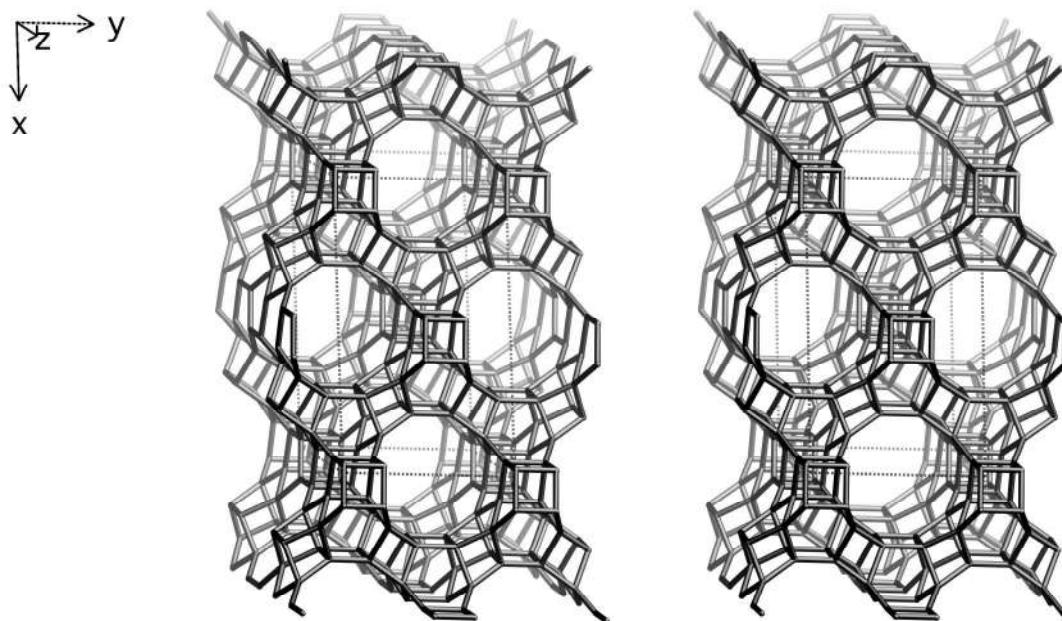
**Channels:**  $<100> \textbf{20} 4.0 \times 13.2^{***} | <100> \textbf{8} 3.8 \times 3.8^{***}$



20-ring viewed along  $<100>$



8-ring viewed along  $<100>$



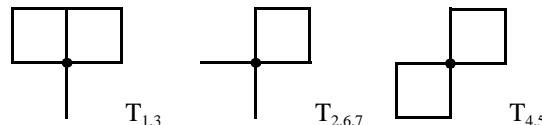
framework viewed along [001]

**Idealized cell constants:** monoclinic, C2/m,  $a = 22.7\text{\AA}$ ,  $b = 13.4\text{\AA}$ ,  $c = 12.6\text{\AA}$ ,  $\beta = 69.5^\circ$

Coordination sequences and vertex symbols:	T <sub>1</sub> (8, 1)	4 10 19 32 50 73 101 132 164 199	4·5·4·10 <sub>4</sub> ·5·6 <sub>2</sub>
	T <sub>2</sub> (8, 1)	4 11 18 31 52 77 98 126 164 205	4·5·5·6·5·10 <sub>2</sub>
	T <sub>3</sub> (8, 1)	4 10 21 32 47 74 105 134 159 196	4·6·4·10 <sub>4</sub> ·5·6 <sub>2</sub>
	T <sub>4</sub> (8, 1)	4 10 19 32 51 74 100 130 165 203	4·4·5·6 <sub>2</sub> ·5·12 <sub>7</sub>
	T <sub>5</sub> (8, 1)	4 10 19 32 51 74 101 130 164 203	4·4·5·6 <sub>2</sub> ·5·12 <sub>4</sub>
	T <sub>6</sub> (8, 1)	4 11 18 28 49 77 103 126 155 201	4·5 <sub>2</sub> ·5·6 <sub>2</sub> ·6·6 <sub>2</sub>
	T <sub>7</sub> (8, 1)	4 11 19 32 49 75 105 131 159 196	4·6 <sub>2</sub> ·5·6·5·10 <sub>2</sub>

**Secondary building units:** 5-2

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*CIT-1<sup>(1)</sup>  
SSZ-26 (contains major structural units of CON)<sup>(1,2)</sup>  
SSZ-33 (contains major structural units of CON)<sup>(1,2)</sup>

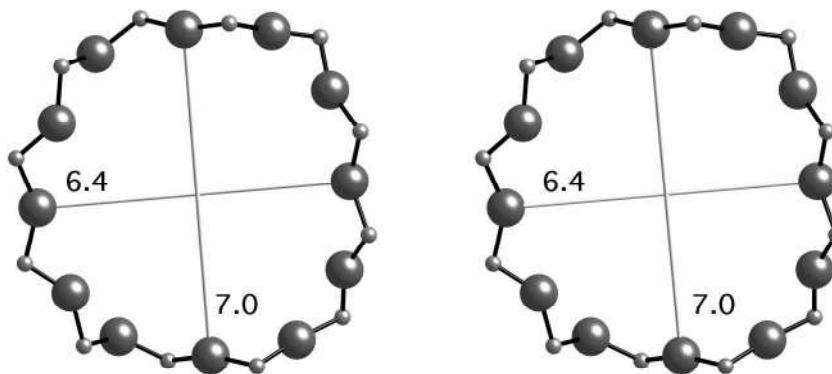
(1) Lobo, R.F. and Davis, M.E. *J. Am. Chem. Soc.*, **117**, 3764-3779 (1995)

(2) Lobo, R.F., Pan, M., Chan, I., Li, H.X., Medrud, R.C., Zones, S.I., Crozier, P.A. and Davis, M.E. *Science*, **262**, 1543-1546 (1993)

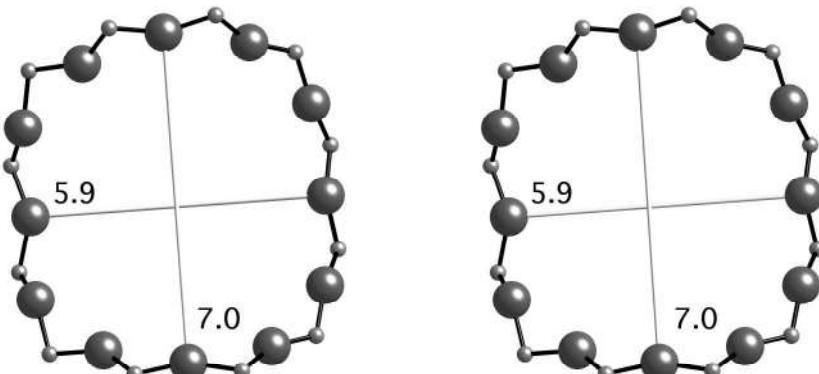
**Crystal chemical data:**  $[\text{H}^+]_2[\text{B}_2\text{Si}_{54}\text{O}_{112}]\text{-CON}$   
monoclinic, C12/m1  
 $a = 22.624\text{\AA}$ ,  $b = 13.350\text{\AA}$ ,  $c = 12.364\text{\AA}$ ,  $\beta = 68.91^\circ$  <sup>(1)</sup>

**Framework density:** 16.1 T/1000 $\text{\AA}^3$

**Channels:** [001] **12** 6.4 x 7.0\*  $\leftrightarrow$  [100] **12** 7.0 x 5.9\*  $\leftrightarrow$  [010] **10** 5.1 x 4.5\*

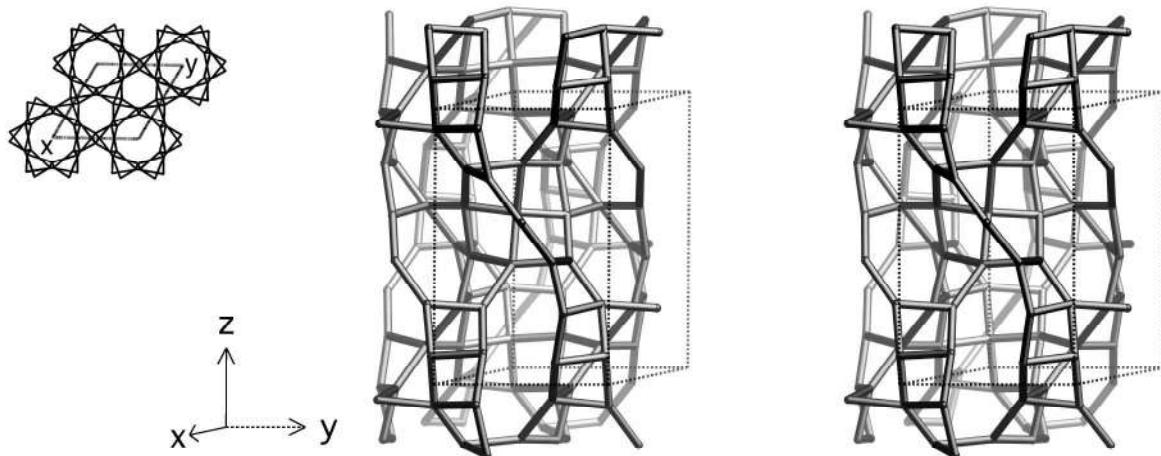


12-ring viewed along [001]



12-ring viewed along [100]

See Appendix A for 10-ring viewed along [010]



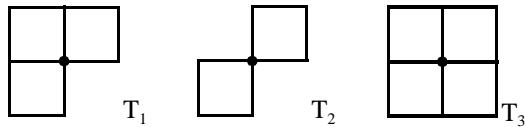
framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** hexagonal, P6<sub>1</sub>22,  $a = 9.4\text{\AA}$ ,  $c = 15.3\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (12, 1)    4    9    18    32    54    83    113    149    191    234	4·4·4·8 <sub>6</sub> ·8·8
	T <sub>2</sub> (6, 2)    4    10    20    33    56    85    114    144    192    242	4·4·8 <sub>3</sub> ·8 <sub>3</sub> ·8 <sub>6</sub> ·8 <sub>6</sub>
	T <sub>3</sub> (6, 2)    4    8    16    33    52    73    112    160    190    214	4·4·4·4·8·8

**Secondary building units:** 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*Chiral Zincophosphate<sup>(1,2)</sup>

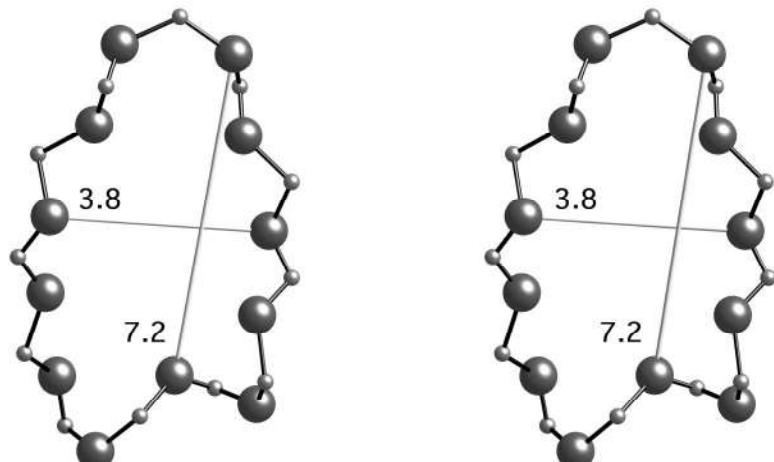
### References:

- (1) Rajic, N., Logar, N.Z. and Kaucic, V. *Zeolites*, **15**, 672-678 (1995)
- (2) Harrison, W.T.A., Gier, T.E., Stucky, G.D., Broach, R.W. and Bedard, R.A. *Chem. Mater.*, **8**, 145-151 (1996)

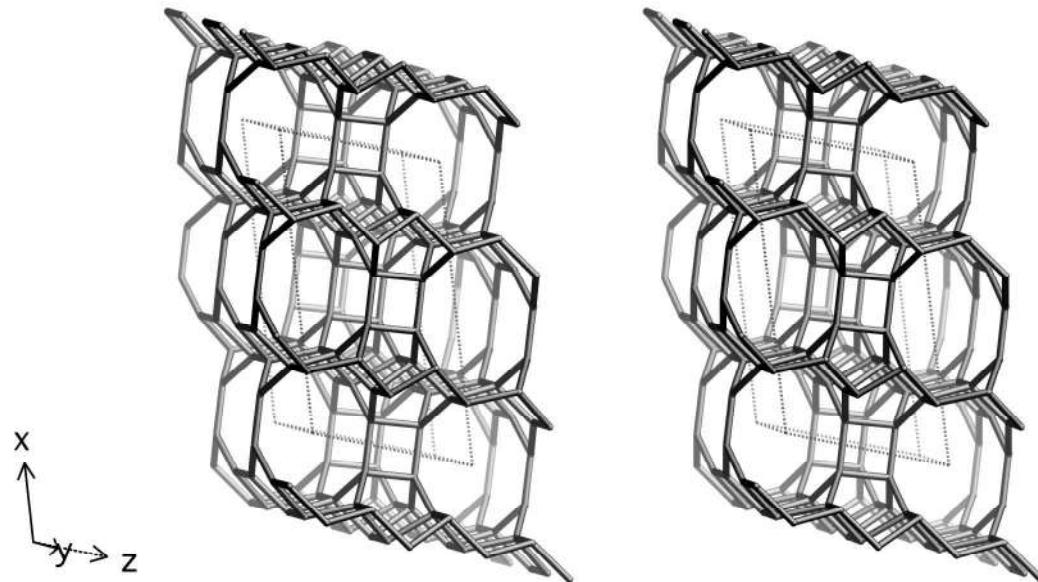
**Crystal chemical data:**  $[\text{Na}^+_{12} (\text{H}_2\text{O})_{12}] [\text{Zn}_{12}\text{P}_{12} \text{O}_{48}]$ -CZP  
hexagonal, P6<sub>1</sub>22,  $a = 10.480\text{\AA}$ ,  $c = 15.089\text{\AA}$ <sup>(2)</sup>

**Framework density:** 16.7 T/1000 $\text{\AA}^3$

**Channels:** [001] **12** 3.8 x 7.2\* (highly distorted 12-ring)



*distorted 12-ring viewed along [001]*



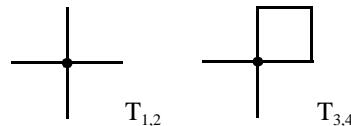
framework viewed along [001]

**Idealized cell constants:** monoclinic, C2/m,  $a = 18.6\text{\AA}$ ,  $b = 7.5\text{\AA}$ ,  $c = 10.4\text{\AA}$ ,  $\beta = 108.9^\circ$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (8, 1)    4 12 22 39 65 91 121 163 208 250	5·5·5·5 <sub>2</sub> ·8·10 <sub>2</sub>
	T <sub>2</sub> (8, 1)    4 12 20 37 63 91 118 164 212 245	5·5·5·5 <sub>2</sub> ·5·8
	T <sub>3</sub> (4, m)    4 11 24 41 59 99 130 155 202 262	4·5 <sub>2</sub> ·5·8·5·8
	T <sub>4</sub> (4, m)    4 11 24 39 63 95 132 156 199 266	4·5 <sub>2</sub> ·5·8·5·8

**Secondary building units:** 5-1

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Dachiardite<sup>(1,2)</sup>

**Alternate designation:** Svetlozarite (discredited)<sup>(3)</sup>

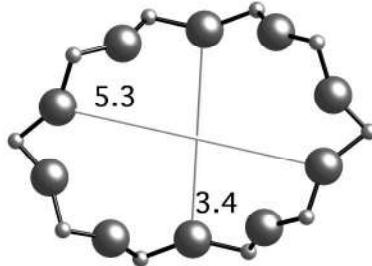
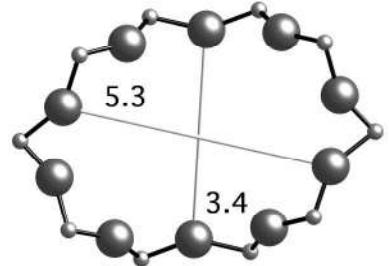
#### References:

- (1) Gottardi, G. and Meier, W.M. *Z. Kristallogr.*, **119**, 53-64 (1963)
- (2) Vezzalini, G. *Z. Kristallogr.*, **166**, 63-71 (1984)
- (3) Gellens, L.R., Price, G.D. and Smith, J.V. *Mineral. Mag.*, **45**, 157-161 (1982)

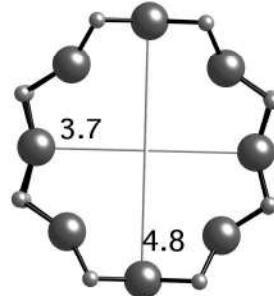
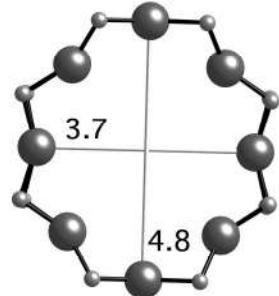
**Crystal chemical data:**  $[(\text{Ca}^{2+}_{0.5}, \text{K}^+, \text{Na}^+)_5 (\text{H}_2\text{O})_{12}] [\text{Al}_5\text{Si}_{19}\text{O}_{48}]$ -DAC  
monoclinic, C12/m1  
 $a = 18.676\text{\AA}$ ,  $b = 7.518\text{\AA}$ ,  $c = 10.246\text{\AA}$ ,  $\beta = 107.87^\circ$  <sup>(2)</sup>

**Framework density:** 17.5 T/1000 $\text{\AA}^3$

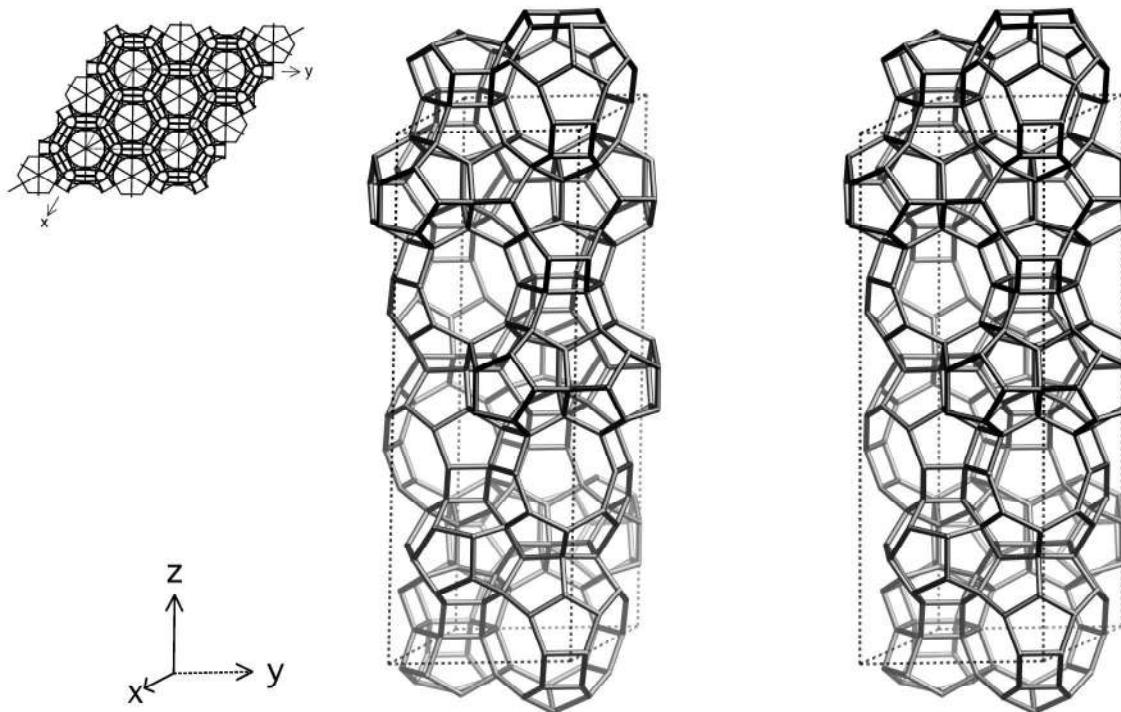
**Channels:** [010] **10** 3.4 x 5.3\*  $\leftrightarrow$  [001] **8** 3.7 x 4.8\*



10-ring viewed along [010]



8-ring viewed along [001]

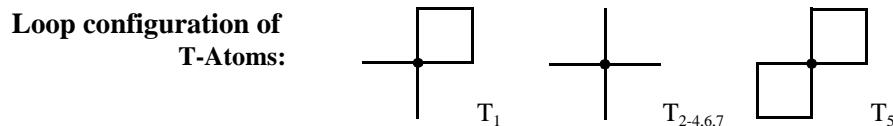


*framework viewed normal to [001] (top left: projection down [001])*

**Idealized cell constants:** trigonal, R̄3m,  $a = 13.8\text{\AA}$ ,  $c = 40.8\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (36, 1)	4 11 23 39 62 91 124 159 203 251	4.5.5.5.5.8
	T <sub>2</sub> (18, m)	4 12 22 37 59 93 127 158 193 251	5.5.5.5.5.5
	T <sub>3</sub> (18, m)	4 12 25 40 61 86 119 164 212 253	5.5.5.5.5.6
	T <sub>4</sub> (18, m)	4 12 24 40 63 87 121 165 208 255	5.5.5.5.5.8
	T <sub>5</sub> (18, 2)	4 10 21 37 62 94 124 158 196 252	4.4.5.5.6.8
	T <sub>6</sub> (6, 2)	4 12 24 39 57 93 121 157 210 240	5.5.5.5.5.5
	T <sub>7</sub> (6, 2)	4 12 24 33 60 97 136 150 192 264	5.5.5.5.5.5

**Secondary building units:** 5-1

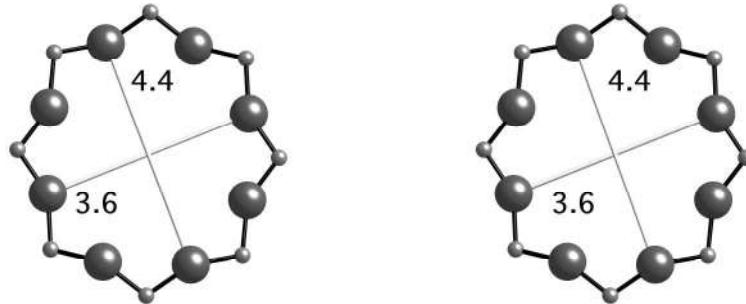


**Isotypic framework  
structures:** \*Deca-dodecasil 3R<sup>(1)</sup>  
Sigma-1<sup>(2)</sup>  
ZSM-58<sup>(3,4)</sup>

**Crystal chemical data:**  $[(C_{10}H_{17}N)_6(N_2)_9][Si_{120}O_{240}]$ -DDR  
 $C_{10}H_{17}N$  = 1-aminoadamantane  
trigonal,  $R\bar{3}m$ ,  $a = 13.860\text{\AA}$ ,  $c = 40.891\text{\AA}$  <sup>(1)</sup>

**Framework density:** 17.6 T/1000 $\text{\AA}^3$

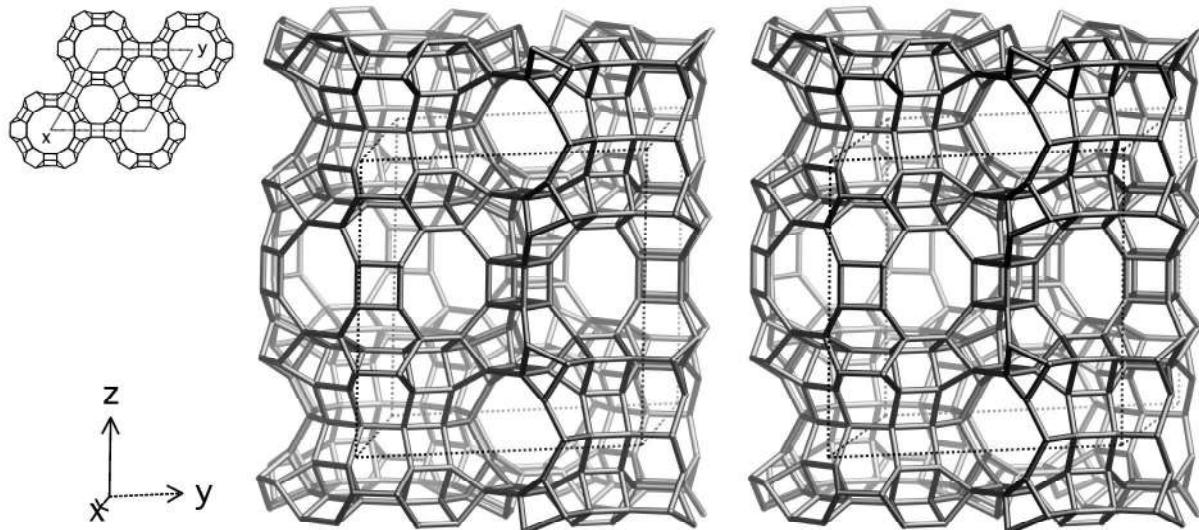
**Channels:**  $\perp [001]$  8 3.6 x 4.4\*\*



8-ring viewed normal to [001]

#### References:

- (1) Gies, H. Z. *Kristallogr.*, **175**, 93-104 (1986)
- (2) Stewart, A., Johnson, D.W. and Shannon, M.D. *Stud. Surf. Sci. Catal.*, **37**, 57-64 (1988)
- (3) Valyocsik, E.W. *U.S. Patent 4,698,217* (1987)
- (4) Ernst, S., Chen, C.Y., Lindner, D. and Weitkamp, J. In *Zeolites for the Nineties, Recent Progress Reports - Abstracts*, (eds. J.C. Jansen, L. Moscou and M.F.M. Post), pp. 55-56 (1989), 8th IZC, Amsterdam



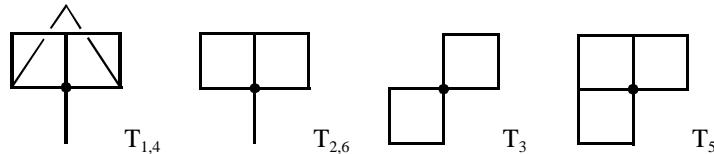
framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** hexagonal, P6/mmm,  $a = 22.0\text{\AA}$ ,  $c = 21.2\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (24, 1)	4    9    16    26    41    60    82    107    135    167	4·6 <sub>3</sub> ·4·6 <sub>3</sub> ·4·12
	T <sub>2</sub> (24, 1)	4    10    18    27    42    62    84    109    135    167	4·6 <sub>2</sub> ·4·8·6·6 <sub>2</sub>
	T <sub>3</sub> (24, 1)	4    10    18    28    45    65    84    106    134    173	4·4·6·6 <sub>3</sub> ·6 <sub>3</sub> ·10
	T <sub>4</sub> (24, 1)	4    9    17    29    45    63    82    106    136    168	4·6·4·6 <sub>2</sub> ·4·8
	T <sub>5</sub> (24, 1)	4    9    17    29    44    62    85    112    139    169	4·4·4·6·6·6 <sub>2</sub>
	T <sub>6</sub> (12, m)	4    10    18    28    45    66    89    115    141    171	4·6 <sub>2</sub> ·4·6 <sub>2</sub> ·10·12

**Secondary building units:** 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*DAF-1<sup>(1)</sup>

## References:

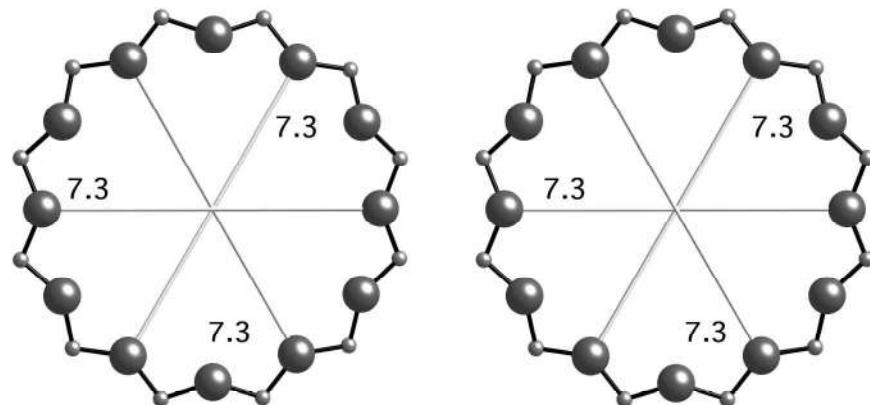
- (1) Wright, P.A., Jones, R.H., Natarajan, S., Bell, R.G., Chen, J.S., Hursthouse, M.B. and Thomas, J.M. *Chem. Commun.*, 633-635 (1993)

**Crystal chemical data:**  $[(C_{16}H_{38}N_2^{2+})_7(H_2O)_{40}] [Mg_{14}Al_{52}P_{66}O_{264}]$ -DFO  
 $C_{16}H_{38}N_2^{2+}$  = decamethonium  
hexagonal, P6/mmm,  $a = 22.351\text{\AA}$ ,  $c = 21.693\text{\AA}$  <sup>(1)</sup>

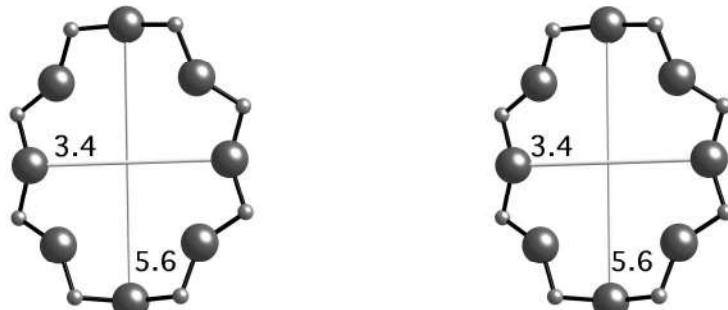
**Framework density:** 14.1 T/1000 $\text{\AA}^3$

**Channels:**  $\{[001] \textbf{12} 7.3 \times 7.3 \leftrightarrow \perp [001] \textbf{8} 3.4 \times 5.6\}^{***} \leftrightarrow \{[001] \textbf{12} 6.2 \times 6.2 \leftrightarrow \perp [001] \textbf{10} 5.4 \times 6.4\}^{***}$

**Stability:** Transforms to AlPO<sub>4</sub>-5 and AlPO<sub>4</sub>-tridymite on heating to 500°C <sup>(1)</sup>

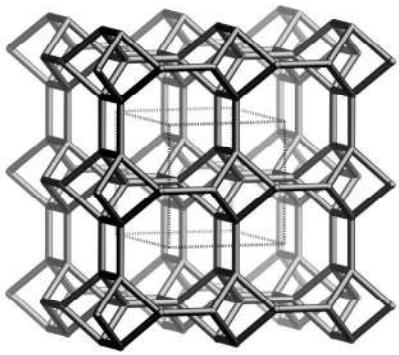


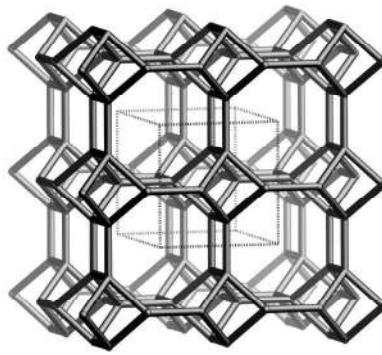
12-ring viewed along [001]



8-ring viewed normal to [001]

See Appendix A for 2nd 12-ring viewed along [001] and 10-ring viewed normal to [001]

 A 3D wireframe model of a zeolite framework viewed along the 001 direction. The framework consists of a dense network of interconnected hexagonal and square rings. A coordinate system is shown in the top-left corner, with the z-axis pointing upwards, the y-axis pointing to the right, and the x-axis pointing downwards.



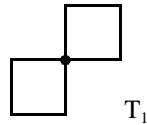
*framework viewed along 001]*

**Idealized cell constants:** tetragonal, P4<sub>2</sub>/mmc,  $a = 7.1\text{\AA}$ ,  $c = 9.0\text{\AA}$

**Coordination sequences  
and vertex symbols:** T<sub>1</sub> (8, m)    4    10    21    36    55    79    106    138    175    215    4·4·6<sub>2</sub>·8<sub>3</sub>·6<sub>2</sub>·8<sub>3</sub>

**Secondary building units:** 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*DAF-2<sup>(1)</sup>  
ACP-3<sup>(2)</sup>  
UCSB-3GaGe<sup>(3)</sup>  
UCSB-3ZnAs<sup>(2)</sup>  
UiO-20<sup>(4)</sup>

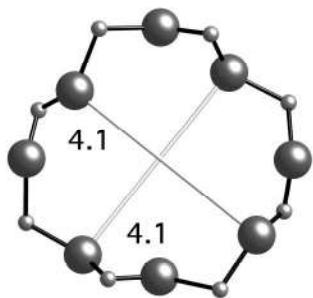
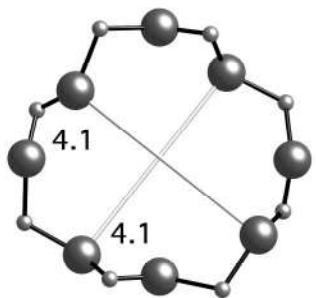
### References:

- (1) Chen, J., Jones, R.H., Natarajan, S., Hursthouse, M.B. and Thomas, J.M. *Angew. Chem., Int. Ed.*, **33**, 639-640 (1994)
- (2) Bu, X., Feng, P., Gier, T.E. and Stucky, G.D. *J. Solid State Chem.*, **136**, 210-215 (1998)
- (3) Bu, X., Feng, P., Gier, T.E., Zhao, D. and Stucky, G.D. *J. Am. Chem. Soc.*, **120**, 13389-13397 (1998)
- (4) Kongshaug, K.O., Fjellvag, H. and Lillerud, K.P. *Chem. Mater.*, **12**, 1095-1099 (2000)

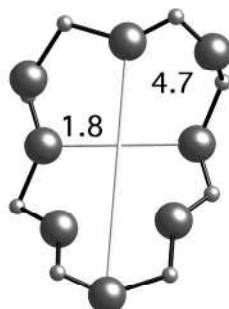
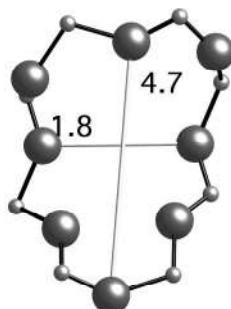
**Crystal chemical data:**  $[(C_2H_{10}N_2^{2+})_2 | [Co_4P_4 O_{16}]$ -DFT  
 $C_2H_8N_2$  = ethylenediamine  
monoclinic, I112/b  
 $a = 14.719\text{\AA}$ ,  $b = 14.734\text{\AA}$ ,  $c = 17.891\text{\AA}$ ,  $\gamma = 90.02^\circ$  <sup>(1)</sup>  
(Relationship to unit cell of Framework Type:  $a' = 2a$ ,  $b' = 2b$ ,  $c' = 2c$ )

**Framework density:** 16.5 T/1000 $\text{\AA}^3$

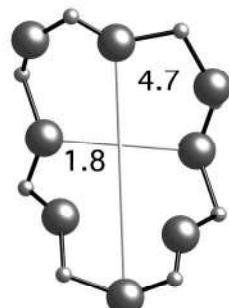
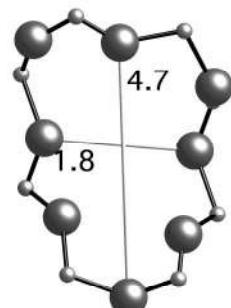
**Channels:** [001] 8 4.1 x 4.1\*  $\leftrightarrow$  [100] 8 1.8 x 4.7\*  $\leftrightarrow$  [010] 8 1.8 x 4.7\*



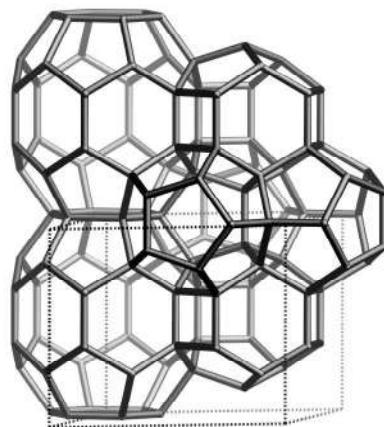
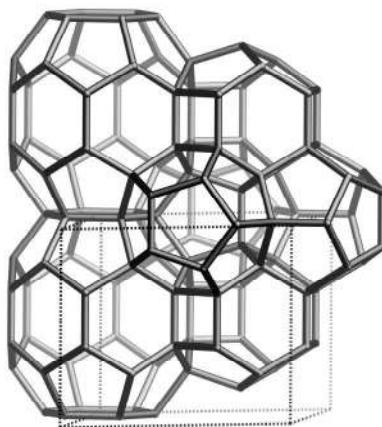
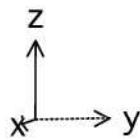
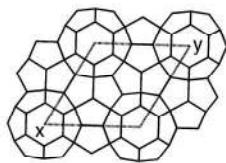
8-ring viewed along [001]



8-ring viewed along [100]



8-ring viewed along [010]



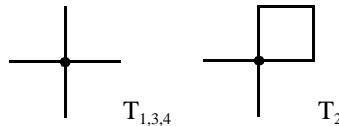
*framework viewed normal to [001] (top left: projection down [001])*

**Idealized cell constants:** hexagonal, P6/mmm,  $a = 14.2\text{\AA}$ ,  $c = 11.5\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (12, m) 4 12 23 41 64 92 128 167 207 259	5·5·5·5·5·6
	T <sub>2</sub> (12, m) 4 11 24 41 63 91 128 171 214 259	4·5·5·6·5·6
	T <sub>3</sub> (6, mm2) 4 12 25 42 68 90 122 167 210 268	5·5·5·5·5·6
	T <sub>4</sub> (4, 3m) 4 12 24 36 61 101 133 156 204 256	5·5·5·5·5·5

**Secondary building units:** combinations only

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*Dodecasil 1H<sup>(1)</sup>

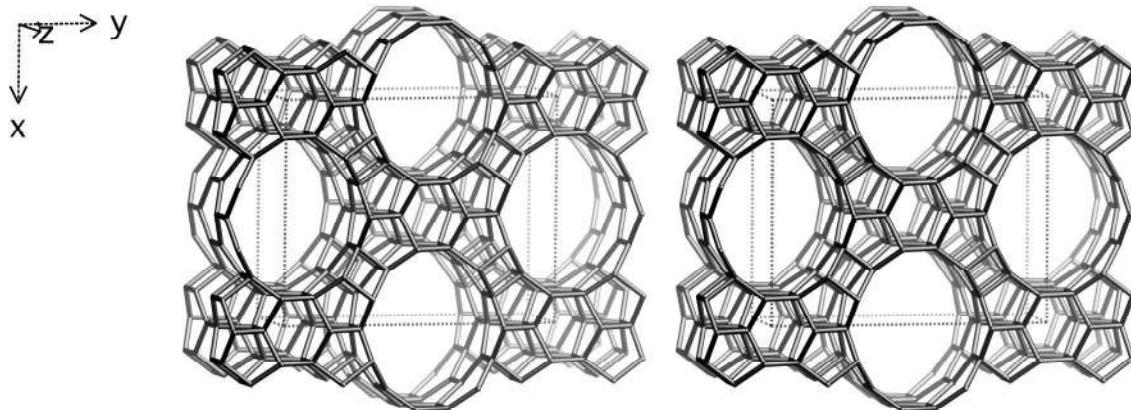
### References:

- (1) Gerke, H. and Gies, H. Z. Kristallogr., **166**, 11-22 (1984)

**Crystal chemical data:**  $[\text{C}_5\text{H}_{11}\text{N} (\text{N}_2)_5] [\text{Si}_{34} \text{O}_{68}]$ -DOH  
 $\text{C}_5\text{H}_{11}\text{N}$  = piperidine  
hexagonal, P6/mmm,  $a = 13.783\text{\AA}$ ,  $c = 11.190\text{\AA}$ <sup>(1)</sup>

**Framework density:** 18.5 T/1000 $\text{\AA}^3$

**Channels:** apertures formed by 6-rings only

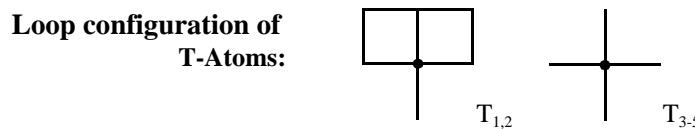


framework viewed along [001]

**Idealized cell constants:** orthorhombic, Cmcm,  $a = 18.9\text{\AA}$ ,  $b = 23.4\text{\AA}$ ,  $c = 8.5\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (16, 1)    4    10    20    34    54    77    107    140    175    218	4·6·4·6 <sub>2</sub> ·5·6
	T <sub>2</sub> (16, 1)    4    10    20    35    54    77    106    138    177    221	4·6·4·6 <sub>2</sub> ·5·6
	T <sub>3</sub> (16, 1)    4    12    24    38    55    76    105    143    184    223	5·6 <sub>2</sub> ·6·6 <sub>2</sub> ·6·6 <sub>2</sub>
	T <sub>4</sub> (8, m)    4    12    22    33    53    80    109    143    179    217	5·6·5·6·6·6 <sub>2</sub>
	T <sub>5</sub> (8, m)    4    12    23    37    52    74    107    143    183    223	5·6 <sub>2</sub> ·5·6 <sub>2</sub> ·6 <sub>2</sub> ·6 <sub>2</sub>

**Secondary building units:** 5-3



**Isotypic framework structures:** \*UTD-1F<sup>(1)</sup>  
UTD-1<sup>(2)</sup>

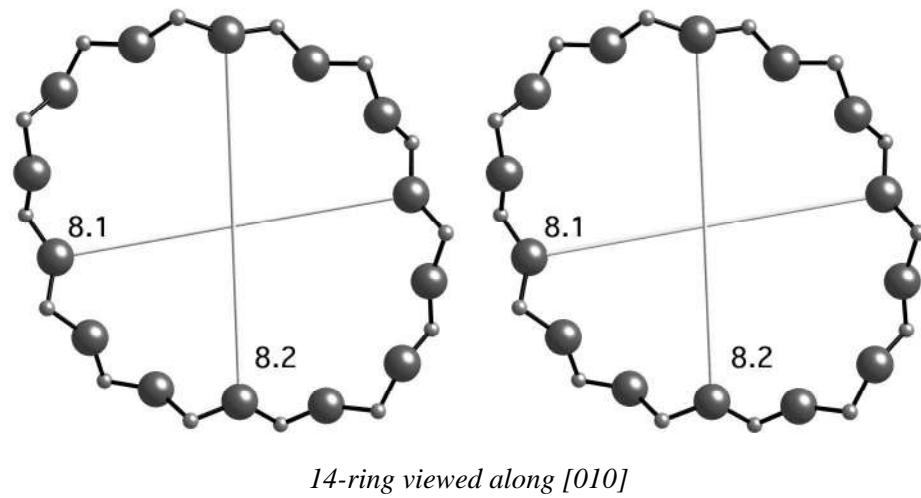
### References:

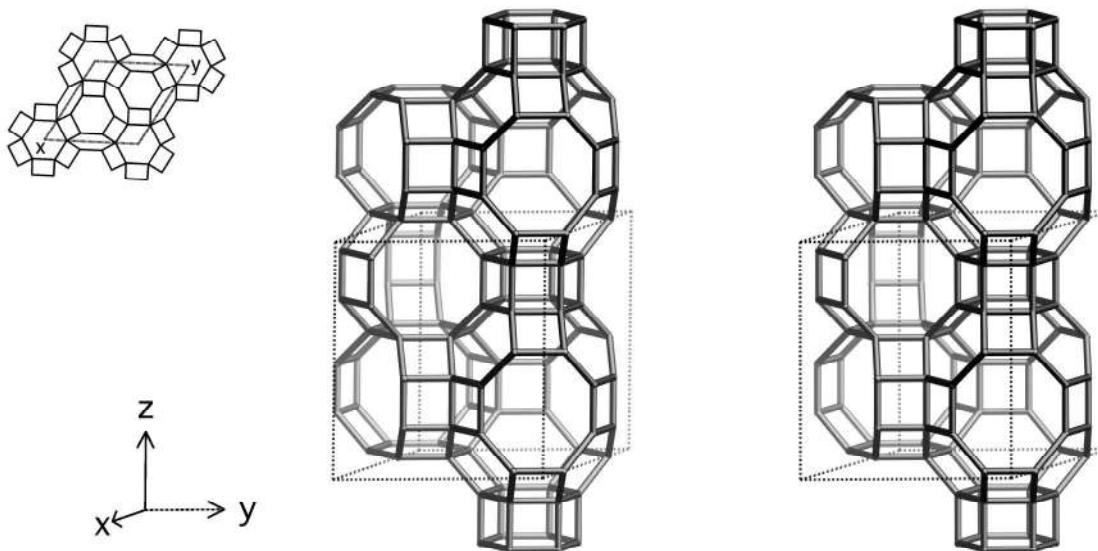
- (1) Wessels, T., Baerlocher, C., McCusker, L.B. and Creyghton, E.J. *J. Am. Chem. Soc.*, **121**, 6242-6247 (1999)
- (2) Lobo, R.F., Tsapatsis, M., Freyhardt, C.C., Khodabandeh, S., Wagner, P., Chen, C.Y., Balkus, K.J., Zones, S.I. and Davis, M.E. *J. Am. Chem. Soc.*, **119**, 8474-8484 (1997)

**Crystal chemical data:**  $|((\text{Cp}^*)_2\text{Co})^+ \text{F}_{1.5}(\text{OH})_{0.5}| [\text{Si}_{64}\text{O}_{128}]$ -DON  
 $\text{Cp}^*$  = pentamethylcyclopentadiene  
monoclinic, P1c1,  $a = 14.970\text{\AA}$ ,  $b = 8.476\text{\AA}$ ,  $c = 30.028\text{\AA}$ ,  $\beta = 102.65^\circ$  <sup>(1)</sup>

**Framework density:** 17.2 T/1000 $\text{\AA}^3$

**Channels:** [010] **14** 8.1 x 8.2\*





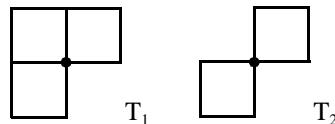
framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** hexagonal, P<sub>6</sub><sub>3</sub>/mmc,  $a = 13.2\text{\AA}$ ,  $c = 15.0\text{\AA}$

**Coordination sequences and vertex symbols:** T<sub>1</sub> (24, 1) 4 9 17 30 49 71 92 115 147 190 4·4·4·6·6·8  
T<sub>2</sub> (12, 2) 4 10 20 32 46 66 94 128 162 192 4·4·6·6·8·8

**Secondary building units:** 6 or 4

**Loop configuration of T-Atoms:**



**Framework description:** ABBACC sequence of 6-rings

**Isotypic framework structures:** \*TMA-E (Aiello and Barrer)<sup>(1,2)</sup>  
Bellbergite<sup>(3)</sup>

### References:

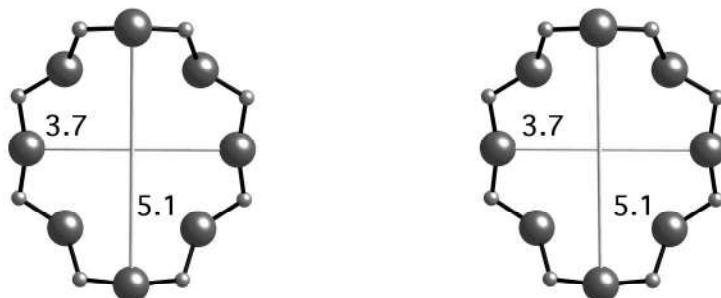
- (1) Aiello, R. and Barrer, R.M. *J. Chem. Soc. (A)*, 1470-1475 (1970)
- (2) Meier, W.M. and Groner, M. *J. Solid State Chem.*, **37**, 204-218 (1981)
- (3) Rüdinger, B., Tillmanns, E. and Hentschel, G. *Miner. Petrol.*, **48**, 147-152 (1993)

**Crystal chemical data:**  $[(C_4H_{12}N^+)_2Na^+ \cdot (H_2O)_{26}] [Al_9Si_{27} O_{72}]$ -EAB

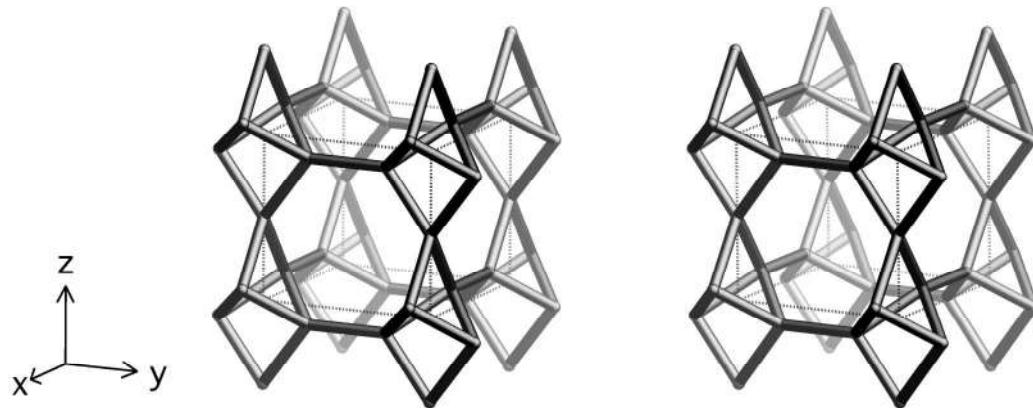
$C_4H_{12}N^+$  = tetramethylammonium  
hexagonal,  $P6_3/mmc$ ,  $a = 13.28\text{\AA}$ ,  $c = 15.21\text{\AA}$  <sup>(2)</sup>

**Framework density:**  $15.5 \text{ T}/1000\text{\AA}^3$

**Channels:**  $\perp [001] \mathbf{8} \ 3.7 \times 5.1^{**}$



*8-ring viewed normal to [001]*



framework viewed normal to [001]

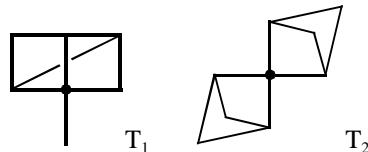
**Idealized cell constants:** tetragonal, P $\bar{4}$ m2,  $a = 6.9\text{\AA}$ ,  $c = 6.4\text{\AA}$

**Coordination sequences and vertex symbols:**

$T_1$ (4, m)	4	9	19	35	52	72	100	131	163	201	$4 \cdot 8_3 \cdot 4 \cdot 8_3 \cdot 4_2 \cdot 8_4$
$T_2$ (1, $\bar{4}$ m2)	4	8	18	32	52	74	100	128	162	204	$4_2 \cdot 4_2 \cdot 8_4 \cdot 8_4 \cdot 8_4 \cdot 8_4$

**Secondary building units:** 4=1

**Loop configuration of T-Atoms:**



**Isotypic framework structures:**

\*Edingtonite<sup>(1-3)</sup>  
[Co-Al-P-O]-EDI<sup>(4)</sup>  
[Co-Ga-P-O]-EDI<sup>(4)</sup>  
K-F<sup>(5,6)</sup>

Linde F<sup>(7)</sup>  
Synthetic edingtonite<sup>(8)</sup>  
Tetragonal edingtonite<sup>(9)</sup>  
Zeolite N<sup>(10)</sup> (not to be confused with Linde N)

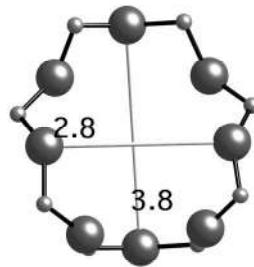
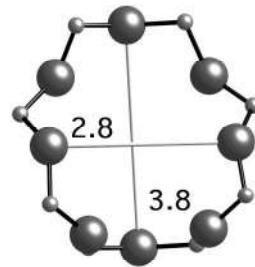
## References:

- (1) Taylor, W.H. and Jackson, R. Z. *Kristallogr.*, **86**, 53-64 (1933)
- (2) Galli, E. *Acta Crystallogr.*, **B32**, 1623-1627 (1976)
- (3) Kvick, Å. and Smith J.V. *J. Chem. Phys.*, **79**, 2356-2362 (1983)
- (4) Bu, X., Gier, T.E., Feng, P. and Stucky, G.D. *Chem. Mater.*, **10**, 2546-2551 (1998)
- (5) Barrer, R.M. and Baynham, J.W. *J. Chem. Soc.*, 2882-2891 (1956)
- (6) Baerlocher, Ch. and Barrer, R.M. Z. *Kristallogr.*, **140**, 10-26 (1974)
- (7) Sherman, J.D. *ACS Sym. Ser.*, **40**, 30-42 (1977)
- (8) Ghobarkar, H. and Schaeff, O. *Cryst. Res. Technol.*, **32**, 653-657 (1997)
- (9) Mazzi, F., Galli, E. and Gottardi, G. *N. Jb. Miner. Mh.*, 373-382 (1984)
- (10) Christensen, A.N. and Fjellvåg, H. *Acta Chemica Scand.*, **51**, 969-973 (1997)

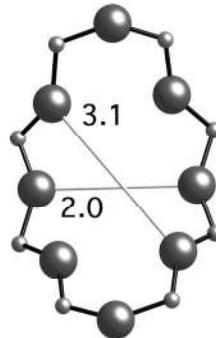
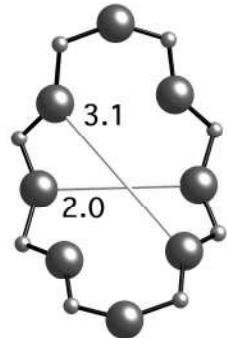
**Crystal chemical data:**  $[\text{Ba}^{2+}_2 (\text{H}_2\text{O})_8] [\text{Al}_4\text{Si}_6 \text{O}_{20}]$ -EDI  
orthorhombic,  $\text{P}2_1\text{2}_1\text{2}$ ,  $a = 9.550\text{\AA}$ ,  $b = 9.665\text{\AA}$ ,  $c = 6.523\text{\AA}$  <sup>(2)</sup>  
(Relationship to unit cell of Framework Type:  
 $\mathbf{a}' = \mathbf{a} \cdot \text{sqrt}(2)$ ,  $\mathbf{b}' = \mathbf{b} \cdot \text{sqrt}(2)$ ,  $\mathbf{c}' = \mathbf{c}$   
or, as vectors,  $\mathbf{a}' = \mathbf{a} + \mathbf{b}$ ,  $\mathbf{b}' = \mathbf{b} - \mathbf{a}$ ,  $\mathbf{c}' = \mathbf{c}$ )

**Framework density:**  $16.6 \text{ T}/1000\text{\AA}^3$

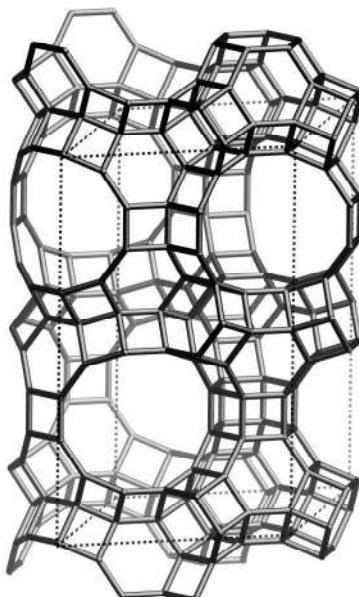
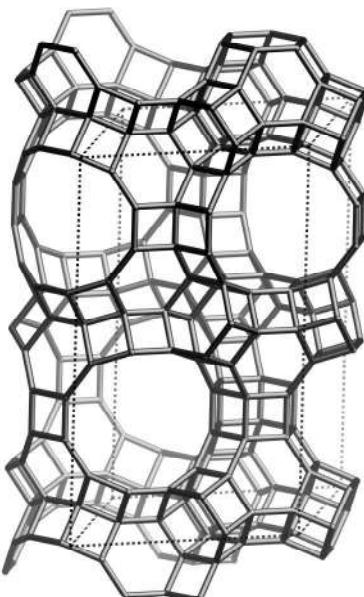
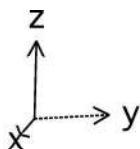
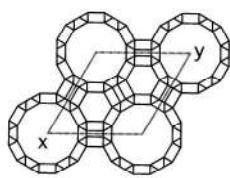
**Channels:**  $<110> \text{ 8 } 2.8 \times 3.8^{**} \leftrightarrow [001] \text{ 8 } 2.0 \times 3.1^*$  (variable due to considerable flexibility of the framework)



8-ring viewed along  $<110>$



8-ring viewed along  $[001]$  (variable)



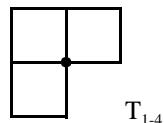
framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** hexagonal, P<sub>6</sub><sub>3</sub>/mmc,  $a = 17.2\text{\AA}$ ,  $c = 28.1\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (24, 1)    4    9    16    25    37    53    73    96    121    148	4.4.4.6.6.12
	T <sub>2</sub> (24, 1)    4    9    16    25    37    53    73    96    121    148	4.4.4.6.6.12
	T <sub>3</sub> (24, 1)    4    9    16    25    37    53    73    97    124    152	4.4.4.6.6.12
	T <sub>4</sub> (24, 1)    4    9    16    25    37    53    73    96    120    145	4.4.4.6.6.12

**Secondary building units:** 6-6 or 6 or 4

**Loop configuration of  
T-Atoms:**



**Framework description:** structural derivative of hexagonal diamond and tridymite, respectively

**Isotypic framework  
structures:**

\*EMC-2<sup>(1,2)</sup>  
CSZ-1 (**EMT-FAU** structural intermediate)<sup>(3)</sup>  
ECR-30 (**EMT-FAU** structural intermediate)<sup>(4)</sup>  
ZSM-20 (**EMT-FAU** structural intermediate)<sup>(5)</sup>  
ZSM-3 (**EMT-FAU** structural intermediate)<sup>(6)</sup>

**Alternate designation:** Breck structure six (improper)

BSS (improper)

hexagonal faujasite

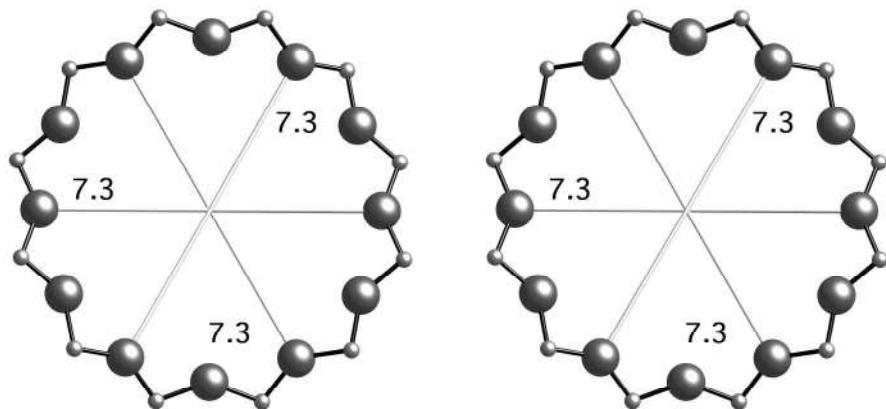
**Crystal chemical data:**  $[\text{Na}^+_{21} (\text{C}_{12}\text{H}_{24}\text{O}_6)_{41} [\text{Al}_{21}\text{Si}_{75} \text{O}_{192}]]\text{-EMT}$

$\text{C}_{12}\text{H}_{24}\text{O}_6$  = 18-crown-6

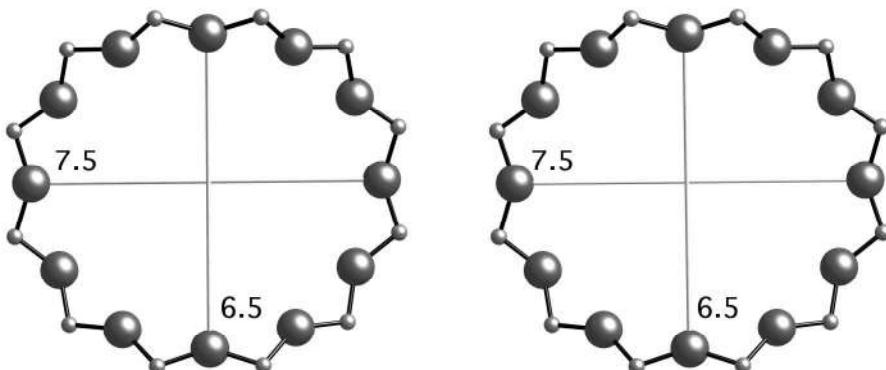
hexagonal,  $P6_3/mmc$ ,  $a = 17.374\text{\AA}$ ,  $c = 28.365\text{\AA}$  <sup>(2)</sup>

**Framework density:**  $12.9 \text{ T}/1000\text{\AA}^3$

**Channels:** [001] **12**  $7.3 \times 7.3^*$   $\leftrightarrow \perp$  [001] **12**  $6.5 \times 7.5^{**}$



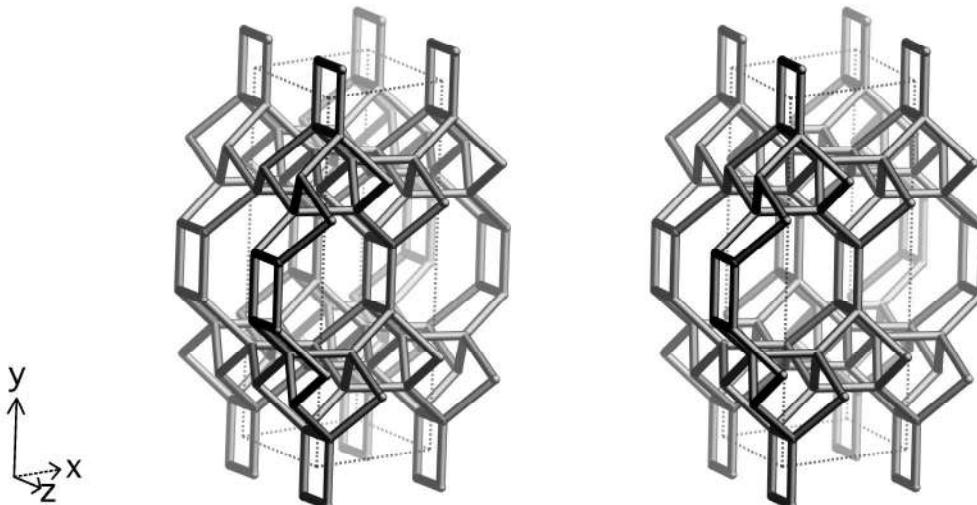
12-ring viewed along [001]



12-ring viewed normal to [001]

#### References:

- (1) Delprato, F., Delmotte, L., Guth, J.L. and Huve, L. *Zeolites*, **10**, 546-552 (1990)
- (2) Baerlocher, Ch., McCusker, L.B. and Chiappetta, R. *Microporous Materials*, **2**, 269-280 (1994)
- (3) Barrett, M.G. and Vaughan, D.E.W. *UK Patent GB 2,076,793* (1981)
- (4) Vaughan, D.E.W. *E. Patent 0,351,461* (1989)
- (5) Newsam, J.M., Treacy, M.M.J., Vaughan, D.E.W., Strohmaier, K.G. and Mortier, W.J. *Chem. Commun.*, 493-495 (1989)
- (6) Kokotailo, G.T. and Ceric, J. *Adv. Chem. Ser.* , **101**, 109-121 (1971)



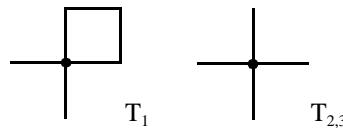
framework viewed along [001]

**Idealized cell constants:** monoclinic, C2/m,  $a = 9.1\text{\AA}$ ,  $b = 17.5\text{\AA}$ ,  $c = 10.4\text{\AA}$ ,  $\beta = 124.9^\circ$

<b>Coordination sequences and vertex symbols:</b>	$T_1(8, 1)$	4 11 24 42 63 93 127 160 206 262	$4\cdot5_2\cdot5\cdot8\cdot5\cdot8$
	$T_2(8, 1)$	4 12 22 37 64 94 119 161 204 252	$5\cdot5\cdot5_2\cdot8\cdot10_2$
	$T_3(8, 1)$	4 12 20 39 66 90 118 164 214 245	$5\cdot5\cdot5_2\cdot5\cdot8$

**Secondary building units:** 5-1

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Epistilbite<sup>(1-4)</sup>  
Synthetic epistilbite<sup>(5)</sup>

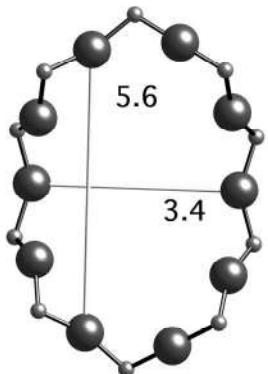
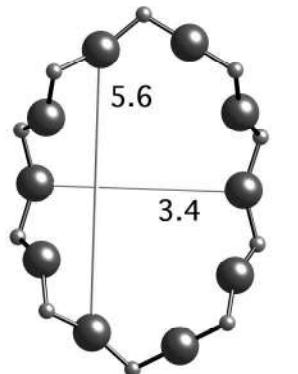
### References:

- (1) Kerr, I.S. *Nature*, **202**, 589 (1964)
- (2) Perrotta, A.J. *Mineral. Mag.*, **36**, 480-490 (1967)
- (3) Alberti, A., Galli, E. and Vezzalini, G. Z. *Kristallogr.*, **173**, 257-265 (1985)
- (4) Yang, P. and Armbruster, T. *Eur. J. Mineral.*, **8**, 263-271 (1996)
- (5) Ghobarkar, H. *Cryst. Res. Technol.*, 151-1573 (1984)

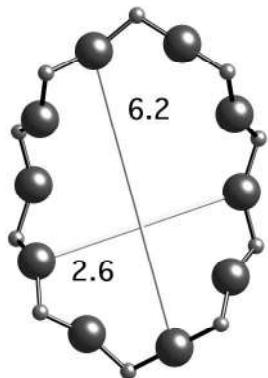
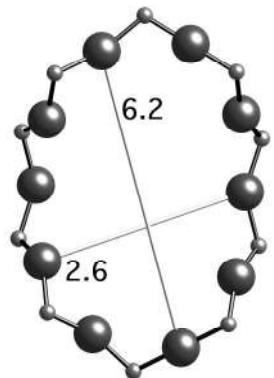
**Crystal chemical data:**  $[\text{Ca}^{2+}_3 (\text{H}_2\text{O})_{16}] [\text{Al}_6\text{Si}_{18} \text{O}_{48}]$ -EPI  
monoclinic, C2/m,  $a = 9.08\text{\AA}$ ,  $b = 17.74\text{\AA}$ ,  $c = 10.25\text{\AA}$ ,  $\beta = 124.54^\circ$  <sup>(2)</sup>

**Framework density:** 17.6 T/1000 $\text{\AA}^3$

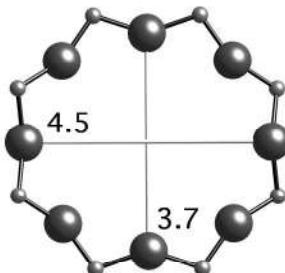
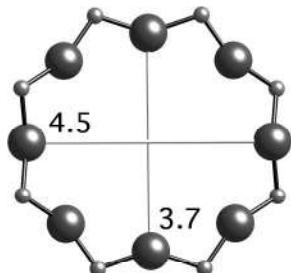
**Channels:** [100] **10** 3.4 x 5.6\*  $\leftrightarrow$  [001] **8** 3.7 x 4.5\*



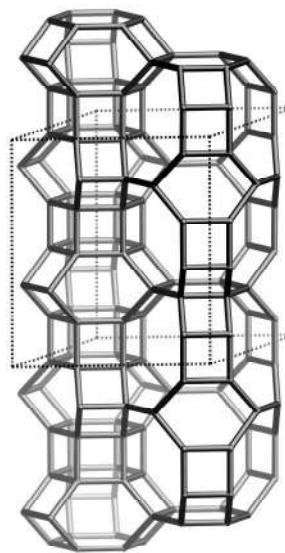
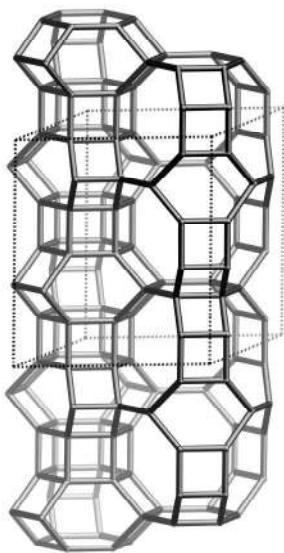
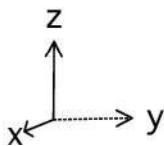
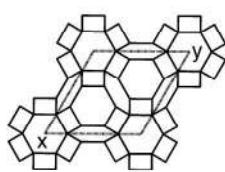
*10-ring viewed along [100] associated*



*with second 10-ring along [100]*



*8-ring viewed along [001]*



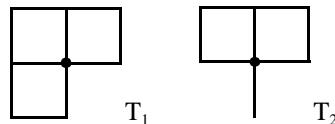
framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** hexagonal, P<sub>6</sub><sub>3</sub>/mmc,  $a = 13.1\text{\AA}$ ,  $c = 15.2\text{\AA}$

**Coordination sequences and vertex symbols:** T<sub>1</sub> (24, 1) 4 9 17 30 50 75 98 118 144 185  
T<sub>2</sub> (12, m) 4 10 20 32 46 64 90 126 164 196 4·4·4·6·6·8  
4·8·4·8·6·6

**Secondary building units:** 6 or 4

**Loop configuration of T-Atoms:**



**Framework description:** AABAAC sequence of 6-rings

**Isotypic framework structures:** \*Erionite<sup>(1-3)</sup>  
AlPO-17 plus numerous compositional variants<sup>(4-6)</sup>  
LZ-220<sup>(7)</sup>  
Linde T (**ERI-OFF** structural intermediate)<sup>(8)</sup>

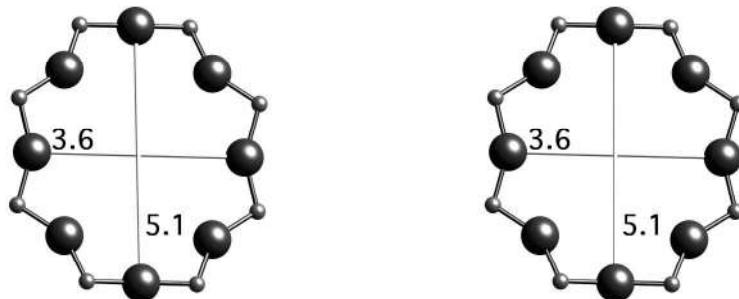
### References:

- (1) Staples, L.W. and Gard, J.A. *Mineral. Mag.*, **32**, 261-281 (1959)
- (2) Kawahara, A. and Curien, H. *Bull. Soc. fr. Minéral. Cristallogr.*, **92**, 250-256 (1969)
- (3) Gard, J.A. and Tait, J.M. In *Proc. 3rd Int. Conf. Molecular Sieves*, (ed. J.B. Uytterhoeven), pp. 94-99 (1973), Leuven University Press, Leuven
- (4) Pluth, J.J., Smith, J.V. and Bennett, J.M. *Acta Crystallogr.*, **C42**, 283-286 (1986)
- (5) Flanigen, E.M., Lok, B.M., Patton, R.L. and Wilson, S.T. *Pure Appl. Chem.*, **58**, 1351-1358 (1986)

**Crystal chemical data:**  $[(\text{Ca}^{2+}, \text{Na}^+)_2 \text{K}^+_2 (\text{H}_2\text{O})_{27}] [\text{Al}_9\text{Si}_{27} \text{O}_{72}]$ -ERI  
hexagonal,  $\text{P}6_3/\text{mmc}$ ,  $a = 13.27\text{\AA}$ ,  $c = 15.05\text{\AA}$ <sup>(3)</sup>

**Framework density:** 15.7 T/1000 $\text{\AA}^3$

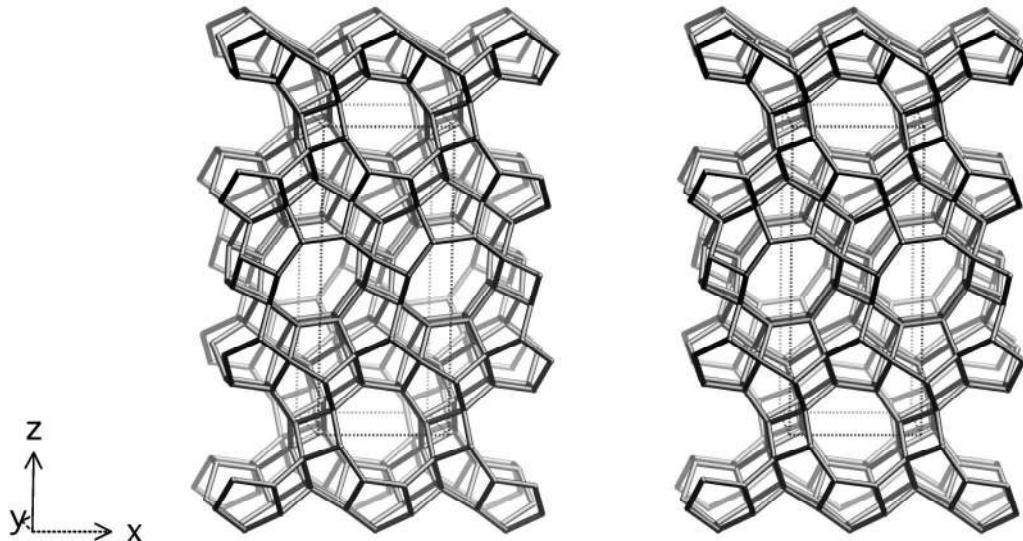
**Channels:**  $\perp [001]$  8 3.6 x 5.1\*\*\*



8-ring viewed normal to [001]

#### References (cont.):

- (6) Flanigen, E.M., Lok, B.M., Patton, R.L. and Wilson, S.T. In *Proc. 7th Int. Zeolite Conf.*, (eds. Y. Murakami, A. Iijima and J.W. Ward), pp. 103-112 (1986), Kodansha, Tokyo
- (7) Breck, D.W. and Skeels, G.W. *U.S. Patent 4,503,023* (1985)
- (8) Breck, D.W. *Zeolite Molecular Sieves*, p. 173 (1974), Wiley, New York



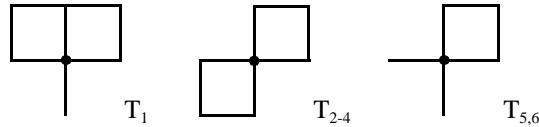
framework viewed along [010]

**Idealized cell constants:** orthorhombic, Pnma,  $a = 9.7\text{\AA}$ ,  $b = 12.2\text{\AA}$ ,  $c = 22.8\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (8, 1)    4 10 20 35 56 82 111 143 180 228 T <sub>2</sub> (8, 1)    4 10 20 36 58 82 109 144 186 230 T <sub>3</sub> (8, 1)    4 10 21 36 57 82 111 145 183 231 T <sub>4</sub> (8, 1)    4 10 21 36 56 82 113 145 180 224 T <sub>5</sub> (8, 1)    4 11 22 36 56 78 110 148 184 225 T <sub>6</sub> (8, 1)    4 11 20 37 54 82 112 142 182 226	4.5·4.5·6·8 4.4·5·6·6·8 4.4·5·8·6·6 4.4·5·6·6·8 4.6·5·6·5·6 4.5·5·5·6·6
---	--	--

**Secondary building units:** 6-2

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*ERS-7<sup>(1,2)</sup>

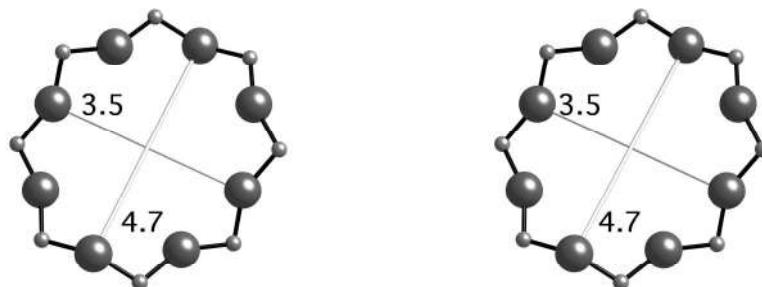
## References:

- (1) Campbell, B.J., Bellussi, G., Carluccio, L., Perego, G., Cheetham, A.K., Cox, D.E. and Millini, R. *Chem. Commun.*, 1725-1726 (1998)
- (2) Millini, R., Perego, G., Carluccio, L., Bellussi, G., Cox, D.E., Campbell, B.J. and Cheetham, A.K. In *Proc. 12th Int. Zeolite Conf.*, (eds. M.M.J. Treacy, B.K. Marcus, M.E. Bisher and J.B. Higgins), pp. 541-548 (1999), MRS, Warrendale, PA

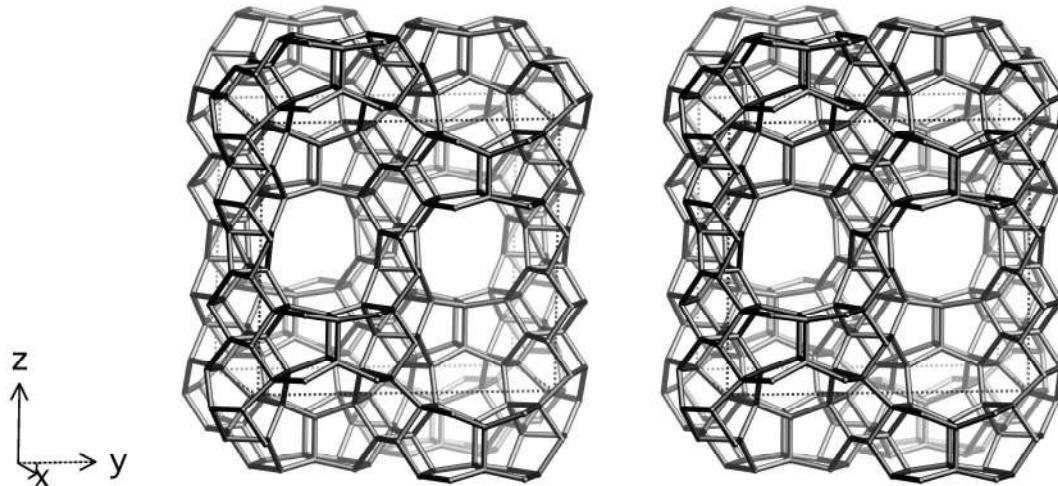
**Crystal chemical data:**  $[H^{+}_{5.06}Na^{+}_{0.07}] [Al_{5.13}Si_{42.87} O_{96}]$ -ESV  
orthorhombic, Pnma,  $a = 9.780\text{\AA}$ ,  $b = 12.412\text{\AA}$ ,  $c = 22.861\text{\AA}$  <sup>(2)</sup>

**Framework density:** 17.3 T/1000 $\text{\AA}^3$

**Channels:** [010] 8 3.5 x 4.7\*



*8-ring viewed along [010]*



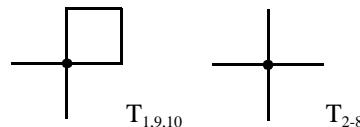
framework viewed along [100]

**Idealized cell constants:** orthorhombic, Cmma,  $a = 13.9\text{\AA}$ ,  $b = 22.9\text{\AA}$ ,  $c = 20.6\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (16, 1)	4 11 21 36 59 92 129 167 197 246	4·6·5·5 <sub>2</sub> ·10
	T <sub>2</sub> (16, 1)	4 12 25 41 64 88 122 160 202 256	5·6·5·6·5·6
	T <sub>3</sub> (16, 1)	4 12 23 38 60 89 124 159 194 248	5·5 <sub>2</sub> ·5·6·5·6
	T <sub>4</sub> (16, 1)	4 12 22 39 59 91 124 160 206 257	5·5·5·6 <sub>2</sub> ·5·10
	T <sub>5</sub> (8, m)	4 12 20 31 61 88 120 159 197 248	5·5·5·5·6 <sub>2</sub>
	T <sub>6</sub> (8, m)	4 12 20 34 57 92 131 164 202 236	5·5 <sub>2</sub> ·5·5 <sub>2</sub> ·12 <sub>2</sub> ·*
	T <sub>7</sub> (8, m)	4 12 24 39 60 91 126 161 195 243	5·6·5·6·5·6 <sub>2</sub>
	T <sub>8</sub> (8, m)	4 12 24 36 56 90 127 157 197 236	5·5·5·5·12 <sub>4</sub> ·*
	T <sub>9</sub> (8, m)	4 11 23 45 67 88 115 162 218 261	4·6·5·5·5·5
	T <sub>10</sub> (8, m)	4 11 24 39 68 92 118 156 213 268	4·10·5·5·5·5

**Secondary building units:** combinations only

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*EU-1<sup>(1,2)</sup>  
TPZ-3<sup>(3)</sup>  
ZSM-50<sup>(4)</sup>

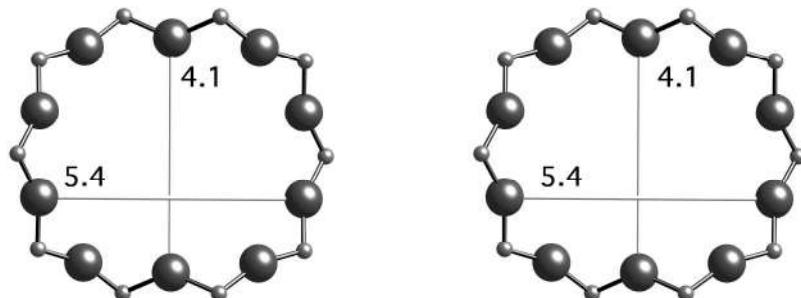
### References:

- (1) Casci, J.L., Lowe, B.M. and Whittam, T.V. U.S. Patent 4,537,754, (1985)
- (2) Briscoe, N.A., Johnson, D.W., Shannon, M.D., Kokotailo, G.T. and McCusker, L.B. *Zeolites*, **8**, 74-76 (1988)
- (3) Sumitani, K., Sakai, T., Yamasaki, Y. and Onodera, T. E. Patent EP 51318 (1982)
- (4) Rohrbaugh, W.J. *private communication*

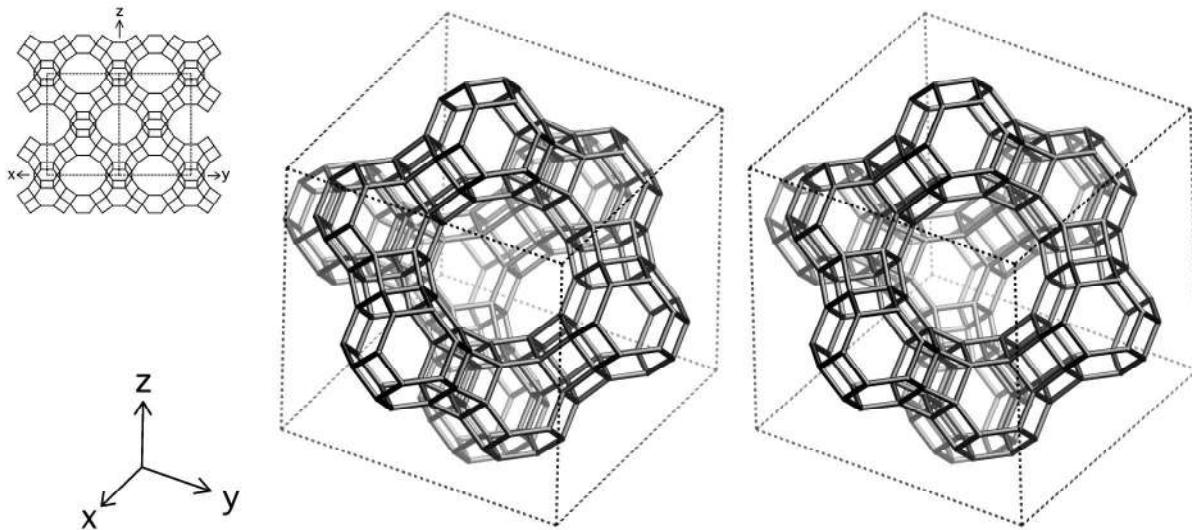
**Crystal chemical data:**  $[\text{Na}^+ \text{n} (\text{H}_2\text{O})_{26}] [\text{Al}_n \text{Si}_{112-n} \text{O}_{224}]$ -EUO , n < 19, typically n ~ 3.6  
orthorhombic, Cmma, a = 13.695Å, b = 22.326Å, c = 20.178Å <sup>(2)</sup>

**Framework density:** 18.2 T/1000Å<sup>3</sup>

**Channels:** [100] **10** 4.1 x 5.4\* (with large side pockets)



10-ring viewed along [100]



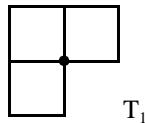
*framework viewed along [111] (top left: projection down [110])*

**Idealized cell constants:** cubic, Fd $\bar{3}$ m (origin choice 2),  $a = 24.3\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(192, 1) \quad 4 \quad 9 \quad 16 \quad 25 \quad 37 \quad 53 \quad 73 \quad 96 \quad 120 \quad 145$       4.4.4.6.6.12

**Secondary building units:** 6-6 or 6-2 or 6 or 4

**Loop configuration of T-Atoms:**



**Framework description:** structural derivative of diamond and cristobalite, respectively

**Isotypic framework structures:**

- \*Faujasite<sup>(1,2)</sup>
- [Al-Ge-O]-FAU<sup>(3)</sup>
- [Co-Al-P-O]-FAU<sup>(4)</sup>
- [Ga-Ge-O]-FAU<sup>(3)</sup>
- Beryllophosphate X<sup>(5)</sup>
- CSZ-1 (**EMT-FAU** structural intermediate)<sup>(6)</sup>
- ECR-30 (**EMT-FAU** structural intermediate)<sup>(7)</sup>
- LZ-210<sup>(8)</sup>
- Linde X<sup>(9,10)</sup>
- Linde Y<sup>(11,12)</sup>
- SAPO-37<sup>(13)</sup>
- Siliceous Na-Y<sup>(14)</sup>
- ZSM-20 (**EMT-FAU** structural intermediate)<sup>(15)</sup>
- ZSM-3 (**EMT-FAU** structural intermediate)<sup>(16)</sup>
- Zincophosphate X<sup>(5)</sup>

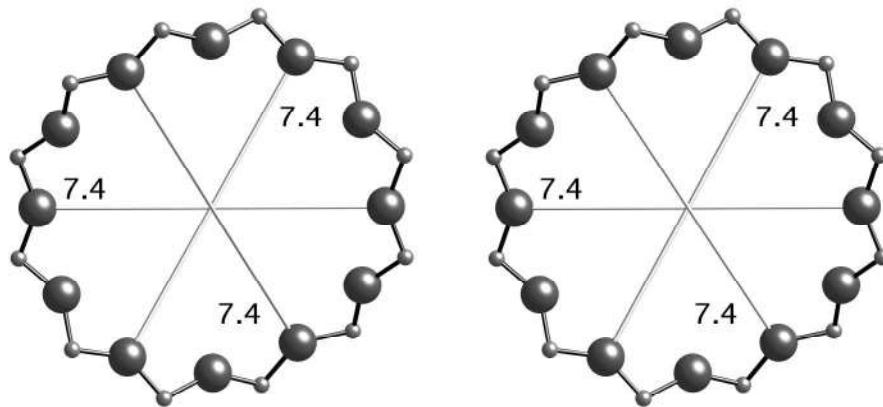
#### References:

- (1) Bergerhoff, G., Baur, W.H. and Nowacki, W. *N. Jb. Miner. Mh.*, 193-200 (1958)
- (2) Baur, W.H. *Am. Mineral.*, **49**, 697-704 (1964)

**Crystal chemical data:**  $[(\text{Ca}^{2+}, \text{Mg}^{2+}, \text{Na}^+)_2]_{29} (\text{H}_2\text{O})_{240} [\text{Al}_{58}\text{Si}_{134}\text{O}_{384}]$ -FAU  
cubic,  $\text{Fd}\bar{3}\text{m}$ ,  $a = 24.74\text{\AA}$ <sup>(2)</sup>

**Framework density:** 12.7 T/1000 $\text{\AA}^3$

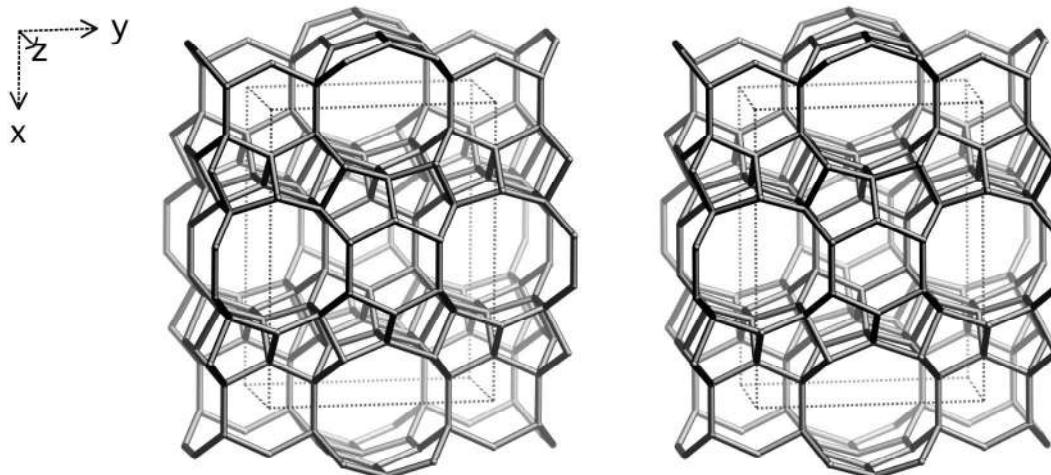
**Channels:** <111> **12** 7.4 x 7.4\*\*\*



12-ring viewed along <111>

#### References (cont.):

- (3) Barrer, R.M., Baynham, J.W., Bultitude, F.W. and Meier, W.M. *J. Chem. Soc.*, 195-208 (1959)
- (4) Feng, P.Y., Bu, X.H. and Stucky, G.D. *Nature*, **388**, 735-741 (1997)
- (5) Gier, T.E. and Stucky, G.D. *Zeolites*, **12**, 770-775 (1992)
- (6) Barrett, M.G. and Vaughan, D.E.W. *UK Patent GB 2,076,793* (1981)
- (7) Vaughan, D.E.W. *E. Patent 0,351,461* (1989)
- (8) Breck, D.W. and Skeels, G.W. *U.S. Patent 4,503,023* (1985)
- (9) Milton, R.M. *U.S. Patent 2,882,244* (1959)
- (10) Olson, D.H. *J. Phys. Chem.*, **74**, 2758-2764 (1970)
- (11) Breck, D.W. *U.S. Patent 3,130,007* (1964)
- (12) Costenoble, M.L., Mortier, W.J. and Uytterhoeven, J.B. *J. Chem. Soc., Faraday Trans. I*, **72**, 1877-1883 (1976)
- (13) Lok, B.M., Messina, C.A., Patton, R.L., Gajek, R.T., Cannan, T.R. and Flanigen, E.M. *J. Am. Chem. Soc.*, **106**, 6092-6093 (1984)
- (14) Hriljac, J.J., Eddy, M.M., Cheetham, A.K., Donohue, J.A. and Ray, G.J. *J. Solid State Chem.*, **106**, 66-72 (1993)
- (15) Newsam, J.M., Treacy, M.M.J., Vaughan, D.E.W., Strohmaier, K.G. and Mortier, W.J. *Chem. Commun.*, 493-495 (1989)
- (16) Kokotailo, G.T. and Ceric, J. *Adv. Chem. Ser.*, **101**, 109-121 (1971)



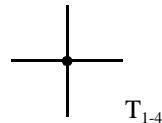
framework viewed along [001]

**Idealized cell constants:** orthorhombic, Immm,  $a = 19.0\text{\AA}$ ,  $b = 14.3\text{\AA}$ ,  $c = 7.5\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(16, 1)$	4 12 21 39 66 95 126 169 221 265	$5\cdot5\cdot5\cdot5_2\cdot5\cdot8$
	$T_2(8, m)$	4 12 27 43 62 97 139 172 206 264	$5\cdot8\cdot5\cdot8\cdot5_2\cdot6$
	$T_3(8, m)$	4 12 20 35 67 104 121 157 223 276	$5\cdot5_2\cdot5\cdot5_2\cdot10_2\cdot12_2$
	$T_4(4, 2mm)$	4 12 23 40 66 96 131 164 214 272	$5\cdot5\cdot5\cdot5_2\cdot6$

**Secondary building units:** 5-1

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*Ferrierite<sup>(1)</sup>  
[Ga-Si-O]-FER<sup>(2)</sup>  
[Si-O]-FER<sup>(3,4)</sup>  
FU-9<sup>(5)</sup>  
ISI-6<sup>(6)</sup>

Monoclinic ferrierite<sup>(7)</sup>  
NU-23<sup>(8)</sup>  
Sr-D<sup>(9)</sup>  
ZSM-35<sup>(10)</sup>

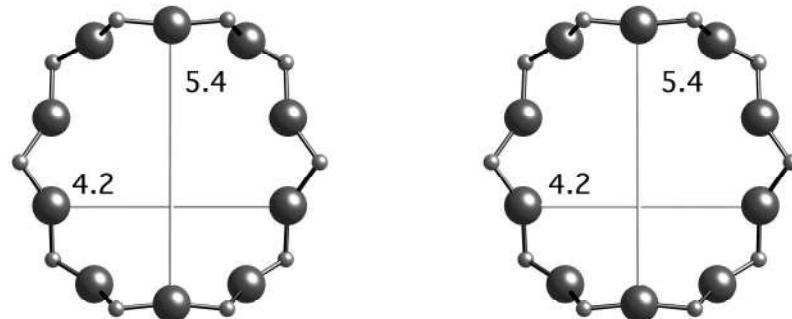
## References:

- (1) Vaughan, P.A. *Acta Crystallogr.*, **21**, 983-990 (1966)
- (2) Jacob, N.E., Joshi, P.N., Shaikh, A.A. and Shiralkar, V.P. *Zeolites*, **13**, 430-434 (1993)
- (3) Gies, H. and Gunawardane, R.P. *Zeolites*, **7**, 442-445 (1987)
- (4) Morris, R.E., Weigel, S.J., Henson, N.J., Bull, L.M., Janicke, M.T., Chmelka, B.F. and Cheetham, A.K. *J. Am. Chem. Soc.*, **116**, 11849-11855 (1994)
- (5) Seddon, D. and Whittam, T.V. *E. Patent B-55,529* (1985)
- (6) Morimoto, N., Takatsu, K. and Sugimoto, M. *U.S. Patent 4,578,259* (1986)
- (7) Gramlich-Meier, R., Gramlich, V. and Meier, W.M. *Am. Mineral.*, **70**, 619-623 (1985)

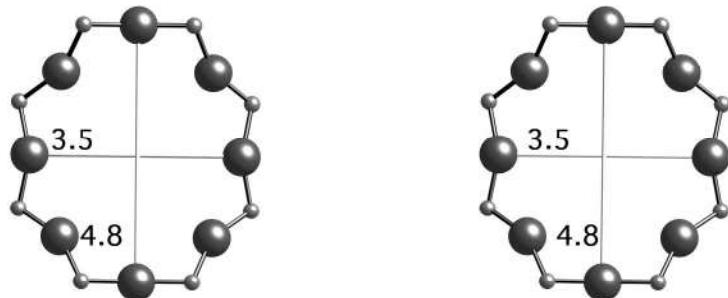
**Crystal chemical data:**  $[\text{Mg}^{2+}_2\text{Na}^+_2(\text{H}_2\text{O})_{18}] [\text{Al}_6\text{Si}_{30}\text{O}_{72}]$ -FER  
orthorhombic, Imm̄,  $a = 19.156\text{\AA}$ ,  $b = 14.127\text{\AA}$ ,  $c = 7.489\text{\AA}$  <sup>(1)</sup>

**Framework density:** 17.8 T/1000 $\text{\AA}^3$

**Channels:** [001] **10** 4.2 x 5.4\*  $\leftrightarrow$  [010] **8** 3.5 x 4.8\*



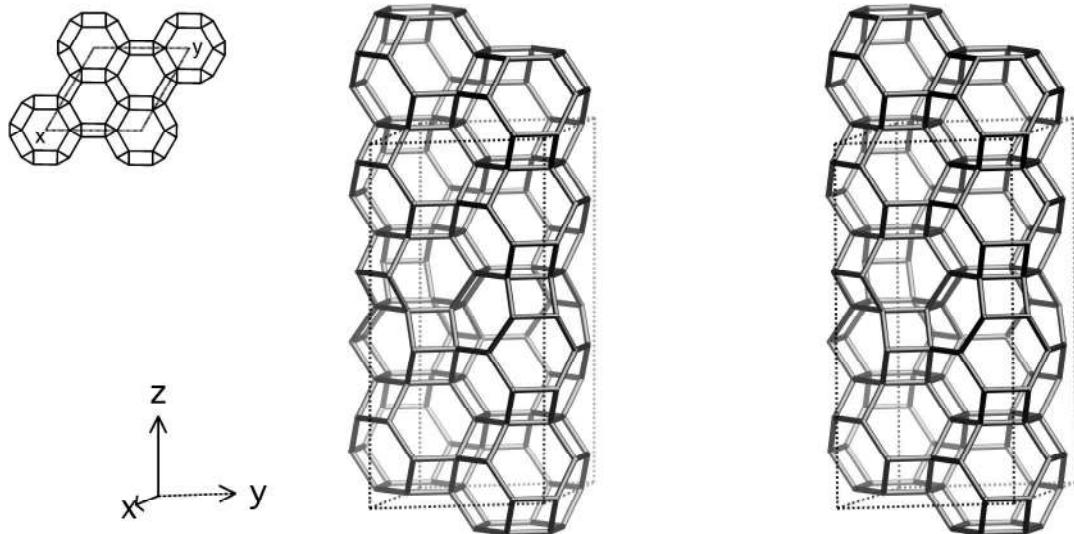
10-ring viewed along [001]



8-ring viewed along [010]

#### References (cont.):

- (8) Whittam, T.V. *E. Patent A-103,981* (1984)
- (9) Barrer, R.M. and Marshall, D.J. *J. Chem. Soc.*, 2296-2305 (1964)
- (10) Plank, C.J., Rosinski, E.J. and Rubin, M.K. *U.S. Patent 4,016,245* (1977)



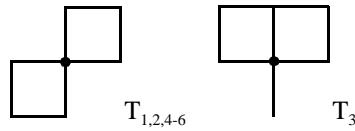
*framework viewed normal to [001] (top left: projection down [001])*

**Idealized cell constants:** trigonal, P $\bar{3}$ m1,  $a = 12.7\text{\AA}$ ,  $c = 25.3\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (12, 1)    4    10    20    34    52    74    100    130    165    205	4·4·6·6·6·6
	T <sub>2</sub> (12, 1)    4    10    20    34    52    74    101    133    168    206	4·4·6·6·6·6
	T <sub>3</sub> (12, 1)    4    10    20    34    52    74    101    133    168    206	4·6·4·6·6·6
	T <sub>4</sub> (12, 1)    4    10    20    34    53    76    102    132    166    206	4·4·6·6·6·6
	T <sub>5</sub> (6, 2)    4    10    20    34    52    74    100    130    164    202	4·4·6·6·6·6
	T <sub>6</sub> (6, 2)    4    10    20    34    54    78    104    134    168    208	4·4·6·6·6·6

**Secondary building units:** 6

**Loop configuration of  
T-Atoms:**



**Framework description:** ABCABACABC sequence of 6-rings

**Isotypic framework  
structures:**

\*Franzinit<sup>(1)</sup>

## References:

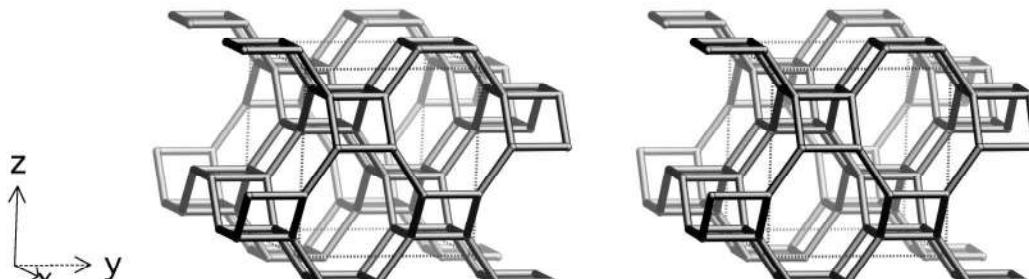
- (1) Ballirano, P., Bonaccorsi, E., Maras, A. and Merlino, S. *Can. Mineral.*, **38**, 657-668 (2000)

**Franzinite****Type Material****FRA**

**Crystal chemical data:**  $[(\text{Na},\text{K})^{+}_{30} \text{Ca}^{2+}_{10} (\text{SO}_4)^{2-}_{10} (\text{H}_2\text{O})_2] [\text{Al}_{30}\text{Si}_{30}\text{O}_{120}]$ -**FRA**  
trigonal, P321,  $a = 12.916\text{\AA}$ ,  $c = 26.543\text{\AA}$ <sup>(1)</sup>

**Framework density:** 15.6 T/1000 $\text{\AA}^3$

**Channels:** apertures formed by 6-rings only



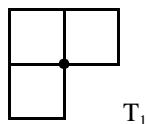
framework viewed along [100]

**Idealized cell constants:** tetragonal, I4<sub>1</sub>/amd (origin choice 2),  $a = 9.8\text{\AA}$ ,  $c = 10.2\text{\AA}$

**Coordination sequences and vertex symbols:** T<sub>1</sub> (16, 2)    4    9    18    32    48    67    92    120    150    185                  4·4·4·8<sub>2</sub>·8·8

**Secondary building units:** 8 or 4

**Loop configuration of T-Atoms:**



**Isotypic framework structures:**

- \*Gismondine<sup>(1)</sup>
- [Al-Co-P-O]-GIS<sup>(2)</sup>
- [Co-Al-P-O]-GIS<sup>(3)</sup>
- [Co-Ga-P-O]-GIS<sup>(4)</sup>
- [Co-P-O]-GIS<sup>(5)</sup>
- [Ga-Si-O]-GIS<sup>(6)</sup>
- [Mg-Al-P-O]-GIS<sup>(3)</sup>
- [Zn-Ga-P-O]-GIS<sup>(7)</sup>
- |(NH<sub>4</sub>)<sub>4</sub>||[Zn<sub>4</sub>B<sub>4</sub>P<sub>8</sub>O<sub>32</sub>]-GIS<sup>(8)</sup>
- |Cs<sub>4</sub>||[Zn<sub>4</sub>B<sub>4</sub>P<sub>8</sub>O<sub>32</sub>]-GIS<sup>(8)</sup>
- |Rb<sub>4</sub>||[Zn<sub>4</sub>B<sub>4</sub>P<sub>8</sub>O<sub>32</sub>]-GIS<sup>(8)</sup>
- Amicite<sup>(9)</sup>
- Garronite<sup>(10,11)</sup>
- Gobbinsite<sup>(12)</sup>
- High-silica Na-P<sup>(13)</sup>
- Low-silica Na-P (MAP)<sup>(14)</sup>
- MAPO-43<sup>(15)</sup>
- MAPSO-43<sup>(16,17)</sup>
- Na-P1<sup>(18)</sup>
- Na-P2<sup>(19)</sup>
- SAPO-43<sup>(20)</sup>
- Synthetic Ca-garronite<sup>(21)</sup>
- Synthetic amicite<sup>(22)</sup>
- Synthetic garronite<sup>(22)</sup>
- Synthetic gobbinsite<sup>(22)</sup>
- TMA-gismondine<sup>(23)</sup>

**Alternate designation:** Gismondite (discredited)  
synthetic zeolite B (disused)

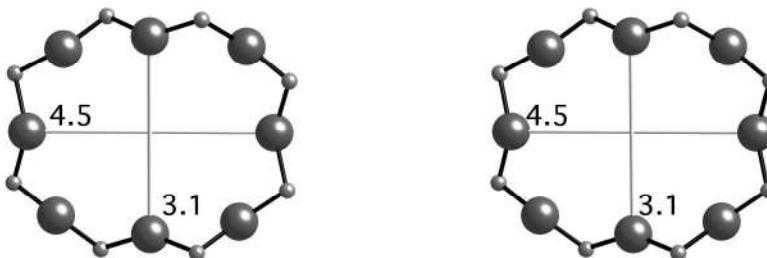
#### References:

- (1) Fischer, K. and Schramm, V. *Adv. Chem. Ser.*, **101**, 250-258 (1971)
- (2) Feng, P.Y., Bu, X.H. and Stucky, G.D. *Nature*, **388**, 735-741 (1997)
- (3) Feng, P., Bu, X., Gier, T.E. and Stucky, G.D. *Microporous and Mesoporous Materials*, **23**, 221-229 (1998)
- (4) Cowley, A.R. and Chippindale, A.M. *Chem. Commun.*, 673-674 (1996)
- (5) Yuan, H.M., Chen, J.S., Zhu, G.S., Li, J.Y., Yu, J.H., Yang, G.D. and Xu, R. *Inorg. Chem.*, **39**, 1476-1479 (2000)

**Crystal chemical data:**  $[\text{Ca}^{2+}_4 (\text{H}_2\text{O})_{16}] [\text{Al}_8 \text{Si}_8 \text{O}_{32}]$ -GIS  
 monoclinic,  $P112_1/a$   
 $a = 9.843\text{\AA}$ ,  $b = 10.023\text{\AA}$ ,  $c = 10.616\text{\AA}$ ,  $\gamma = 92.417^\circ$  <sup>(1)</sup>  
 (Relationship to unit cell of Framework Type:  $a' = a$ ,  $b' = b$ ,  $c' = c$ )

**Framework density:**  $15.3 \text{ T}/1000\text{\AA}^3$

**Channels:**  $\{[100] 8 \text{ } 3.1 \times 4.5 \leftrightarrow [010] 8 \text{ } 2.8 \times 4.8\}$ \*\*\* (variable due to considerable flexibility of the framework)

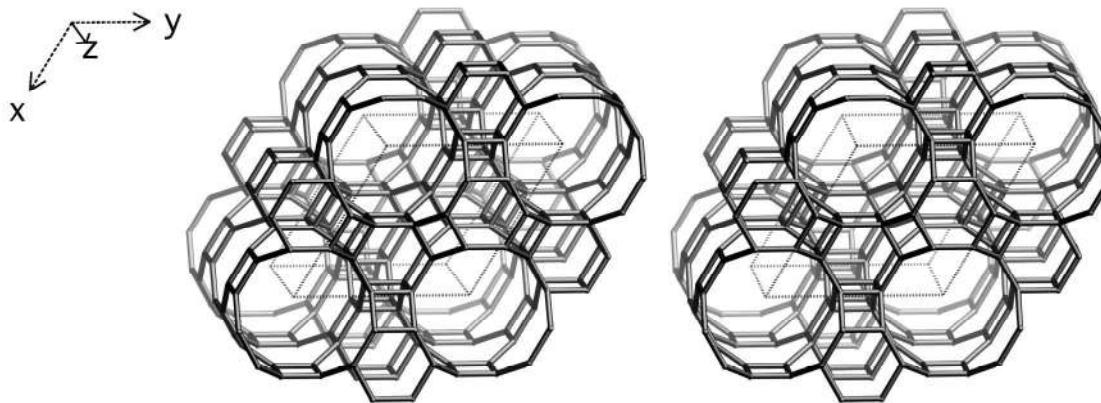


8-ring viewed along [100]

See Appendix A for 8-ring viewed along [010]

#### References (cont.):

- (6) Cho, H.H., Kim, S.H., Kim, Y.G., Kim, Y.C., Koller, H., Camblor, M.A. and Hong, S.B. *Chem. Mater.*, **12**, 2292-2300 (2000)
- (7) Chippindale, A.M., Cowley, A.R. and Peacock, K.J. *Microporous and Mesoporous Materials*, **24**, 133-141 (1998)
- (8) Knipe, R., Schäfer, G., Engelhardt, H. and Boy, I. *Angew. Chem. Int. Ed.*, **38**, 3642-3644 (1999)
- (9) Alberti, A. and Vezzalini, G. *Acta Crystallogr.*, **B35**, 2866-2869 (1979)
- (10) Artioli, G. *Am. Mineral.*, **77**, 189-196 (1992)
- (11) Artioli, G. and Marchi, M. *Powder Diffraction*, **14**, 190-194 (1999)
- (12) McCusker, L.B., Baerlocher, Ch. and Nawaz, R. Z. *Kristallogr.*, **171**, 281-289 (1985)
- (13) Håkansson, U., Fälth, L. and Hansen, S. *Acta Crystallogr.*, **C46**, 1363-1364 (1990)
- (14) Albert, B.R., Cheetham, A.K., Stuart, J.A. and Adams, C.J. *Microporous and Mesoporous Materials*, **21**, 133-142 (1998)
- (15) Pluth, J.J., Smith, J.V. and Bennett, J.M. *J. Am. Chem. Soc.*, **111**, 1692-1698 (1989)
- (16) Flanigen, E.M., Lok, B.M., Patton, R.L. and Wilson, S.T. *Pure Appl. Chem.*, **58**, 1351-1358 (1986)
- (17) Flanigen, E.M., Lok, B.M., Patton, R.L. and Wilson, S.T. In *Proc. 7th Int. Zeolite Conf.*, (eds. Y. Murakami, A. Iijima and J.W. Ward), pp. 103-112 (1986), Kodansha, Tokyo
- (18) Baerlocher, Ch. and Meier, W.M. Z. *Kristallogr.*, **135**, 339-354 (1972)
- (19) Hansen, S., Håkansson, U. and Fälth, L. *Acta Crystallogr.*, **C46**, 1361-1362 (1990)
- (20) Helliwell, M., Kaucic, V., Cheetham, G.M.T., Harding, M.M., Kariuki, B.M. and Rizkallah, P.J. *Acta Crystallogr.*, **B49**, 413-420 (1993)
- (21) Schropfer, L and Joswig, W. *Eur. J. Mineral.*, **9**, 53-65 (1997)
- (22) Ghobarkar, H. and Schaef, O. *Mater. Res. Bull.*, **34**, 517-525 (1999)
- (23) Baerlocher, Ch. and Meier, W.M. *Helv. Chim. Acta*, **53**, 1285-1293 (1970)



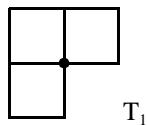
framework viewed along [001]

**Idealized cell constants:** hexagonal, P6<sub>3</sub>/mmc,  $a = 13.7\text{\AA}$ ,  $c = 9.9\text{\AA}$

**Coordination sequences and vertex symbols:** T<sub>1</sub> (24, 1) 4 9 17 29 45 65 89 116 144 175 4·4·4·8·6·8

**Secondary building units:** 6-6 or 8 or 4-2 or 6 or 4

**Loop configuration of T-Atoms:**



**Framework description:** AABB sequence of 6-rings

**Isotypic framework structures:** \*Gmelinite<sup>(1)</sup>  
K-rich gmelinite<sup>(2)</sup>  
Synthetic fault-free gmelinite<sup>(3)</sup>

**Alternate designation:** sarcolite (discredited)

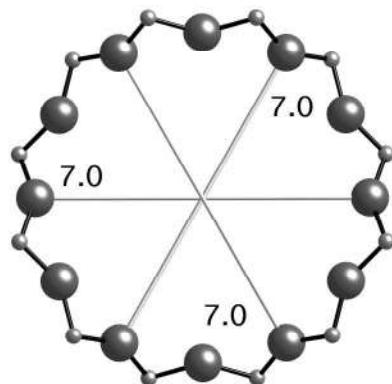
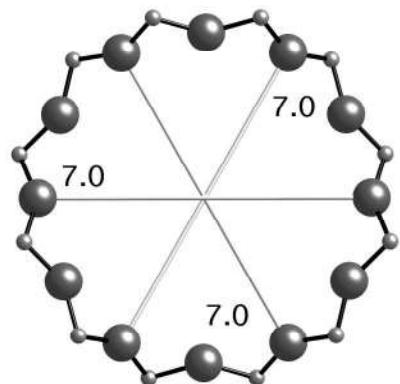
### References:

- (1) Fischer, K. *N. Jb. Miner. Mh.*, 1-13 (1966)
- (2) Vezzalini, G., Quartieri, S. and Passaglia, E. *N. Jb. Miner. Mh.*, 504-516 (1990)
- (3) Daniels, R.H., Kerr, G.T. and Rollmann, L.D. *J. Am. Chem. Soc.*, **100**, 3097-3100 (1978)

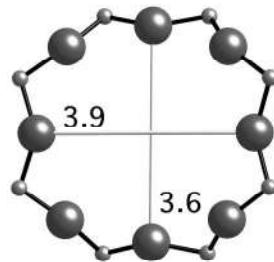
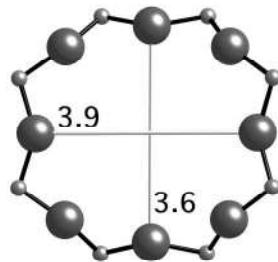
**Crystal chemical data:**  $[(\text{Ca}^{2+}, \text{Na}^+)_4 (\text{H}_2\text{O})_{24}] [\text{Al}_8\text{Si}_{16} \text{O}_{48}]$ -GME  
hexagonal, P6<sub>3</sub>/mmc,  $a = 13.75\text{\AA}$ ,  $c = 10.05\text{\AA}$ <sup>(1)</sup>

**Framework density:** 14.6 T/1000 $\text{\AA}^3$

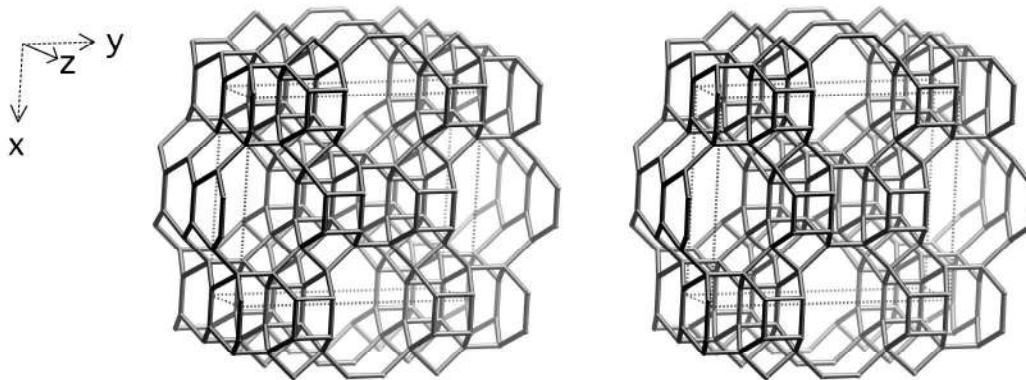
**Channels:** [001] **12** 7.0 x 7.0\*  $\leftrightarrow$   $\perp$  [001] **8** 3.6 x 3.9\*\*



12-ring viewed along [001]



8-ring viewed normal to [001]



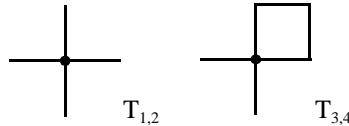
framework viewed along [001]

**Idealized cell constants:** orthorhombic, Cmmm,  $a = 16.9\text{\AA}$ ,  $b = 20.4\text{\AA}$ ,  $c = 5.3\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(8, 1)$	4 12 21 39 63 85 117 154 192 242	$5 \cdot 6 \cdot 5 \cdot 6 \cdot 5_2 \cdot 6$
	$T_2(8, 1)$	4 12 25 38 57 86 119 158 194 233	$5 \cdot 6 \cdot 5 \cdot 6 \cdot 6 \cdot 12_6$
	$T_3(8, 1)$	4 11 22 38 58 86 121 156 191 229	$4 \cdot 6_2 \cdot 5 \cdot 6 \cdot 5 \cdot 6$
	$T_4(8, 1)$	4 11 22 38 63 91 115 147 195 244	$4 \cdot 6_2 \cdot 5 \cdot 6 \cdot 5 \cdot 6$

**Secondary building units:** 5-3

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*GUS-1<sup>(1)</sup>

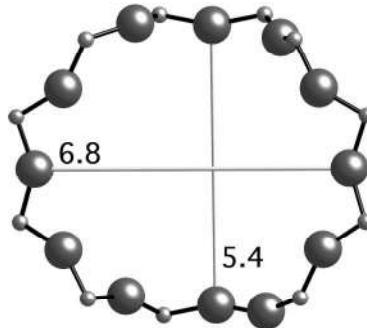
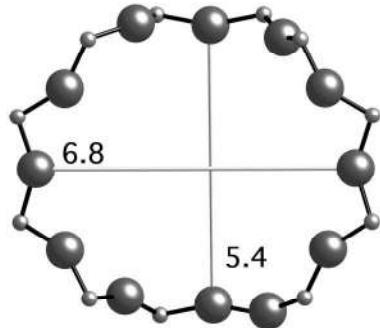
### References:

- (1) Plévert, J., Kubota, Y., Honda, T., Okubo, T. and Sugi, Y. *Chem. Commun.*, 2363-2364 (2000)

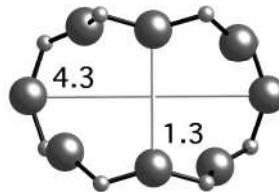
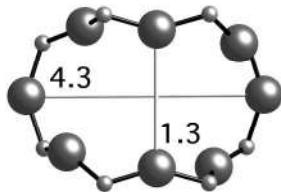
**Crystal chemical data:**  $[\text{Si}_{32} \text{O}_{64}]$ -GON  
orthorhombic, C222,  $a = 16.421\text{\AA}$ ,  $b = 20.054\text{\AA}$ ,  $c = 5.046\text{\AA}$ <sup>(1)</sup>

**Framework density:** 19.3 T/1000 $\text{\AA}^3$

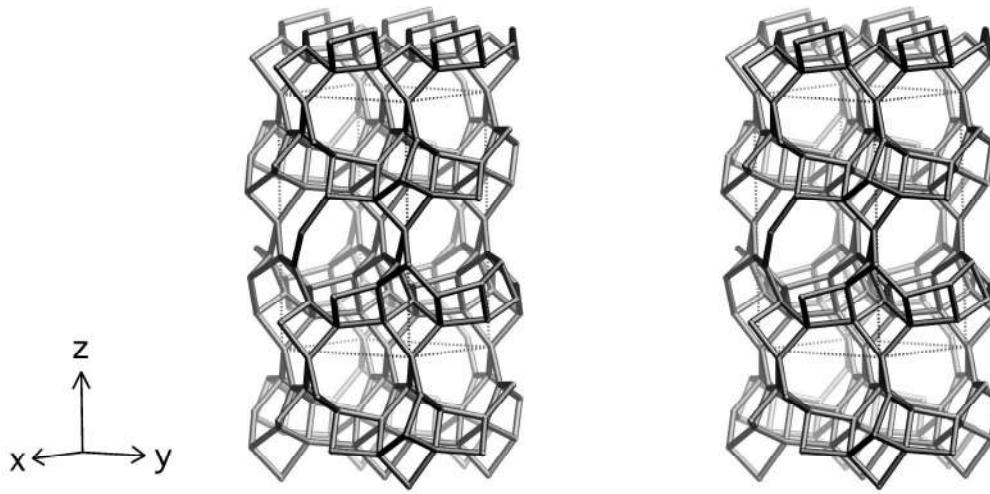
**Channels:** [001] 12 5.4 x 6.8\*



12-ring viewed along [001]



8-ring viewed along [001]



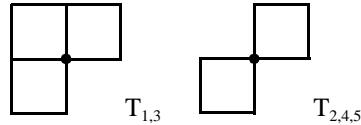
framework viewed along [110]

**Idealized cell constants:** orthorhombic, C222<sub>1</sub>,  $a = 8.7\text{\AA}$ ,  $b = 11.0\text{\AA}$ ,  $c = 17.5\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (8, 1)	4 9 19 36 56 80 102 132 180 220	4·4·4·8 <sub>5</sub> ·6 <sub>2</sub> ·8 <sub>2</sub>
	T <sub>2</sub> (8, 1)	4 10 20 36 55 77 108 140 174 219	4·4·6·8 <sub>3</sub> ·6 <sub>3</sub> ·8 <sub>2</sub>
	T <sub>3</sub> (8, 1)	4 9 19 36 56 76 106 142 172 217	4·4·4·8 <sub>2</sub> ·6·8 <sub>2</sub>
	T <sub>4</sub> (4, 2)	4 10 20 34 58 82 102 136 176 220	4·4·6 <sub>4</sub> ·8 <sub>4</sub> ·8 <sub>2</sub> ·8 <sub>2</sub>
	T <sub>5</sub> (4, 2)	4 10 20 34 52 78 110 140 176 212	4·4·8 <sub>2</sub> ·10 <sub>6</sub> ·8 <sub>3</sub> ·8 <sub>3</sub>

**Secondary building units:** 6-2

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Goosecreekite<sup>(1)</sup>

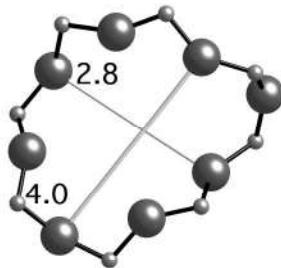
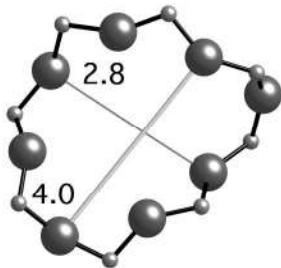
### References:

- (1) Rouse, R.C. and Peacor, D.R. *Am. Mineral.*, **71**, 1494-1501 (1986)

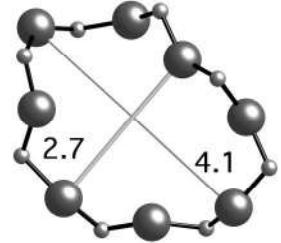
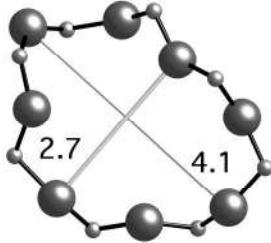
**Crystal chemical data:**  $[\text{Ca}^{2+}_2 (\text{H}_2\text{O})_{10}] [\text{Al}_4\text{Si}_{12} \text{O}_{32}]$ -GOO  
monoclinic, P12<sub>1</sub>1,  $a = 7.401\text{\AA}$ ,  $b = 17.439\text{\AA}$ ,  $c = 7.293\text{\AA}$ ,  $\beta = 105.44^\circ$  <sup>(1)</sup>  
(Relationship to unit cell of Framework type:  
 $\mathbf{a}' = \mathbf{a}/2\cos(\beta'/2)$ ,  $\mathbf{b}' = \mathbf{c}$ ,  $\mathbf{c}' = \mathbf{a}/2\cos(\beta'/2)$   
or, as vectors,  $\mathbf{a}' = (\mathbf{a} + \mathbf{b})/2$ ,  $\mathbf{b}' = \mathbf{c}$ ,  $\mathbf{c}' = (\mathbf{a} - \mathbf{b})/2$ )

**Framework density:** 17.6 T/1000 $\text{\AA}^3$

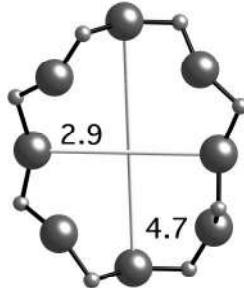
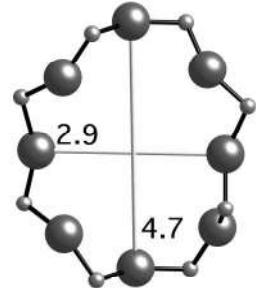
**Channels:** [100] 8 2.8 x 4.0\*  $\leftrightarrow$  [010] 8 2.7 x 4.1\*  $\leftrightarrow$  [001] 8 2.9 x 4.7\*



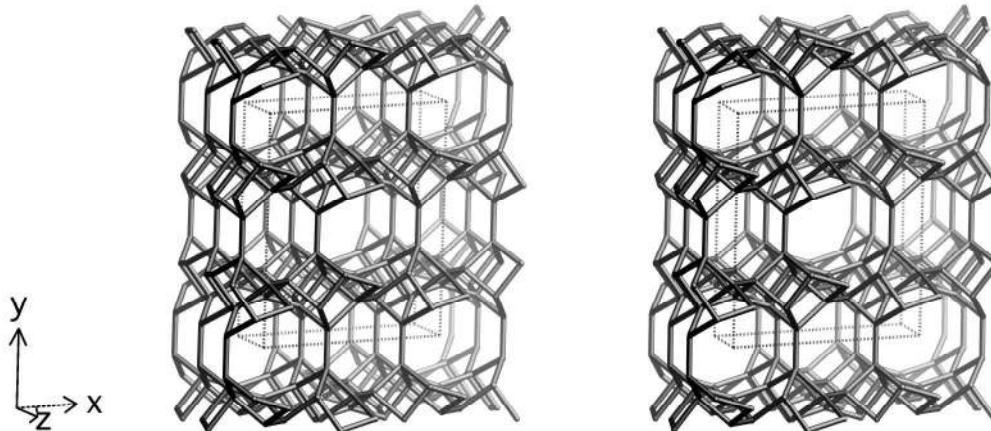
8-ring viewed along [100]



8-ring viewed along [010]



8-ring viewed along [001]



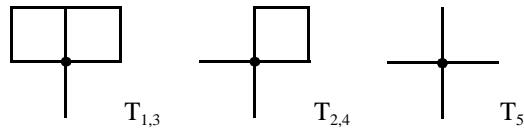
framework viewed along [001]

**Idealized cell constants:** monoclinic, C2/m,  $a = 17.5\text{\AA}$ ,  $b = 17.6\text{\AA}$ ,  $c = 7.4\text{\AA}$ ,  $\beta = 116.1^\circ$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (8, 1)    4 10 20 34 62 85 104 148 201 241 T <sub>2</sub> (8, 1)    4 11 23 39 55 82 127 158 178 221 T <sub>3</sub> (8, 1)    4 10 19 37 58 84 109 149 201 236 T <sub>4</sub> (8, 1)    4 11 21 35 61 89 111 146 194 243 T <sub>5</sub> (4, 2)    4 12 18 34 62 88 110 132 196 254	4·5·4·5·5·8 4·8·5·8·5·8 4·5·4·8·5·5 4·5·5·5·5·8 5·5·5 <sub>2</sub> ·5 <sub>2</sub> ·10·10
---	---	---

**Secondary building units:** 4-4=1

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Heulandite<sup>(1,2)</sup>  
Clinoptilolite<sup>(3)</sup>  
Dehydrated Ca,NH<sub>4</sub>-Heulandite<sup>(4)</sup>  
LZ-219<sup>(5)</sup>

**Alternate designation:** stilbite (mistake in older literature)

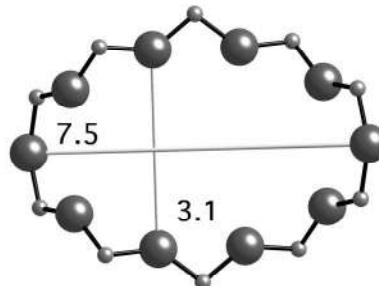
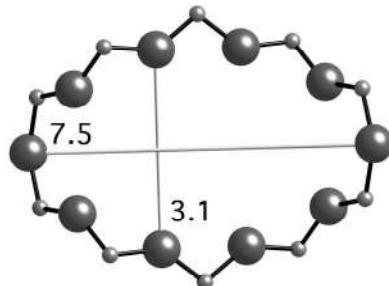
### References:

- (1) Merkle, A.B. and Slaughter, M. *Am. Mineral.*, **52**, 273-276 (1967)
- (2) Alberti, A. *Tschermaks Min. Petr. Mitt.*, **18**, 129-146 (1972)
- (3) Koyama, K. and Takeuchi, Y. *Z. Kristallogr.*, **145**, 216-239 (1977)
- (4) Mortier, W.J. and Pearce, J.R. *Am. Mineral.*, **66**, 309-314 (1981)
- (5) Breck, D.W. and Skeels, G.W. *U.S. Patent 4,503,023* (1985)

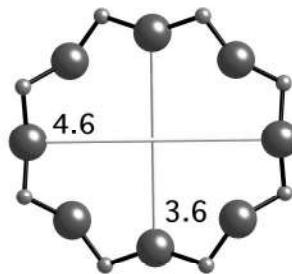
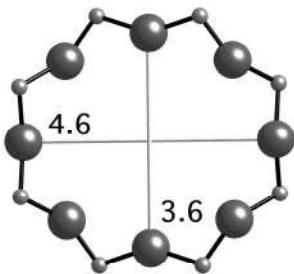
**Crystal chemical data:**  $[\text{Ca}^{2+}_4 (\text{H}_2\text{O})_{24}] [\text{Al}_8\text{Si}_{28} \text{O}_{72}]$ -HEU  
monoclinic, Cm,  $a = 17.718\text{\AA}$ ,  $b = 17.897\text{\AA}$ ,  $c = 7.428\text{\AA}$ ,  $\beta = 116.42^\circ$  <sup>(2)</sup>

**Framework density:** 17.1 T/1000 $\text{\AA}^3$

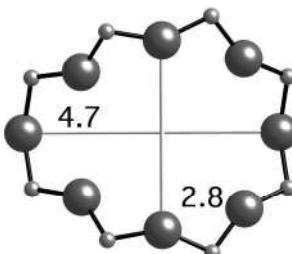
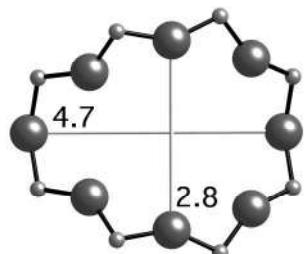
**Channels:** {[001] **10** 3.1 x 7.5\* + **8** 3.6 x 4.6\*}  $\leftrightarrow$  [100] **8** 2.8x 4.7\* (variable due to considerable flexibility of the framework)



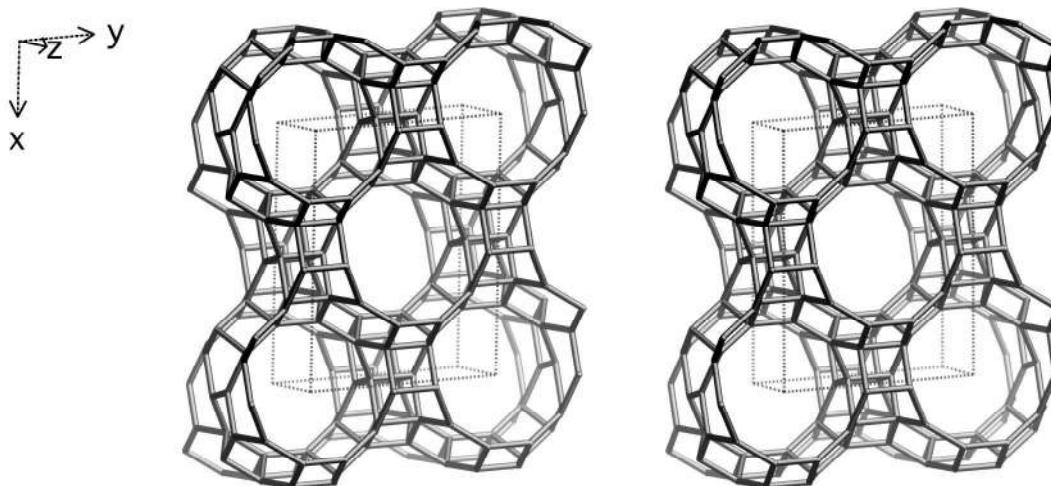
10-ring viewed along [001]



8-ring, also along [001]



8-ring viewed along [100]



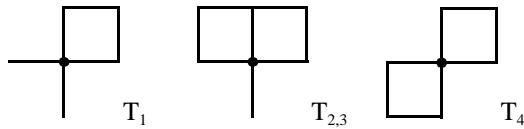
framework viewed along [001]

**Idealized cell constants:** monoclinic, C2/m,  $a = 18.6\text{\AA}$ ,  $b = 13.4\text{\AA}$ ,  $c = 7.6\text{\AA}$ ,  $\beta = 102.3^\circ$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (8, 1)    4 11 20 32 53 77 100 135 166 199	4·6 <sub>2</sub> ·5·6·6·12 <sub>3</sub>
	T <sub>2</sub> (8, 1)    4 10 18 31 52 77 103 127 159 210	4·5·4·6·5·6 <sub>2</sub>
	T <sub>3</sub> (8, 1)    4 10 19 32 53 78 102 126 162 209	4·6·4·6·5·6 <sub>2</sub>
	T <sub>4</sub> (8, 1)    4 10 21 35 50 74 105 133 165 206	4·4·5·12 <sub>5</sub> ·6·6 <sub>2</sub>

**Secondary building units:** 6-2

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*ITQ-4<sup>(1)</sup>  
MCM-58<sup>(2)</sup>  
SSZ-42<sup>(3)</sup>

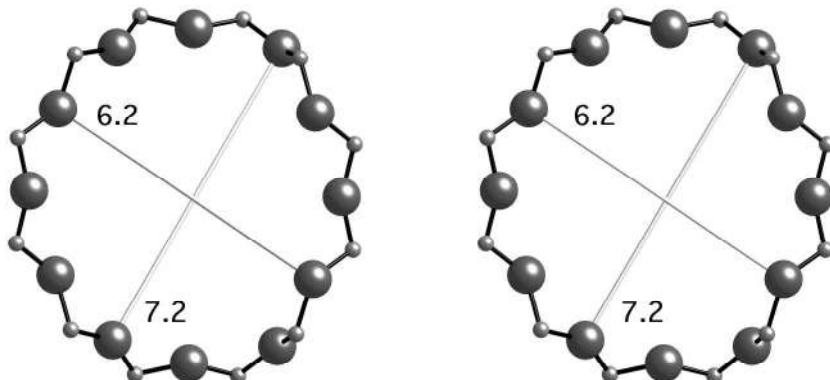
### References:

- (1) Barrett, P.A., Camblor, M.A., Corma, A., Jones, R.H. and Villaescusa, L.A. *Chem. Mater.*, **9**, 1713-1715 (1997)
- (2) Valyocsik, E.W. *WOP 9511196* (1995)
- (3) Chen, C.Y., Finger, L.W., Medrud, R.C., Crozier, P.A., Chan, I.Y., Harris, T.V. and Zones, S.I. *Chem. Commun.*, 1775-1776 (1997)

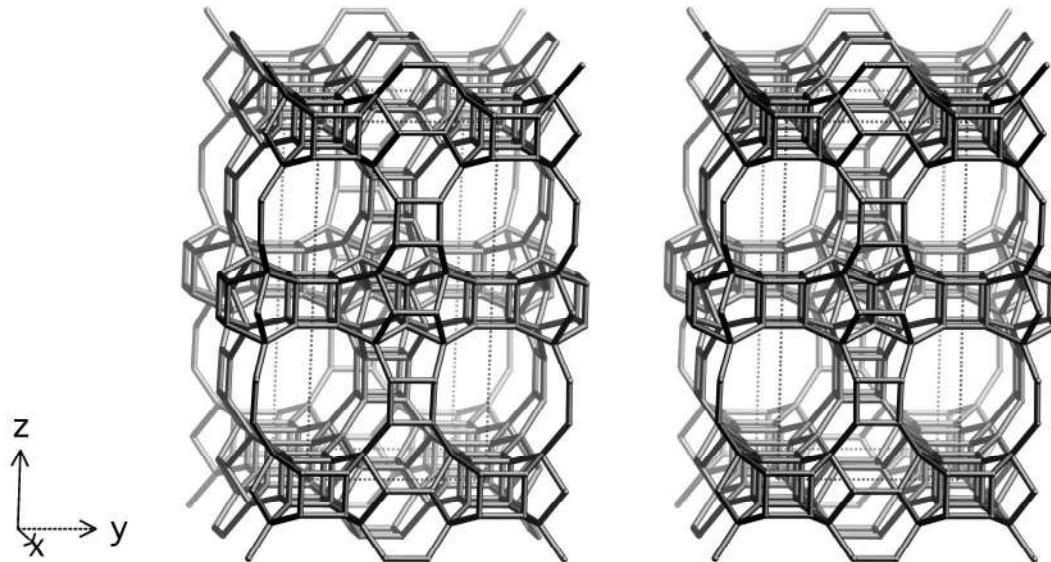
**Crystal chemical data:**  $[\text{Si}_{32} \text{O}_{64}]$ -IFR  
monoclinic, I12/m1  
 $a = 18.652\text{\AA}$ ,  $b = 13.496\text{\AA}$ ,  $c = 7.631\text{\AA}$ ,  $\beta = 101.98^\circ$ <sup>(1)</sup>  
(Relationship to unit cell of Framework Type:  
as vectors,  $\mathbf{a}' = \mathbf{a} + \mathbf{c}$ ,  $\mathbf{b}' = \mathbf{b}$ ,  $\mathbf{c}' = -\mathbf{c}$ )

**Framework density:** 17 T/1000 $\text{\AA}^3$

**Channels:** [001] **12** 6.2 x 7.2\*



12-ring viewed along [001]



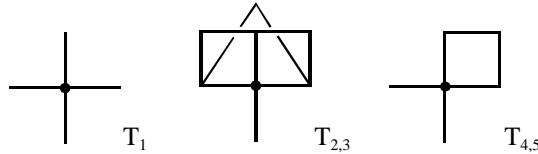
framework viewed along [100]

**Idealized cell constants:** tetragonal, P<sub>4</sub><sub>2</sub>/mmc,  $a = 12.9\text{\AA}$ ,  $c = 25.7\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (16, 1)	4 12 17 30 48 72 99 128 160 199	5·5·5 <sub>2</sub> ·12 <sub>5</sub> ·6·6
	T <sub>2</sub> (16, 1)	4 9 18 32 50 71 96 129 167 200	4·5·4·6·4·12 <sub>7</sub>
	T <sub>3</sub> (16, 1)	4 9 18 32 50 72 97 128 167 203	4·5·4·6·4·12 <sub>4</sub>
	T <sub>4</sub> (8, m)	4 11 20 28 42 74 110 132 150 195	4·5 <sub>2</sub> ·5·6·5·6
	T <sub>5</sub> (8, m)	4 11 20 28 41 70 105 131 154 188	4·5 <sub>2</sub> ·5·6·5·6

**Secondary building units:** 6-2

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*ITQ-7<sup>(1)</sup>

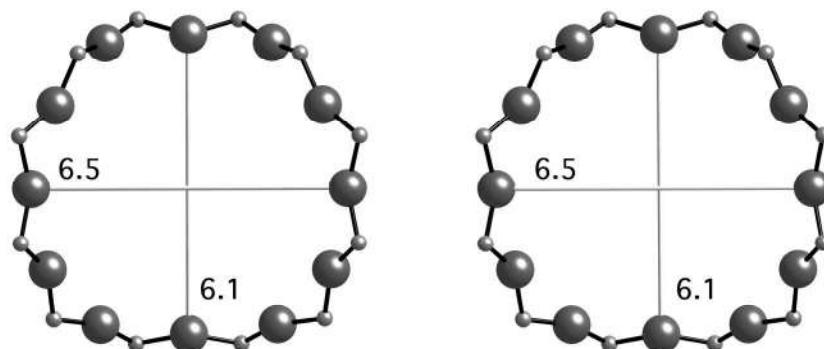
### References:

- (1) Villaescusa, L.A., Barrett, P.A. and Camblor, M.A. *Angew. Chem., Int. Ed.*, **38**, 1997-2000 (1999)

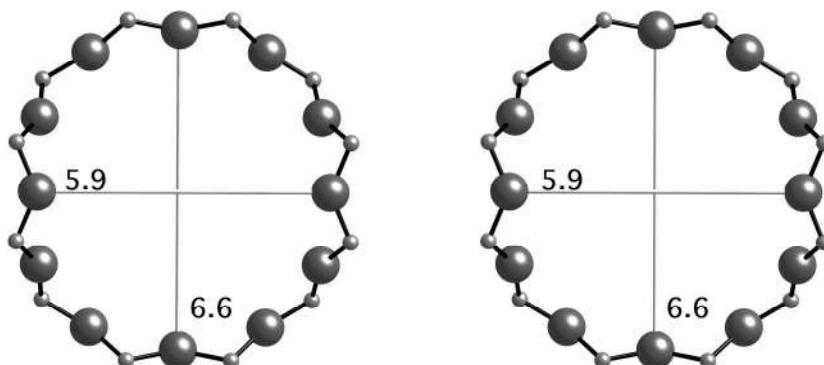
**Crystal chemical data:**  $[\text{Si}_{64} \text{O}_{128}]$ -ISV  
tetragonal,  $\text{P}4_2/\text{mmc}$ ,  $a = 12.853 \text{ \AA}$ ,  $c = 25.214 \text{ \AA}$ <sup>(1)</sup>

**Framework density:**  $15.4 \text{ T}/1000\text{\AA}^3$

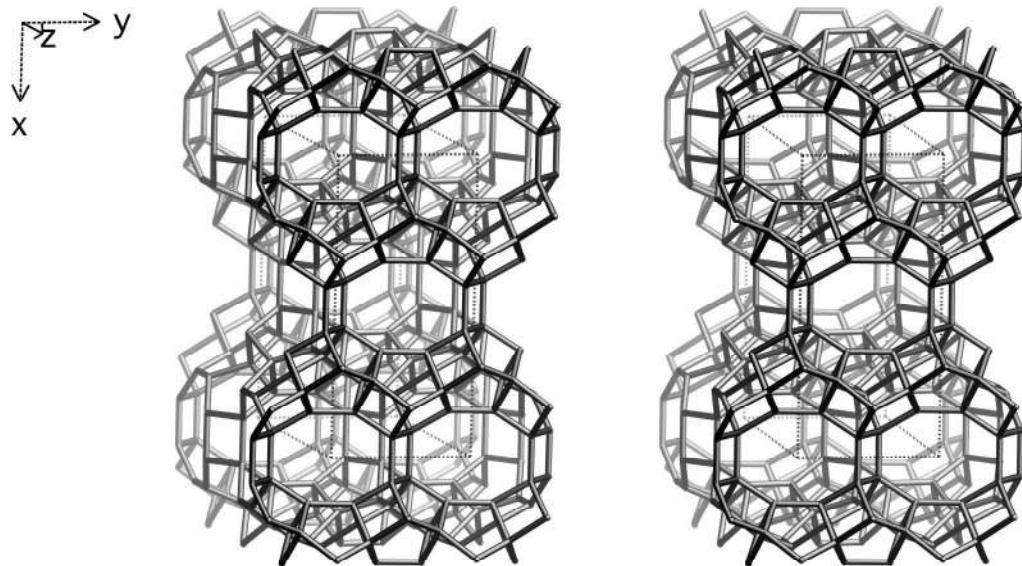
**Channels:**  $<100> \textbf{12 } 6.1 \times 6.5^{**} \leftrightarrow [001] \textbf{12 } 5.9 \times 6.6^*$



*12-ring viewed along  $<100>$*



*12-ring viewed along  $[001]$*



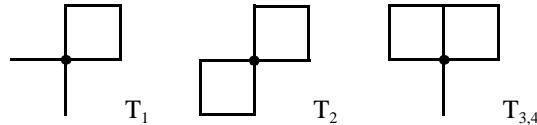
framework viewed along [001]

**Idealized cell constants:** orthorhombic, Cmcm,  $a = 20.8\text{\AA}$ ,  $b = 9.8\text{\AA}$ ,  $c = 20.0\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(16, 1)$	4 11 21 34 53 78 108 138 168 211	4·6·5·6·5·8
	$T_2(16, 1)$	4 10 21 36 54 77 102 135 181 217	4·4·5·8·5·8
	$T_3(16, 1)$	4 10 19 31 50 82 107 132 168 209	4·5·4·6·5·5
	$T_4(16, 1)$	4 10 18 31 55 77 105 136 166 216	4·5·4·8·5·5

**Secondary building units:** 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*ITQ-3<sup>(1)</sup>

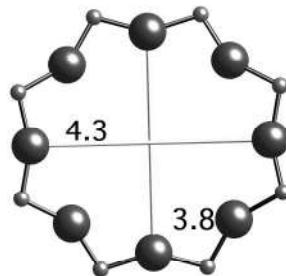
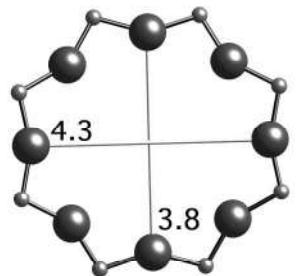
## References:

- (1) Cambor, M.A., Corma, A., Lightfoot, P., Villaescusa, L.A. and Wright, P.A. *Angew. Chem., Int. Ed.*, **36**, 2659-2661 (1997)

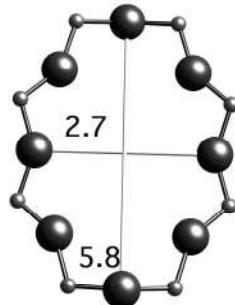
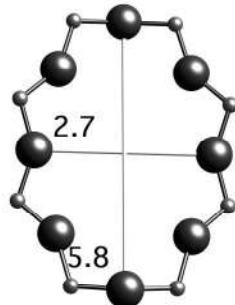
**Crystal chemical data:**  $[\text{Si}_{64} \text{O}_{128}]\text{-ITE}$   
orthorhombic, Cmcm,  $a = 20.622\text{\AA}$ ,  $b = 9.724\text{\AA}$ ,  $c = 19.623\text{\AA}$ <sup>(1)</sup>

**Framework density:** 16.3 T/1000 $\text{\AA}^3$

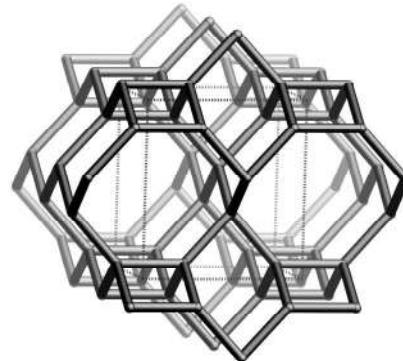
**Channels:** [010] 8 3.8 x 4.3\*  $\leftrightarrow$  [001] 8 2.7 x 5.8\*



8-ring viewed along [010]



8-ring viewed along [001]



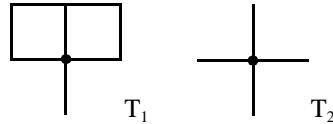
framework viewed along [100]

**Idealized cell constants:** orthorhombic, Pmma,  $a = 5.3\text{\AA}$ ,  $b = 7.5\text{\AA}$ ,  $c = 8.2\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(4, m) \quad 4 \ 10 \ 21 \ 39 \ 61 \ 81 \ 107 \ 148 \ 192 \ 228$        $4\cdot6_2\cdot4\cdot6\cdot8_2$   
 $T_2(2, mm2) \quad 4 \ 12 \ 24 \ 36 \ 56 \ 86 \ 118 \ 146 \ 176 \ 228$        $6\cdot6\cdot6\cdot6_2\cdot6_2$

**Secondary building units:** 6

**Loop configuration of T-Atoms:**



**Isotypic framework structures:**

\*Na-J (Barrer and White)<sup>(1)</sup>  
 Nepheline hydrate<sup>(2)</sup> (not related to nepheline)  
 Synthetic |Na-[Al-Si-O]-JBW<sup>(3)</sup>

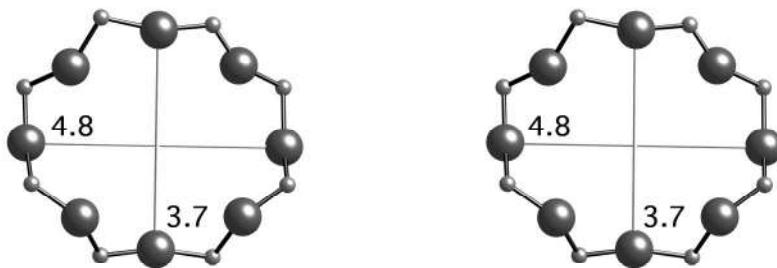
### References:

- (1) Hansen, S. and Fälth, L. *Zeolites*, **2**, 162-166 (1982)
- (2) Rheinhardt, A., Hellner, E. and Ahsbahs, H. *Fortsch. Mineral.*, **60**, 175-176 (1982)
- (3) Ragimov, K.G., Chiragove, M.I., Mustafaev, N.M. and Mamedov, Kh.S. *Sov. Phys. Dokl.*, **23**, 697-698 (1978)

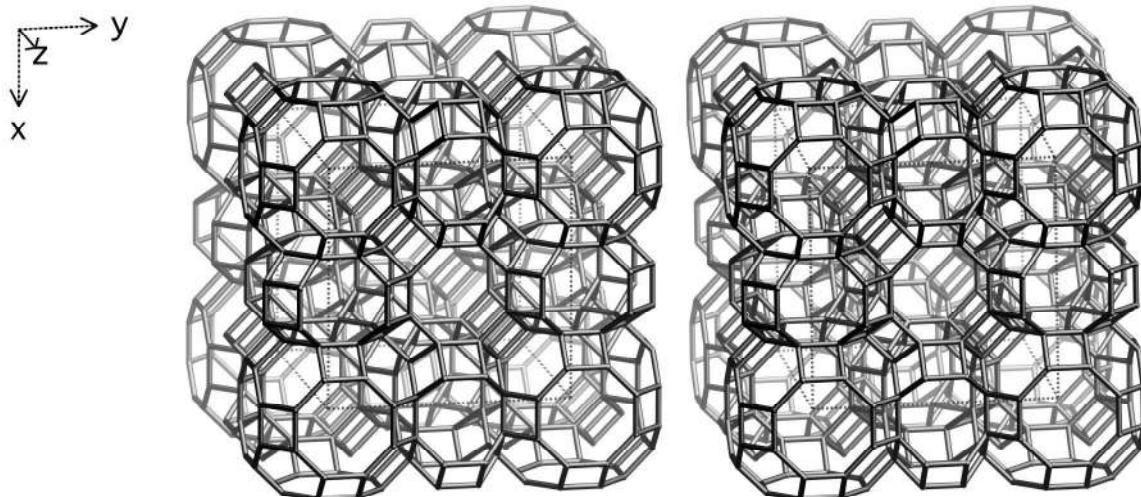
**Crystal chemical data:**  $[\text{Na}^+ \cdot (\text{H}_2\text{O})_{1.5}] \cdot [\text{Al}_3\text{Si}_3\text{O}_{12}]$ -JBW  
orthorhombic, Pna<sub>2</sub><sub>1</sub>,  $a = 16.426\text{\AA}$ ,  $b = 15.014\text{\AA}$ ,  $c = 5.224\text{\AA}$ <sup>(1)</sup>  
(Relationship to unit cell of Framework Type:  $a' = 2c$ ,  $b' = 2b$ ,  $c' = a$ )

**Framework density:** 18.6 T/1000 $\text{\AA}^3$

**Channels:** [001] 8 3.7 x 4.8\*



8-ring along [001]



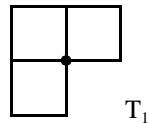
framework viewed along [001]

**Idealized cell constants:** cubic, Im $\bar{3}$ m,  $a = 18.6\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(96, 1) \quad 4 \quad 9 \quad 17 \quad 29 \quad 45 \quad 64 \quad 86 \quad 112 \quad 141 \quad 173$       4·4·4·8·6·8

**Secondary building units:** 6-6 or 6-2 or 8 or 6 or 4

**Loop configuration of T-Atoms:**



**Isotypic framework structures:** \*ZK-5<sup>(1)</sup>  
 (Cs,K)-ZK-5<sup>(2,3)</sup>  
 $P^{(4)}$   
 $Q^{(4)}$

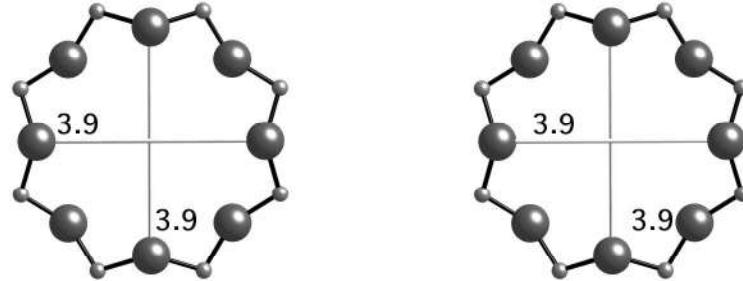
### References:

- (1) Meier, W.M. and Kokotailo, G.T. *Z. Kristallogr.*, **121**, 211-219 (1965)
- (2) Robson, H.E. *U.S. Patent 3,720,753* (1973)
- (3) Parise, J.B., Shannon, R.D., Prince, E. and Cox, D.E. *Z. Kristallogr.*, **165**, 175-190 (1983)
- (4) Barrer, R.M. and Robinson, D. *Z. Kristallogr.*, **135**, 374-390 (1972)

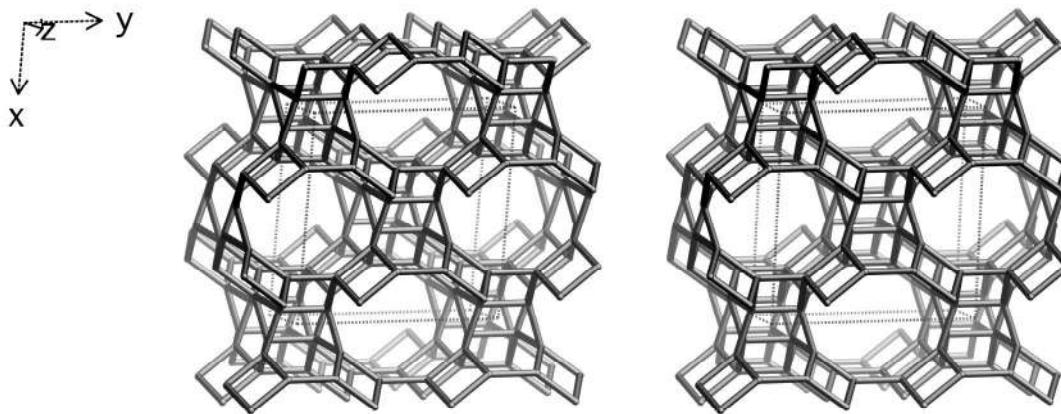
**Crystal chemical data:**  $[\text{Na}^+_{30} (\text{H}_2\text{O})_{98}] [\text{Al}_{30}\text{Si}_{66} \text{O}_{192}]$ -KFI  
cubic,  $\text{Im}\bar{3}\text{m}$ ,  $a = 18.75\text{\AA}$ <sup>(1)</sup>

**Framework density:** 14.6 T/1000 $\text{\AA}^3$

**Channels:**  $<100> 8 3.9 \times 3.9^{***} | <100> 8 3.9 \times 3.9^{***}$



*8-ring viewed along <100>*



framework viewed along [001]

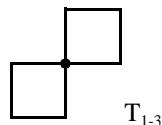
**Idealized cell constants:** monoclinic, C2/m,  $a = 14.6\text{\AA}$ ,  $b = 12.9\text{\AA}$ ,  $c = 7.6\text{\AA}$ ,  $\beta = 111.2^\circ$

**Coordination sequences and vertex symbols:**

$T_1(8, 1)$	4	10	20	33	51	74	99	128	161	199	$4 \cdot 4 \cdot 6 \cdot 6_2 \cdot 6 \cdot 10_4$
$T_2(8, 1)$	4	10	19	32	52	74	99	126	162	203	$4 \cdot 4 \cdot 6 \cdot 6_2 \cdot 6 \cdot 10_2$
$T_3(8, 1)$	4	10	19	33	53	74	96	127	166	201	$4 \cdot 4 \cdot 6 \cdot 6_3 \cdot 6 \cdot 6_3$

**Secondary building units:** 6-2 or 6

**Loop configuration of T-Atoms:**



**Isotypic framework structures:**

- \*Laumontite<sup>(1-4)</sup>
- [Co-Ga-P-O]-LAU<sup>(5,6)</sup>
- [Fe-Ga-P-O]-LAU<sup>(6)</sup>
- [Mn-Ga-P-O]-LAU<sup>(6)</sup>
- Synthetic laumontite<sup>(8)</sup>

**Alternate designation:** Leonhardite<sup>(7)</sup> (discredited)

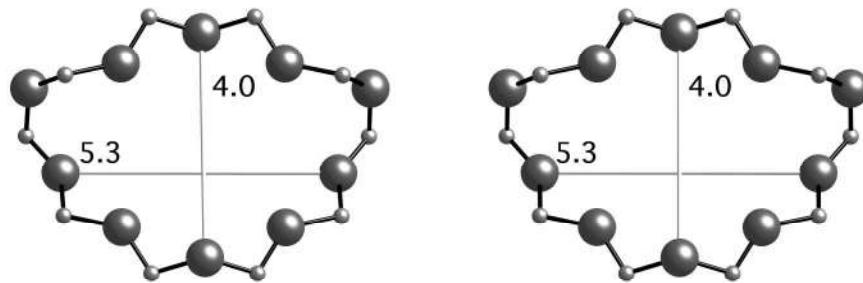
### References:

- (1) Bartl, H. and Fischer, K. *N. Jb. Miner. Mh.*, 33-42 (1967)
- (2) Amirov, S.T., Ilyukhin, V.V. and Belov, N.V. *Dokl. Akad. Nauk SSSR*, **174**, 667- (1967)
- (3) Schramm, V. and Fischer *Adv. Chem. Ser.*, **101**, 259-265 (1971)
- (4) Artioli, G. and Stähli, K. *Zeolites*, **17**, 249-255 (1993)
- (5) Chippindale, A.M. and Walton, R.I. *Chem. Commun.*, 2453-2454 (1994)
- (6) Bond, A.D., Chippindale, A., Cowley, A.R., Readman, J.E. and Powell, A.V. *Zeolites*, **19**, 326-333 (1997)
- (7) Lapham, D.L. *Am. Mineral.*, **48**, 683-689 (1963)
- (8) Ghobarkar, H. and Schaef, O. *Microporous and Mesoporous Materials*, **23**, 55-60 (1998)

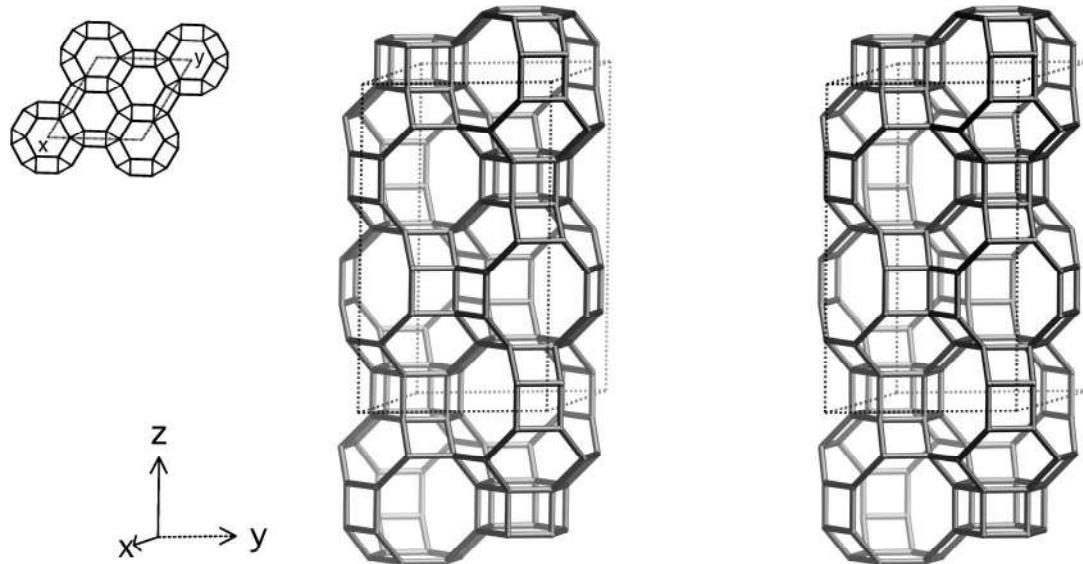
**Crystal chemical data:**  $[\text{Ca}^{2+}_4 (\text{H}_2\text{O})_{16}] [\text{Al}_8\text{Si}_{16} \text{O}_{48}]$ -LAU  
monoclinic, Am,  $a = 7.549\text{\AA}$ ,  $b = 14.740\text{\AA}$ ,  $c = 13.072\text{\AA}$ ,  $\gamma = 111.9^\circ$  <sup>(3)</sup>  
(Relationship to unit cell of Framework Type:  $a' = c$ ,  $b' = a$ ,  $c' = b$ ,  $\gamma' = \beta$ )

**Framework density:** 17.8 T/1000 $\text{\AA}^3$

**Channels:** [100] **10** 4.0 x 5.3\* (contracts upon dehydration)



10-ring viewed along [100]



framework viewed normal to [001] (top left: projection down [001])

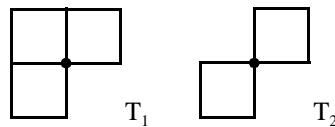
**Idealized cell constants:** trigonal, R $\bar{3}$ m,  $a = 13.2\text{\AA}$ ,  $c = 22.6\text{\AA}$

**Coordination sequences and vertex symbols:**

T <sub>1</sub> (36, 1)	4	9	17	30	49	71	92	114	143	183	4·4·4·6·6·8
T <sub>2</sub> (18, 2)	4	10	20	32	46	64	90	124	156	184	4·4·6·6·8·8

**Secondary building units:** 6

**Loop configuration of T-Atoms:**



**Framework description:** AABCCABC sequence of 6-rings

**Isotypic framework structures:**

- \*Levyne<sup>(1,2)</sup>
- AlPO-35<sup>(3)</sup>
- CoDAF-4<sup>(4)</sup>
- LZ-132<sup>(5)</sup>

NU-3<sup>(6)</sup>  
SAPO-35<sup>(7)</sup>  
ZK-20<sup>(8)</sup>

**Alternate designation:** Levynite (obsolete)

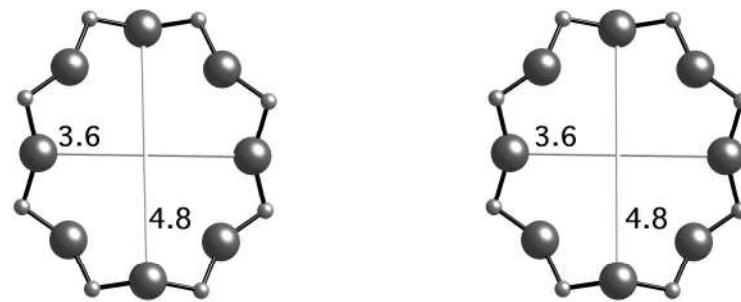
#### References:

- (1) Barrer, R.M. and Kerr, I.S. *Trans. Faraday Soc.*, **55**, 1915-1923 (1959)
- (2) Merlini, S., Galli, E. and Alberti, A. *Tschermaks Min. Petr. Mitt.*, **22**, 117-129 (1975)
- (3) Zhu, G.S., Xiao, F.S., Qiu, S.L., Hun, P.C., Xu, R.R., Ma, S.J. and Terasaki, O. *Microporous Materials*, **11**, 269-273 (1997)
- (4) Barrett, P.A. and Jones, R.H. *Phys. Chem. Chem. Phys.*, **2**, 407-412 (2000)

**Crystal chemical data:**  $[\text{Ca}^{2+}_9 (\text{H}_2\text{O})_{50}] [\text{Al}_{18}\text{Si}_{36} \text{O}_{108}]$ -LEV  
trigonal,  $\text{R}\bar{3}\text{m}$ ,  $a = 13.338\text{\AA}$ ,  $c = 23.014\text{\AA}$  <sup>(2)</sup>

**Framework density:** 15.2 T/1000 $\text{\AA}^3$

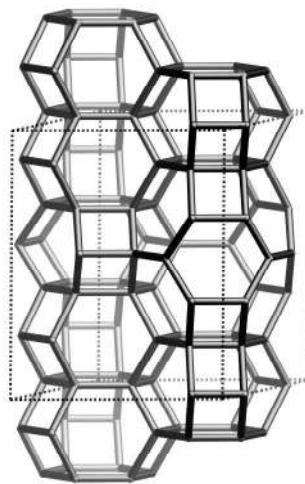
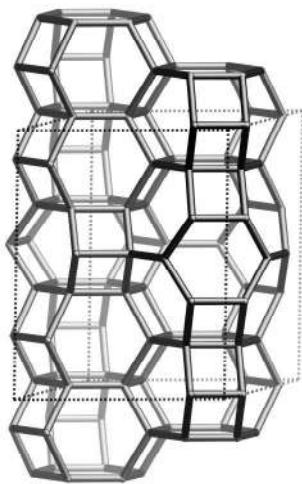
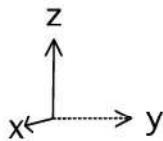
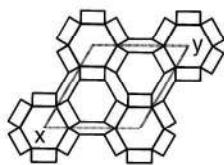
**Channels:**  $\perp [001]$  8 3.6 x 4.8\*\*



8-ring viewed normal to [001]

#### References (cont.):

- (5) Tvaruzkova, Z., Tupa, M., Kiru, P., Nastro, A., Giordano, G. and Trifiro, F. *Int. Zeolite Sym., Wurzburg, Extended Abstracts* (1988)
- (6) McCusker, L.B. *Mater. Sci. Forum*, **133-136**, 423-433 (1993)
- (7) Lok, B.M., Messina, C.A., Patton, R.L., Gajek, R.T., Cannan, T.R. and Flanigen, E.M. *J. Am. Chem. Soc.*, **106**, 6092-6093 (1984)
- (8) Kerr, G.T. *U.S. Patent 3,459,676* (1969)



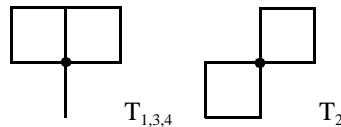
framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** hexagonal, P $\bar{6}$ m2,  $a = 12.3\text{\AA}$ ,  $c = 15.6\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (12, 1)	4 10 20 34 53 76 103 135 170 209	4·6·4·6·6·6
	T <sub>2</sub> (12, 1)	4 10 20 34 54 78 104 134 168 210	4·4·6·6·6·6
	T <sub>3</sub> (6, m)	4 10 20 34 52 74 102 136 172 208	4·6·4·6·6·6
	T <sub>4</sub> (6, m)	4 10 20 34 54 78 104 134 168 210	4·6·4·6·6·6

**Secondary building units:** 6

**Loop configuration of  
T-Atoms:**



**Framework description:** ABABAC sequence of 6-rings

**Isotypic framework  
structures:** \*Liottite<sup>(1,2)</sup>

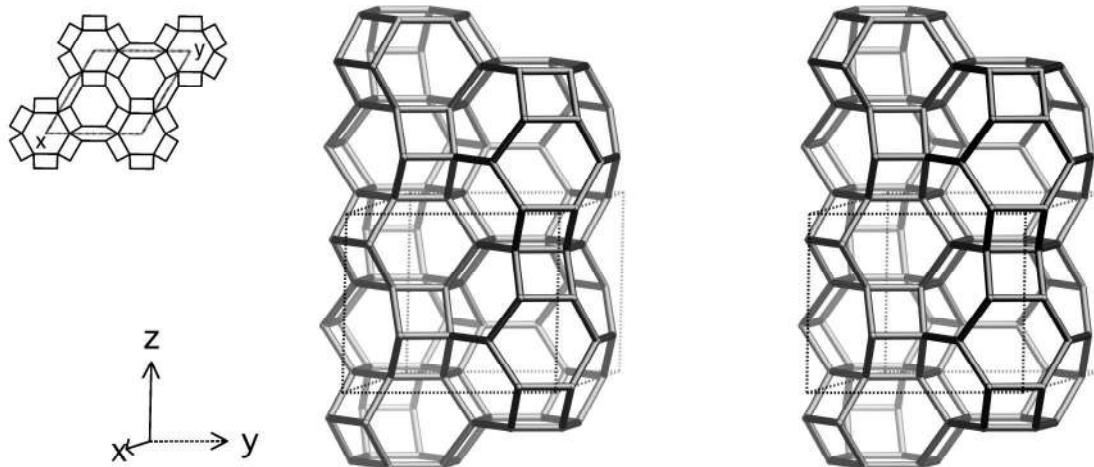
## References:

- (1) Merlini, S. and Orlandi, P. *Am. Mineral.*, **62**, 321-326 (1977)
- (2) Ballirano, P., Merlini, S. and Bonaccorsi, E. *Can. Mineral.*, **34**, 1021-1030 (1996)

**Crystal chemical data:**  $|\text{Ca}^{2+}_8(\text{K}^+, \text{Na}^+)_1\text{6} (\text{SO}_4^{2-})_5\text{Cl}^-_4| [\text{Al}_{18}\text{Si}_{18} \text{O}_{72}]$ -**LIO**  
hexagonal,  $\text{P}\bar{6}$ ,  $a = 12.870\text{\AA}$ ,  $c = 16.096\text{\AA}$  <sup>(2)</sup>

**Framework density:**  $17.6 \text{ T}/1000\text{\AA}^3$

**Channels:** apertures formed by 6-rings only



framework viewed normal to [001] (top left: projection down [001])

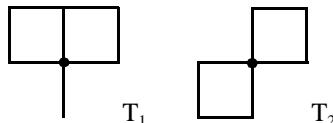
**Idealized cell constants:** hexagonal, P<sub>6</sub><sub>3</sub>/mmc,  $a = 12.6\text{\AA}$ ,  $c = 10.3\text{\AA}$

**Coordination sequences and vertex symbols:**

T <sub>1</sub> (12, m)	4	10	20	34	52	74	102	136	172	210	4·6·4·6·6·6
T <sub>2</sub> (12, 2)	4	10	20	34	54	78	104	134	168	210	4·4·6·6·6·6

**Secondary building units:** 6-2 or 6

**Loop configuration of T-Atoms:**



**Framework description:** ABAC sequence of 6-rings

**Isotypic framework structures:**

\*Losod<sup>(1,2)</sup>  
[Al-Ge-O]-**LOS**<sup>(3)</sup>  
|Li-|[Be-P-O]-**LOS**<sup>(4)</sup>  
Bystrite<sup>(5)</sup>

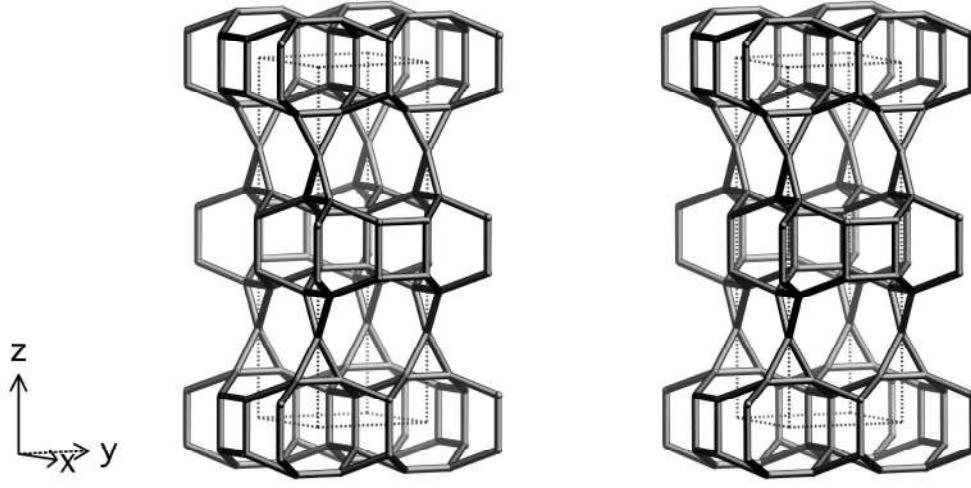
## References:

- (1) Sieber, W. and Meier, W.M. *Helv. Chim. Acta*, **57**, 1533-1549 (1974)
- (2) Schicker, P. *Ph.D. Thesis, ETH, Zürich, Switzerland*, (1988)
- (3) Sokolov, Yu.A., Maksimov, B.A., Ilyukhin, V.V. and Belov, N.V. *Sov. Phys. Dokl.*, **23**, 789-791 (1978)
- (4) Harrison, W.T.A., Gier, T.E. and Stucky, G.D. *Zeolites*, **13**, 242-248 (1993)
- (5) Pobedimskaya, E.A., Terent'eva, L.F., Sapozhnikov, A.N., Kashaev, A.A. and Dorokhova, G.I. *Sov. Phys. Dokl.*, **36**, 553-555 (1991)

**Crystal chemical data:**  $[\text{Na}^+_{12} (\text{H}_2\text{O})_{18}] [\text{Al}_{12}\text{Si}_{12} \text{O}_{48}]$ -**LOS**  
hexagonal,  $\text{P}6_3\text{mc}$ ,  $a = 12.906\text{\AA}$ ,  $c = 10.541\text{\AA}$  <sup>(2)</sup>

**Framework density:**  $15.8 \text{ T}/1000\text{\AA}^3$

**Channels:** apertures formed by 6-rings only



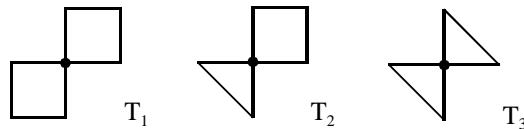
framework viewed normal to [001]

**Idealized cell constants:** tetragonal, P<sub>4</sub><sub>2</sub>/mmc,  $a = 7.2\text{\AA}$ ,  $c = 20.9\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (8, m)    4    10    21    37    58    87    111    138    187    232	4·4·6 <sub>2</sub> ·8·6 <sub>2</sub> ·8
	T <sub>2</sub> (8, m)    4    9    19    39    55    79    113    149    177    229	3·4·8 <sub>3</sub> ·9 <sub>4</sub> ·8 <sub>3</sub> ·9 <sub>4</sub>
	T <sub>3</sub> (2, 42m)    4    8    20    40    54    76    116    144    200    210	3·3·9 <sub>4</sub> ·9 <sub>4</sub> ·9 <sub>4</sub> ·9 <sub>4</sub>

**Secondary building units:** combinations only

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Lovdarite<sup>(1,2)</sup>  
Synthetic lovdarite<sup>(3)</sup>

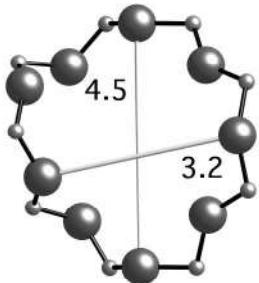
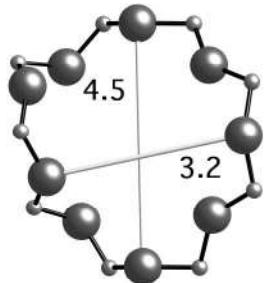
### References:

- (1) Merlino, S. *Acta Crystallogr. (Suppl.)*, **A37**, C189 (1981)
- (2) Merlino, S. *Eur. J. Mineral.*, **2**, 809-817 (1990)
- (3) Ueda, S., Koizumi, M., Baerlocher, Ch., McCusker, L.B. and Meier, W.M. *Preprints of Poster Papers, 7th Int. Zeolite Conf.*, pp. 23-24 (1986), Jap. Assoc. Zeolite, Tokyo

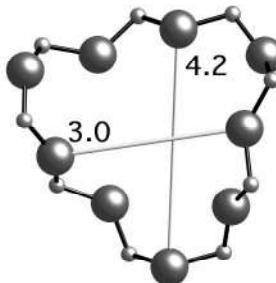
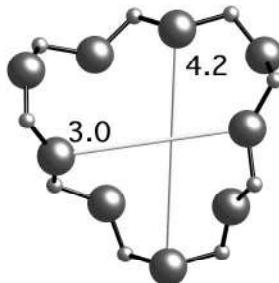
**Crystal chemical data:**  $[\text{K}^+ \text{Na}^+]_{12} (\text{H}_2\text{O})_{18} | [\text{Be}_8\text{Si}_{28}\text{O}_{72}]$ -LOV  
orthorhombic, Pma2,  $a = 39.576\text{\AA}$ ,  $b = 6.931\text{\AA}$ ,  $c = 7.153\text{\AA}$ <sup>(2)</sup>  
(Relationship to unit cell of Framework Type:  $a' = 2c$ ,  $b' = c' = a$ )

**Framework density:** 18.3 T/1000 $\text{\AA}^3$

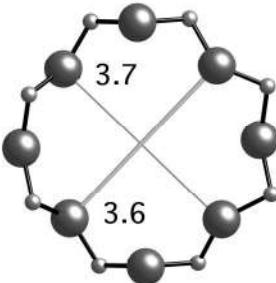
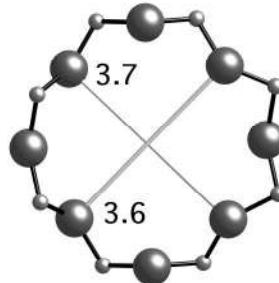
**Channels:** [010] **9** 3.2 x 4.5\*  $\leftrightarrow$  [001] **9** 3.0 x 4.2\*  $\leftrightarrow$  [100] **8** 3.6 x 3.7\*



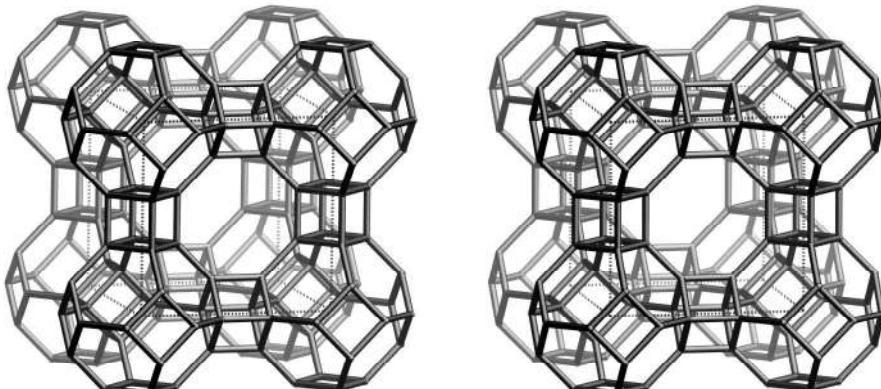
9-ring viewed along [010]



9-ring viewed along [001]



8-ring viewed along [100]



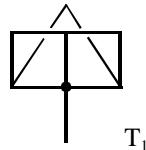
framework viewed along [001]

**Idealized cell constants:** cubic, Pm $\bar{3}$ m,  $a = 11.9\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(24, m)$  4 9 17 28 42 60 81 105 132 162 4·6·4·6·4·8

**Secondary building units:** 8 or 4-4 or 6-2 or 4-2 or 4

**Loop configuration of T-Atoms:**



**Isotypic framework structures:**

\*Linde Type A<sup>(1,2)</sup>  
 $[\text{Al-Ge-O}]$ -LTA<sup>(3)</sup>  
 $[\text{Ga-P-O}]$ -LTA<sup>(4)</sup>  
Alpha<sup>(5)</sup>  
LZ-215<sup>(6)</sup>

N-A<sup>(7)</sup>  
SAPO-42<sup>(8)</sup>  
ZK-21<sup>(9)</sup>  
ZK-22<sup>(9)</sup>  
ZK-4<sup>(10)</sup>

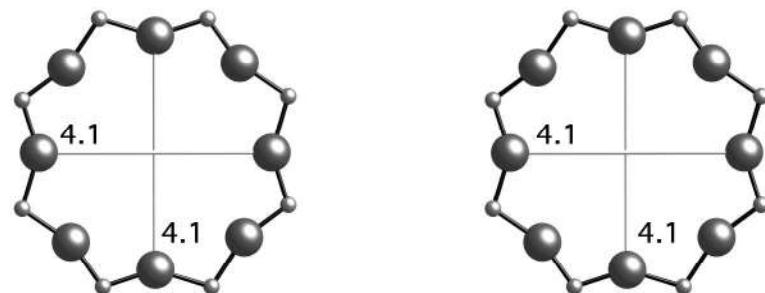
### References:

- (1) Reed, T.B. and Breck, D.W. *J. Am. Chem. Soc.*, **78**, 5972-5977 (1956)
- (2) Gramlich, V. and Meier, W.M. *Z. Kristallogr.*, **133**, 134-149 (1971)
- (3) Barrer, R.M., Baynham, J.W., Bultitude, F.W. and Meier, W.M. *J. Chem. Soc.*, 195-208 (1959)
- (4) Simmen, A., Patarin, J. and Baerlocher, Ch. In *Proc. 9th Int. Zeolite Conf.*, (eds. R. von Ballmoos, J.B. Higgins and M.M.J. Treacy), pp. 433-440 (1993), Butterworth-Heinemann, Boston
- (5) Wadlinger, R.L., Rosinski, E.J. and Plank, C.J. *U.S. Patent 3,375,205* (1968)
- (6) Breck, D.W. and Skeels, G.W. *U.S. Patent 4,503,023* (1985)
- (7) Barrer, R.M. and Denny, P.J. *J. Chem. Soc.*, 971-982 (1961)
- (8) Lok, B.M., Messina, C.A., Patton, R.L., Gajek, R.T., Cannan, T.R. and Flanigen, E.M. *J. Am. Chem. Soc.*, **106**, 6092-6093 (1984)
- (9) Kuehl, G.H. *Inorg. Chem.*, **10**, 2488-2495 (1971)
- (10) Kerr, G.T. *Inorg. Chem.*, **5**, 1537-1539 (1966)

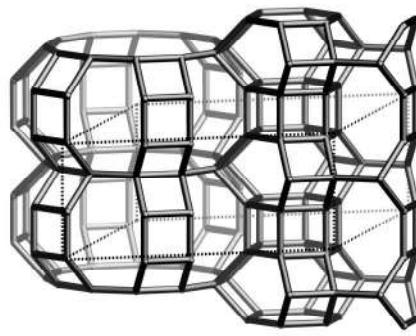
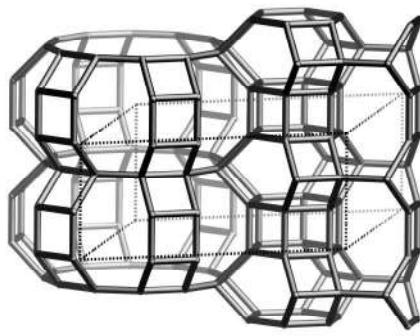
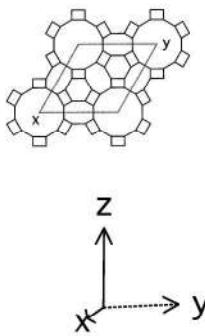
**Crystal chemical data:**  $[\text{Na}^+_{12} (\text{H}_2\text{O})_{27}]_8 [\text{Al}_{12} \text{Si}_{12} \text{O}_{48}]_8$  -LTA  
cubic,  $\text{Fm}\bar{3}\text{c}$ ,  $a = 24.61\text{\AA}$ <sup>(2)</sup>  
(Relationship to unit cell of Framework Type:  $a' = b' = c' = 2a$ )

**Framework density:**  $12.9 \text{ T}/1000\text{\AA}^3$

**Channels:**  $<100> 8 \text{ } 4.1 \times 4.1***$



*8-ring viewed along <100>*



*framework viewed normal to [001]*

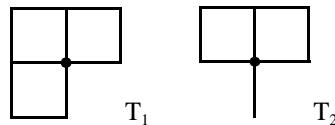
**Idealized cell constants:** hexagonal, P6/mmm,  $a = 18.1\text{\AA}$ ,  $c = 7.6\text{\AA}$

**Coordination sequences and vertex symbols:**

$T_1(24, 1)$	4	9	17	29	46	69	98	131	162	187	4·4·4·6·6·8
$T_2(12, m)$	4	10	21	35	49	66	89	117	150	190	4·8 <sub>3</sub> ·4·8 <sub>3</sub> ·6·12

**Secondary building units:** 8 or 6

**Loop configuration of T-Atoms:**



**Isotypic framework structures:**

\*Linde Type L<sup>(1)</sup>  
(K,Ba)-G,L<sup>(2)</sup>  
Gallosilicate L<sup>(3,4)</sup>  
LZ-212<sup>(5)</sup>  
Periallite<sup>(6,7)</sup>

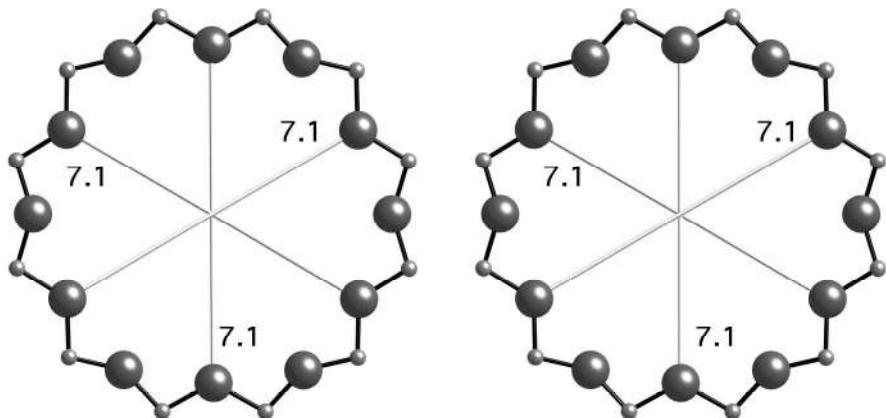
## References:

- (1) Barrer, R.M. and Villiger, H. Z. *Kristallogr.*, **128**, 352-370 (1969)
- (2) Baerlocher, Ch. and Barrer, R.M. Z. *Kristallogr.*, **136**, 245-254 (1972)
- (3) Wright, P.A., Thomas, J.M., Cheetham, A.K. and Nowak, A.K. *Nature*, **318**, 611-614 (1985)
- (4) Newsam, J.M. *Mater. Res. Bull.*, **21**, 661-672 (1986)
- (5) Breck, D.W. and Skeels, G.W. U.S. Patent 4,503,023 (1985)
- (6) Menshikov, Y.P. *Zap. Vses. Mineral. O-va*, **113**, 607-612 (1984)
- (7) Artioli, G. and Kvick, Å. *Eur. J. Mineral.*, **2**, 749-759 (1990)

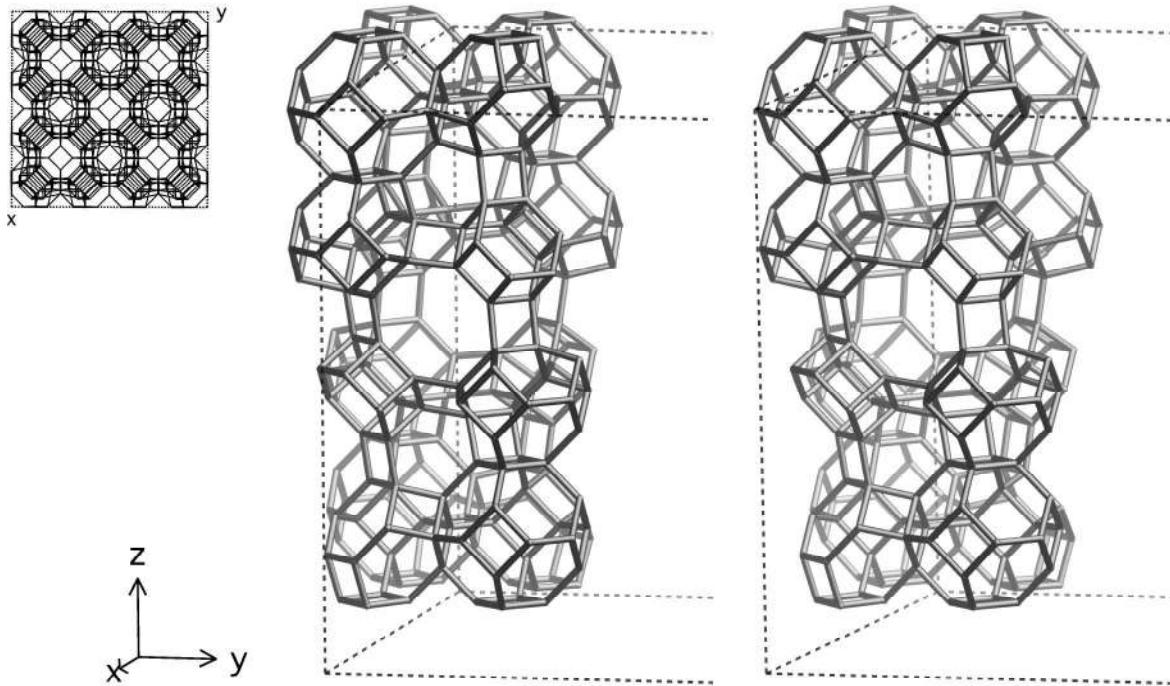
**Crystal chemical data:**  $|\text{K}^+ \cdot \text{Na}^+ \cdot (\text{H}_2\text{O})_{21}| [\text{Al}_9\text{Si}_{27}\text{O}_{72}]$ -LTL  
hexagonal, P6/mmm,  $a = 18.40\text{\AA}$ ,  $c = 7.52\text{\AA}$  <sup>(2)</sup>

**Framework density:** 16.3 T/1000 $\text{\AA}^3$

**Channels:** [001] 12 7.1 x 7.1\*



12-ring viewed along [001]



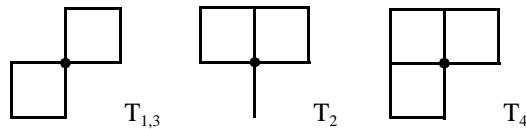
framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** cubic, Fd $\bar{3}$ m,  $a = 35.6\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (192, 1) 4 10 20 34 53 76 102 132 166 205	4·4·6·6·6·6
	T <sub>2</sub> (192, 1) 4 10 20 34 51 71 96 126 162 202	4·6·4·6·6·6
	T <sub>3</sub> (192, 1) 4 10 20 33 50 71 97 129 163 200	4·4·6·6·6·8
	T <sub>4</sub> (192, 1) 4 9 17 30 49 72 97 125 158 197	4·4·4·8·6·6

**Secondary building units:** 6-2 or 6 or 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Linde Type N<sup>(1)</sup>  
NaZ-21<sup>(2)</sup>

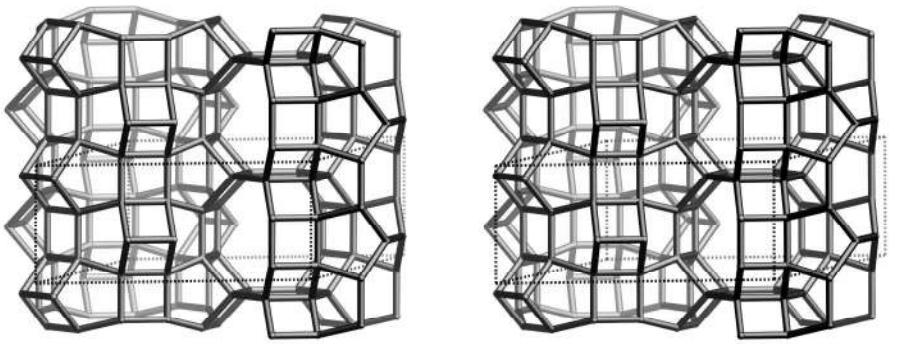
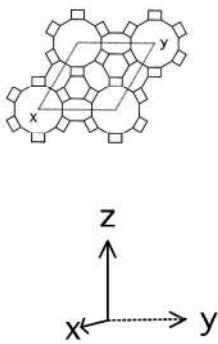
#### References:

- (1) Fälth, L. and Andersson, S. Z. *Kristallogr.*, **160**, 313-316 (1982)
- (2) Shepelev, Yu.F., Smolin, Yu.I., Butikova, I.K. and Tarasov, V.I. *Dokl. Akad. Nauk SSSR*, **272**, 1133-1137 (1983)

**Crystal chemical data:**  $[\text{Na}^+_{384} (\text{H}_2\text{O})_{518}] [\text{Al}_{384}\text{Si}_{384} \text{O}_{1536}]$ -LTN  
cubic,  $\text{Fd}\bar{3}$ ,  $a = 36.93\text{\AA}$  <sup>(1)</sup>

**Framework density:** 15.2 T/1000 $\text{\AA}^3$

**Channels:** apertures formed by 6-rings only



framework viewed normal to [001] (top left: projection down [001])

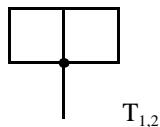
**Idealized cell constants:** hexagonal, P<sub>6</sub><sub>3</sub>/mmc,  $a = 18.1\text{\AA}$ ,  $c = 7.6\text{\AA}$

**Coordination sequences and vertex symbols:**

T <sub>1</sub> (24, 1)	4	10	20	35	54	78	104	134	171	210	4·5·4·5·8·12
T <sub>2</sub> (12, m)	4	10	21	36	53	74	104	138	174	212	4·8 <sub>2</sub> ·4·8 <sub>2</sub> ·5·6

**Secondary building units:** 5-1 or 4-2 or 4

**Loop configuration of T-Atoms:**



**Isotypic framework structures:**

- \*Mazzite<sup>(1,2)</sup>
- [Ga-Si-O]-MAZ<sup>(3)</sup>
- LZ-202<sup>(4)</sup>
- Omega<sup>(5)</sup>
- ZSM-4<sup>(6)</sup>

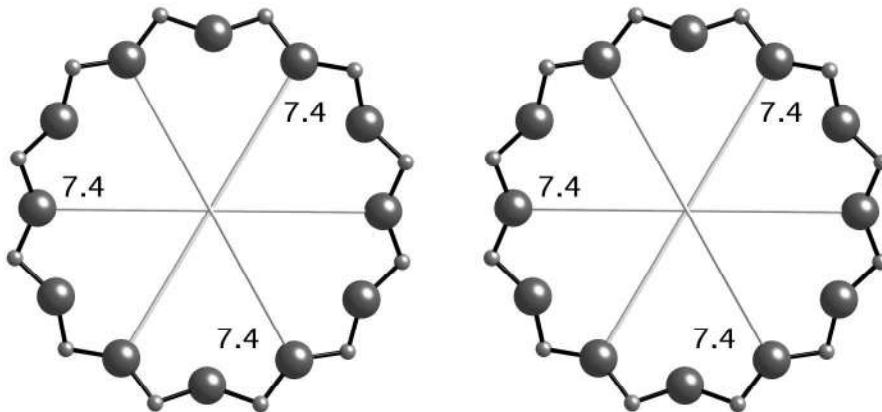
## References:

- (1) Galli, E. *Cryst. Struct. Comm.*, **3**, 339-344 (1974)
- (2) Galli, E. *Rend. Soc. Ital. Mineral. Petrol.*, **31**, 599-612 (1975)
- (3) Newsam, J.M., Jarman, R.H. and Jacobson, A.J. *Mater. Res. Bull.*, **20**, 125-136 (1985)
- (4) Breck, D.W. and Skeels, G.W. *U.S. Patent 4,503,023* (1985)
- (5) Galli, E. *Cryst. Struct. Comm.*, **3**, 339-344 (1974)
- (6) Rubin, M.K., Plank, C.J. and Rosinski, E.J. *U.S. Patent 4,021,447* (1977)

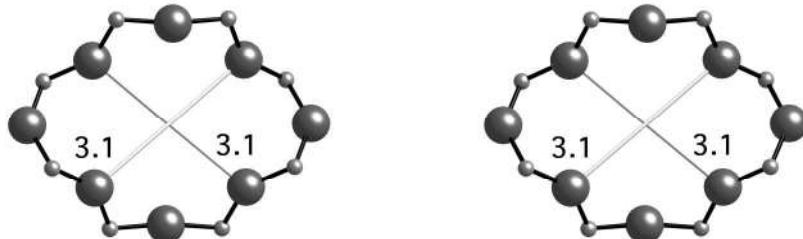
**Crystal chemical data:**  $[(\text{Na}^+, \text{K}^+, \text{Ca}^{2+}, \text{Mg}^{2+})_5 (\text{H}_2\text{O})_{28}] [\text{Al}_{10}\text{Si}_{26}\text{O}_{72}]$ -MAZ  
hexagonal, P6<sub>3</sub>/mmc,  $a = 18.392\text{\AA}$ ,  $c = 7.646\text{\AA}$ <sup>(2)</sup>

**Framework density:** 16.1 T/1000 $\text{\AA}^3$

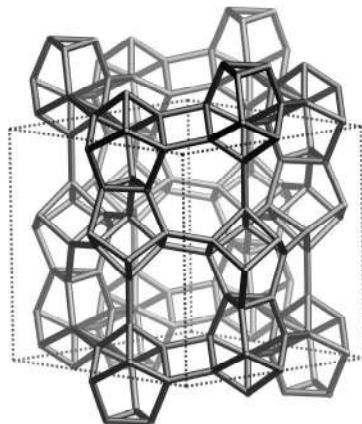
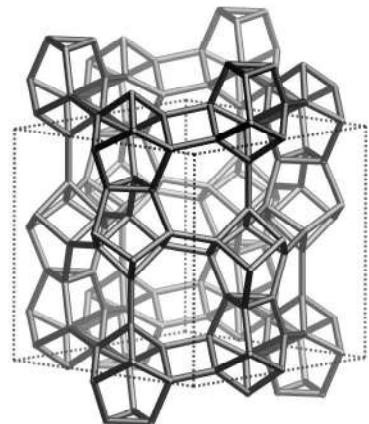
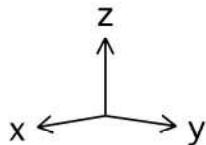
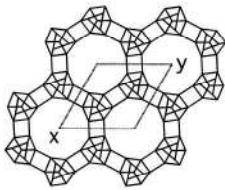
**Channels:** [001] **12** 7.4 x 7.4\* | [001] **8** 3.1 x 3.1\*\*\*



*12-ring viewed along [001]*



*limiting 8-ring along [001]*



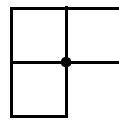
framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** hexagonal, P6<sub>3</sub>/m,  $a = 13.1\text{\AA}$ ,  $c = 15.6\text{\AA}$

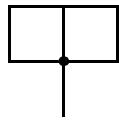
<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (12, 1)    4    9    18    30    46    63    94    125    152    183	4·4·4·12·5·7
	T <sub>2</sub> (12, 1)    4    10    17    30    46    67    91    123    153    190	4·5·4·7·5·12
	T <sub>3</sub> (6, m)    4    10    16    26    46    66    94    114    158    194	3·7·5·5·5·5
	T <sub>4</sub> (4, 3)    4    9    18    30    39    67    98    121    147    189	4·7·4·7·4·7

**Secondary building units:** combinations only

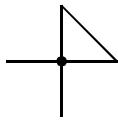
**Loop configuration of  
T-Atoms:**



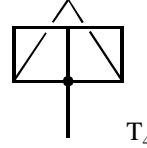
T<sub>1</sub>



T<sub>2</sub>



T<sub>3</sub>



T<sub>4</sub>

**Isotypic framework  
structures:** \*ZSM-18<sup>(1)</sup>

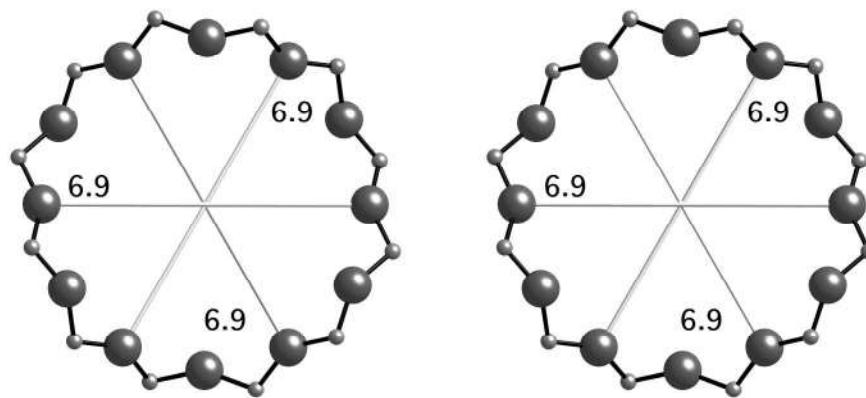
## References:

- (1) Lawton, S.L. and Rohrbaugh, W.J. *Science*, **247**, 1319-1321 (1990)

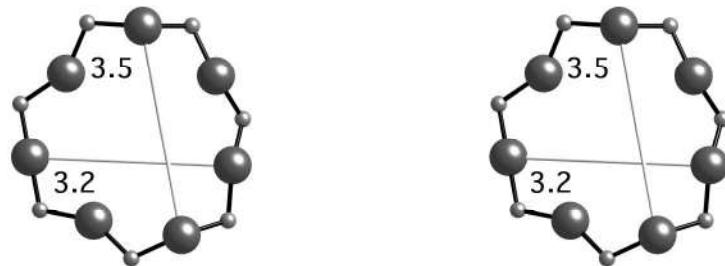
**Crystal chemical data:**  $[\text{Na}^+ \text{n} (\text{H}_2\text{O})_{28}] [\text{Al}_n \text{Si}_{34-n} \text{O}_{68}]$ -MEI,  $n = 2.1 - 5.7$   
hexagonal,  $P6_3/m$ ,  $a = 13.175\text{\AA}$ ,  $c = 15.848\text{\AA}$ <sup>(1)</sup>

**Framework density:**  $14.3 \text{ T}/1000\text{\AA}^3$

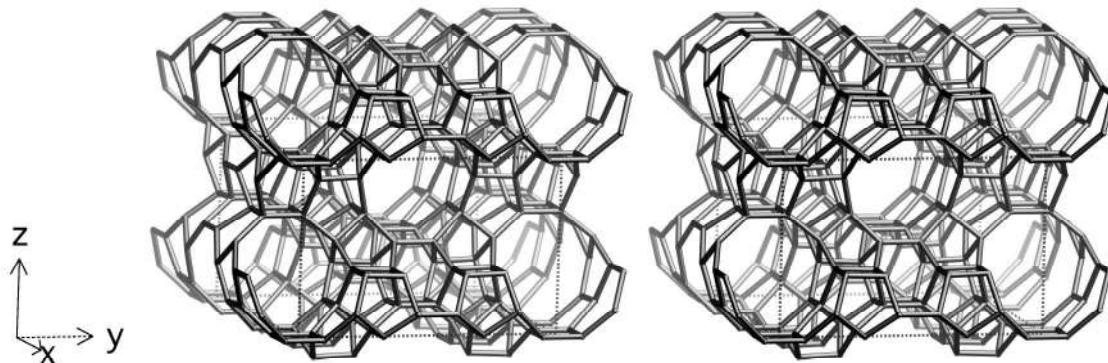
**Channels:** [001] **12**  $6.9 \times 6.9^* \leftrightarrow \perp$  [001] **7**  $3.2 \times 3.5^{**}$



12-ring viewed along [001]



7-ring viewed normal to [001]



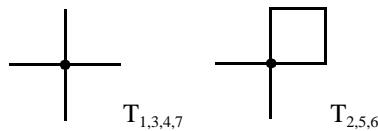
framework viewed along [100]

**Idealized cell constants:** tetragonal, I<sup>4</sup>m2,  $a = 20.3\text{\AA}$ ,  $c = 13.5\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (16, 1)	4 12 21 36 63 88 121 153 192 249	5·5·5·5·5·6
	T <sub>2</sub> (16, 1)	4 11 23 38 61 93 121 153 198 246	4·5·5·6 <sub>2</sub> ·5·10 <sub>2</sub>
	T <sub>3</sub> (16, 1)	4 12 23 38 62 91 116 155 203 244	5·5·5·5 <sub>2</sub> ·5·10 <sub>3</sub>
	T <sub>4</sub> (16, 1)	4 12 23 38 59 87 122 158 198 243	5·6 <sub>2</sub> ·5·10 <sub>2</sub> ·5 <sub>2</sub> ·6
	T <sub>5</sub> (16, 1)	4 11 22 36 57 90 127 157 194 244	4·5·5·6·5·8 <sub>2</sub>
	T <sub>6</sub> (8, 2)	4 11 23 38 57 88 126 158 191 236	4·5 <sub>2</sub> ·6 <sub>2</sub> ·10·10
	T <sub>7</sub> (8, 2)	4 12 21 40 63 83 124 155 197 244	5·5·5·5 <sub>2</sub> ·10 <sub>2</sub>

**Secondary building units:** 5-1

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*ZSM-11<sup>(1-3)</sup>  
Bor-D (MFI/MEL intergrowth)<sup>(4)</sup>  
Boralite D<sup>(5)</sup>

SSZ-46<sup>(6,7)</sup>  
Silicalite 2<sup>(8)</sup>  
TS-2<sup>(9)</sup>

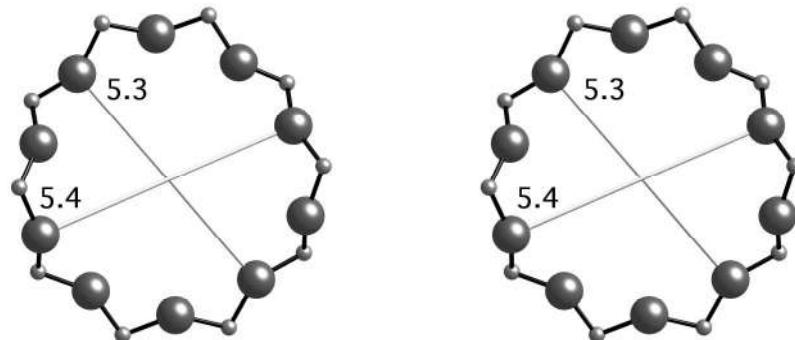
### References:

- (1) Kokotailo, G.T., Chu, P., Lawton, S.L. and Meier, W.M. *Nature*, **275**, 119-120 (1978)
- (2) Fyfe, C.A., Gies, H., Kokotailo, G.T., Pasztor, C., Strobl, H. and Cox, D.E. *J. Am. Chem. Soc.*, **111**, 2470-2474 (1989)
- (3) van Koningsveld, H., den Exter, M.J., Koegler, J.H., Laman, C.D., Njo, S.L. and Graafsma, H. In *Proc. 12th Int. Zeolite Conf.*, (eds. M.M.J. Treacy, B.K. Marcus, M.E. Bisher and J.B. Higgins), pp. 2419-2424 (1999), MRS, Warrendale, PA
- (4) Perego, G. and Cesari, M. *J. Appl. Crystallogr.*, **17**, 403-410 (1984)
- (5) Taramasso, M., Manara, G., Fattore, V. and Notari, B. *GB Patent 2,024,790*, (1980)
- (6) Terasaki, O., Ohsuna, T., Sakuma, H., Watanabe, D., Nakagawa, Y. and Medrud, R.C. *Chem. Mater.*, **8**, 463-468 (1996)
- (7) Nakagawa, Y. and Dartt, C. *U.S. Patent 5,968,474* (1999)

**Crystal chemical data:**  $[\text{Na}^+ \text{n} (\text{H}_2\text{O})_{16}] [\text{Al}_n \text{Si}_{96-n} \text{O}_{192}]$ -MEL,  $n < 16$   
tetragonal,  $\bar{I}\bar{4}m2$ ,  $a = 20.12\text{\AA}$ ,  $c = 13.44\text{\AA}$ <sup>(1)</sup>

**Framework density:** 17.6 T/1000 $\text{\AA}^3$

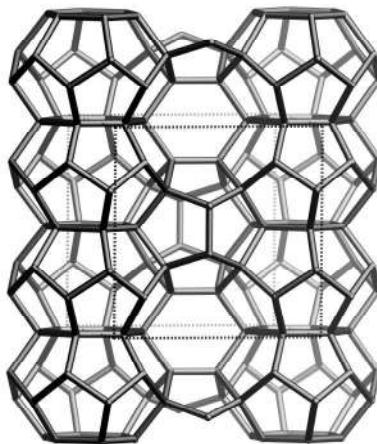
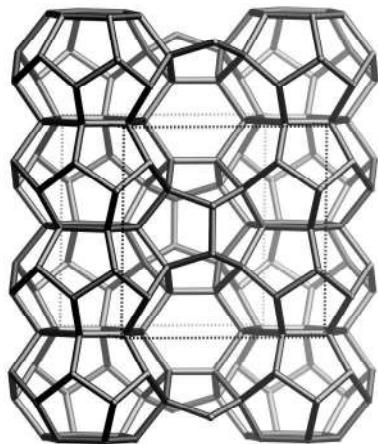
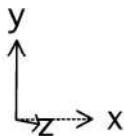
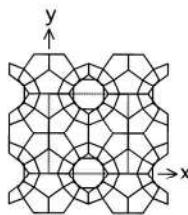
**Channels:**  $<100> \textbf{10} \text{ } 5.3 \times 5.4^{***}$



10-ring viewed along  $<100>$

**References (cont.):**

- (8) Bibby, D.M., Milestone, N.B. and Aldridge, L.P. *Nature*, **280**, 664-665 (1979)
- (9) Reddy, J.S. and Kumar, R. *Zeolites*, **12**, 95-100 (1992)



*framework viewed along [001] (top left: projection down [001])*

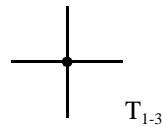
**Idealized cell constants:** cubic, Pm $\bar{3}$ n,  $a = 13.7\text{\AA}$

**Coordination sequences  
and vertex symbols:**

T <sub>1</sub> (24, m)	4	12	25	42	69	100	129	176	229	277	5·5·5·5·5·6
T <sub>2</sub> (16, 3)	4	12	24	42	67	95	133	177	219	277	5·5·5·5·5
T <sub>3</sub> (6, 42m)	4	12	26	44	64	98	144	172	222	272	5·5·5·5·6·6

**Secondary building units:** combinations only

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Melanophlogite<sup>(1)</sup>  
Synthetic melanophlogite<sup>(2)</sup>

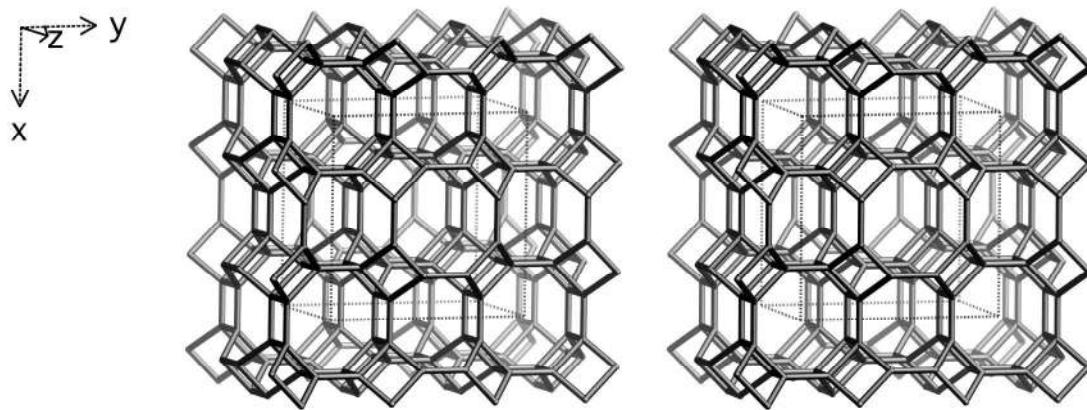
### References:

- (1) Gies, H. Z. *Kristallogr.*, **164**, 247-257 (1983)
- (2) Gies, H., Gerke, H. and Liebau, F. N. *Jb. Miner. Mh.*, 119-124 (1982)

**Crystal chemical data:**  $[(\text{CH}_4, \text{N}_2, \text{CO}_2)_x | [\text{Si}_{46} \text{O}_{92}]]$ -MEP  
cubic,  $\text{Pm}\bar{3}\text{n}$ ,  $a = 13.436 \text{\AA}$ <sup>(1)</sup>  
(Data refer to structure at 200°C; tetragonal at 25°C)

**Framework density:** 19 T/1000 $\text{\AA}^3$

**Channels:** apertures formed by 6-rings only



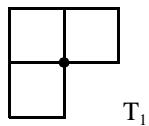
framework viewed along [001]

**Idealized cell constants:** tetragonal, I4/mmm,  $a = 14.0\text{\AA}$ ,  $c = 10.0\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(32, 1) \quad 4 \quad 9 \quad 18 \quad 32 \quad 49 \quad 69 \quad 93 \quad 121 \quad 153 \quad 189$        $4\cdot4\cdot4\cdot8_2\cdot8\cdot8$

**Secondary building units:** 8-8 or 8 or 4

**Loop configuration of T-Atoms:**



**Isotypic framework structures:**

\*Merlinoite<sup>(1,2)</sup>

[Al-Co-P-O]-MER<sup>(3)</sup>

|Ba-[Al-Si-O]-MER<sup>(4)</sup>

|Ba-Cl-[Al-Si-O]-MER<sup>(5)</sup>

|NH<sub>4</sub>-|[Be-P-O]-MER<sup>(6)</sup>

K-M<sup>(4,7)</sup>

Linde W<sup>(4,8)</sup>

Synthetic merlinoite<sup>(9)</sup>

Zeolite W<sup>(10)</sup>

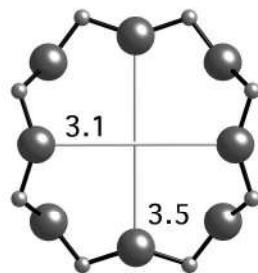
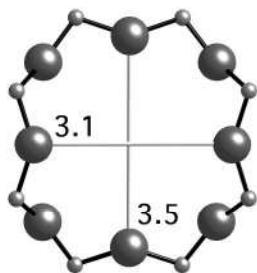
## References:

- (1) Passaglia, E., Pongiluppi, D. and Rinaldi, R. *N. Jb. Miner. Mh.*, 355-364 (1977)
- (2) Galli, E., Gottardi, G. and Pongiluppi, D. *N. Jb. Miner. Mh.*, 1-9 (1979)
- (3) Feng, P.Y., Bu, X.H. and Stucky, G.D. *Nature*, **388**, 735-741 (1997)
- (4) Gottardi, G. and Galli, E. *Natural Zeolites*, p. 157 (1985), Springer-Verlag, Berlin
- (5) Solov'eva, L.P., Borisov, S.V. and Bakakin, V.V. *Sov. Phys. Crystallogr.*, **16**, 1035-1038 (1972)
- (6) Bu, X., Gier, T.E. and Stucky, G.D. *Microporous and Mesoporous Materials*, **26**, 61-66 (1998)
- (7) Barrer, R.M. and Baynham, J.W. *J. Chem. Soc.*, 2882-2891 (1956)
- (8) Sherman, J.D. *ACS Sym. Ser.*, **40**, 30-42 (1977)
- (9) Barrett, P.A., Valencia, S. and Camblor, M.A. *J. Mater. Chem.*, **8**, 2263-2268 (1998)
- (10) Bieniok, A., Bornholdt, K., Brendel, U. and Baur, W.H. *J. Mater. Chem.*, **6**, 271-275 (1996)

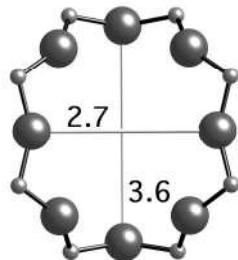
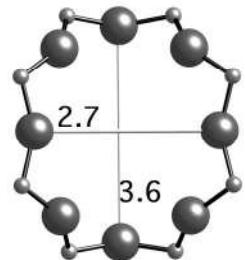
**Crystal chemical data:**  $[\text{K}^+ \text{Ca}^{2+}]_{24} [\text{Al}_9\text{Si}_{23} \text{O}_{64}]$ -MER  
orthorhombic, Imm̄,  $a = 14.116\text{\AA}$ ,  $b = 14.229\text{\AA}$ ,  $c = 9.946\text{\AA}$ <sup>(2)</sup>  
(Relationship to unit cell of Framework Type:  $a' = b' = a$ ,  $c' = c$ )

**Framework density:** 16 T/1000 $\text{\AA}^3$

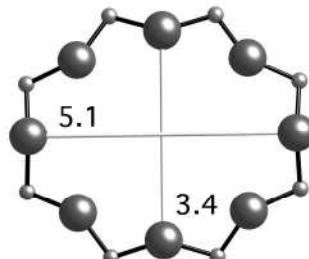
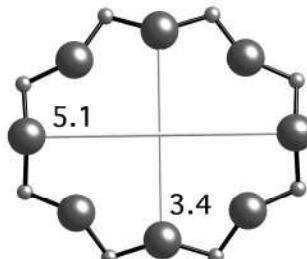
**Channels:** [100] 8 3.1 x 3.5\* ↔ [010] 8 2.7 x 3.6\*  
↔ [001] {8 3.4 x 5.1\* + 8 3.3 x 3.3\*}



8-ring viewed along [100]

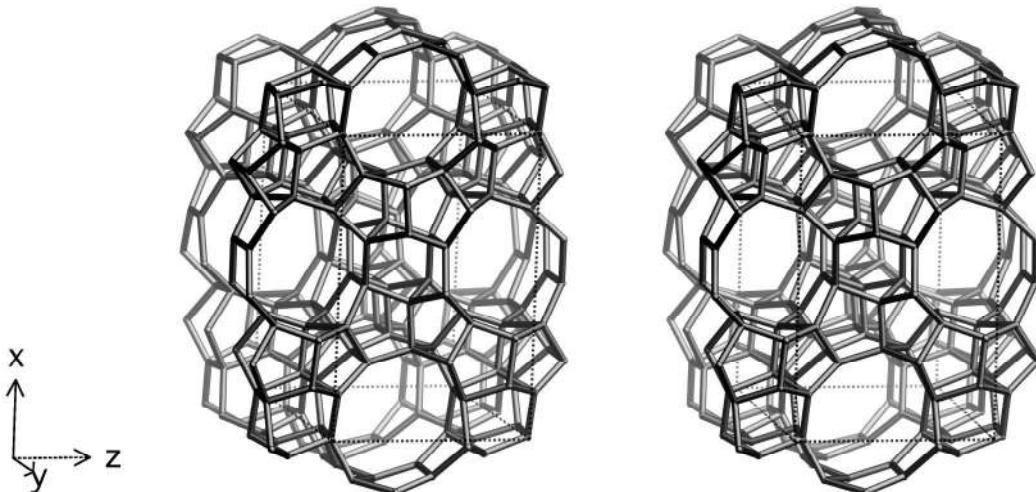


8-ring viewed along [010]



8-ring viewed along [001]

See Appendix A for 2nd 8-ring viewed along [001]



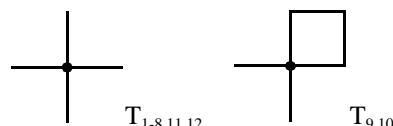
framework viewed along [010]

**Idealized cell constants:** orthorhombic, Pnma,  $a = 20.1\text{\AA}$ ,  $b = 19.7\text{\AA}$ ,  $c = 13.1\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(8, 1)$	4 12 22 41 61 88 125 159 198 250	$5 \cdot 5 \cdot 5 \cdot 10_2 \cdot 5_2 \cdot 6$
	$T_2(8, 1)$	4 12 22 39 64 91 117 158 209 247	$5 \cdot 5 \cdot 5 \cdot 5_2 \cdot 5 \cdot 10_3$
	$T_3(8, 1)$	4 12 23 37 62 91 120 157 206 250	$5 \cdot 5 \cdot 5 \cdot 5_2 \cdot 5 \cdot 10_3$
	$T_4(8, 1)$	4 12 21 36 61 90 122 159 196 251	$5 \cdot 5 \cdot 5 \cdot 5_2 \cdot 5 \cdot 6$
	$T_5(8, 1)$	4 12 24 38 63 93 123 157 206 247	$5 \cdot 5 \cdot 5 \cdot 6_2 \cdot 5 \cdot 10_3$
	$T_6(8, 1)$	4 12 22 40 61 88 124 156 197 253	$5 \cdot 5 \cdot 5 \cdot 5_2 \cdot 10_3$
	$T_7(8, 1)$	4 12 24 38 56 90 132 164 193 241	$5 \cdot 6_2 \cdot 5_2 \cdot 6_2 \cdot 10 \cdot 10$
	$T_8(8, 1)$	4 12 21 37 63 90 121 155 201 253	$5 \cdot 5 \cdot 5 \cdot 5 \cdot 5 \cdot 6$
	$T_9(8, 1)$	4 11 23 39 62 93 119 153 204 254	$4 \cdot 5 \cdot 5 \cdot 6_2 \cdot 5 \cdot 10_3$
	$T_{10}(8, 1)$	4 11 22 36 61 93 120 154 200 255	$4 \cdot 5 \cdot 5 \cdot 6_2 \cdot 5 \cdot 10_3$
	$T_{11}(8, 1)$	4 12 22 38 59 92 125 159 202 250	$5 \cdot 5 \cdot 5 \cdot 6 \cdot 5 \cdot 6_2$
	$T_{12}(8, 1)$	4 12 23 38 59 89 126 161 196 246	$5 \cdot 6_2 \cdot 5 \cdot 10_2 \cdot 5_2 \cdot 6_2$

**Secondary building units:** 5-1

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*ZSM-5<sup>(1-3)</sup>

(See Appendix A for additional structures and references)

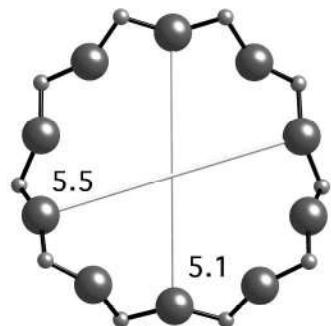
#### References:

- (1) Kokotailo, G.T., Lawton, S.L., Olson, D.H. and Meier, W.M. *Nature*, **272**, 437-438 (1978)
- (2) Olson, D.H., Kokotailo, G.T., Lawton, S.L. and Meier, W.M. *J. Phys. Chem.*, **85**, 2238-2243 (1981)
- (3) van Koningsveld, H., van Bekkum, H. and Jansen, J.C. *Acta Crystallogr.*, **B43**, 127-132 (1987)

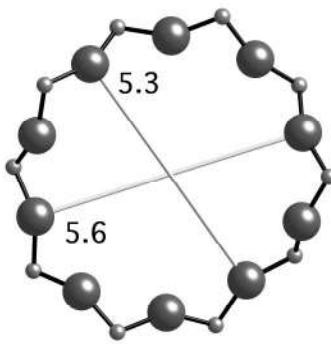
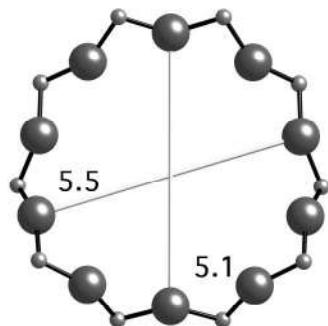
**Crystal chemical data:**  $[\text{Na}^+ \text{n} (\text{H}_2\text{O})_{16}] [\text{Al}_n \text{Si}_{96-n} \text{O}_{192}]$ -MFI,  $n < 27$   
orthorhombic, Pnma,  $a = 20.07\text{\AA}$ ,  $b = 19.92\text{\AA}$ ,  $c = 13.42\text{\AA}$  <sup>(2)</sup>

**Framework density:** 17.9 T/1000 $\text{\AA}^3$

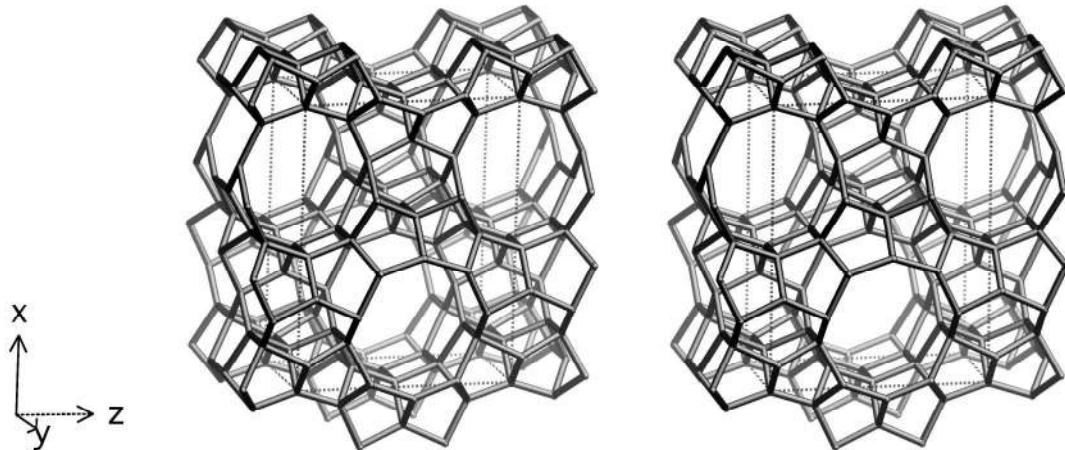
**Channels:** {[100] **10** 5.1 x 5.5 ↔ [010] **10** 5.3 x 5.6}\*\*\*



10-ring viewed along [100]



10-ring viewed along [010]



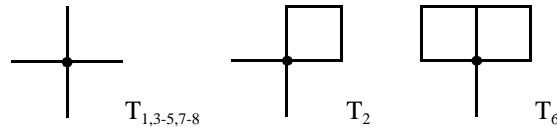
framework viewed along [010]

**Idealized cell constants:** orthorhombic, Imm2,  $a = 7.5\text{\AA}$ ,  $b = 14.4\text{\AA}$ ,  $c = 19.0\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (8, 1) 4 12 23 39 62 93 127 165 210 260	5·5·5·5·8 <sub>2</sub>
	T <sub>2</sub> (8, 1) 4 11 20 39 66 92 124 163 215 257	4·5 <sub>2</sub> ·5·5·8·10
	T <sub>3</sub> (4, m) 4 12 24 42 64 90 131 168 206 259	5·5·5·5·10 <sub>2</sub>
	T <sub>4</sub> (4, m) 4 12 26 40 60 94 136 168 200 259	5 <sub>2</sub> ·5 <sub>2</sub> ·6·8·6·8
	T <sub>5</sub> (4, m) 4 12 19 35 64 96 123 155 207 272	5·5 <sub>2</sub> ·5·5 <sub>2</sub> 6·*
	T <sub>6</sub> (4, m) 4 10 22 36 64 98 124 158 213 260	4·5·4·5·8 <sub>2</sub> ·10
	T <sub>7</sub> (2, mm2) 4 12 22 40 62 92 138 160 196 262	5·5·5·5·6 <sub>2</sub> ·*
	T <sub>8</sub> (2, mm2) 4 12 24 38 66 100 118 162 220 262	5·5·5·5 <sub>2</sub> ·10 <sub>2</sub>

**Secondary building units:** 5-1

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*ZSM-57<sup>(1)</sup>

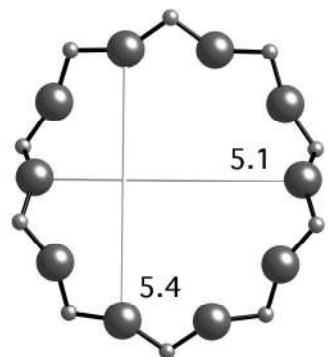
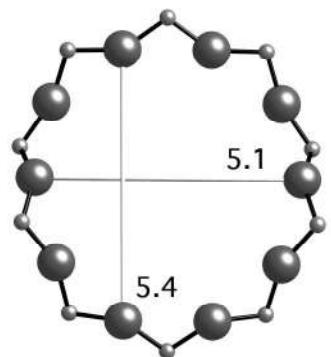
## References:

- (1) Schlenker, J.L., Higgins, J.B. and Valyocsik, E.W. *Zeolites*, **10**, 293-296 (1990)

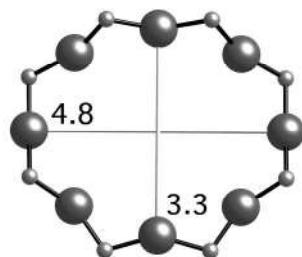
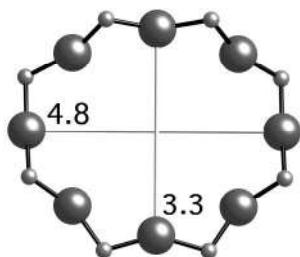
**Crystal chemical data:**  $[\text{H}^{+}_{1.5}] [\text{Al}_{1.5}\text{Si}_{34.5}\text{O}_{72}]\text{-MFS}$   
orthorhombic, Imm2,  $a = 7.451\text{\AA}$ ,  $b = 14.171\text{\AA}$ ,  $c = 18.767\text{\AA}$ <sup>(1)</sup>

**Framework density:** 18.2 T/1000 $\text{\AA}^3$

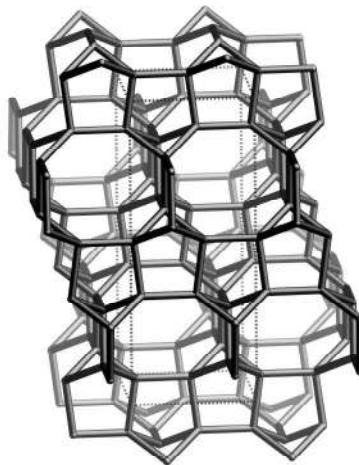
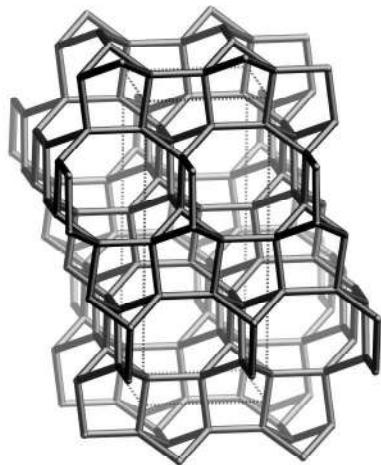
**Channels:** [100] **10** 5.1 x 5.4\*  $\leftrightarrow$  [010] **8** 3.3 x 4.8\*



10-ring viewed along [100]



8-ring viewed along [010]



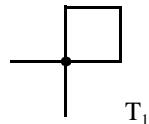
*framework viewed along [100]*

**Idealized cell constants:** tetragonal, I4<sub>1</sub>/amd (origin choice 2),  $a = 7.1\text{\AA}$ ,  $c = 17.8\text{\AA}$

**Coordination sequences and vertex symbols:** T<sub>1</sub> (16, m) 4 11 23 44 67 95 134 168 215 271 4·5<sub>2</sub>·5·8<sub>2</sub>·5·8<sub>2</sub>

**Secondary building units:** 4

**Loop configuration of T-Atoms:**



**Isotypic framework structures:**

\*Montesommaite<sup>(1)</sup>

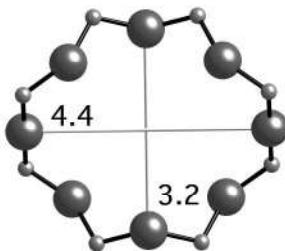
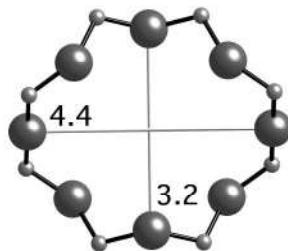
## References:

- (1) Rouse, R.C., Dunn, P.J., Grice, J.D., Schlenker, J.L. and Higgins, J.B. *Am. Mineral.*, **75**, 1415-1420 (1990)

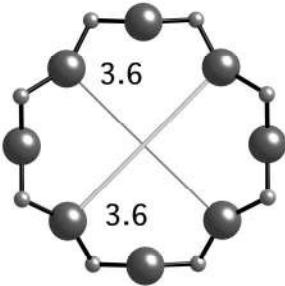
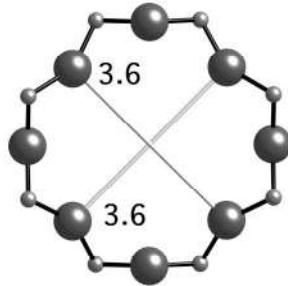
**Crystal chemical data:**  $[(\text{K}^+, \text{Na}^+)_{4.5} (\text{H}_2\text{O})_5] [\text{Al}_{4.5} \text{Si}_{11.5} \text{O}_{32}]$ -MON  
tetragonal, I4<sub>1</sub>/amd,  $a = 7.141\text{\AA}$ ,  $c = 17.307\text{\AA}$ <sup>(1)</sup>

**Framework density:** 18.1 T/1000 $\text{\AA}^3$

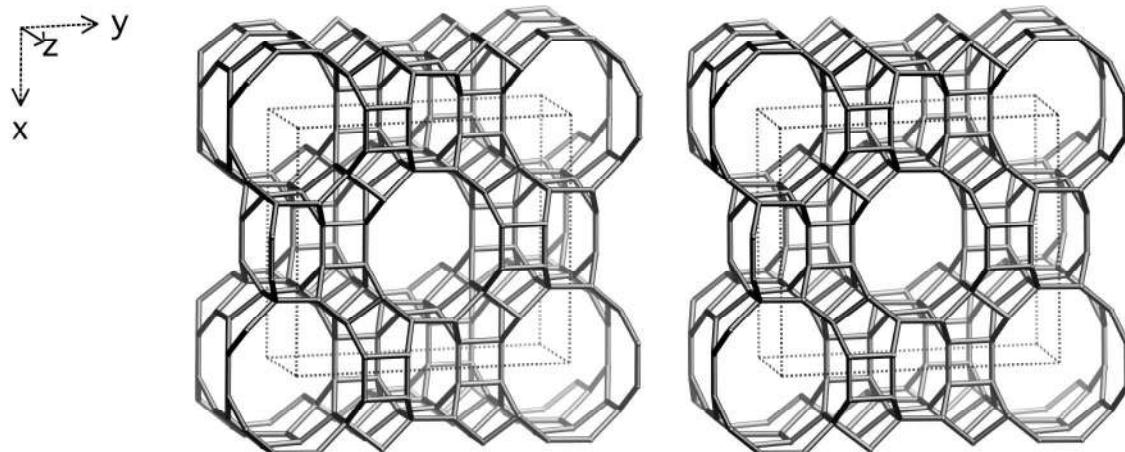
**Channels:** [100] 8 3.2 x 4.4\*  $\leftrightarrow$  [001] 8 3.6 x 3.6\*



8-ring viewed along [100]



8-ring viewed along [001]

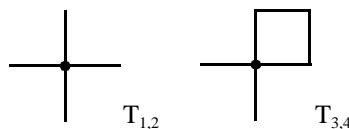
**MOR****Framework Type****Cmcm***framework viewed along [001]*

**Idealized cell constants:** orthorhombic, Cmcm,  $a = 18.3\text{\AA}$ ,  $b = 20.5\text{\AA}$ ,  $c = 7.5\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (16, 1)    4    12    22    38    60    88    115    155    204    242	5·5·5·5 <sub>2</sub> ·8·12
	T <sub>2</sub> (16, 1)    4    12    20    37    64    87    114    154    198    241	5·5·5·5 <sub>2</sub> ·5·8
	T <sub>3</sub> (8, m)    4    11    24    39    54    86    126    156    195    242	4·5 <sub>2</sub> ·5·8 <sub>2</sub> ·5·8 <sub>2</sub>
	T <sub>4</sub> (8, m)    4    11    24    39    60    92    122    148    195    250	4·5 <sub>2</sub> ·5·8·5·8

**Secondary building units:** 5-1

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Mordenite<sup>(1)</sup>  
[Ga-Si-O]-MOR<sup>(2)</sup>  
Ca-Q<sup>(3)</sup>  
LZ-211<sup>(4)</sup>  
Large port mordenite<sup>(5)</sup>  
Maricopaite (interrupted framework)<sup>(6)</sup>  
Na-D<sup>(7)</sup>

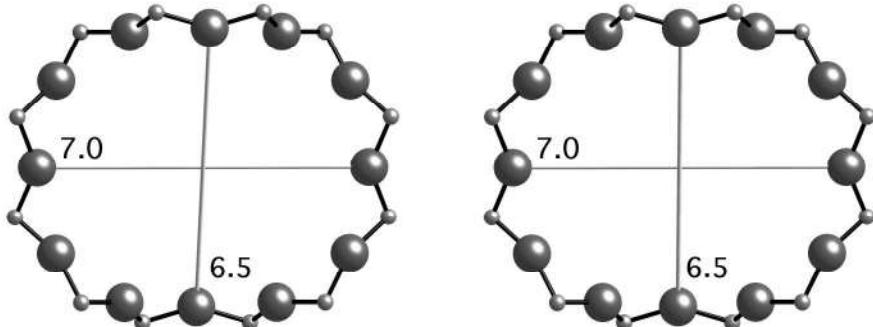
**Alternate designation:** Ptilolite (discredited)  
Arduinitite (discredited)  
Flokitite (discredited)

**References:**

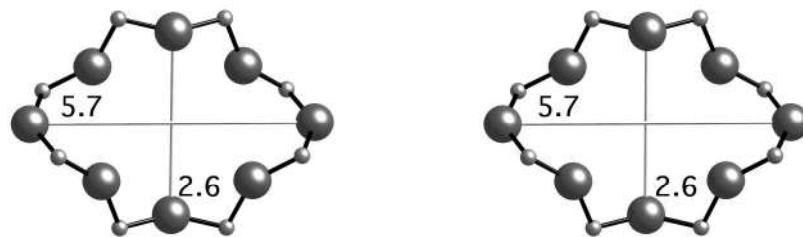
- (1) Meier, W.M. *Z. Kristallogr.*, **115**, 439-450 (1961)
- (2) Eapen, M.J., Reddy, K.S.N., Joshi, P.N. and Shiralkar, V.P. *J. Incl. Phenom.*, **14**, 119-129 (1992)

**Crystal chemical data:**  $[\text{Na}^+ \cdot (\text{H}_2\text{O})_{24}] \cdot [\text{Al}_8\text{Si}_{40}\text{O}_{96}]$ -MOR  
orthorhombic, Cmcm,  $a = 18.1\text{\AA}$ ,  $b = 20.5\text{\AA}$ ,  $c = 7.5\text{\AA}$ <sup>(1)</sup>

**Framework density:** 17.2 T/1000 $\text{\AA}^3$

**Channels:** [001] **12** 6.5 x 7.0\*  $\leftrightarrow$  {[010] **8** 3.4 x 4.8  $\leftrightarrow$  [001] **8** 2.6 x 5.7}\*  


12-ring viewed along [001]

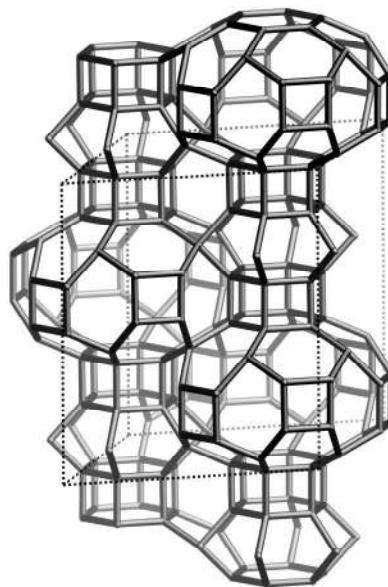
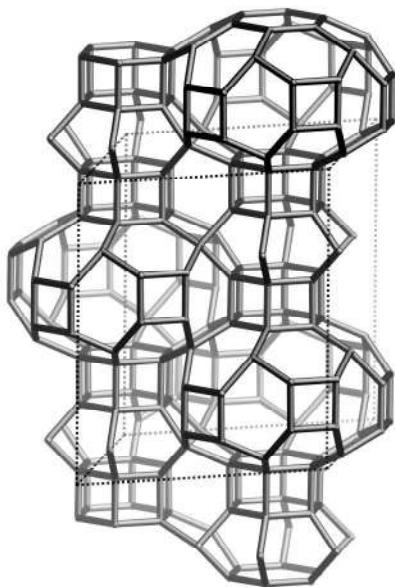
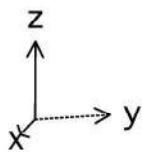
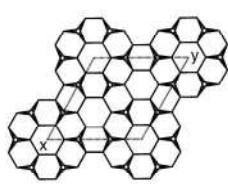


limiting 8 ring along [001]  
between 12-ring channels

See Appendix A for 8-ring viewed along [010]

#### References (cont.):

- (3) Koizumi, M. and Roy, R. *J. Geol.*, **68**, 41-53 (1960)
- (4) Breck, D.W. and Skeels, G.W. *U.S. Patent 4,503,023* (1985)
- (5) Sand, L.B. In *Molecular Sieves*, (ed. R.M. Barrer), pp. 71-77 (1968), Soc. Chem. Indus., London
- (6) Rouse, R.C. and Peacor, D.R. *Am. Mineral.*, **79**, 175-184 (1994)
- (7) Barrer, R.M. and White, E.A.D. *J. Chem. Soc.*, 1561-1571 (1952)



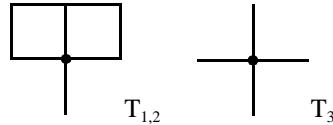
*framework viewed normal to [001] (top left: projection down [001])*

**Idealized cell constants:** trigonal,  $R\bar{3}m$ ,  $a = 17.2\text{\AA}$ ,  $c = 19.8\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(36, 1)$	4 10 21 37 55 75 101 136 175 211	4·6·4·6 <sub>2</sub> ·6·6
	$T_2(36, 1)$	4 10 20 34 53 77 106 138 170 206	4·6 <sub>2</sub> ·4·6 <sub>2</sub> ·6·6
	$T_3(18, 2)$	4 12 21 32 51 80 110 132 164 212	6·6·6 <sub>2</sub> ·6 <sub>2</sub> ·6 <sub>2</sub>

**Secondary building units:** 2-6-2

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*MCM-61<sup>(1,2)</sup>

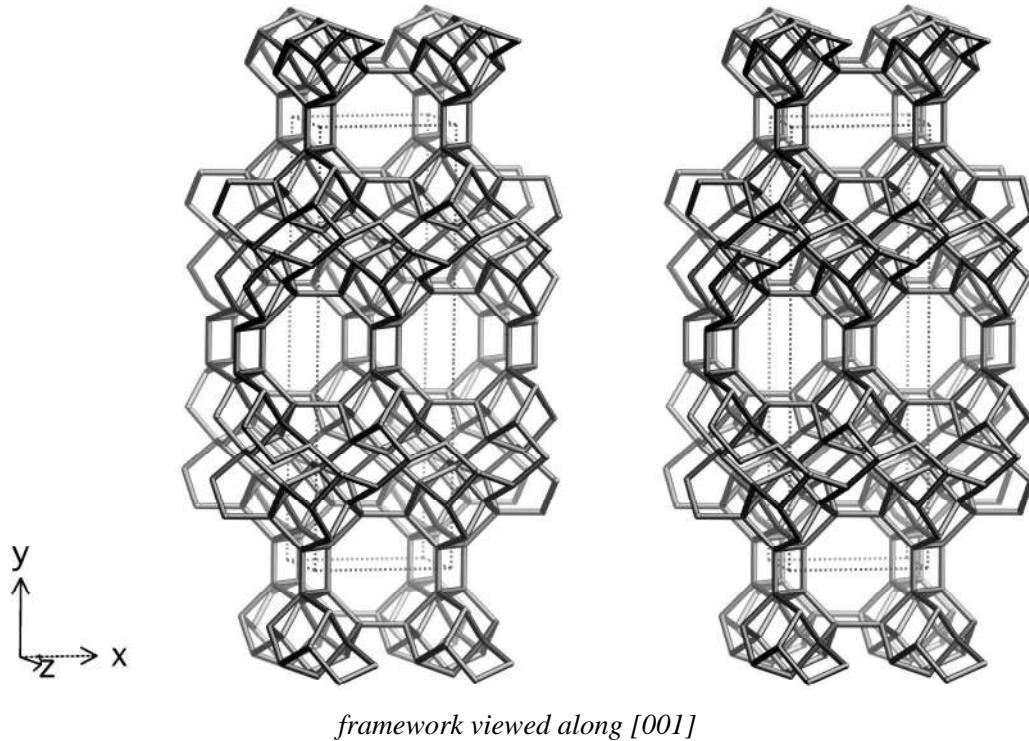
### References:

- (1) Valyosik, E.W. U.S. Patent 5,670,131 (1997)
- (2) Shantz, D.F., Burton, A. and Lobo, R.F. *Mesoporous and Mesoporous Materials*, **31**, 61-73 (1999)

**Crystal chemical data:**  $[K^{+}_{2.1} C_{12}H_{24}O_6] [Si_{27.9}Al_{2.1} O_{60}]$ -MSO  
 $C_{12}H_{24}O_6$  = 18-crown-6  
rhombohedral,  $R\bar{3}m$ ,  $a = 11.841\text{\AA}$ ,  $\alpha = 93.29^\circ$ <sup>(2)</sup>  
(hexagonal setting:  $a = 17.220\text{\AA}$ ,  $c = 19.296\text{\AA}$ )

**Framework density:**  $18.2 \text{ T}/1000\text{\AA}^3$

**Channels:** apertures formed by 6-rings only

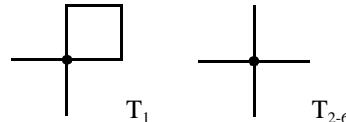


**Idealized cell constants:** monoclinic, C12/m1,  $a = 9.6\text{\AA}$ ,  $b = 30.4\text{\AA}$ ,  $c = 7.2\text{\AA}$ ,  $\beta = 90.5^\circ$

<b>Coordination sequences and vertex symbols:</b>	$T_1(8, 1)$	4 11 24 45 77 109 137 174 224 280	4·5·5·6·5·8
	$T_2(8, 1)$	4 12 24 42 70 95 136 184 227 277	5·5 <sub>2</sub> ·5·6·5·8
	$T_3(8, 1)$	4 12 27 47 69 99 142 184 227 281	5·6·5·7·5·8
	$T_4(8, 1)$	4 12 26 45 64 96 134 186 230 290	5·6·5·6·5·6
	$T_5(8, 1)$	4 12 24 42 64 93 133 179 234 290	5·6·5·6 <sub>2</sub> ·5·7
	$T_6(4, 2)$	4 12 21 44 74 106 138 172 226 284	5·5·5·5·5 <sub>2</sub> ·6

**Secondary building units:** 5-5=1

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*MCM-35<sup>(1)</sup>  
UTM-1<sup>(2)</sup>

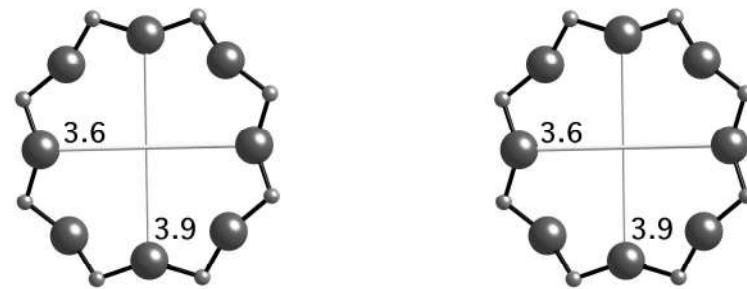
### References:

- (1) Barrett, P.A., Díaz-Cabañas, M.J. and Camblor, M.A. *Chem. Mater.*, **11**, 2919-2927 (1999)
- (2) Plévert, J., Yamamoto, K. Chiari, G. and Tatsumi, T. *J. Phys. Chem. B*, **103**, 8647-8649 (1999)

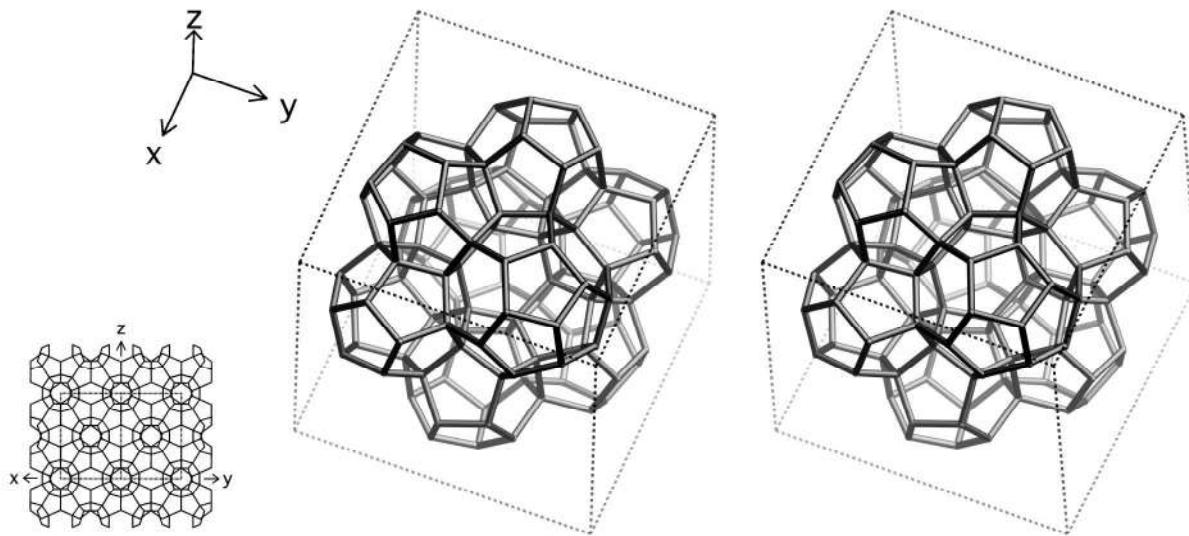
**Crystal chemical data:**  $[\text{Si}_{44} \text{O}_{88}]$ -MTF  
monoclinic, C12/m1  
 $a = 9.500 \text{ \AA}$ ,  $b = 30.710 \text{ \AA}$ ,  $c = 7.313 \text{ \AA}$ ,  $\beta = 91.71^\circ$  <sup>(1)</sup>

**Framework density:** 20.6 T/1000 $\text{\AA}^3$

**Channels:** [001] 8 3.6 x 3.9\*



8-ring viewed along [001]



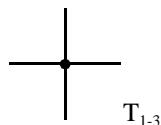
framework viewed along [111] (bottom left: projection down [110])

**Idealized cell constants:** cubic, Fd $\bar{3}m$  (origin choice 2),  $a = 19.9\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (96, m) 4 12 25 43 68 95 133 177 223 274	5·5·5·5·5·6
	T <sub>2</sub> (32, 3m) 4 12 24 39 66 103 130 168 216 274	5·5·5·5·5·5
	T <sub>3</sub> (8, 43m) 4 12 24 36 64 112 132 156 222 264	5·5·5·5·5·5

**Secondary building units:** combinations only

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*ZSM-39<sup>(1)</sup>, CF-4<sup>(2)</sup>, Dodecasil-3C<sup>(3)</sup>, Holdstite<sup>(4)</sup>

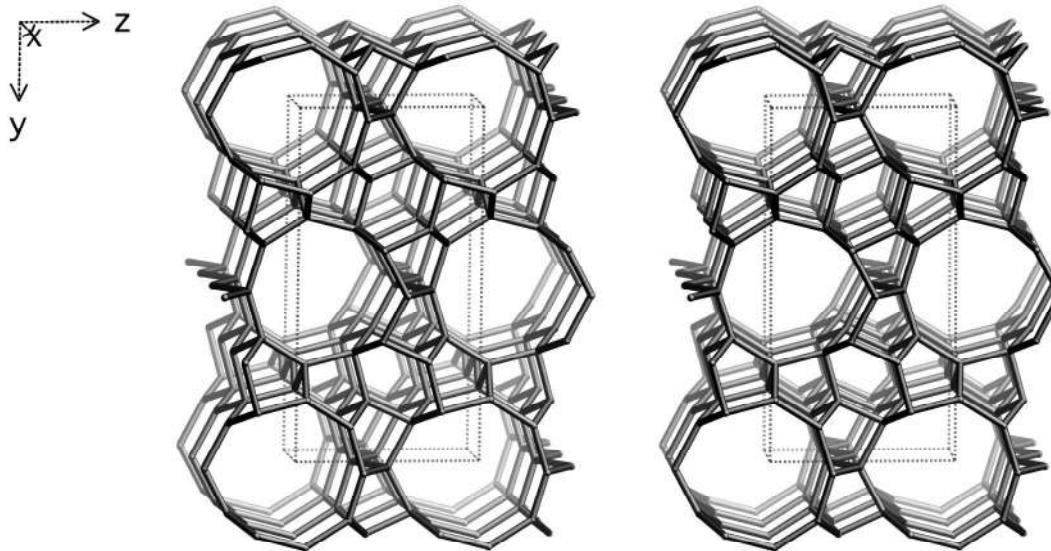
### References:

- (1) Schlenker, J.L., Dwyer, F.G., Jenkins, E.E., Rohrbaugh, W.J., Kokotailo G.T. and Meier, W.M. *Nature*, **294**, 340-342 (1981)
- (2) Long, Y., He, H., Zheng, P., Guang, W. and Wang, B. *J. Incl. Phenom.*, **5**, 355-362 (1987)
- (3) Gies, H. Z. *Kristallogr.*, **167**, 73-82 (1984)
- (4) Smith, J.V. and Blackwell, C.S. *Nature*, **303**, 223-225 (1983)

**Crystal chemical data:**  $[(C_8H_{20}N^+)_n(OH)^-_n][Si_{136}O_{272}]$ -MTN  
 $C_8H_{20}N^+$  = tetraethylammonium  
cubic,  $Fd\bar{3}m$ ,  $a = 19.36\text{\AA}$ <sup>(1)</sup>

**Framework density:** 18.7 T/1000 $\text{\AA}^3$

**Channels:** apertures formed by 6-rings only



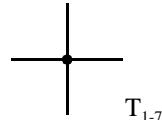
framework viewed along [100]

**Idealized cell constants:** orthorhombic, Pmmn (origin choice 1),  $a = 5.3\text{\AA}$ ,  $b = 22.0\text{\AA}$ ,  $c = 11.4\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(4, m)$	4 12 23 43 66 92 130 170 213 261	$5 \cdot 5 \cdot 5 \cdot 6 \cdot 10_2$
	$T_2(4, m)$	4 12 24 40 64 96 136 167 207 258	$5_2 \cdot 6_2 \cdot 6 \cdot 6 \cdot 6_2$
	$T_3(4, m)$	4 12 22 40 67 97 124 165 219 265	$5 \cdot 5 \cdot 5 \cdot 6_2 \cdot 10_2$
	$T_4(4, m)$	4 12 22 40 65 98 132 159 206 278	$5 \cdot 5 \cdot 5 \cdot 6_2 \cdot *$
	$T_5(4, m)$	4 12 24 41 62 97 129 170 212 262	$5_2 \cdot 6_2 \cdot 6 \cdot 6_2 \cdot 6 \cdot 6_2$
	$T_6(2, mm2)$	4 12 22 42 66 94 126 164 220 270	$5 \cdot 5 \cdot 5 \cdot 6_2 \cdot 10_2$
	$T_7(2, mm2)$	4 12 24 44 66 88 132 174 214 258	$5 \cdot 5 \cdot 5 \cdot 10_2 \cdot *$

**Secondary building units:** 5-1

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*ZSM-23<sup>(1-3)</sup>  
EU-13<sup>(4)</sup>  
ISI-4<sup>(5)</sup>  
KZ-1<sup>(6)</sup>

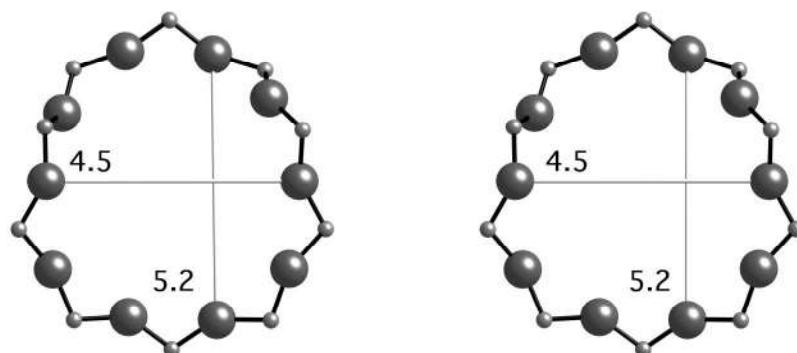
### References:

- (1) Schlenker, J.L., Higgins, J.B. and Cox, D.E. *private communication*
- (2) Rohrman Jr., A.C., LaPierre, R.B., Schlenker, J.L., Wood, J.D., Valyocsik, E.W., Rubin, M.K., Higgins, J.B. and Rohrbaugh, W.J. *Zeolites*, **5**, 352-354 (1985)

**Crystal chemical data:**  $[\text{Na}^+ \text{n} (\text{H}_2\text{O})_4] [\text{Al}_n \text{Si}_{24-n} \text{O}_{48}]$ -MTT,  $n < 2$   
orthorhombic,  $\text{Pmn}2_1$ ,  $a = 21.5\text{\AA}$ ,  $b = 11.1\text{\AA}$ ,  $c = 5.0 \text{\AA}$ <sup>(1)</sup>

**Framework density:** 20.1 T/1000 $\text{\AA}^3$

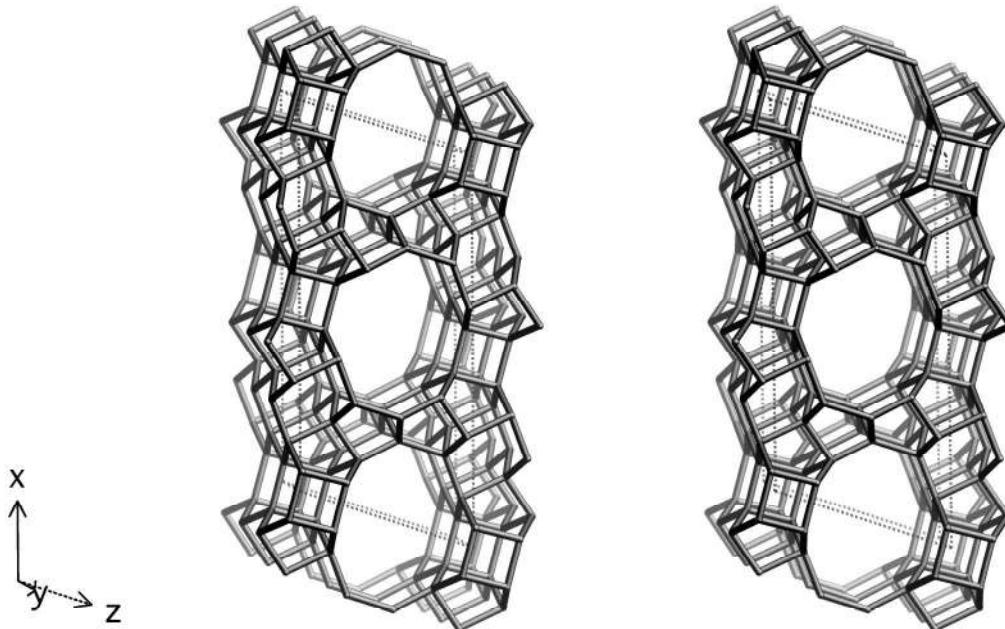
**Channels:** [001] **10** 4.5 x 5.2\*



10-ring viewed along [001]

#### References (cont.):

- (3) Marler, B., Deroche, C., Gies, H., Fyfe, C.A., Grondy, H., Kokotailo, G.T., Feng, Y., Ernst, S., Weitkamp, J. and Cox, D.E. *J. Appl. Crystallogr.*, **26**, 636-644 (1993)
- (4) Araya, A. and Lowe, B.M. *U.S. Patent 4,581,211* (1986)
- (5) Kakatsu, K. and Kawata, N. *Eur. Pat. Appl. EPA 102,497* (1984)
- (6) Parker, L.M. and Bibby, D.M. *Zeolites*, **3**, 8-11 (1983)



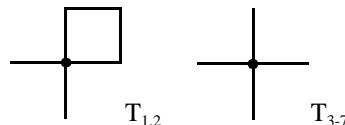
framework viewed along [010]

**Idealized cell constants:** monoclinic, C2/m,  $a = 25.6\text{\AA}$ ,  $b = 5.3\text{\AA}$ ,  $c = 12.1\text{\AA}$ ,  $\beta = 109.3^\circ$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (4, m)    4 11 22 38 60 88 113 147 190 243 T <sub>2</sub> (4, m)    4 11 22 38 60 86 115 147 191 238 T <sub>3</sub> (4, m)    4 12 21 37 62 84 119 147 188 239 T <sub>4</sub> (4, m)    4 12 23 37 59 85 120 154 184 231 T <sub>5</sub> (4, m)    4 12 21 37 58 87 119 154 182 227 T <sub>6</sub> (4, m)    4 12 24 39 55 85 122 156 188 225 T <sub>7</sub> (4, m)    4 12 23 38 59 83 115 155 192 233	4·6 <sub>2</sub> ·5·6·5·6 4·6 <sub>2</sub> ·5·6·5·6 5·6·5·6·5 <sub>2</sub> ·6 5·6·5·6·6·6 <sub>2</sub> 5·6 <sub>2</sub> ·5·6 <sub>2</sub> ·5 <sub>2</sub> ·6 5·6 <sub>2</sub> ·5·6 <sub>2</sub> ·6 <sub>2</sub> ·12 <sub>6</sub> 5·5·5·5·6·12 <sub>6</sub>
---	---	--

**Secondary building units:** 5-2

**Loop configuration of  
T-Atoms:**



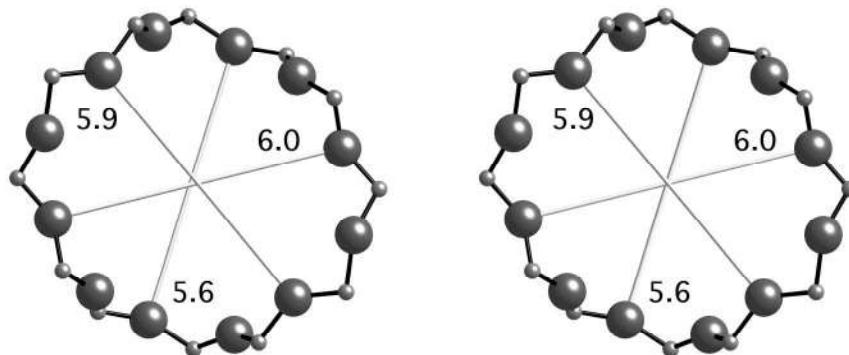
**Isotypic framework  
structures:**

*ZSM-12 <sup>(1,2)</sup>	TPZ-12 <sup>(6)</sup>
[Ga-Si-O]-MTW <sup>(3)</sup>	Theta-3 <sup>(7)</sup>
CZH-5 <sup>(4)</sup>	VS-12 <sup>(8)</sup>
NU-13 <sup>(5)</sup>	

**Crystal chemical data:**  $[\text{Na}^+ \text{n} (\text{H}_2\text{O})_8] [\text{Al}_n \text{Si}_{56-n} \text{O}_{112}]$ -MTW,  $n < 5$   
 monoclinic, C12/c1  
 $a = 24.863\text{\AA}$ ,  $b = 5.012\text{\AA}$ ,  $c = 24.328\text{\AA}$ ,  $\beta = 107.72^\circ$  <sup>(2)</sup>

**Framework density:** 19.4 T/1000 $\text{\AA}^3$

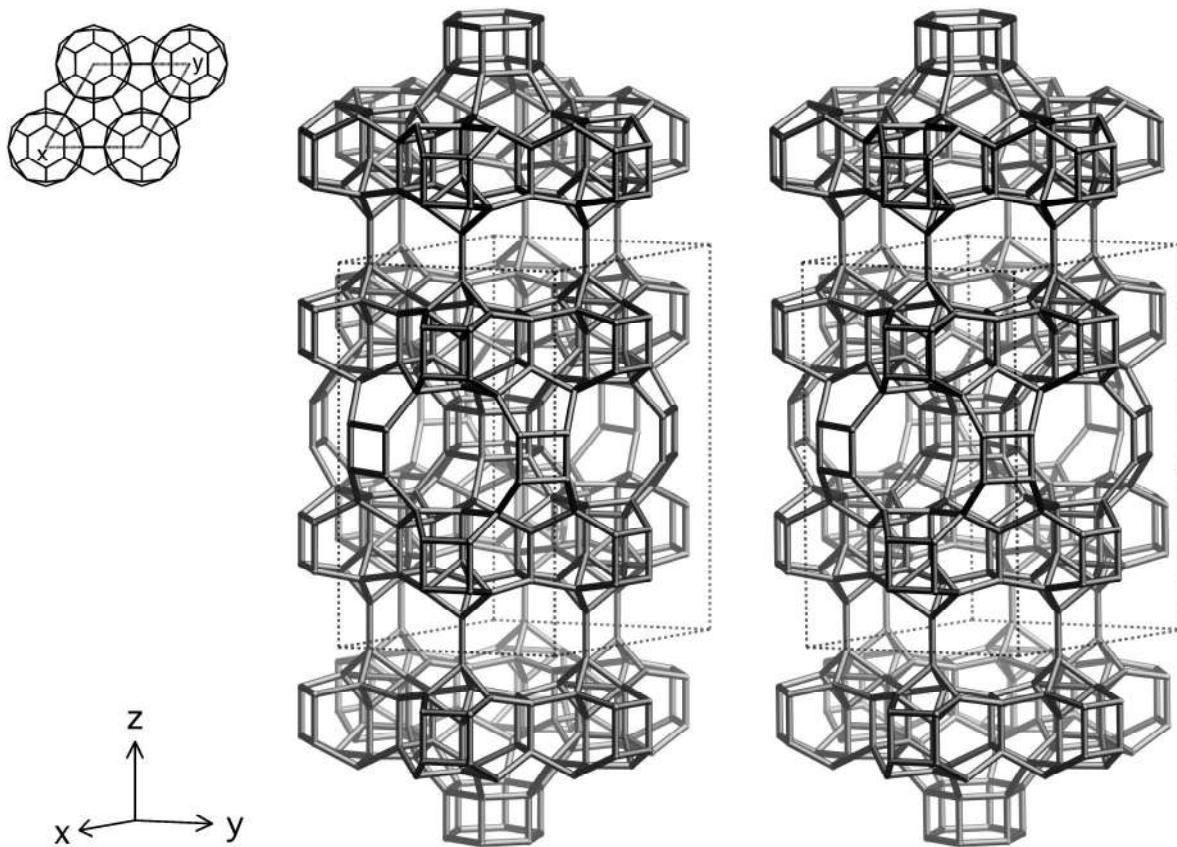
**Channels:** [010] 12 5.6 x 6.0\*



12-ring viewed along [010]

### References:

- (1) LaPierre, R.B., Rohrman Jr., A.C., Schlenker, J.L., Wood, J.D., Rubin, M.K. and Rohrbaugh, W.J. *Zeolites*, **5**, 346-348 (1985)
- (2) Fyfe, C.A., Gies, H., Kokotailo, G.T., Marler, B. and Cox, D.E. *J. Phys. Chem.*, **94**, 3718-3721 (1990)
- (3) Zhi, Y.X., Tuel, A., Bentaarit, Y. and Naccache, C. *Zeolites*, **12**, 138-141 (1992)
- (4) Hickson, D.A. *UK Pat. Appl. GB 2079735A* (1981)
- (5) Whittam, T.V. *Eur. Pat. Appl. EPA 0059059* (1982)
- (6) Sumitani, K., Sakai, T., Yamasaki, Y. and Onodera, T. *U.S. Patent 4,557,919* (1985)
- (7) Barlow, T.M. *E. Patent A-162,719* (1985)
- (8) Reddy, K.M., Moudrakovski, I. and Sayari, A. *Chem. Commun.*, 1491-1492 (1994)



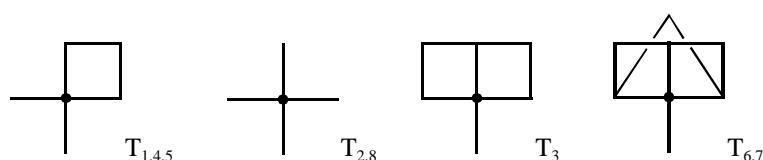
*framework viewed normal to [001] (top left: projection down [001])*

**Idealized cell constants:** hexagonal, P6/mmm,  $a = 14.4\text{\AA}$ ,  $c = 25.2\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (12, .m.) 4 10 20 32 52 76 111 146 185 225	4 <sub>2</sub> ·6·5·5·5·5
	T <sub>2</sub> (12, .m.) 4 12 22 35 51 81 109 137 175 218	5·5·5·5·6·10 <sub>2</sub>
	T <sub>3</sub> (12, .m.) 4 10 21 40 62 78 95 128 177 228	4·5·4·5·6·10 <sub>2</sub>
	T <sub>4</sub> (12, ..m) 4 11 18 32 52 78 107 147 187 215	4·10 <sub>2</sub> ·5 <sub>2</sub> ·6 <sub>2</sub> ·5 <sub>2</sub> ·6 <sub>2</sub>
	T <sub>5</sub> (12, ..m) 4 11 22 32 53 79 113 144 176 220	4·5·5·6 <sub>2</sub> ·5·6 <sub>2</sub>
	T <sub>6</sub> (4, 3m) 4 10 20 34 54 87 114 139 188 244	4·10 <sub>4</sub> ·4·10 <sub>4</sub> ·4·10 <sub>4</sub>
	T <sub>7</sub> (4, 3m) 4 10 16 30 49 77 100 138 181 214	4·6 <sub>2</sub> ·4·6 <sub>2</sub> ·4·6 <sub>2</sub>
	T <sub>8</sub> (4, 3m) 4 12 22 31 52 74 112 142 166 204	5·6 <sub>2</sub> ·5·6 <sub>2</sub> ·5·6 <sub>2</sub>

**Secondary building units:** combinations only

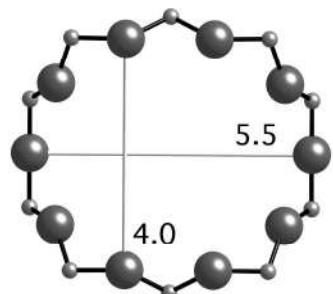
**Loop configuration of  
T-Atoms:**



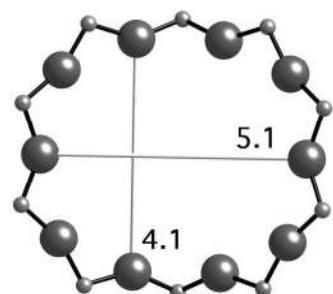
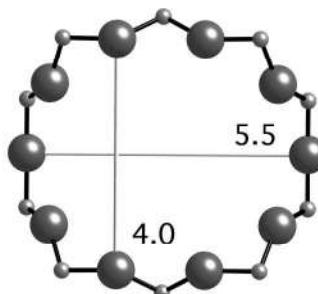
**Crystal chemical data:**  $[\text{H}^{+}_{2.4}\text{Na}^{+}_{3.1}] [\text{Al}_{0.4}\text{B}_{5.1}\text{Si}_{66.5}\text{O}_{144}]\text{-MWW}$   
hexagonal, P6/mmm,  $a = 14.208\text{\AA}$ ,  $c = 24.945\text{\AA}$ <sup>(4)</sup>

**Framework density:** 16.5 T/1000 $\text{\AA}^3$

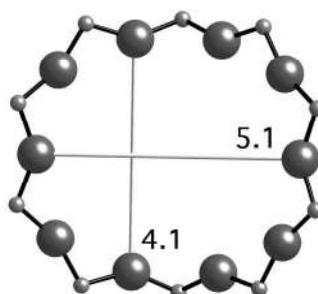
**Channels:**  $\perp [001]$  **10** 4.0 x 5.5\*\* |  $\perp [001]$  **10** 4.1 x 5.1\*\*



10-ring viewed normal to [001]  
between layers



10-ring viewed normal to [001]  
within layers



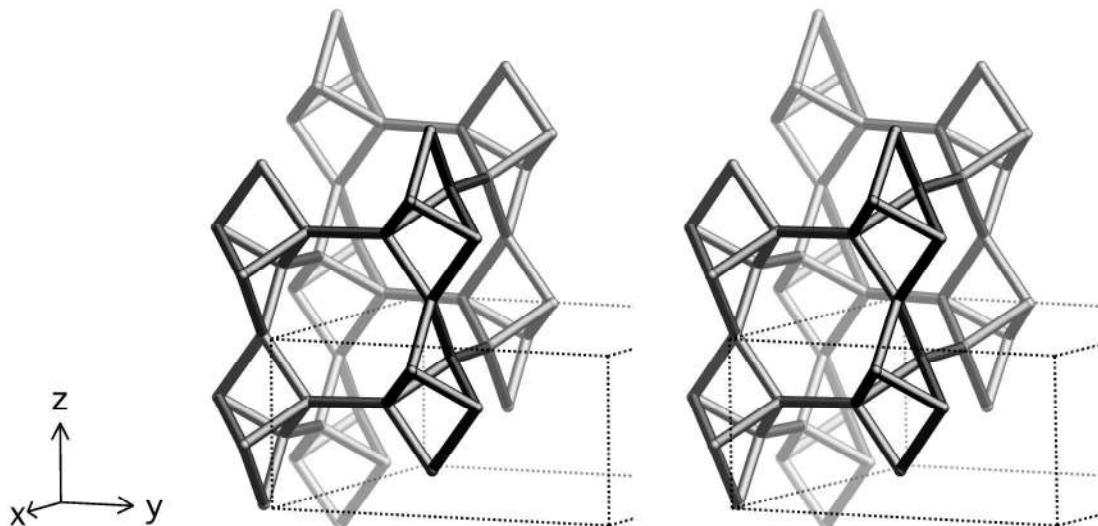
**Isotypic framework structures:**

\*MCM-22<sup>(1)</sup>  
ERB-1<sup>(2)</sup>  
ITQ-1<sup>(3,4)</sup>

PSH-3<sup>(5)</sup>  
SSZ-25<sup>(6)</sup>

#### References:

- (1) Leonowicz, M.E., Lawton, J.A., Lawton, S.L. and Rubin, M.K. *Science*, **264**, 1910-1913 (1994)
- (2) Belussi, G., Perego, G., Clerici, M.G. and Giusti, A. *Eur. Pat. Appl. EPA 293032* (1988)
- (3) Cambor, M.A., Corell, C., Corma, A., Díaz-Cabañas, M.J., Nicolopoulos, S., González-Calbet, J.M. and Vallet-Regí, M. *Chem. Mater.*, **8**, 2415-2417 (1996)
- (4) Cambor, M.A., Corma, A., Díaz-Cabañas, M.J. and Baerlocher, Ch. *J. Phys. Chem. B*, **102**, 44-51 (1998)
- (5) Puppe, L. and Weisser, J. *U.S. Patent 4,439,409* (1984)
- (6) Zones, S.I. *E. Patent 231,860* (1987)



framework viewed normal to [001]

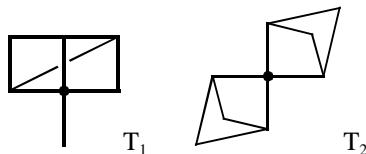
**Idealized cell constants:** tetragonal, I4<sub>1</sub>/amd (origin choice 2),  $a = 13.9\text{\AA}$ ,  $c = 6.4\text{\AA}$

**Coordination sequences and vertex symbols:**

T <sub>1</sub> (16, m)	4    9    19    35    52    78    106    139    179    213	4·8 <sub>2</sub> ·4·8 <sub>2</sub> ·4 <sub>2</sub> ·8 <sub>4</sub>
T <sub>2</sub> (4, 4m2)	4    8    18    36    56    66    116    140    154    232	4 <sub>2</sub> ·4 <sub>2</sub> ·8 <sub>4</sub> ·8 <sub>4</sub> ·8 <sub>4</sub>

**Secondary building units:** 4=1

**Loop configuration of T-Atoms:**



**Isotypic framework structures:**

- \*Natrolite<sup>(1,2)</sup>
- [Al-Ge-O]-NAT<sup>(3)</sup>
- [Ga-Si-O]-NAT<sup>(4)</sup>
- |Rb-|[Ga-Ge-O]-NAT<sup>(5)</sup>
- Gonnardite<sup>(6)</sup>
- High natrolite<sup>(7)</sup>
- Mesolite<sup>(8)</sup>

- Metanatrolite<sup>(9)</sup>
- Scolecite<sup>(10-12)</sup>
- Synthetic gonnardite<sup>(13)</sup>
- Synthetic mesolite<sup>(14)</sup>
- Synthetic natrolite<sup>(14)</sup>
- Synthetic scolecite<sup>(14)</sup>

**Alternate designation:** Laubanite (discredited)

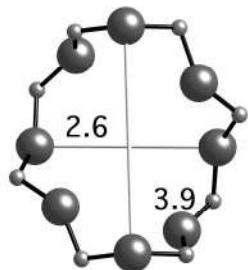
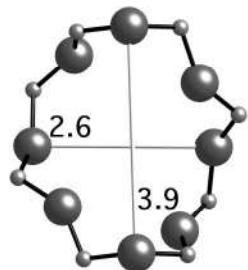
#### References:

- (1) Pauling, L. *Proc. Natl. Acad. Sci.*, **16**, 453-459 (1930)
- (2) Meier, W.M. *Z. Kristallogr.*, **113**, 430-444 (1960)
- (3) Tripathi, A., Johnson, G.M., Kim, S.J. and Parise, J.B. *J. Mater. Chem.*, **10**, 451-455 (2000)
- (4) Xie, D., Newsam, J.M., Yang, J. and Yelong, W.B. In *MRS Sym. Proc.*, (eds. M.M.J. Treacy, J.M. White and J.M. Thomas), **111**, pp. 147-154 (1988), Materials Research Society, Pittsburgh, PA

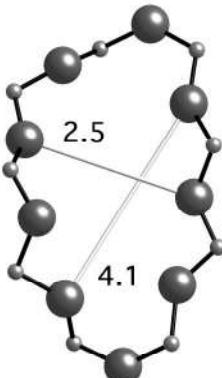
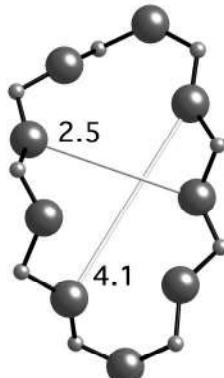
**Crystal chemical data:**  $[\text{Na}^+_{16} (\text{H}_2\text{O})_{16}] [\text{Al}_{16}\text{Si}_{24} \text{O}_{80}]$ -NAT  
 orthorhombic, Fdd2,  $a = 18.30\text{\AA}$ ,  $b = 18.63\text{\AA}$ ,  $c = 6.60\text{\AA}$ <sup>(2)</sup>  
 (Relationship to unit cell of Framework Type:  $\mathbf{a}' = \mathbf{b}' = \mathbf{a} \cdot \sqrt{2}$ ,  $\mathbf{c}' = \mathbf{c}$   
 or, as vectors,  $\mathbf{a}' = \mathbf{a} + \mathbf{b}$ ,  $\mathbf{b}' = \mathbf{b} - \mathbf{a}$ ,  $\mathbf{c}' = \mathbf{c}$ )

**Framework density:** 17.8 T/1000 $\text{\AA}^3$

**Channels:**  $<100>$  8 2.6 x 3.9\*\*  $\leftrightarrow$  [001] 9 2.5 x 4.1\* (variable due to considerable flexibility of framework)



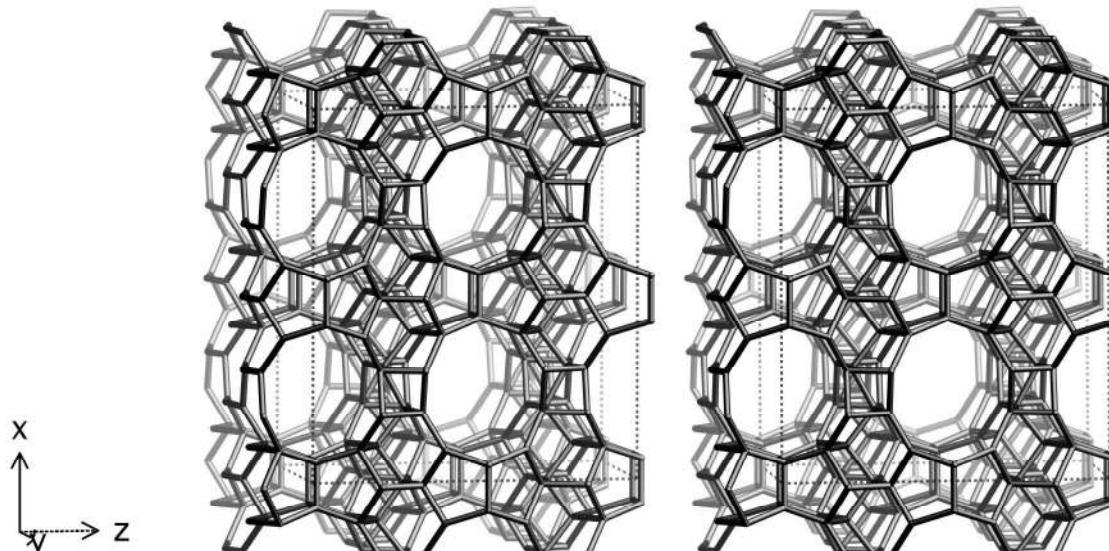
8-ring viewed along  $<100>$



9-ring viewed along [001] (variable)

#### References (cont.):

- (5) Klaska, K.H. and Jarchow, O. Z. *Kristallogr.*, **172**, 167-174 (1985)
- (6) Mazzi, F., Larsen, A.O., Gottardi, G. and Galli, E. N. *Jb. Miner. Mh.*, 219-228 (1986)
- (7) Baur, W.H. and Joswig, W. N. *Jb. Miner. Mh.*, 171-187 (1996)
- (8) Artioli, G., Smith, J.V. and Pluth, J.J. *Acta Crystallogr.*, **C42**, 937-942 (1986)
- (9) Joswig, W. and Baur, W.H. N. *Jb. Miner. Mh.*, 26-38 (1995)
- (10) Taylor, W.H. and Jackson, R. Z. *Kristallogr.*, **86**, 53-64 (1933)
- (11) Fälth, L. and Hansen, S. *Acta Crystallogr.*, **B35**, 1877-1880 (1979)
- (12) Smith, J.V., Pluth, J.J., Artioli, G. and Ross, F.K. In *Proc. 6th Int. Zeolite Conf.*, (eds. D.H. Olson and A. Bisio), pp. 842-850 (1984), Butterworths, Guildford, Surrey
- (13) Ghobarkar, H. and Schaef, O. *Zeolites*, **19**, 259-261 (1997)
- (14) Ghobarkar, H. and Schaef, O. *Cryst. Res. Technol.*, **31**, K67-69 (1996)



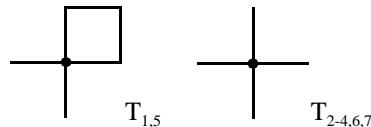
framework viewed along [010]

**Idealized cell constants:** orthorhombic, Fmmm,  $a = 26.1\text{\AA}$ ,  $b = 13.9\text{\AA}$ ,  $c = 22.9\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (32, 1) 4 11 21 36 58 89 123 157 187 237	4·6·5·5·5 <sub>2</sub> ·10
	T <sub>2</sub> (32, 1) 4 12 22 38 57 86 118 152 196 245	5·5·5·6 <sub>2</sub> ·5·10
	T <sub>3</sub> (16, m) 4 12 20 34 57 88 125 158 192 224	5·5 <sub>2</sub> ·5·5 <sub>2</sub> ·12 <sub>2</sub> ·*
	T <sub>4</sub> (16, m) 4 12 20 31 57 84 118 150 187 237	5·5·5·5·6 <sub>2</sub>
	T <sub>5</sub> (16, m) 4 11 24 40 63 86 114 158 208 255	4·10·5·5·5·5
	T <sub>6</sub> (16, m) 4 12 22 35 55 83 119 151 184 237	5·5 <sub>2</sub> ·5·6·5·6
	T <sub>7</sub> (8, mm2) 4 12 24 32 50 88 120 152 180 226	5·5·5·5·12 <sub>6</sub> ·*

**Secondary building units:** 5-3

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*NU-87<sup>(1)</sup>  
Gottardiite<sup>(2)</sup>

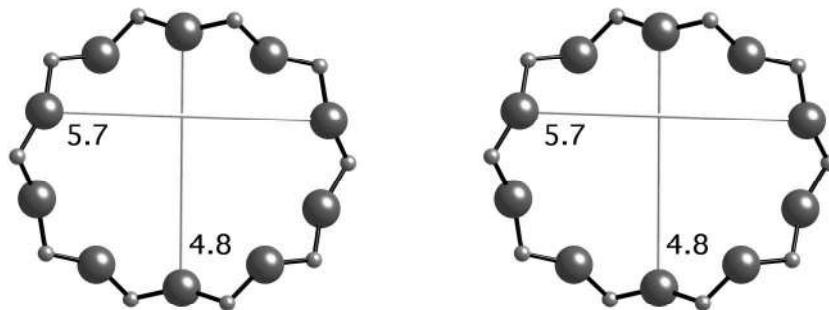
### References:

- (1) Shannon, M.D., Casci, J.L., Cox, P.A. and Andrews, S.J. *Nature*, **353**, 417-420 (1991)
- (2) Alberti, A., Vezzalini, G., Galli, E. and Quartieri, S. *Eur. J. Mineral.*, **8**, 69-75 (1996)

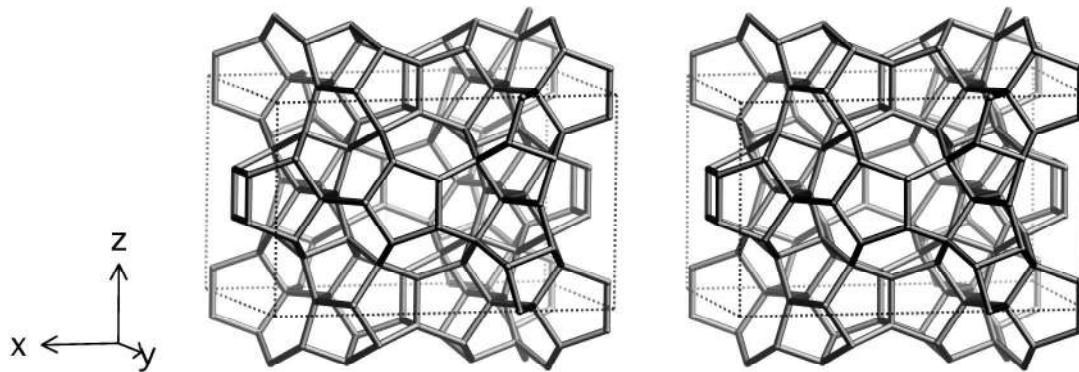
**Crystal chemical data:**  $[H^+_{\cdot 4} (H_2O)_n] [Al_4Si_{64} O_{136}]$ -NES  
monoclinic,  $P12_1/c1$   
 $a = 14.324\text{\AA}$ ,  $b = 22.376\text{\AA}$ ,  $c = 25.092\text{\AA}$ ,  $\beta = 151.51^\circ$ <sup>(1)</sup>  
(Relationship to unit cell of Framework Type:  
 $a' = b/2\sin\beta$ ,  $b' = c$ ,  $c' = a$   
or, as vectors,  $\mathbf{a}' = (\mathbf{b} - \mathbf{a})/2$ ,  $\mathbf{b}' = \mathbf{c}$ ,  $\mathbf{c}' = \mathbf{a}$ )

**Framework density:** 17.7 T/1000 $\text{\AA}^3$

**Channels:** [100] **10** 4.8 x 5.7\*\*



10-ring along [100]



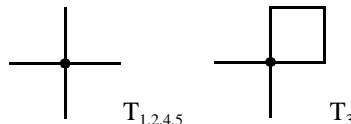
framework viewed along [010]

**Idealized cell constants:** orthorhombic, Fmmm,  $a = 22.9\text{\AA}$ ,  $b = 15.7\text{\AA}$ ,  $c = 13.9\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(32, 1)$	4 12 25 42 67 95 133 174 219 273	5·6·5·6·5·6
	$T_2(16, m)$	4 12 24 39 64 99 130 174 217 262	5·6·5·6·5·6 <sub>2</sub>
	$T_3(16, m)$	4 11 23 44 72 95 124 170 229 279	4·6·5·5·5·5
	$T_4(16, m)$	4 12 24 41 65 97 133 173 212 267	5·5 <sub>2</sub> ·5·6·5·6
	$T_5(8, 2mm)$	4 12 24 40 62 92 142 166 214 262	5·5·5·5·12 <sub>2</sub> ·*

**Secondary building units:** combinations only

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Nonasil<sup>(1)</sup>  
[B-Si-O]-NON<sup>(2)</sup>  
|(Co(C<sub>5</sub>H<sub>5</sub>)<sub>2</sub>)<sub>4</sub> F<sub>4</sub>| [Si<sub>88</sub>O<sub>176</sub>]-NON<sup>(3)</sup>  
CF-3<sup>(4)</sup>  
ZSM-51<sup>(5)</sup>

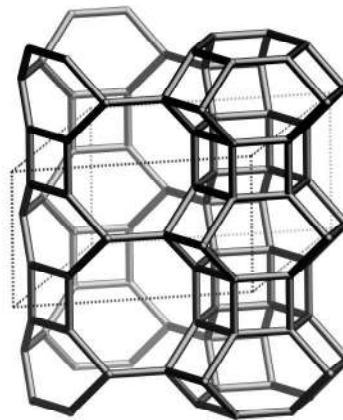
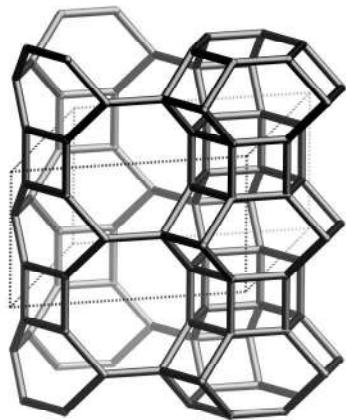
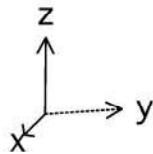
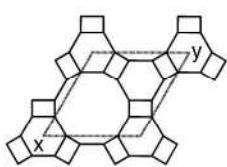
## References:

- (1) Marler, B., Dehnhostel, N., Eulert, H.-H., Gies, H. and Liebau, F. *J. Incl. Phenom.*, **4**, 339-349 (1986)
- (2) Marler, B. and Gies, H. *Zeolites*, **15**, 517-525 (1995)
- (3) Vandegoor, G., Freyhardt, C.C. and Behrens, P. *Z. anorg. allg. Chemie*, **621**, 311-322 (1999)
- (4) Long, Y.-C., Zhong, W. and Shen, X. *J. Incl. Phenom.*, **4**, 121-127 (1986)
- (5) Rohrbaugh, W.J. *private communication*

**Crystal chemical data:**  $[(C_5H_{13}N)_4| [Si_{88} O_{176}]]$ -NON  
 $C_5H_{13}N$  = 2-aminopentane  
orthorhombic, Fmmm,  $a = 22.232\text{\AA}$ ,  $b = 15.058\text{\AA}$ ,  $c = 13.627\text{\AA}$  <sup>(1)</sup>

**Framework density:** 19.3 T/1000 $\text{\AA}^3$

**Channels:** apertures formed by 6-rings only



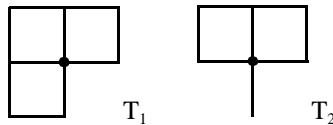
framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** hexagonal, P $\bar{6}$ m2,  $a = 13.1\text{\AA}$ ,  $c = 7.6\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(12, 1)$ 4    9    17    30    50    75    98    118    144    185	4·4·4·6·6·8
	$T_2(6, m)$ 4    10    20    32    46    66    94    128    162    192	4·8·4·8·6·6

**Secondary building units:** 6

**Loop configuration of  
T-Atoms:**



**Framework description:** AAB sequence of 6-rings

**Isotypic framework  
structures:**

\*Offretite<sup>(1,2)</sup>  
LZ-217<sup>(3)</sup>  
Linde T (**ERI-OFF** structural intermediate)<sup>(4)</sup>  
Synthetic offretite<sup>(5)</sup>  
TMA-O<sup>(6)</sup>

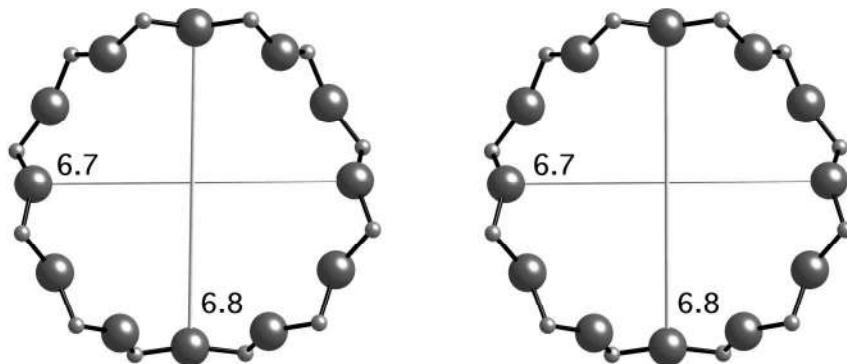
### References:

- (1) Bennett, J.M. and Gard, J.A. *Nature*, **214**, 1005-1006 (1967)
- (2) Gard, J.A. and Tait, J.M. *Acta Crystallogr.*, **B28**, 825-834 (1972)
- (3) Breck, D.W. and Skeels, G.W. *U.S. Patent 4,503,023* (1985)
- (4) Breck, D.W. *Zeolite Molecular Sieves*, p. 173 (1974), Wiley, New York
- (5) Ghobarkar, H. and Schaeff, O. *Cryst. Res. Technol.*, **31**, K29-31 (1996)
- (6) Aiello, R., Barrer, R.M., Davies, J.A. and Kerr, I.S. *Trans. Faraday Soc.*, **66**, 1610-1617 (1970)

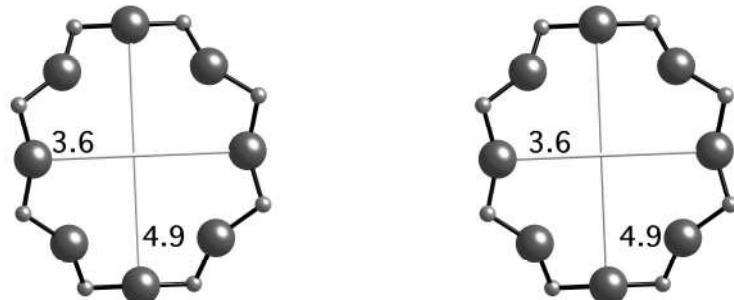
**Crystal chemical data:**  $[(\text{Ca}^{2+}, \text{Mg}^{2+})_{1.5}\text{K}^+(\text{H}_2\text{O})_{14}] [\text{Al}_4\text{Si}_{14}\text{O}_{36}]$ -OFF  
hexagonal,  $\text{P}\bar{6}\text{m}2$ ,  $a = 13.291\text{\AA}$ ,  $c = 7.582\text{\AA}$ <sup>(2)</sup>

**Framework density:** 15.5 T/1000 $\text{\AA}^3$

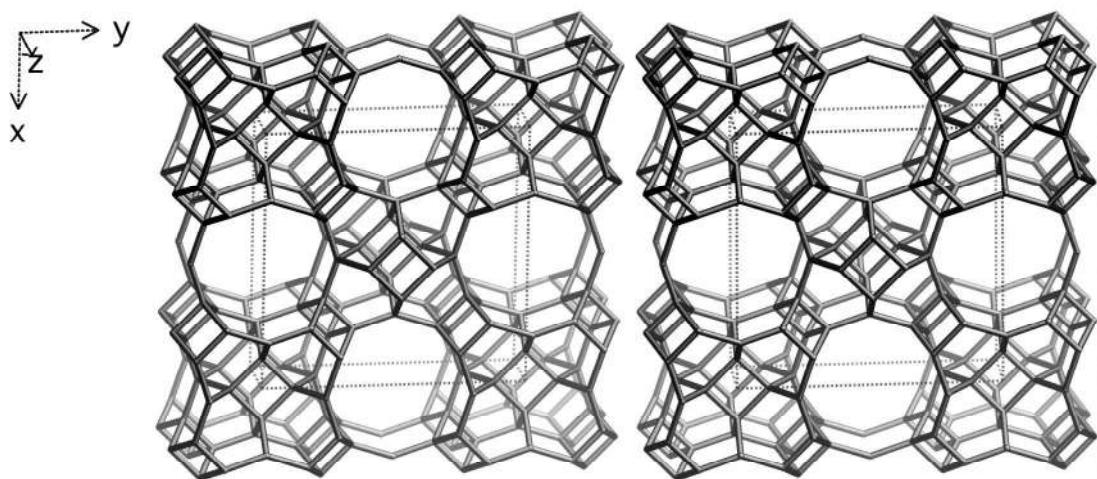
**Channels:** [001] **12** 6.7 x 6.8\*  $\leftrightarrow$   $\perp$  [001] **8** 3.6 x 4.9\*\*



12-ring viewed along [001]



8-ring viewed normal to [001]



framework viewed along [001]

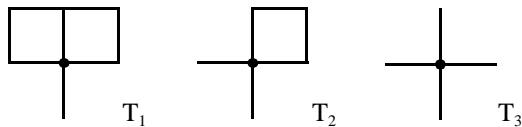
**Idealized cell constants:** tetragonal, I4/mmm,  $a = 18.5\text{\AA}$ ,  $c = 5.3\text{\AA}$

**Coordination sequences and vertex symbols:**

$T_1$ (16, m)	4 10 21 37 57 82 111 145 189 236	4·6·4·6·6·6 <sub>2</sub>
$T_2$ (8, m2m)	4 11 22 34 52 84 120 149 180 220	4·6 <sub>2</sub> ·6 <sub>2</sub> ·6 <sub>2</sub> ·6 <sub>2</sub>
$T_3$ (8, m2m)	4 12 22 37 60 81 112 154 192 230	6 <sub>2</sub> ·6 <sub>2</sub> ·6 <sub>2</sub> ·6 <sub>2</sub> ·12 <sub>6</sub>

**Secondary building units:** 6-2

**Loop configuration of T-Atoms:**



**Isotypic framework structures:** \*UIO-6<sup>(1)</sup>

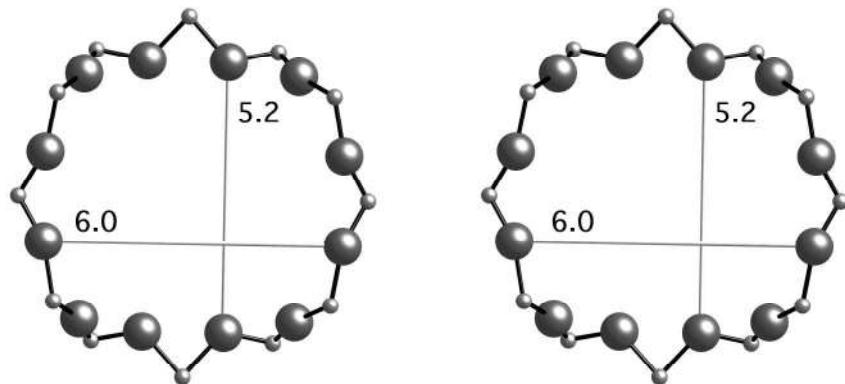
### References:

- (1) Akporiaye, D.E., Fjellvåg, H., Halvorsen, E.N., Haug, T., Karlsson, A. and Lillerud, K.P. *Chem. Commun.*, 1553-1554 (1996)

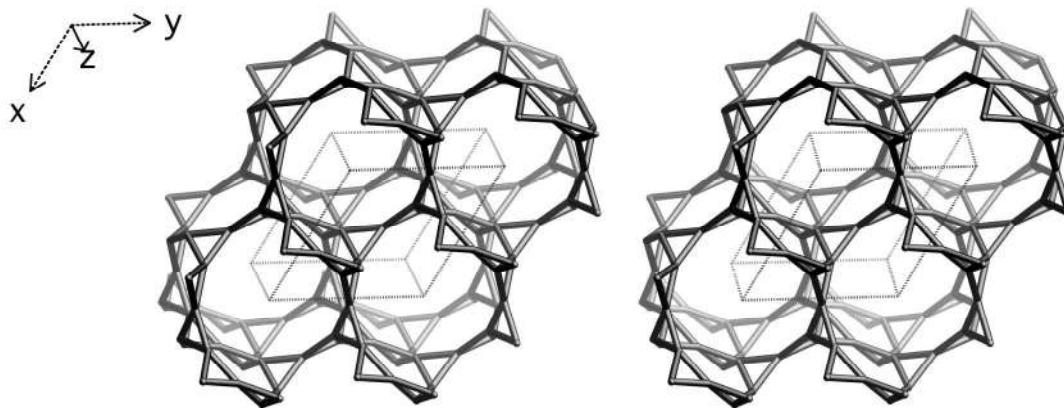
**Crystal chemical data:**  $[Al_{16}P_{16}O_{64}]$ -OSI  
orthorhombic, Imm2,  $a = 18.355\text{\AA}$ ,  $b = 18.321\text{\AA}$ ,  $c = 5.053\text{\AA}$ <sup>(1)</sup>

**Framework density:** 18.8 T/1000 $\text{\AA}^3$

**Channels:** [001] **12** 5.2 x 6.0\*



12-ring viewed along [001]



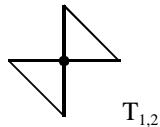
framework viewed along [001]

**Idealized cell constants:** hexagonal, P6<sub>2</sub>22,  $a = 10.1\text{\AA}$ ,  $c = 7.6\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(6, 2)$	4    8    16    29    46    70    101    118    162    190	3·3·8·8·8·14 <sub>10</sub>
	$T_2(3, 222)$	4    8    16    30    44    76    92    130    148    202	3·3·8·8·14 <sub>7</sub> ·14 <sub>7</sub>

**Secondary building units:** 3

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*OSB-1<sup>(1)</sup>

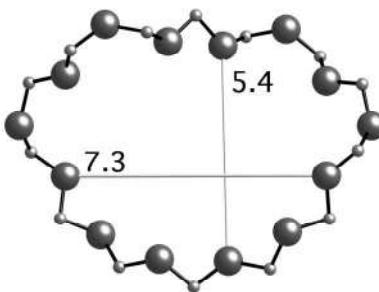
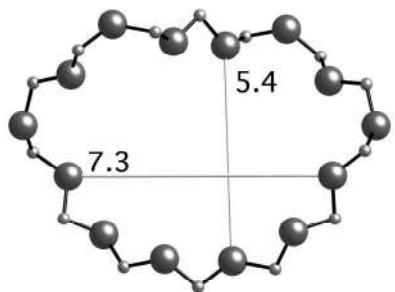
### References:

- (1) Kongshaug, K.O., Fjellvåg, H., Lillerud, K.P., Gier, T.E., Stucky, G.D. and Cheetham, A.K. *private communication*

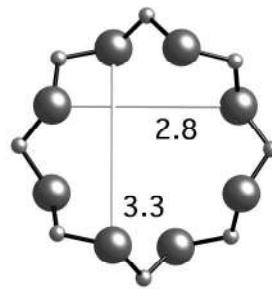
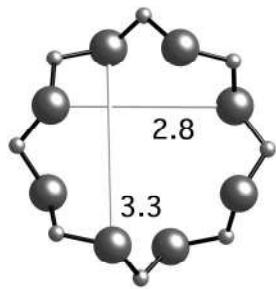
**Crystal chemical data:**  $[\text{K}^+ \cdot (\text{H}_2\text{O})_9] \cdot [\text{Be}_3\text{Si}_6\text{O}_{18}]$ -OSO  
trigonal,  $\text{P}3_2$ ,  $a = 10.093\text{\AA}$ ,  $c = 7.626\text{\AA}$ <sup>(1)</sup>

**Framework density:** 13.4 T/1000 $\text{\AA}^3$

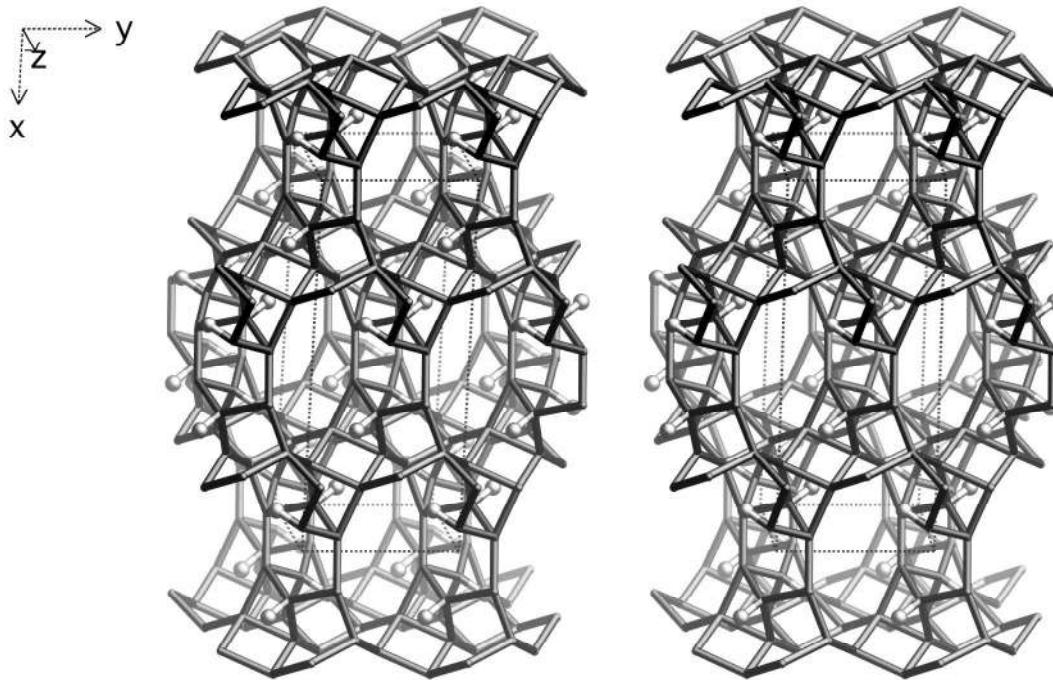
**Channels:** [001] **14** 5.4 x 7.3\*  $\leftrightarrow$   $\perp$  [001] **8** 2.8 x 3.3\*\*



*puckered 14-ring viewed along [001]*



*8-ring viewed normal to [001]*

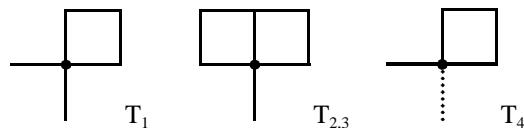
*framework viewed along [001]*

**Idealized cell constants:** monoclinic, C2/c,  $a = 20.9\text{\AA}$ ,  $b = 9.2\text{\AA}$ ,  $c = 8.6\text{\AA}$ ,  $\beta = 89.7^\circ$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (8, 1)    4    9    18    31    48    71    99    132    162    197	4·6·6·6·6·10 <sub>2</sub>
	T <sub>2</sub> (8, 1)    4    9    19    33    51    77    96    126    162    203	4·6·4·8 <sub>2</sub> ·6·6 <sub>2</sub>
	T <sub>3</sub> (8, 1)    4    10    19    33    52    72    102    126    161    204	4·6 <sub>2</sub> ·4·8 <sub>2</sub> ·8 <sub>2</sub> ·10
	T <sub>4</sub> (8, 1)    3    8    16    29    49    68    94    123    162    203	4·6·6 <sub>2</sub>

**Secondary building units:** 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*Partheite<sup>(1)</sup>

### References:

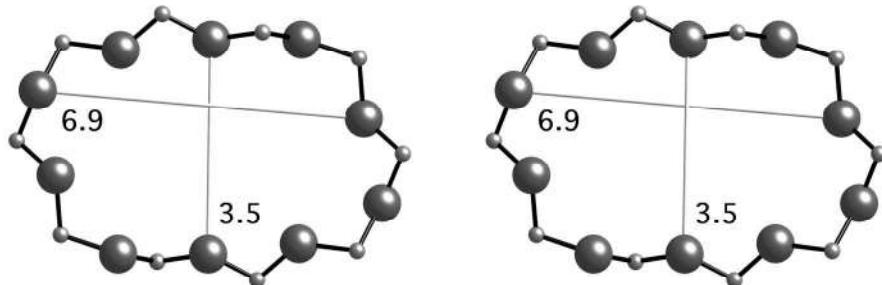
- (1) Engel, N. and Yvon, K. *Z. Kristallogr.*, **169**, 165-175 (1984)

**Crystal chemical data:**  $[\text{Ca}^{2+}_8 (\text{H}_2\text{O})_{16}] [\text{Al}_{16}\text{Si}_{16} \text{O}_{60}(\text{OH})_8]$ -**-PAR**  
monoclinic, C2/c,  $a = 21.555\text{\AA}$ ,  $b = 8.761\text{\AA}$ ,  $c = 9.304\text{\AA}$ ,  $\beta = 91.55^\circ$  <sup>(1)</sup>

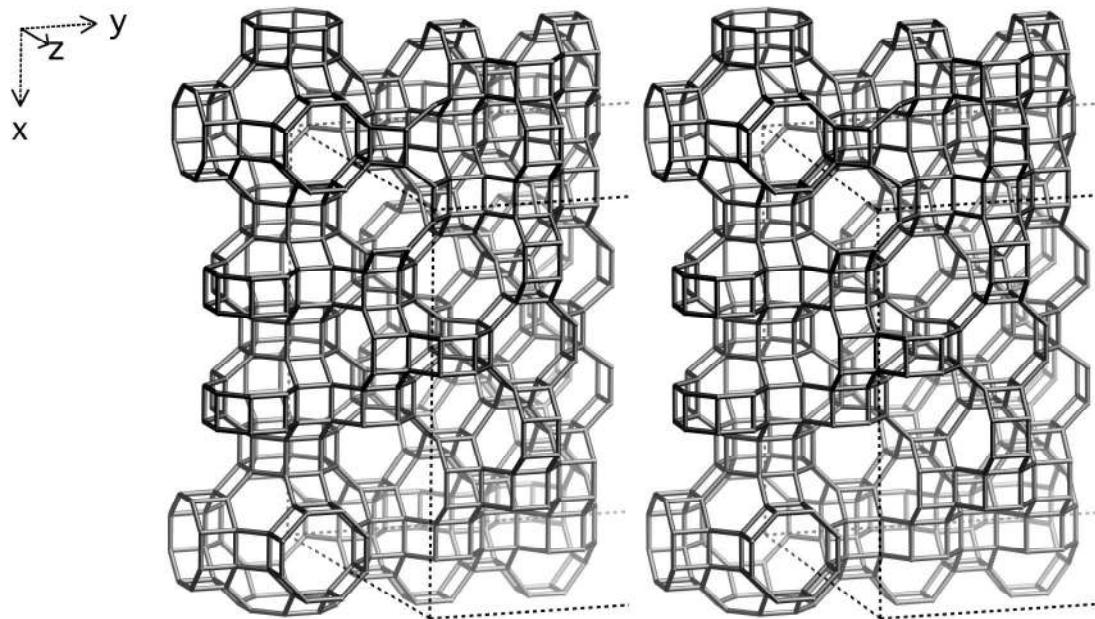
**Framework density:** 18.2 T/1000 $\text{\AA}^3$

**Channels:** [001] **10** 3.5 x 6.9\*

**Stability:** Stable at 150°C, transforms at 400°C <sup>(1)</sup>



*10-ring viewed along [001]*



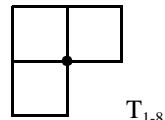
framework viewed along [001]

**Idealized cell constants:** cubic, Im $\bar{3}$ m,  $a = 34.8\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (96, 1)    4    9    18    31    47    68    91    117    151    188	4·4·4·8 <sub>2</sub> ·8·8
	T <sub>2</sub> (96, 1)    4    9    18    32    49    70    95    122    154    191	4·4·4·8 <sub>2</sub> ·8·8
	T <sub>3</sub> (96, 1)    4    9    18    32    49    69    94    123    153    186	4·4·4·8 <sub>2</sub> ·8·8
	T <sub>4</sub> (96, 1)    4    9    18    32    48    67    92    121    152    185	4·4·4·8 <sub>2</sub> ·8·8
	T <sub>5</sub> (96, 1)    4    9    18    31    47    68    92    119    152    188	4·4·4·8 <sub>2</sub> ·8·8
	T <sub>6</sub> (96, 1)    4    9    17    29    45    65    89    117    149    185	4·4·4·6·8·8
	T <sub>7</sub> (48, 2)    4    9    17    30    47    66    88    113    144    183	4·4·4·6·8·8
	T <sub>8</sub> (48, 2)    4    9    18    32    49    69    95    123    152    188	4·4·4·8 <sub>2</sub> ·8·8

**Secondary building units:** 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Paulingite<sup>(1)</sup>  
ECR-18<sup>(2)</sup>

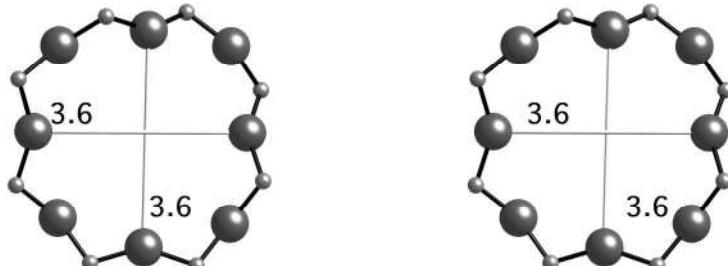
### References:

- (1) Gordon, E.K., Samson, S. and Kamb, W.K. *Science*, **154**, 1004-1007 (1966)
- (2) Vaughan, D.E.W. and Strohmaier, G. *U.S. Patent 4,661,332* (1987)

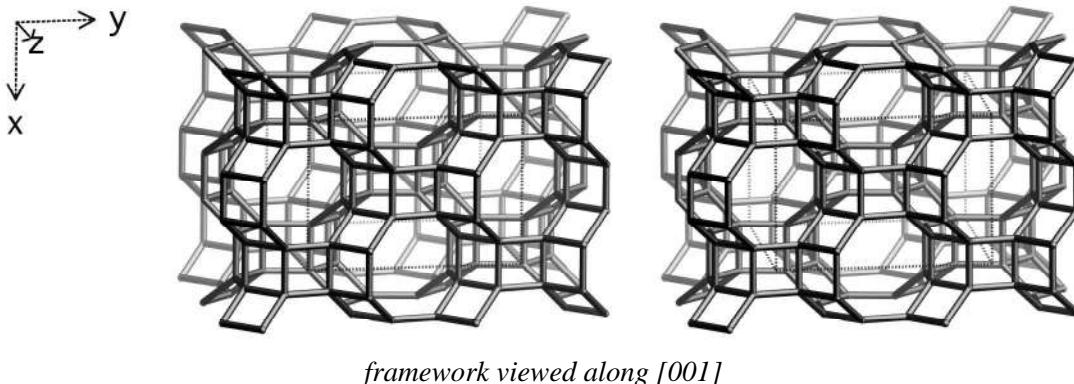
**Crystal chemical data:**  $[(\text{Ca}^{2+}, \text{K}^+, \text{Na}^+)_7 \text{H}_2\text{O}]_{700} [\text{Al}_{152}\text{Si}_{520}\text{O}_{1344}]$ -PAU  
cubic,  $\text{Im}\bar{3}\text{m}$ ,  $a = 35.093\text{\AA}$ <sup>(1)</sup>

**Framework density:** 15.5 T/1000 $\text{\AA}^3$

**Channels:**  $<100> 8 \text{ } 3.6 \times 3.6 \text{***} | <100> 8 \text{ } 3.6 \times 3.6 \text{***}$



*8-ring viewed along <100>*



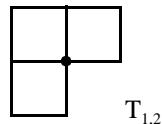
framework viewed along [001]

**Idealized cell constants:** orthorhombic, Cmcm,  $a = 9.9\text{\AA}$ ,  $b = 14.1\text{\AA}$ ,  $c = 14.0\text{\AA}$

**Coordination sequences  
and vertex symbols:**  $T_1(16, 1) \quad 4 \quad 9 \quad 18 \quad 32 \quad 50 \quad 71 \quad 94 \quad 122 \quad 157 \quad 195$        $4 \cdot 4 \cdot 4 \cdot 8_2 \cdot 8 \cdot 8$   
 $T_2(16, 1) \quad 4 \quad 9 \quad 18 \quad 32 \quad 48 \quad 68 \quad 96 \quad 126 \quad 155 \quad 191$        $4 \cdot 4 \cdot 4 \cdot 8_2 \cdot 8 \cdot 8$

**Secondary building units:** 8 or 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Phillipsite<sup>(1,2)</sup>  
 $[\text{Al-Co-P-O}] \text{-PHI}^{(3)}$   
Harmotome<sup>(2,4)</sup>  
ZK-19<sup>(6)</sup>

**Alternate designation:** Wellsite<sup>(5)</sup> (discredited)

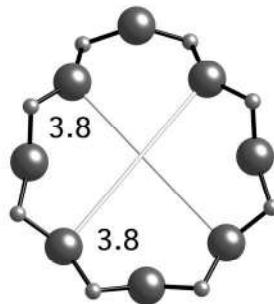
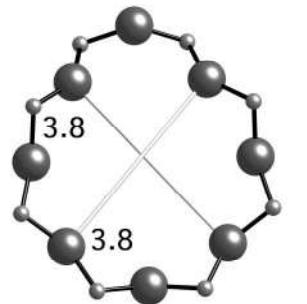
### References:

- (1) Steinfink, H. *Acta Crystallogr.*, **15**, 644-651 (1962)
- (2) Rinaldi, R., Pluth, J.J. and Smith, J.V. *Acta Crystallogr.*, **B30**, 2426-2433 (1974)
- (3) Feng, P.Y., Bu, X.H. and Stucky, G.D. *Nature*, **388**, 735-741 (1997)
- (4) Sadanaga, R., Marumo, F. and Yakéuchi, Y. *Acta Crystallogr.*, **14**, 1153-1163 (1961)
- (5) Cerny, P., Rinaldi, R. and Surdam, R.C. *N. Jb. Miner. Abh.*, **128**, 312-330 (1977)
- (6) Kuehl, G.H. *Am. Mineral.*, **54**, 1607-1612 (1969)

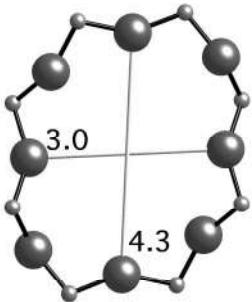
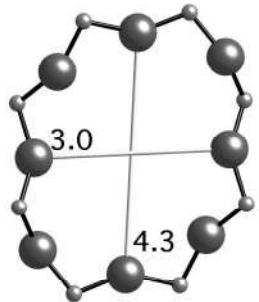
**Crystal chemical data:**  $[\text{K}^+_{2(\text{Ca}^{2+},\text{Na}^+)_2} (\text{H}_2\text{O})_{12}] [\text{Al}_6\text{Si}_{10} \text{O}_{32}]$ -PHI  
monoclinic, P12<sub>1</sub>/m1  
 $a = 9.865\text{\AA}$ ,  $b = 14.300\text{\AA}$ ,  $c = 8.668\text{\AA}$ ,  $\beta = 124.20^\circ$ <sup>(2)</sup>  
(Relationship to unit cell of Framework Type:  
 $\mathbf{a}' = \mathbf{a}$ ,  $\mathbf{b}' = \mathbf{c}$ ,  $\mathbf{c}' = \mathbf{b}/2\sin(\beta)$   
or, as vectors,  $\mathbf{a}' = \mathbf{a}$ ,  $\mathbf{b}' = \mathbf{c}$ ,  $\mathbf{c}' = (\mathbf{b} - \mathbf{a})/2$ )

**Framework density:** 15.8 T/1000 $\text{\AA}^3$

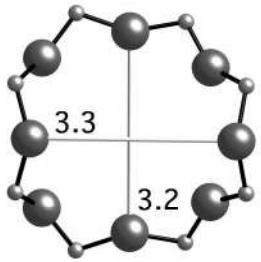
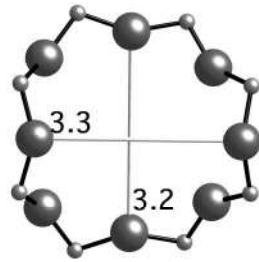
**Channels:** [100] 8 3.8 x 3.8\*  $\leftrightarrow$  [010] 8 3.0 x 4.3\*  $\leftrightarrow$  [001] 8 3.2 x 3.3\*



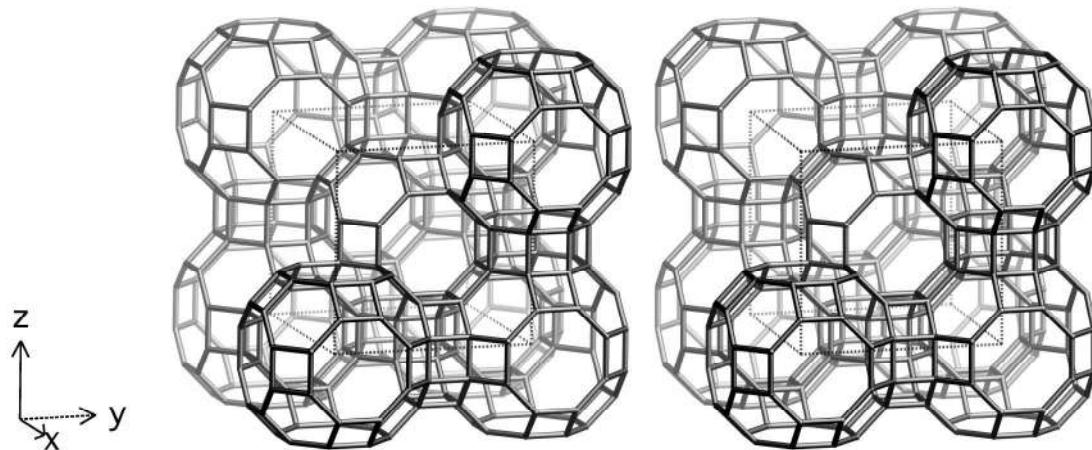
8-ring viewed along [100]



8-ring viewed along [010]



8-ring viewed along [001]



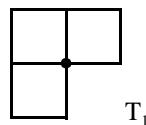
framework viewed along [100]

**Idealized cell constants:** cubic, Im $\bar{3}$ m,  $a = 14.9\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(48, 2)$  4 9 17 28 42 60 81 105 132 162 4·4·4·6·8·8

**Secondary building units:** 8-8 or 6-2 or 8 or 6 or 4

**Loop configuration of T-Atoms:**

 $T_1$ 

**Isotypic framework structures:**

\*Rho<sup>(1,2)</sup>

[Be-As-O]-RHO<sup>(3)</sup>

[Be-P-O]-RHO<sup>(4)</sup>

[Co-Al-P-O]-RHO<sup>(5)</sup>

[Mg-Al-P-O]-RHO<sup>(5)</sup>

[Mn-Al-P-O]-RHO<sup>(5)</sup>

Deuterated Rho<sup>(6)</sup>

Gallosilicate ECR-10<sup>(7)</sup>

LZ-214<sup>(8)</sup>

Pahasapaite<sup>(9,10)</sup>

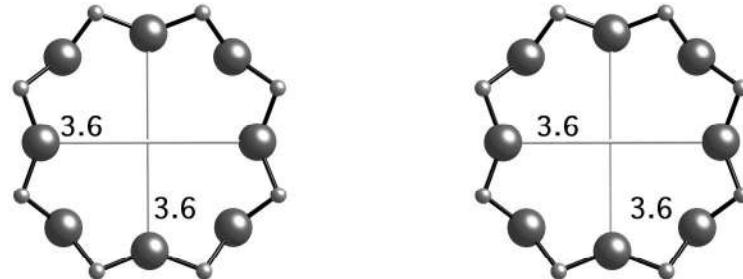
### References:

- (1) Robson, H.E., Shoemaker, D.P., Ogilvie, R.A. and Manor, P.C. *Adv. Chem. Ser.*, **121**, 106-115 (1973)
- (2) McCusker, L.B. and Baerlocher, Ch. In *Proc. 6th Int. Zeolite Conf.*, (eds. D.H. Olson and A. Bisio), pp. 812-822 (1984), Butterworths, Guildford, Surry
- (3) Gier, T.E. and Stucky, G.D. *Nature*, **349**, 508-510 (1991)
- (4) Harvey, G. and Meier, W.M. *Stud. Surf. Sci. Catal.*, **49**, 411-420 (1989)
- (5) Feng, P., Bu, X. and Stucky, G.D. *Microporous and Mesoporous Materials*, **23**, 315-322 (1998)
- (6) Parise, J.B., Gier, T.E., Corbin, D.R. and Cox, D.E. *J. Phys. Chem.*, **88**, 1635-1640 (1984)
- (7) Newsam, J.M., Vaughan, D.E.W. and Strohmaier, K.G. *J. Phys. Chem.*, **99**, 9924-9932 (1995)
- (8) Breck, D.W. and Skeels, G.W. *U.S. Patent 4,503,023*, (1985)
- (9) Rouse, R.C., Peacor, D.R., Dunn, P.J., Campbell, T.J., Roberts, W.L., Wicks, F.J. and Newbury, D. N. *Jb. Miner. Mh.*, 433-440 (1987)
- (10) Rouse, R.C., Peacor, D.R. and Merlini, S. *Am. Mineral.*, **74**, 1195-1202 (1989)

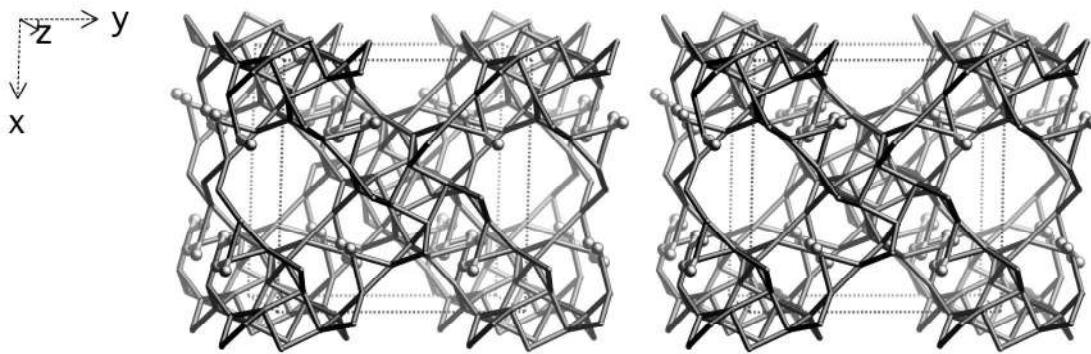
**Crystal chemical data:**  $[(\text{Na}^+, \text{Cs}^+)_{12} (\text{H}_2\text{O})_{44}] [\text{Al}_{12}\text{Si}_{36} \text{O}_{96}]$ -RHO  
cubic,  $\text{Im}\bar{3}\text{m}$ ,  $a = 15.031\text{\AA}$  <sup>(2)</sup>

**Framework density:** 14.1 T/1000 $\text{\AA}^3$

**Channels:**  $<100> 8 3.6 \times 3.6^{***} | <100> 8 3.6 \times 3.6^{***}$



*8-ring viewed along <100>*



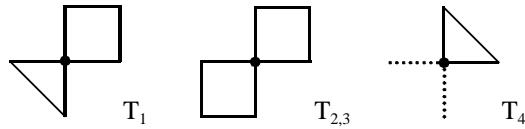
framework viewed along [001]

**Idealized cell constants:** tetragonal, I4/mcm,  $a = 18.1\text{\AA}$ ,  $c = 9.0\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (16, m)	4    7    17    31    49    76    98    125    170    208	3·4·10 <sub>4</sub> *·10 <sub>4</sub> *
	T <sub>2</sub> (16, m)	4    10    20    31    47    78    109    127    162    212	4·4·6·6·6·6
	T <sub>3</sub> (16, 2)	4    10    16    31    56    67    94    146    164    188	4·4·6 <sub>2</sub> ·10 <sub>4</sub> ·10 <sub>2</sub> ·10 <sub>2</sub>
	T <sub>4</sub> (8, m2m)	2    4    10    24    38    58    91    110    138    194	3

**Secondary building units:** combinations only

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Roggianite<sup>(1)</sup>

The previously reported structure (the ROG type) was found to be incomplete and the code -ROG has been discredited based on a decision by the IZA Structure commission.

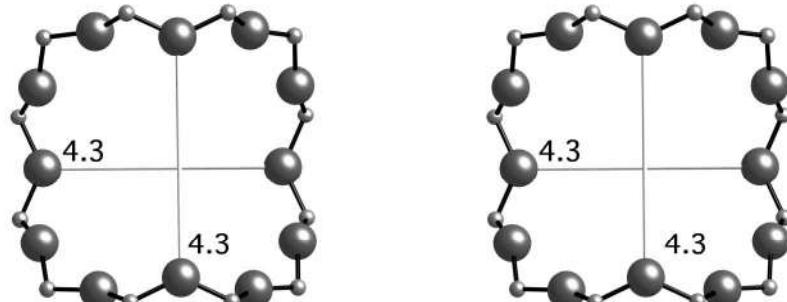
## References:

- (1) Giuseppetti, G., Mazzi, F., Tadini, C. and Galli, E. *N. Jb. Miner. Mh.*, 307-314 (1991)

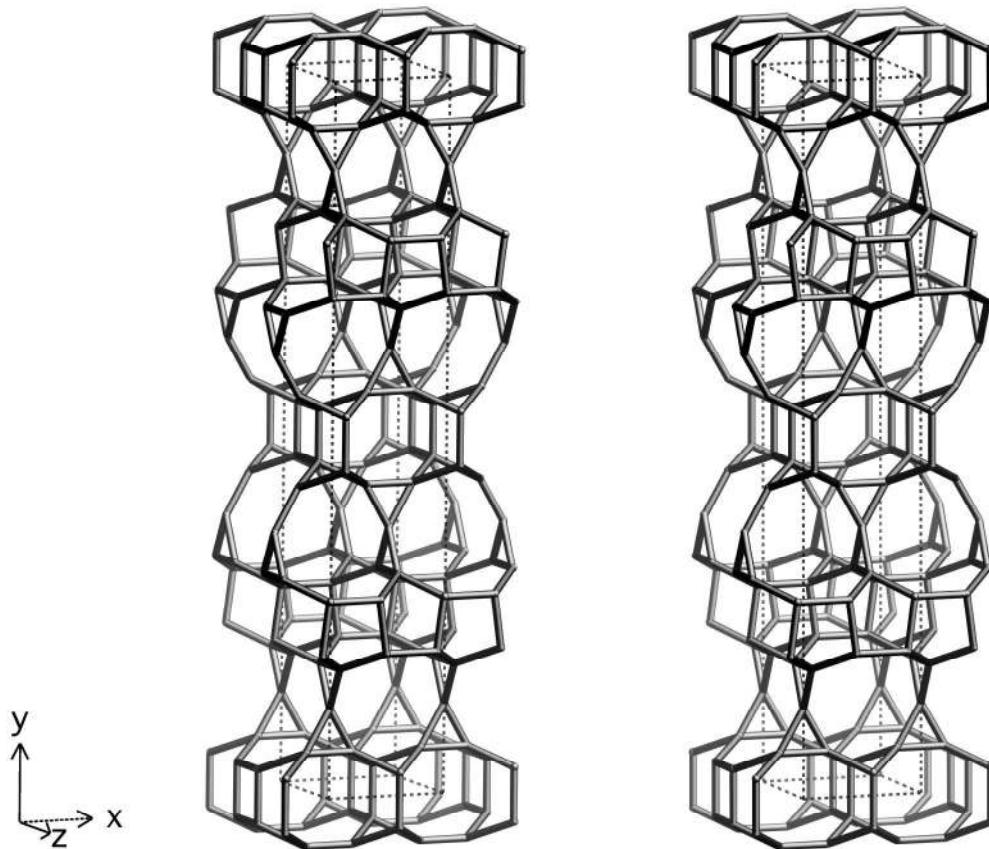
**Crystal chemical data:**  $[\text{Ca}^{2+}]_{16} (\text{H}_2\text{O})_{19} | [\text{Be}_8\text{Al}_{16}\text{Si}_{32}\text{O}_{104}(\text{OH})_{16}] \cdot \text{-RON}$   
tetragonal, I4/mcm,  $a = 18.33\text{\AA}$ ,  $c = 9.16\text{\AA}$  <sup>(1)</sup>

**Framework density:** 18.2 T/1000 $\text{\AA}^3$

**Channels:** [001] **12** 4.3 x 4.3\*



12-ring viewed along [001]

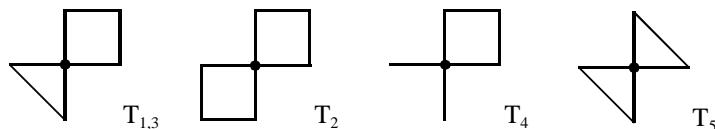


**Idealized cell constants:** monoclinic, C2/m,  $a = 7.2\text{\AA}$ ,  $b = 41.8\text{\AA}$ ,  $c = 7.2\text{\AA}$ ,  $\gamma = 90.0^\circ$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (8, 1)    4    9    19    39    59    78    117    155    189    235 T <sub>2</sub> (8, 1)    4    10    21    37    58    91    117    144    194    241 T <sub>3</sub> (8, 1)    4    9    21    42    57    82    119    151    188    239 T <sub>4</sub> (8, 1)    4    11    21    40    61    89    116    145    191    239 T <sub>5</sub> (4, 2)    4    8    20    44    55    80    118    152    204    228	3·4·8 <sub>3</sub> ·9 <sub>4</sub> ·8 <sub>3</sub> ·9 <sub>4</sub> 4·4·6 <sub>2</sub> ·8·6 <sub>2</sub> ·8 3·4·8 <sub>2</sub> ·9 <sub>4</sub> ·8 <sub>2</sub> ·9 <sub>4</sub> 4·5 <sub>2</sub> ·5·8·5·8 3·3·9 <sub>4</sub> ·9 <sub>4</sub> ·9 <sub>4</sub>
---	---	--

**Secondary building units:** combinations only

**Loop configuration of  
T-Atoms:**



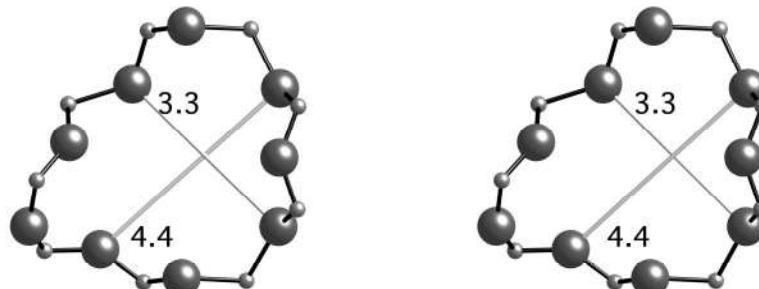
**Isotypic framework  
structures:** \*RUB-17<sup>(1)</sup>

**Crystal chemical data:**  $[\text{K}^+ \text{Na}^{+12} (\text{H}_2\text{O})_{18}] [\text{Zn}_8\text{Si}_{28} \text{O}_{72}]$ -RSN  
monoclinic, C1m1,  $a = 7.238\text{\AA}$ ,  $b = 40.56\text{\AA}$ ,  $c = 7.308\text{\AA}$ ,  $\beta = 91.8^\circ$ <sup>(1)</sup>

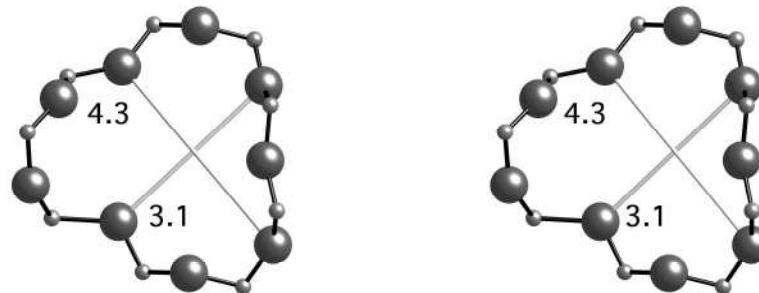
**Framework density:**  $16.8 \text{ T}/1000\text{\AA}^3$

**Channels:** [100] 9 3.3 x 4.4\*  $\leftrightarrow$  [001] 9 3.1 x 4.3\*  $\leftrightarrow$  [010] 8 3.4 x 4.1\*

**Stability:** Complete dehydration leads to destruction of the framework<sup>(1)</sup>



9-ring viewed along [100]

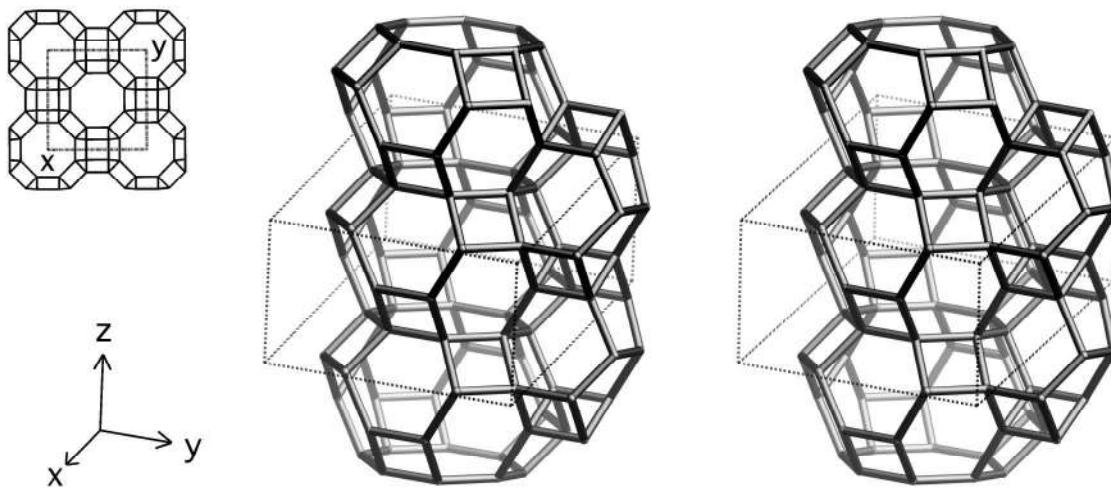


9-ring viewed along [001]

See Appendix A for 8-ring viewed along [010]

### References:

- (1) Röhrig, C. and Gies, H. *Angew. Chem., Int. Ed.*, **34**, 63-65 (1995)

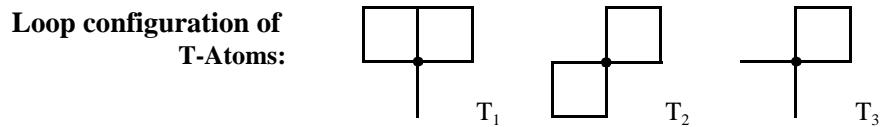


framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** monoclinic, C2/m,  $a = 14.1\text{\AA}$ ,  $b = 13.7\text{\AA}$ ,  $c = 7.4\text{\AA}$ ,  $\beta = 102.4^\circ$

<b>Coordination sequences and vertex symbols:</b>	$T_1(8, 1)$	4 10 19 33 56 81 105 136 175 219	4.5.4.6.5.6
	$T_2(8, 1)$	4 10 22 37 54 79 108 140 176 215	4.4.5.8.6.6
	$T_3(8, 1)$	4 11 21 35 57 80 106 139 176 218	4.5.5.6.6.8

**Secondary building units:** 5-3 or 4



**Isotypic framework structures:** \*RUB-3<sup>(1,2)</sup>

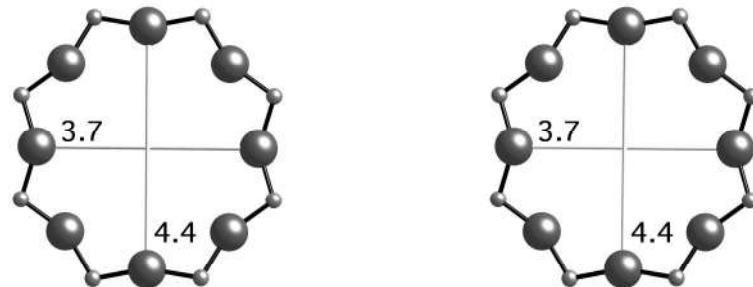
### References:

- (1) Marler, B., Grünwald-Lüke, A. and Gies, H. *Zeolites*, **15**, 388-399 (1995)
- (2) Marler, B., Grünwald-Lüke, A. and Gies, H. *Microporous and Mesoporous Materials*, **26**, 49-59 (1998)

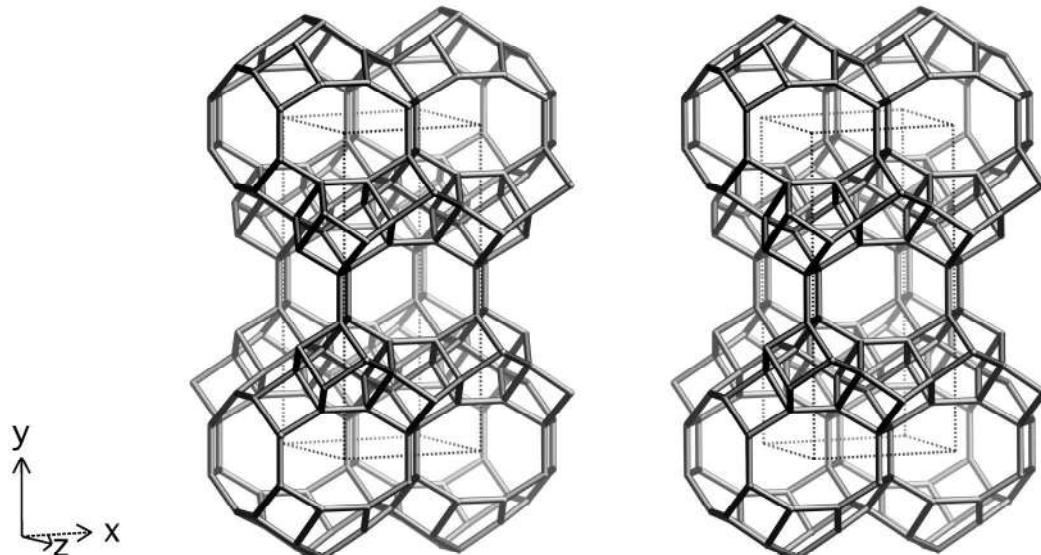
**Crystal chemical data:**  $[(C_8H_{15}N)_2] [Si_{24}O_{48}]$ -RTE  
 $C_8H_{15}N$  = exo-2-aminobicyclo[2.2.1.]heptane  
monoclinic, C12/m1  
 $a = 14.039\text{\AA}$ ,  $b = 13.602\text{\AA}$ ,  $c = 7.428\text{\AA}$ ,  $\beta = 102.22^\circ$  <sup>(1)</sup>

**Framework density:**  $17.3 \text{ T}/1000\text{\AA}^3$

**Channels:** [001] 8 3.7 x 4.4\*



8-ring viewed along [001]



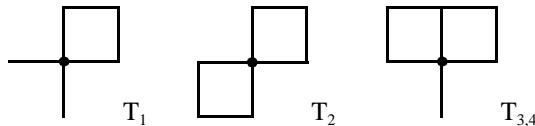
framework viewed along [001]

**Idealized cell constants:** monoclinic, C2/m,  $a = 9.8\text{\AA}$ ,  $b = 20.5\text{\AA}$ ,  $c = 10.0\text{\AA}$ ,  $\beta = 96.9^\circ$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (8, 1)    4    11    21    34    53    78    108    137    165    207 T <sub>2</sub> (8, 1)    4    10    21    36    54    75    100    136    181    214 T <sub>3</sub> (8, 1)    4    10    19    31    50    82    106    130    168    203 T <sub>4</sub> (8, 1)    4    10    18    31    55    77    103    134    165    214	4.6·5·6·5·8 4·4·5·8·5·8 4·5·4·6·5·5 4·5·4·8·5·5
---	--	--

**Secondary building units:** 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*RUB-13<sup>(1)</sup>

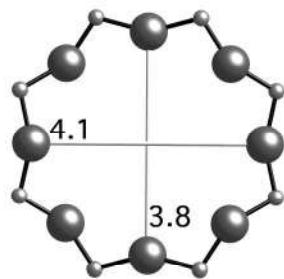
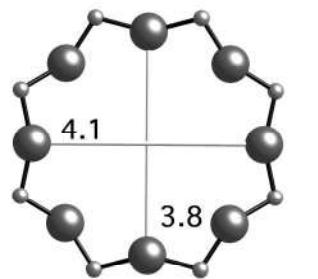
### References:

- (1) Vortmann, S., Marler, B., Gies, H. and Daniels, P. *Microporous Materials*, **4**, 111-121 (1995)

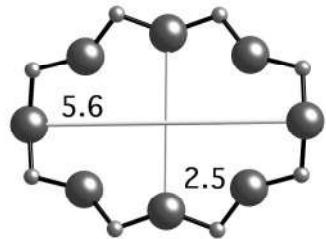
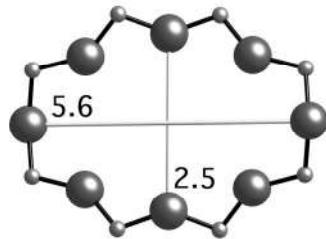
**Crystal chemical data:**  $[(C_{10}H_{21}N^+)_2| [B_2Si_{30}O_{64}]]\text{-RTH}$   
 $C_{10}H_{21}N^+$  = pentamethylpiperidinium  
monoclinic, C12/m1,  $a = 9.659\text{\AA}$ ,  $b = 20.461\text{\AA}$ ,  $c = 9.831\text{\AA}$ ,  $\beta = 96.58^\circ$  <sup>(1)</sup>

**Framework density:** 16.6 T/1000 $\text{\AA}^3$

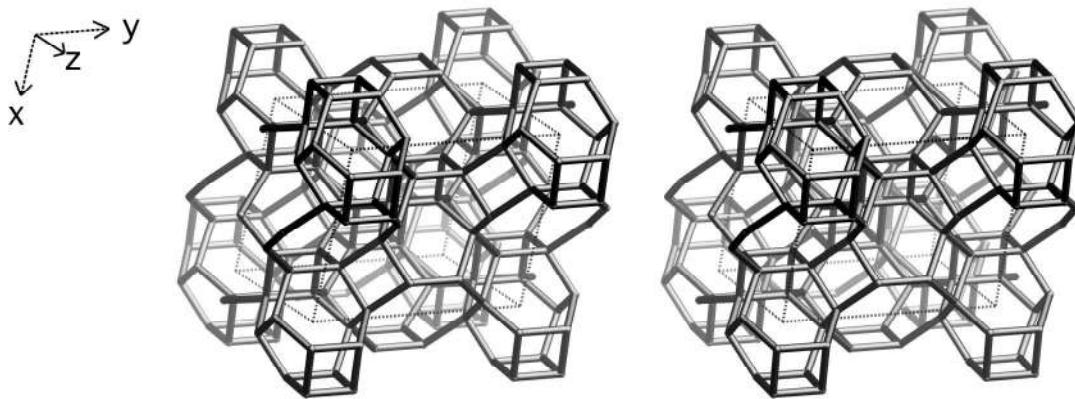
**Channels:** [100] 8 3.8 x 4.1\*  $\leftrightarrow$  [001] 8 2.5 x 5.6\*



8-ring viewed along [100]



8-ring viewed along [001]



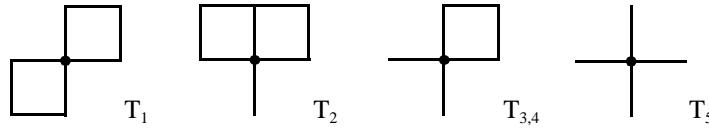
framework viewed along [001]

**Idealized cell constants:** monoclinic, C2/m,  $a = 13.2\text{\AA}$ ,  $b = 13.3\text{\AA}$ ,  $c = 12.5\text{\AA}$ ,  $\beta = 114.8^\circ$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (8, 1)    4 10 21 37 58 87 116 146 185 232	4·4·5·6·5·8
	T <sub>2</sub> (8, 1)    4 10 21 38 60 84 113 148 192 232	4·5·4·6·5·6
	T <sub>3</sub> (8, 1)    4 11 23 38 58 86 114 148 189 234	4·5·5·6·6·6
	T <sub>4</sub> (8, 1)    4 11 21 37 62 85 114 148 185 232	4·5·5·5·6·8
	T <sub>5</sub> (4, 2)    4 12 22 40 58 82 116 154 186 232	5·5·6·6·6·6

**Secondary building units:** combinations only

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*RUB-10<sup>(1)</sup>  
|TMA-|[Si-O]-RUT<sup>(2)</sup>  
B-NU-1<sup>(3)</sup>  
Fe-NU-1<sup>(3)</sup>  
Ga-NU-1<sup>(3)</sup>  
NU-1<sup>(4)</sup>

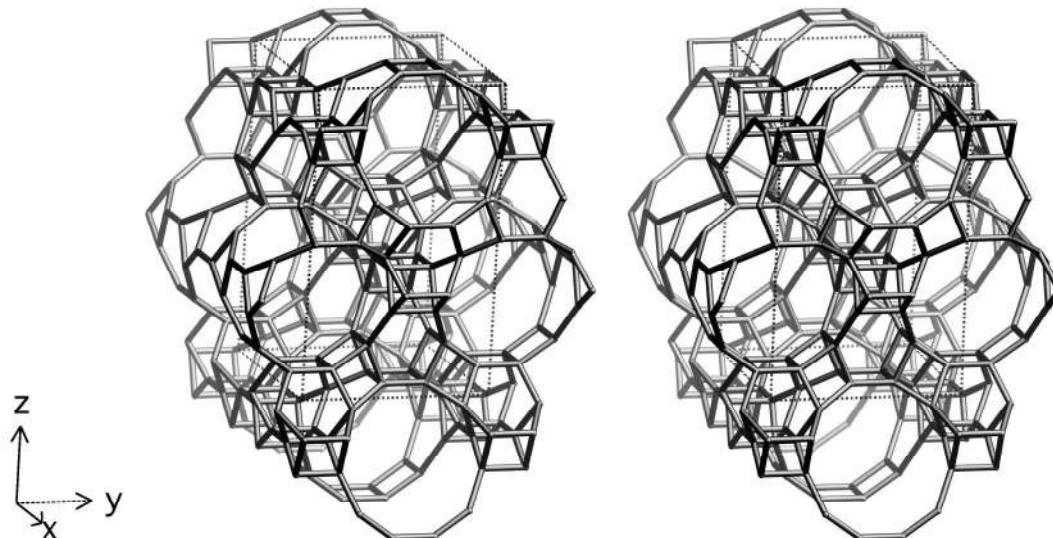
### References:

- (1) Gies, H. and Rius, J. Z. Kristallogr., **210**, 475-480 (1995)
- (2) Broach, R.W., McGuire, N.K., Chao, C.C. and Kirchner, R.M. J. Phys. Chem. Solids, **56**, 1363-1368 (1995)
- (3) Bellussi, G., Millini, R., Carati, A., Maddinelli, G. and Gervasini, A. Zeolites, **10**, 642-649 (1990)
- (4) Whittam, T.V. and Youll, B. U.S. Patent 4,060,590, (1977)

**Crystal chemical data:**  $[(\text{C}_4\text{H}_{12}\text{N}^+)_4] [\text{B}_4\text{Si}_{32}\text{O}_{72}]$ -RUT  
 $\text{C}_4\text{H}_{12}\text{N}^+$  = tetramethylammonium  
monoclinic,  $P12_1/a1$   
 $a = 13.112\text{\AA}$ ,  $b = 12.903\text{\AA}$ ,  $c = 12.407\text{\AA}$ ,  $\beta = 113.50^\circ$  <sup>(1)</sup>

**Framework density:**  $18.7 \text{ T}/1000\text{\AA}^3$

**Channels:** apertures formed by 6-rings only



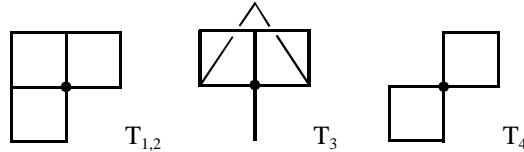
framework viewed along [100]

**Idealized cell constants:** tetragonal, I<sup>4</sup>m2,  $a = 13.4\text{\AA}$ ,  $c = 21.9\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (16, 1)	4    9    16    25    39    61    84    102    124    158	4·4·4·6 <sub>2</sub> ·6 <sub>3</sub> ·12 <sub>4</sub>
	T <sub>2</sub> (16, 1)	4    9    17    27    40    61    85    106    132    167	4·4·4·12 <sub>5</sub> ·6·6 <sub>3</sub>
	T <sub>3</sub> (16, 1)	4    9    16    25    39    58    79    104    130    158	4·6·4·6·4·12 <sub>6</sub>
	T <sub>4</sub> (8, 2)	4    10    16    25    42    61    82    108    132    156	4·4·6·6·6 <sub>2</sub> ·12 <sub>5</sub>

**Secondary building units:** 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*STA-1<sup>(1)</sup>

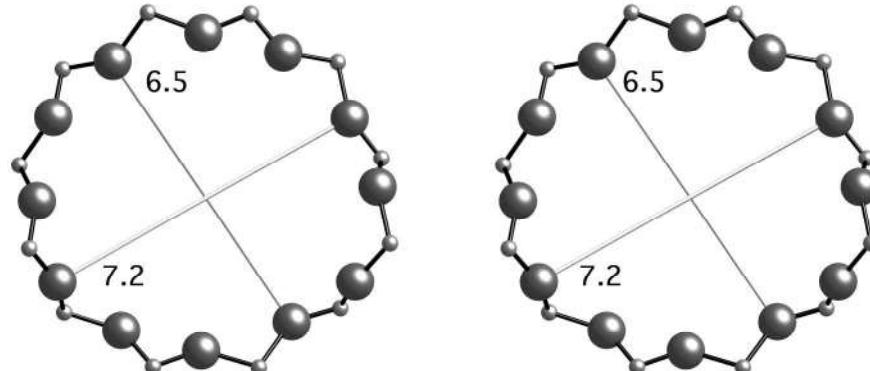
## References:

- (1) Noble, G.W., Wright, P.A., Lightfoot, P., Morris, R.E., Hudson, K.J., Kvick, Å. and Graafsma, H. *Angew. Chem., Int. Ed.*, **36**, 81-83 (1997)

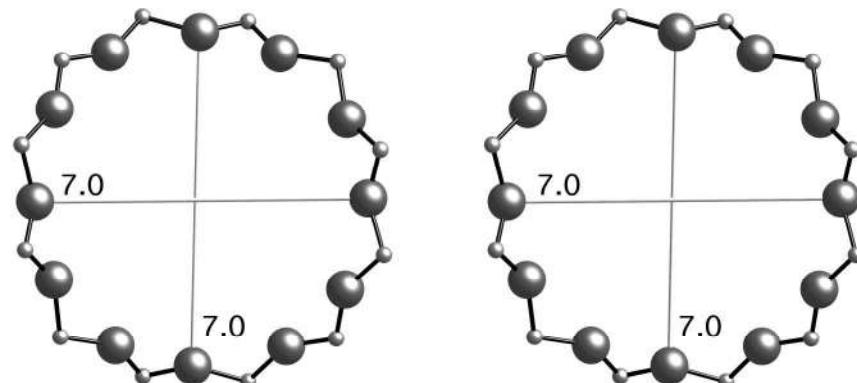
**Crystal chemical data:**  $[(C_{21}H_{40}N_2^{2+})_{2.6} (H_2O)_6] [Mg_5Al_{23}P_{28} O_{112}]$ -SAO  
 $C_{21}H_{40}N_2^{2+} = C_7H_{13}N - (CH_2)_7 - C_7H_{13}N$   
 $C_7H_{13}N$  = quinuclidine  
tetragonal,  $P\bar{4}n2$ ,  $a = 13.810\text{\AA}$ ,  $c = 21.969\text{\AA}$  <sup>(1)</sup>

**Framework density:**  $13.4 \text{ T}/1000\text{\AA}^3$

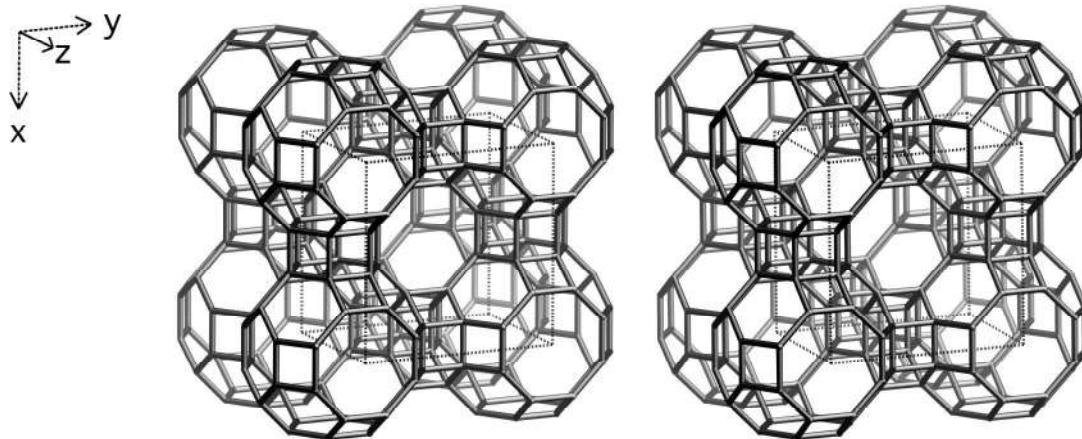
**Channels:**  $<100> \textbf{12} 6.5 \times 7.2^{**} \leftrightarrow [001] \textbf{12} 7.0 \times 7.0^*$



12-ring viewed along  $<100>$



12-ring viewed along  $[001]$



framework viewed along [001]

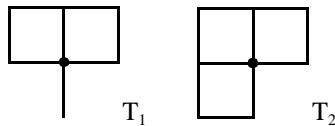
**Idealized cell constants:** tetragonal, I4/mmm,  $a = 14.3\text{\AA}$ ,  $c = 10.4\text{\AA}$

**Coordination sequences and vertex symbols:**

$T_1(16, m)$	4	10	19	30	45	65	90	118	145	175	4·6·4·6·6·8
$T_2(16, 2)$	4	9	17	30	48	68	87	109	142	184	4·4·4·6·6·6

**Secondary building units:** 4

**Loop configuration of T-Atoms:**



**Isotypic framework structures:** \*STA-6<sup>(1)</sup>

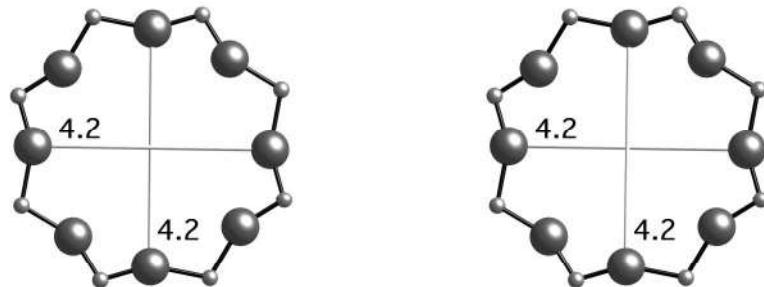
### References:

- (1) Patinec, V., Wright, P.A., Lightfoot, P., Aitken, R.A. and Cox, P.A. *J. Chem. Soc., Dalton Trans.*, 3909-3911 (1999)

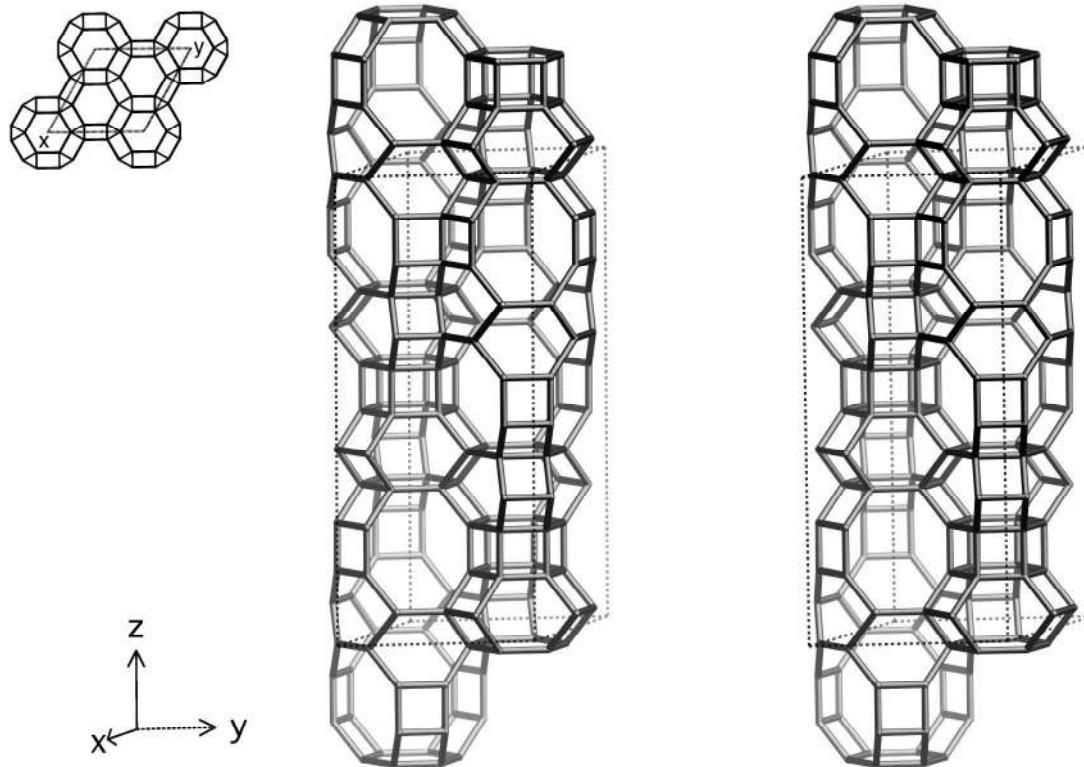
**Crystal chemical data:**  $[(C_{14}H_{34}N_4^{2+})_{1.5} (H_2O)_{2.5}] [Mg_3Al_{13}P_{16} O_{64}]$ -SAS  
 $C_{14}H_{32}N_4 = 1,4,8,11$ -tetramethyl-1,4,8,11-tetraazatetradecane  
tetragonal, P4/mnc,  $a = 14.282\text{\AA}$ ,  $c = 10.249\text{\AA}$  <sup>(1)</sup>

**Framework density:** 15.3 T/1000 $\text{\AA}^3$

**Channels:** [001] 8 4.2 x 4.2\*



8-ring viewed along [001]



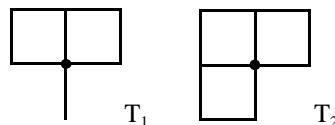
*framework viewed normal to  $[001]$  (top left: projection down  $[001]$ )*

**Idealized cell constants:** trigonal, R $\bar{3}$ m,  $a = 12.9\text{\AA}$ ,  $c = 30.6\text{\AA}$

**Coordination sequences  
and vertex symbols:**  $T_1(36, 1) \quad 4 \quad 10 \quad 20 \quad 33 \quad 50 \quad 71 \quad 95 \quad 124 \quad 158 \quad 197$       4·6·4·8·6·6  
 $T_2(36, 1) \quad 4 \quad 9 \quad 17 \quad 30 \quad 50 \quad 75 \quad 100 \quad 126 \quad 157 \quad 194$       4·4·4·6·6·8

**Secondary building units:** 6

**Loop configuration of  
T-Atoms:**



**Framework description:** ABBCBCCACAAAB sequence of 6-rings

**Isotypic framework  
structures:** \*STA-2<sup>(1)</sup>

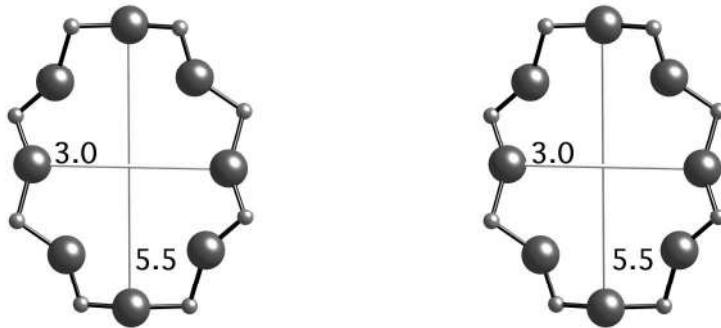
### References:

- (1) Noble, G.W., Wright, P.A. and Kvick, Å. *J. Chem. Soc., Dalton Trans.*, 4485-4490 (1997)

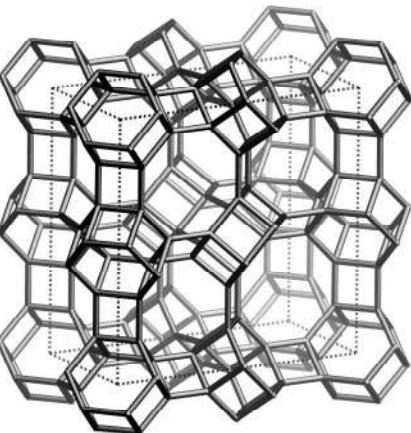
**Crystal chemical data:**  $[(C_{18}H_{34}N_2^{2+})_3(H_2O)_{22.5}] [Mg_{5.4}Al_{30.6}P_{36}O_{144}]$ -SAT  
 $C_{18}H_{34}N_2^{2+} = C_7H_{13}N - (CH_2)_4 - C_7H_{13}N$   
 $C_7H_{13}N = \text{quinuclidine}$   
trigonal,  $R\bar{3}$ ,  $a = 12.726 \text{ \AA}$ ,  $c = 30.939 \text{ \AA}$ <sup>(1)</sup>

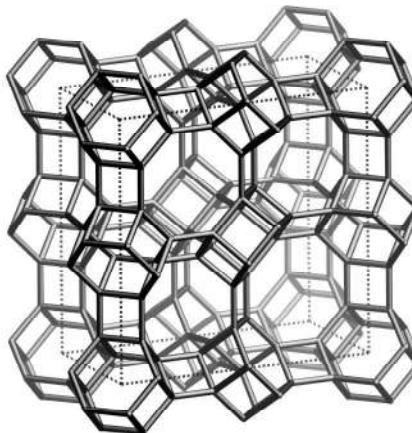
**Framework density:**  $16.6 \text{ T}/1000\text{\AA}^3$

**Channels:**  $\perp [001] 3.0 \times 5.5^{***}$



*8-ring viewed normal to [001]*

 A 3D representation of the framework structure viewed along the [001] direction. The structure is composed of a dense packing of hexagonal and pentagonal rings, forming a complex network of voids and channels. A coordinate system is shown in the top left corner, with the x-axis pointing down, y-axis pointing right, and z-axis pointing diagonally up and to the right.



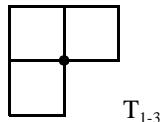
framework viewed along [001]

**Idealized cell constants:** tetragonal, P4/nmm (origin choice 2),  $a = 18.7\text{\AA}$ ,  $c = 9.4\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(16, 1)$	4 9 17 29 45 65 88 113 143 179	4.4.4.8.6.8
	$T_2(16, 1)$	4 9 17 29 45 65 88 114 144 177	4.4.4.8.6.8
	$T_3(16, 1)$	4 9 17 29 45 63 84 112 144 177	4.4.4.8.6.8

**Secondary building units:** 6-6 or 6 or 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Mg-STA-7<sup>(1)</sup>  
Co-STA-7<sup>(1)</sup>  
Zn-STA-7<sup>(1)</sup>

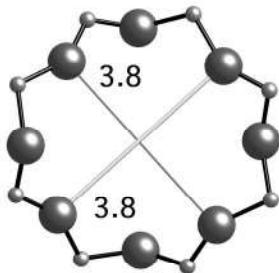
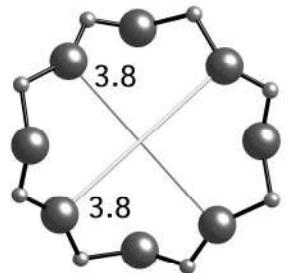
## References:

- (1) Wright, P.A., Maple, M.J., Slawin, A.M.Z., Patinec, V., Aitken, R.A., Welsh, S. and Cox, P.A. *J. Chem. Soc., Dalton Trans.*, 1243-1248 (2000)

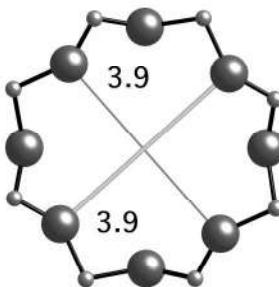
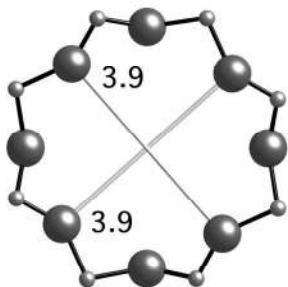
**Crystal chemical data:**  $[(C_{18}H_{42}N_6)_{1.96}(H_2O)_7][Mg_{4.8}Al_{19.2}P_{24}O_{96}]$ -SAV  
 $C_{18}H_{42}N_6 = 1,4,7,10,13,16\text{-hexamethyl-1,4,7,10,13,16-hexaazacyclooctadecane}$   
tetragonal,  $P4/n$ ,  $a = 18.773\text{\AA}$ ,  $c = 9.454\text{\AA}$ <sup>(1)</sup>

**Framework density:**  $14.4 \text{ T}/1000\text{\AA}^3$

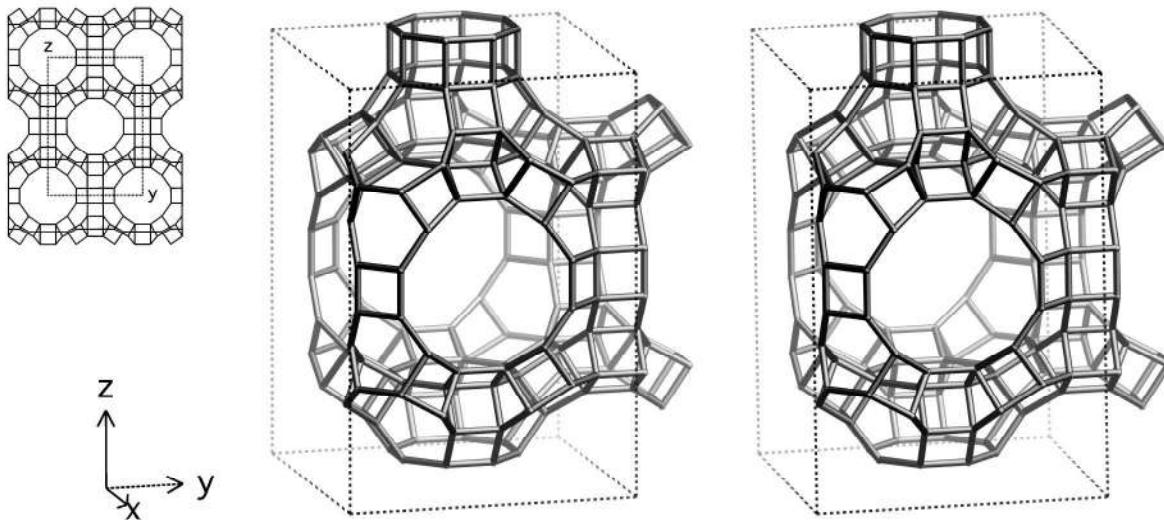
**Channels:**  $<100> \text{ 8 } 3.8 \times 3.8^{**} \leftrightarrow [001] \text{ 8 } 3.9 \times 3.9^*$



8-ring viewed along  $<100>$



8-ring viewed along  $[001]$



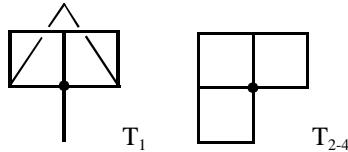
framework viewed along [100] (top left: projection down [100])

**Idealized cell constants:** tetragonal, I4/mmm,  $a = 18.5\text{\AA}$ ,  $c = 27.1\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (32, 1)    4    9    17    27    38    55    78    102    129    157	4·6·4·8·4·8 <sub>7</sub>
	T <sub>2</sub> (32, 1)    4    9    17    28    41    57    77    101    130    162	4·4·4·6·8·12
	T <sub>3</sub> (32, 1)    4    9    17    27    39    56    77    100    126    157	4·4·4·8·6·6 <sub>2</sub>
	T <sub>4</sub> (32, 1)    4    9    17    27    40    59    79    99    126    158	4·4·4·8 <sub>2</sub> ·6 <sub>2</sub> ·8 <sub>4</sub>

**Secondary building units:** 8-8 or 8 or 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**  
 \*UCSB-8Co<sup>(1)</sup>  
 UCSB-8Mg<sup>(1)</sup>  
 UCSB-8Mn<sup>(1)</sup>  
 UCSB-8Zn<sup>(1)</sup>

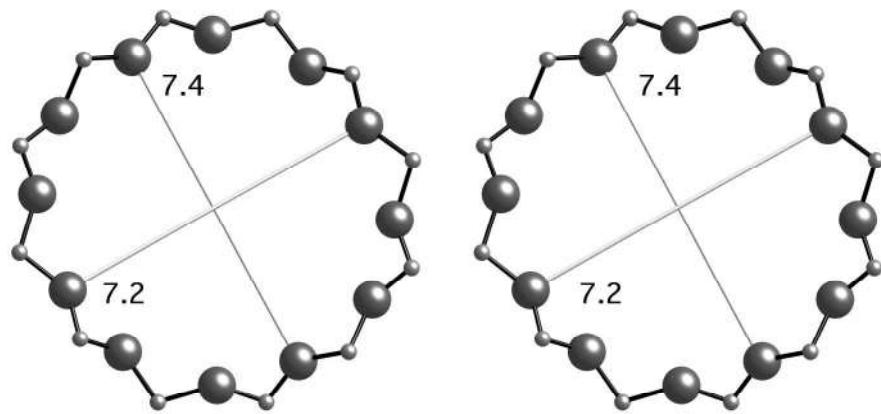
### References:

- (1) Bu, X.H., Feng, P.Y. and Stucky, G.D. *Science*, **278**, 2080-2085 (1997)

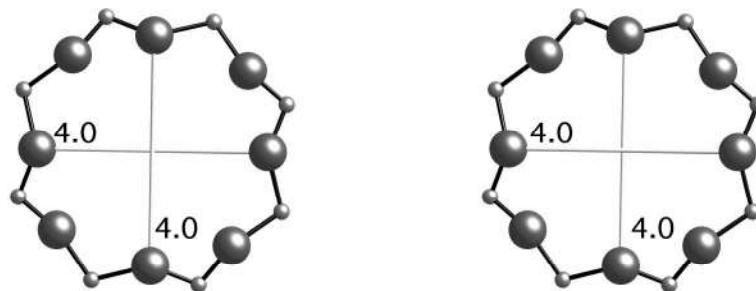
**Crystal chemical data:**  $[(C_9H_{24}N_2^{2+})_{16}] [Al_{32}Co_{32}P_{64} O_{256}]$ -SBE  
 $C_9H_{22}N_2 = 1,9\text{-diaminononane}$   
tetragonal, P4/nnc,  $a = 19.065\text{\AA}$ ,  $c = 27.594\text{\AA}$ <sup>(1)</sup>

**Framework density:** 12.8 T/1000 $\text{\AA}^3$

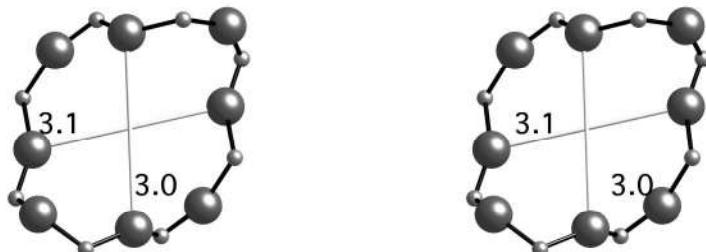
**Channels:**  $<100> \mathbf{12} \ 7.2 \times 7.4^{**} \leftrightarrow [001] \mathbf{8} \ 4.0 \times 4.0^*$



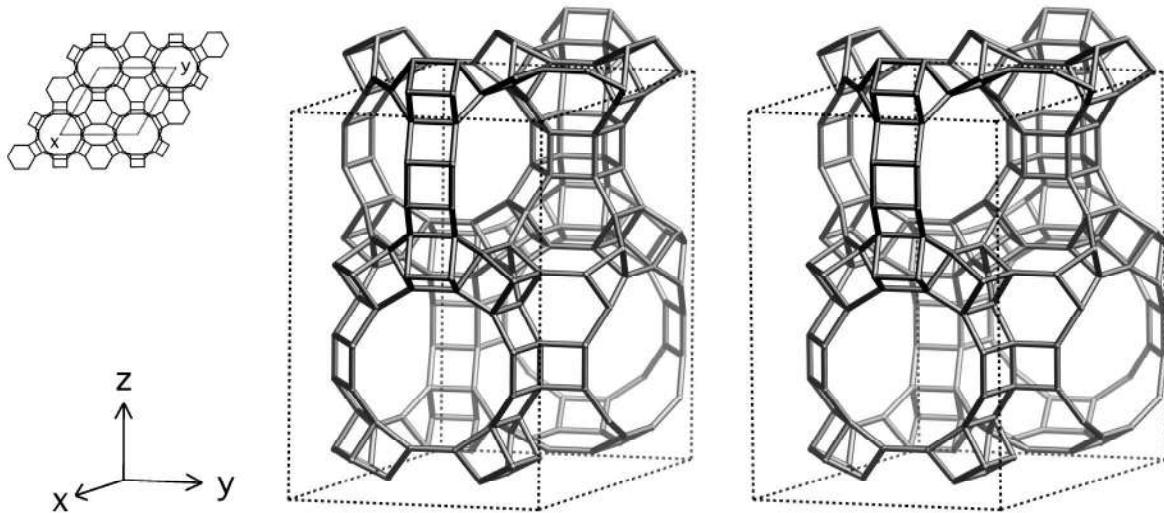
12-ring viewed along  $<100>$



8-ring viewed along  $[001]$  and



second 8-ring along  $[001]$



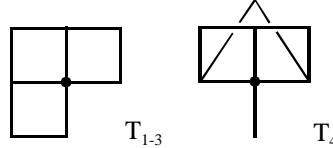
framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** hexagonal, P<sub>6</sub><sub>3</sub>/mmc,  $a = 17.2\text{\AA}$ ,  $c = 27.3\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (24, 1)    4    9    17    28    41    56    75    100    127    157	4·4·4·12 <sub>6</sub> ·6 <sub>2</sub> ·8 <sub>4</sub>
	T <sub>2</sub> (24, 1)    4    9    17    27    39    55    75    100    127    156	4·4·4·8·6·6 <sub>2</sub>
	T <sub>3</sub> (24, 1)    4    9    16    25    38    58    84    111    135    157	4·4·4·6·6·12
	T <sub>4</sub> (24, 1)    4    9    16    24    35    53    77    104    130    153	4·6·4·6·4·8 <sub>7</sub>

**Secondary building units:** 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*UCSB-6GaCo<sup>(1)</sup>  
UCSB-6Co<sup>(1)</sup>  
UCSB-6GaMg<sup>(1)</sup>  
UCSB-6GaZn<sup>(1)</sup>  
UCSB-6Mg<sup>(1)</sup>  
UCSB-6Mn<sup>(1)</sup>  
UCSB-6Zn<sup>(1)</sup>

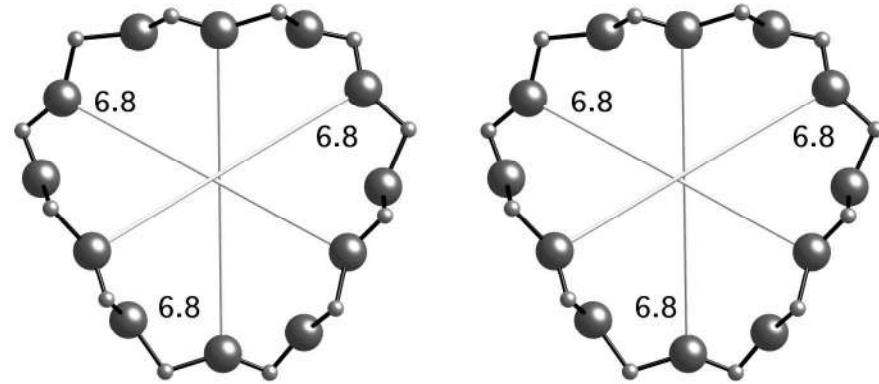
## References:

- (1) Bu, X.H., Feng, P.Y. and Stucky, G.D. *Science*, **278**, 2080-2085 (1997)

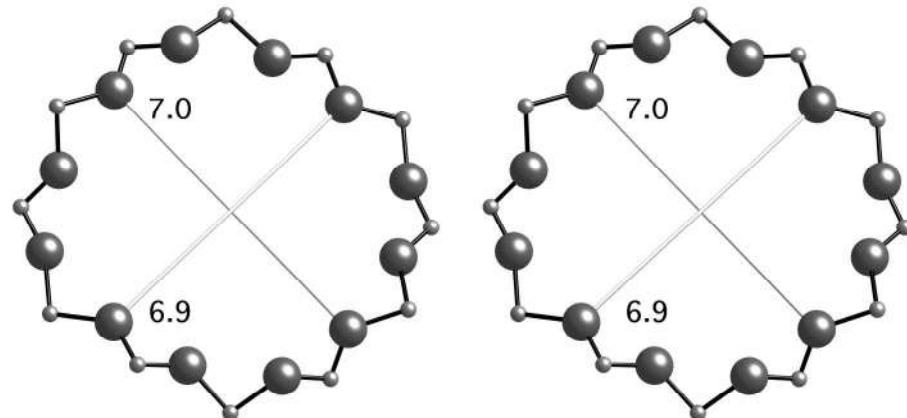
**Crystal chemical data:**  $[(C_9H_{24}N_2)^{2+}]_{12} [Ga_{24}Co_{24}P_{48} O_{192}]$ -SBS  
 $C_9H_{22}N_2 = 1,9\text{-diaminononane}$   
trigonal,  $P\bar{3}1c$ ,  $a = 17.836\text{\AA}$ ,  $c = 27.182\text{\AA}$ <sup>(1)</sup>

**Framework density:** 12.8 T/1000 $\text{\AA}^3$

**Channels:** [001] **12** 6.8 x 6.8\*  $\leftrightarrow \perp$  [001] **12** 6.9 x 7.0\*\*

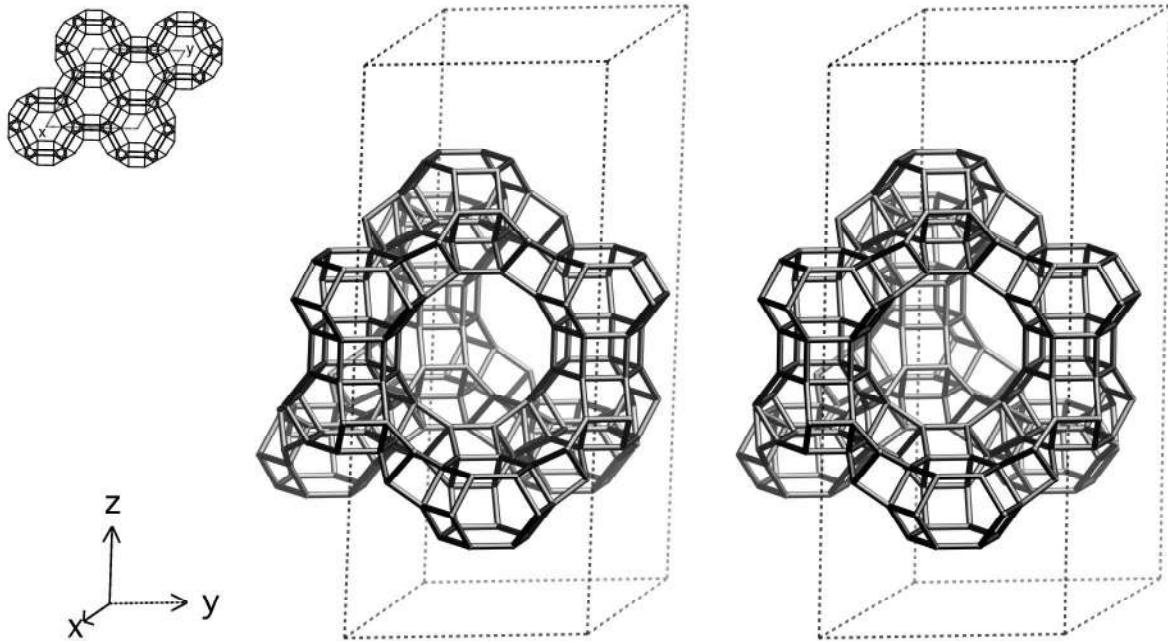


12-ring viewed along [001]



12-ring viewed normal to [001]

See Appendix A for 8-ring along [001]



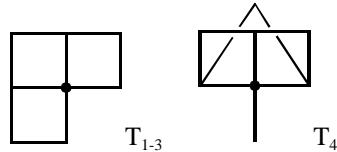
framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** trigonal, R̄3m,  $a = 17.2\text{\AA}$ ,  $c = 41.0\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (36, 1)	4    9    16    25    38    58    84    111    135    157	4·4·4·6·6·12
	T <sub>2</sub> (36, 1)	4    9    17    28    41    56    75    100    127    157	4·4·4·12 <sub>6</sub> ·6 <sub>2</sub> ·8 <sub>4</sub>
	T <sub>3</sub> (36, 1)	4    9    17    27    39    55    75    100    127    156	4·4·4·8·6·6 <sub>2</sub>
	T <sub>4</sub> (36, 1)	4    9    16    24    35    53    77    104    130    153	4·6·4·6·4·8 <sub>7</sub>

**Secondary building units:** 6 or 4·2

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*UCSB-10GaZn<sup>(1)</sup>  
UCSB-10Co<sup>(1)</sup>  
UCSB-10Mg<sup>(1)</sup>  
UCSB-10Zn<sup>(1)</sup>

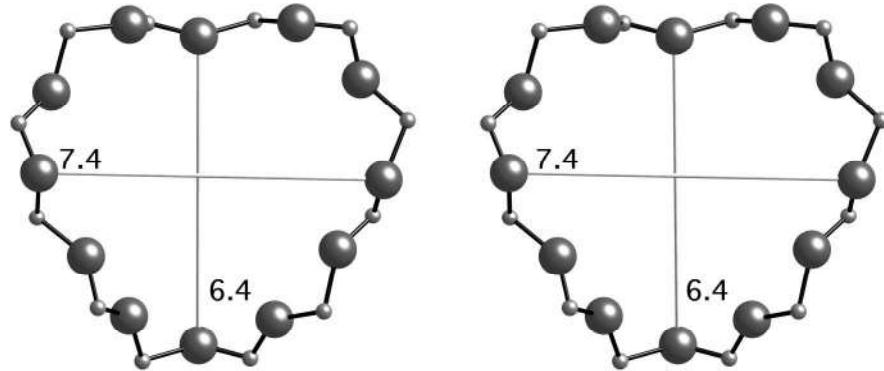
### References:

- (1) Bu, X.H., Feng, P.Y. and Stucky, G.D. *Science*, **278**, 2080-2085 (1997)

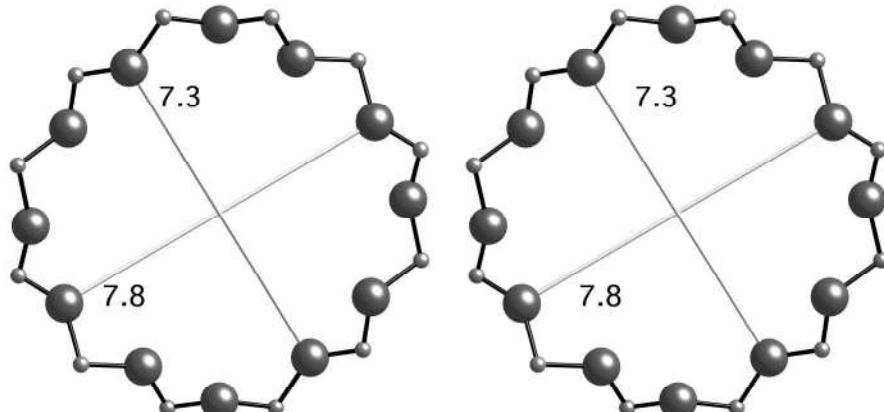
**Crystal chemical data:**  $[(C_{10}H_{26}N_2O_3^{2+})_{18}] [Ga_{36}Zn_{36}P_{72} O_{288}]$ -SBT  
 $C_{10}H_{24}N_2O_3 = 4,7,10\text{-trioxa-1,13-tridecanediamine}$   
trigonal,  $R\bar{3}$ ,  $a = 18.080\text{\AA}$ ,  $c = 41.951\text{\AA}$ <sup>(1)</sup>

**Framework density:**  $12.1 \text{ T}/1000\text{\AA}^3$

**Channels:**  $[001] \mathbf{12} \ 6.4 \times 7.4 \ * \leftrightarrow \perp [001] \mathbf{12} \ 7.3 \times 7.8^{**}$

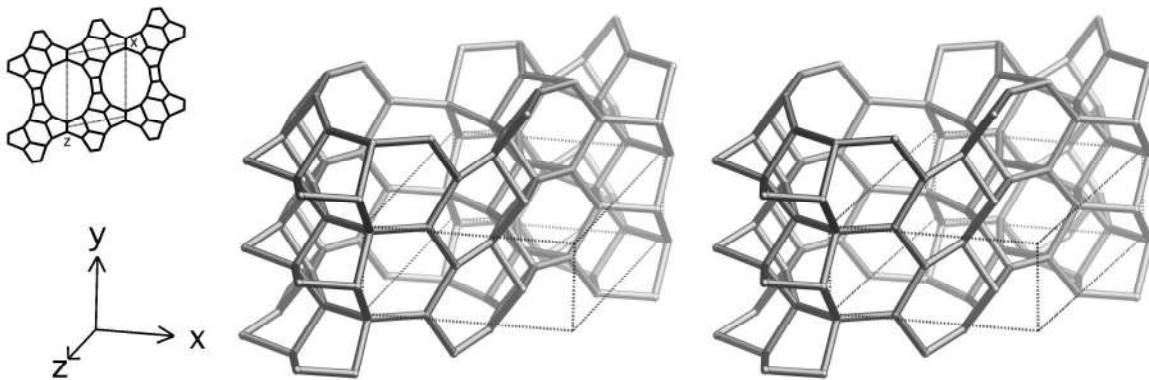


*12-ring viewed along [001]*



*12-ring viewed normal to [001]*

See Appendix A for 8-ring viewed along [102]



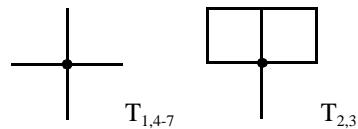
framework viewed normal to [010] (top left: projection down [010])

**Idealized cell constants:** monoclinic, P12<sub>1</sub>1,  $a = 11.4\text{\AA}$ ,  $b = 5.1\text{\AA}$ ,  $c = 13.9\text{\AA}$ ,  $\beta = 100.9^\circ$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (2, 1)    4 12 20 37 62 82 114 142 192 238	5·5·5·5·6·12
	T <sub>2</sub> (2, 1)    4 10 19 35 58 86 108 144 183 233	4·5·4·5·6·12
	T <sub>3</sub> (2, 1)    4 10 20 35 57 83 116 141 181 230	4·5·4·5·12.*
	T <sub>4</sub> (2, 1)    4 12 22 36 56 86 114 150 179 232	5·5·5·5·6.*
	T <sub>5</sub> (2, 1)    4 12 22 37 55 83 120 149 177 227	5·6·6·6·6·6
	T <sub>6</sub> (2, 1)    4 12 24 37 54 80 117 158 184 219	5·6·6·6·6·6
	T <sub>7</sub> (2, 1)    4 12 23 41 58 78 111 154 198 235	5·5·5·5·6·12

**Secondary building units:** 5-2

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*SSZ-48<sup>(1)</sup>

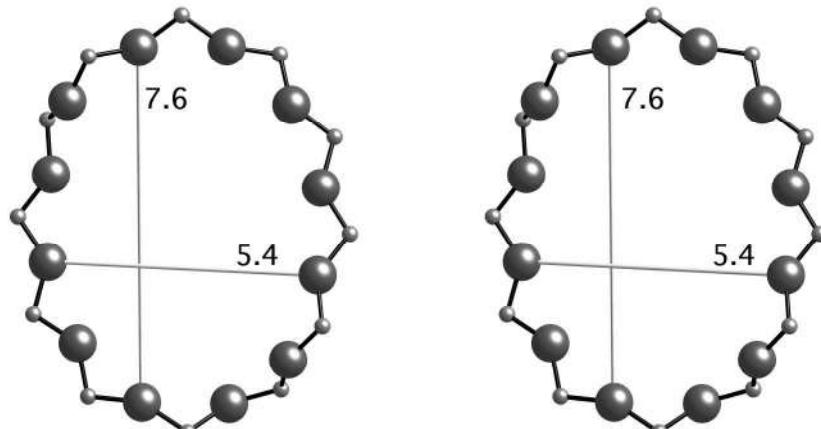
### References:

- (1) Wagner, P., Terasaki, O., Ritsch, S., Nery, J.G., Zones, S.I., Davis, M.E. and Hiraga, K. *J. Phys. Chem. B*, **103**, 8245-8250 (1999)

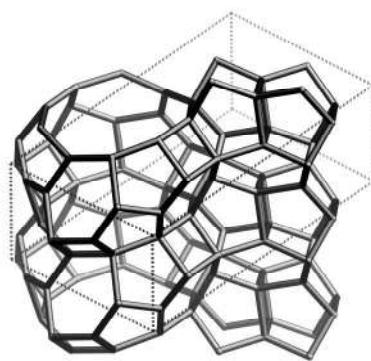
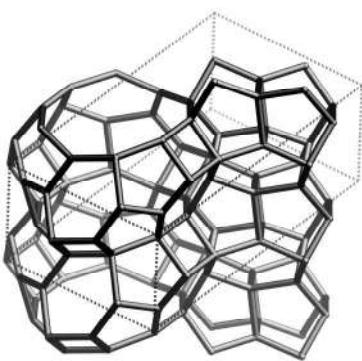
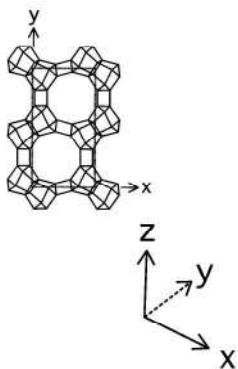
**Crystal chemical data:**  $[\text{Si}_{14} \text{O}_{28}]$ -SFE  
monoclinic,  $P12_11$   
 $a = 11.153\text{\AA}$ ,  $b = 5.002\text{\AA}$ ,  $c = 13.667\text{\AA}$ ,  $\beta = 100.63^\circ$  <sup>(1)</sup>

**Framework density:**  $18.7 \text{ T}/1000\text{\AA}^3$

**Channels:** [010] **12**  $5.4 \times 7.6^*$



*12-ring viewed along [010]*



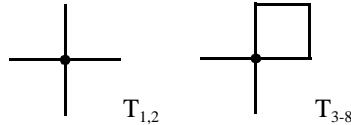
framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** monoclinic, P2<sub>1</sub>/m,  $a = 11.5\text{\AA}$ ,  $b = 21.7\text{\AA}$ ,  $c = 7.2\text{\AA}$ ,  $\beta = 93.2^\circ$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (4, 1)	4 12 20 34 56 87 115 143 176 224	5·5 <sub>2</sub> ·5·6·5·6
	T <sub>2</sub> (4, 1)	4 12 20 34 56 88 115 142 177 225	5·5 <sub>2</sub> ·5·6·5·6
	T <sub>3</sub> (4, 1)	4 11 22 38 57 80 111 148 189 228	4·5·5·6·5·10
	T <sub>4</sub> (4, 1)	4 11 22 39 54 84 110 145 189 234	4·5·5·6·5·10
	T <sub>5</sub> (4, 1)	4 11 23 37 57 82 113 150 184 228	4·5·5·6·5·10
	T <sub>6</sub> (4, 1)	4 11 20 31 58 86 115 142 174 225	4·6 <sub>2</sub> ·5·5·5·5
	T <sub>7</sub> (4, 1)	4 11 19 36 55 82 113 148 181 220	4·6·5·5·5·5
	T <sub>8</sub> (4, 1)	4 11 23 36 59 79 114 147 183 229	4·5·6·5·10

**Secondary building units:** 5-3

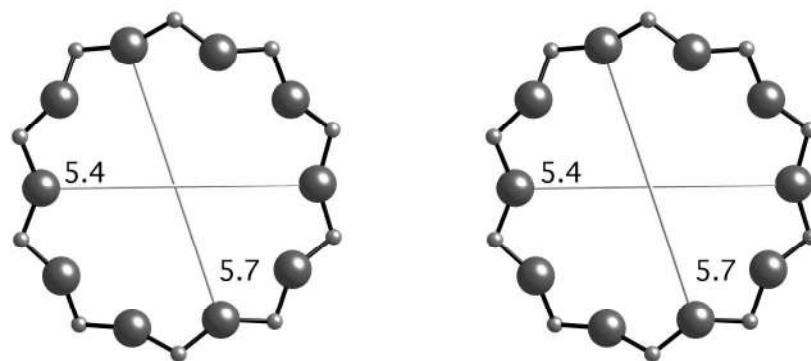
**Loop configuration of  
T-Atoms:**

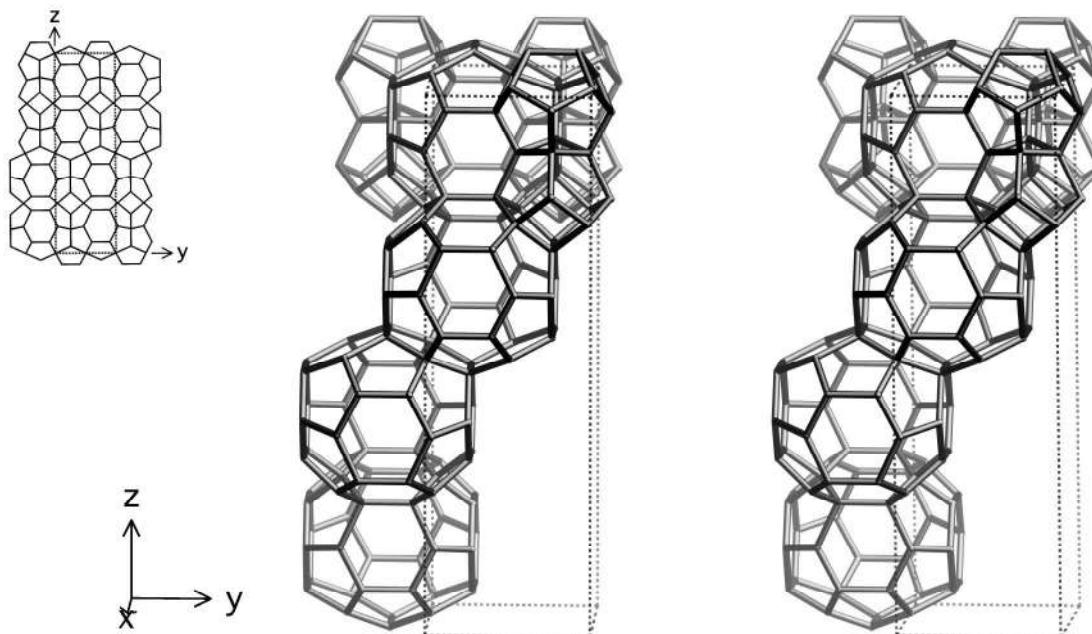


**Isotypic framework  
structures:** \*SSZ-44<sup>(1)</sup>

## References:

- (1) Wagner, P., Zones, S.I., Davis, M.E. and Medrud, R.C. *Angew. Chem., Int. Ed.*, **38**, 1269-1272 (1999)

**Crystal chemical data:** $[\text{Si}_{32} \text{O}_{64}]$ -SFFmonoclinic,  $P2_1/m$ ,  $a = 11.485 \text{ \AA}$ ,  $b = 21.946 \text{ \AA}$ ,  $c = 7.388 \text{ \AA}$ ,  $\beta = 94.70^\circ$  <sup>(1)</sup>**Framework density:**17.2 T/1000 $\text{\AA}^3$ **Channels:**[001] **10** 5.4 x 5.7\**10-ring viewed along [001]*



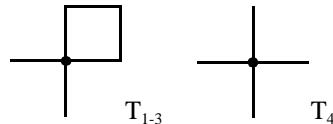
*framework viewed along [100] (top left: projection down [100])*

**Idealized cell constants:** tetragonal, I4<sub>1</sub>/amd (origin choice 2),  $a = 10.3\text{\AA}$ ,  $c = 34.4\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (16, m)    4 11 22 37 62 89 120 155 202 257	4·6·5·5·5·5
	T <sub>2</sub> (16, m)    4 11 21 37 63 86 121 152 196 258	4·6·5·5·5·5
	T <sub>3</sub> (16, 2)    4 11 23 38 62 92 113 159 210 244	4·6·5·5·5·5
	T <sub>4</sub> (16, m)    4 12 24 42 61 87 128 168 205 250	5·6·5·6·5·6

**Secondary building units:** 5-3

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*Sigma-2<sup>(1)</sup>

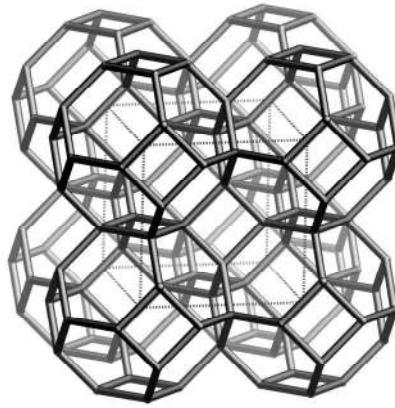
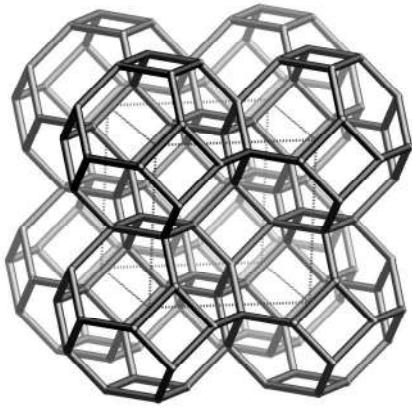
### References:

- (1) McCusker, L.B. *J. Appl. Crystallogr.*, **21**, 305-310 (1988)

**Crystal chemical data:**  $[(C_{10}H_{17}N)_4][Si_{64}O_{128}]$ -SGT  
 $C_{10}H_{17}N$  = 1-aminoadamantane  
tetragonal, I4<sub>1</sub>/amd,  $a = 10.239\text{\AA}$ ,  $c = 34.383\text{\AA}$ <sup>(1)</sup>

**Framework density:** 17.8 T/1000 $\text{\AA}^3$

**Channels:** apertures formed by 6-rings only



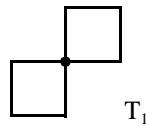
framework viewed along [100]

**Idealized cell constants:** cubic, Im $\bar{3}$ m,  $a = 9.0\text{\AA}$

**Coordination sequences and vertex symbols:**  $T_1(12, \bar{4}2m)4 \quad 10 \quad 20 \quad 34 \quad 52 \quad 74 \quad 100 \quad 130 \quad 164 \quad 202$       4·4·6·6·6·6

**Secondary building units:** 6-2 or 6 or 4

**Loop configuration of T-Atoms:**



**Framework description:** ABC sequence of 6-rings

**Isotypic framework structures:**

\*Sodalite<sup>(1,2)</sup>  
 $[\text{Al-Co-P-O}]\text{-SOD}^{(3)}$   
 $[\text{Al-Ge-O}]\text{-SOD}^{(4)}$   
 $[\text{Be-As-O}]\text{-SOD}^{(5)}$   
 $[\text{Be-P-O}]\text{-SOD}^{(5)}$   
 $[\text{Be-Si-O}]\text{-SOD}^{(6)}$   
 $[\text{Co-Ga-P-O}]\text{-SOD}^{(7)}$   
 $[\text{Ga-Co-P-O}]\text{-SOD}^{(3)}$   
 $[\text{Ga-Ge-O}]\text{-SOD}^{(4)}$   
 $[\text{Ga-Si-O}]\text{-SOD}^{(8)}$   
 $[\text{Zn-As-O}]\text{-SOD}^{(9)}$   
 $[\text{Zn-Ga-As-O}]\text{-SOD}^{(7)}$   
 $[\text{Zn-Ga-P-O}]\text{-SOD}^{(7)}$   
 $[\text{Zn-P-O}]\text{-SOD}^{(9)}$   
 $[\text{Ca}_8(\text{WO}_4)_2[\text{Al}_{12}\text{O}_{24}]]\text{-SOD}^{(10)}$

AlPO-20 plus numerous  
compositional variants<sup>(11,12)</sup>  
Basic sodalite<sup>(13,14)</sup>  
Bicchulite<sup>(15)</sup>  
Danalite<sup>(16)</sup>  
G<sup>(17)</sup>  
Genthelvite<sup>(18)</sup>  
Hauyn<sup>(19)</sup>  
Helvin<sup>(20)</sup>  
Hydroxo sodalite<sup>(21)</sup>  
Nosean<sup>(22)</sup>  
Silica sodalite<sup>(23)</sup>  
TMA sodalite<sup>(24)</sup>  
Tugtupite<sup>(25,26)</sup>

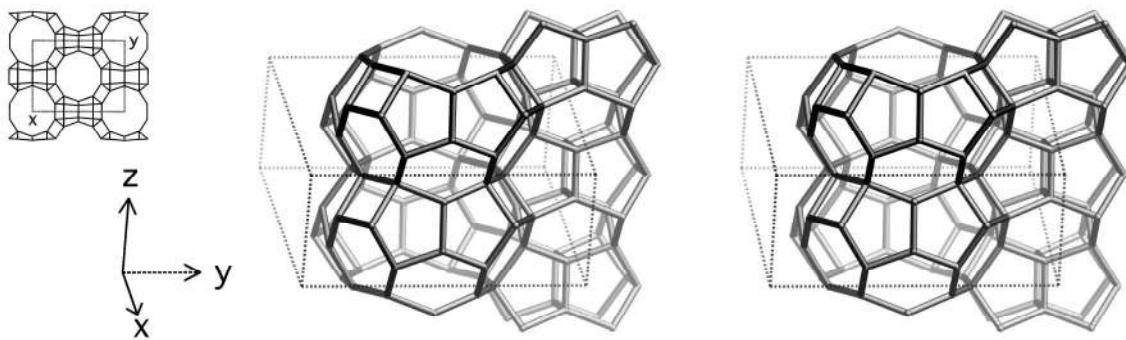
**Crystal chemical data:**  $[\text{Na}^+ \text{Cl}_2] [\text{Al}_6\text{Si}_6 \text{O}_{24}]$ -SOD  
cubic,  $\text{P}\bar{4}3n$ ,  $a = 8.870\text{\AA}$ <sup>(2)</sup>

**Framework density:** 17.2 T/1000 $\text{\AA}^3$

**Channels:** apertures formed by 6-rings only

### References:

- (1) Pauling, L. Z. *Kristallogr.*, **74**, 213-225 (1930)
- (2) Loens, J. and Schulz, H. *Acta Crystallogr.*, **23**, 434-436 (1967)
- (3) Feng, P.Y., Bu, X.H. and Stucky, G.D. *Nature*, **388**, 735-741 (1997)
- (4) Bu, X., Feng, P., Gier, T.E., Zhao, D. and Stucky, G.D. *J. Am. Chem. Soc.*, **120**, 13389-13397 (1998)
- (5) Gier, T.E., Harrison, W.T.A. and Stucky, G.D. *Angew. Chem., Int. Ed.*, **30**, 1169-1171 (1991)
- (6) Dann, S.E. and Weller, M.T. *Inorg. Chem.*, **35**, 555-558 (1996)
- (7) Bu, X., Gier, T.E., Feng, P. and Stucky, G.D. *Microporous and Mesoporous Materials*, **20**, 371-379 (1998)
- (8) McCusker, L.B., Meier, W.M., Suzuki, K. and Shin, S. *Zeolites*, **6**, 388-391 (1986)
- (9) Nenoff, T.M., Harrison, W.T.A., Gier, T.E. and Stucky, G.D. *J. Am. Chem. Soc.*, **113**, 378-379 (1991)
- (10) Depmeier, W. *Acta Crystallogr.*, **C40**, 226-231 (1984)
- (11) Wilson, S.T., Lok, B.M., Messina, C.A., Cannan, T.R. and Flanigen, E.M. *J. Am. Chem. Soc.*, **104**, 1146-1147 (1982)
- (12) Flanigen, E.M., Lok, B.M., Patton, R.L. and Wilson, S.T. In *Proc. 7th Int. Zeolite Conf.*, (eds. Y. Murakami, A. Iijima and J.W. Ward), pp. 103-112 (1986), Kodansha, Tokyo
- (13) Barrer, R.M. and White, E.A.D. *J. Chem. Soc.*, 1267-1278 (1951)
- (14) Hassan, I. and Grundy, H.D. *Acta Crystallogr.*, **C39**, 3-5 (1983)
- (15) Sahl, K. and Chatterjee, N.D. Z. *Kristallogr.*, **146**, 35-41 (1977)
- (16) Glass, J.J., Jahns, R.H. and Stevens, R.E. *Am. Mineral.*, **29**, 163-191 (1944)
- (17) Shishakova, T.N. and Dubinin, M.M. *Izv. Akad. Nauk SSSR*, 1303- (1965)
- (18) Merlin, S. In *Feldspars and Feldspathoids*, (ed. W.L. Brown), pp. 435-470 (1983), Reidel, Dordrecht
- (19) Loehn, J. and Schulz, H. *N. Jb. Miner. Abh.*, **109**, 201-210 (1968)
- (20) Glass, J.J., Jahns, R.H. and Stevens, R.E. *Am. Mineral.*, **29**, 163-191 (1944)
- (21) Felsche, J., Luger, S. and Baerlocher, Ch. *Zeolites*, **6**, 367-372 (1986)
- (22) Schulz, H. and Saalfeld, H. *Tschermaks Min. Petr. Mitt.*, **10**, 225-232 (1965)
- (23) Bibby, D.M. and Dale, M.P. *Nature*, **317**, 157-158 (1985)
- (24) Baerlocher, Ch. and Meier, W.M. *Helv. Chim. Acta*, **52**, 1853-1860 (1969)
- (25) Sorensen, H. *Am. Mineral.*, **48**, 1178 (1963)
- (26) Hassan, I. and Grundy, H.D. *Can. Mineral.*, **29**, 385-390 (1991)



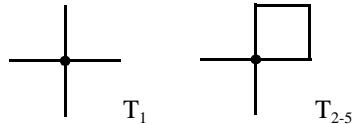
framework viewed normal to [001] (top left: projection down [001])

**Idealized cell constants:** monoclinic, C2/m,  $a = 14.1\text{\AA}$ ,  $b = 18.2\text{\AA}$ ,  $c = 7.5\text{\AA}$ ,  $\beta = 99.0^\circ$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (8, 1)	4 12 20 34 56 88 114 143 173 224	5·5 <sub>2</sub> ·5·6·5·6
	T <sub>2</sub> (8, 1)	4 11 22 39 55 82 111 149 188 223	4·5·5·6·5·10
	T <sub>3</sub> (8, 1)	4 11 23 36 59 80 113 147 183 227	4·5·5·6·5·10
	T <sub>4</sub> (4, m)	4 11 19 36 54 84 110 146 179 226	4·6·5·5·5·5
	T <sub>5</sub> (4, m)	4 11 20 31 58 84 117 137 174 229	4·6 <sub>2</sub> ·5·5·5·5

**Secondary building units:** 5-3

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*SSZ-35<sup>(1)</sup>  
ITQ-9<sup>(2)</sup>

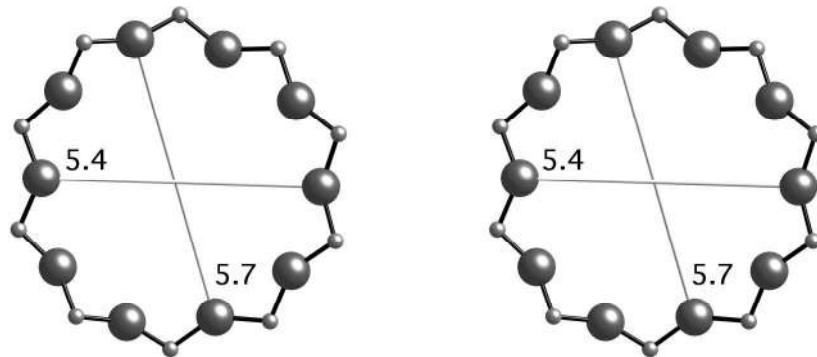
### References:

- (1) Wagner, P., Zones, S.I., Davis, M.E. and Medrud, R.C. *Angew. Chem., Int. Ed.*, **38**, 1269-1272 (1999)
- (2) Villaescusa, L.A., Barrett, P.A. and Camblor, M.A. *Chem. Commun.*, **21**, 2329-2330 (1998)

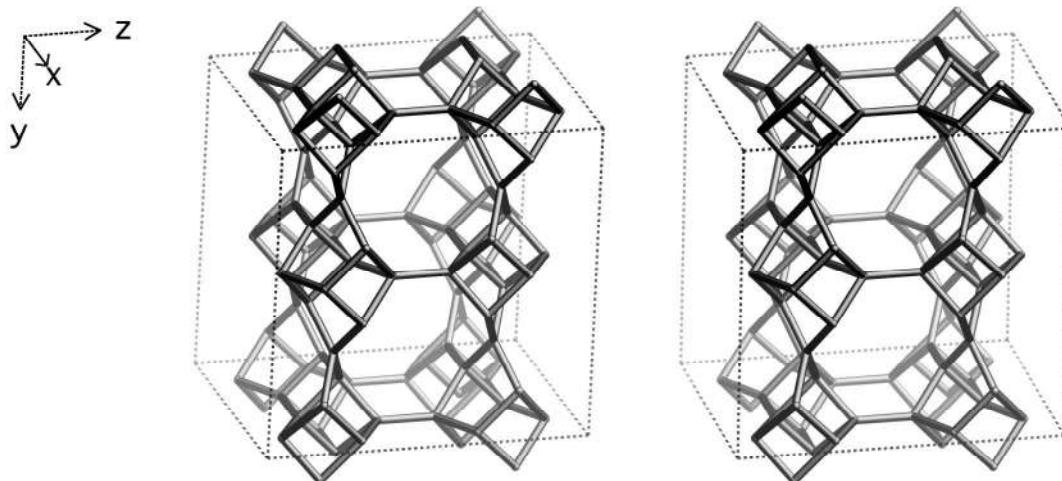
**Crystal chemical data:**  $[\text{Si}_{16} \text{O}_{32}]$ -STF  
triclinic,  $\text{P}\bar{1}$ ,  $a = 11.411\text{\AA}$ ,  $b = 11.527\text{\AA}$ ,  $c = 7.377\text{\AA}$   
 $\alpha = 94.66^\circ$ ,  $\beta = 96.21^\circ$ ,  $\gamma = 104.89^\circ$ <sup>(1)</sup>  
(Relationship to unit cell of Framework Type:  $V' = V/2$ )

**Framework density:**  $17.3 \text{ T}/1000\text{\AA}^3$

**Channels:** [001] **10**  $5.4 \times 5.7^*$



*10-ring viewed along [001]*



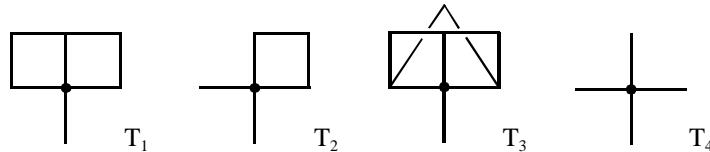
framework viewed along [100]

**Idealized cell constants:** orthorhombic, Fmmm,  $a = 13.5\text{\AA}$ ,  $b = 17.8\text{\AA}$ ,  $c = 17.9\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(32, 1)$	4 10 20 34 57 82 103 138 181 220	4·5·4·6·5·6 <sub>2</sub>
	$T_2(16, m)$	4 11 20 36 57 78 109 140 176 222	4·8·5·8 <sub>2</sub> ·5·8 <sub>2</sub>
	$T_3(16, m)$	4 9 17 35 57 77 103 138 188 225	4·5·4·5·4·8
	$T_4(8, 222)$	4 12 18 34 58 82 112 130 172 228	5 <sub>2</sub> ·5 <sub>2</sub> ·6·6·10·10

**Secondary building units:** 4-4=1

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*Stilbite<sup>(1-3)</sup>  
Barrerite<sup>(4)</sup>  
Stellerite<sup>(5)</sup>  
Synthetic barrerite<sup>(6)</sup>  
Synthetic stellerite<sup>(6)</sup>  
Synthetic stilbite<sup>(7)</sup>

**Alternate designation:** Desmine (discredited)  
Epidesmin (obsolete)

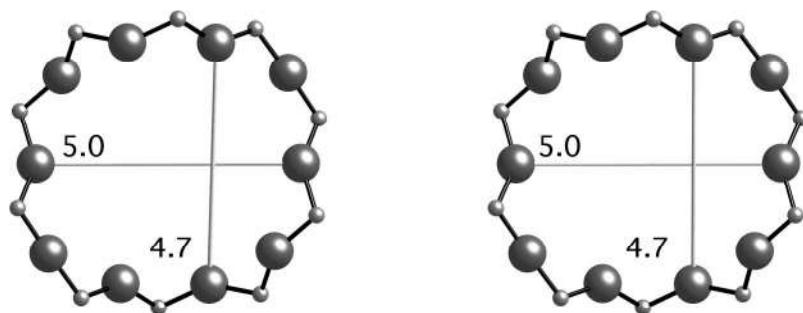
### References:

- (1) Galli, E. and Gottardi, G. *Miner. Petrogr. Acta*, **12**, 1-10 (1966)
- (2) Slaughter, M. *Am. Mineral.*, **55**, 387-397 (1970)

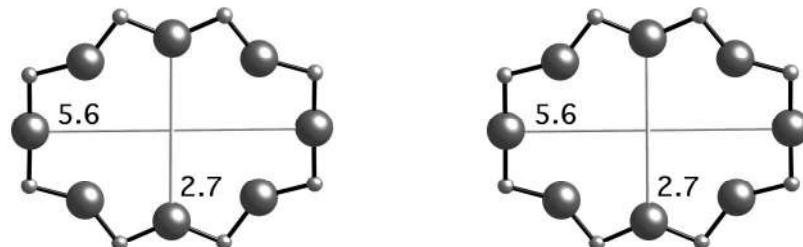
**Crystal chemical data:**  $[\text{Na}^+ \text{Ca}^{2+}_8 (\text{H}_2\text{O})_{56}] [\text{Al}_{20}\text{Si}_{52} \text{O}_{144}]$ -STI  
 monoclinic, C12/m1,  $a = 13.64\text{\AA}$ ,  $b = 18.24\text{\AA}$ ,  $c = 11.27\text{\AA}$ ,  $\beta = 128.0^\circ$ <sup>(3)</sup>  
 (Relationship to unit cell of Framework Type:  
 $\mathbf{a}' = \mathbf{a}$ ,  $\mathbf{b}' = \mathbf{b}$ ,  $\mathbf{c}' = \mathbf{c}/2\sin(\beta)$   
 or, as vectors,  $\mathbf{a}' = \mathbf{a}$ ,  $\mathbf{b}' = \mathbf{b}$ ,  $\mathbf{c}' = (\mathbf{c} - \mathbf{a})/2$ )

**Framework density:** 16.3 T/1000 $\text{\AA}^3$

**Channels:** [100] **10** 4.7 x 5.0\*  $\leftrightarrow$  [001] **8** 2.7 x 5.6\*



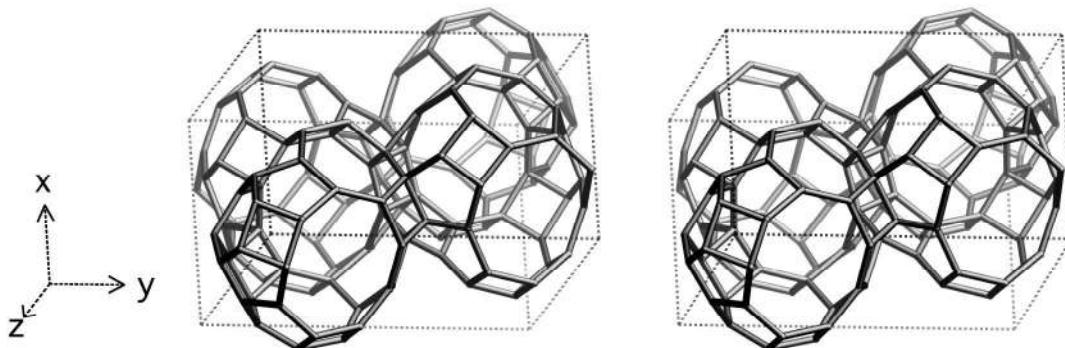
10-ring viewed along [100]



8-ring along [001]

#### References (cont.):

- (3) Galli, E. *Acta Crystallogr.*, **B27**, 833-841 (1971)
- (4) Galli, E. and Alberti, A. *Bull. Soc. fr. Minéral. Cristallogr.*, **98**, 331-340 (1975)
- (5) Galli, E. and Alberti, A. *Bull. Soc. fr. Minéral. Cristallogr.*, **98**, 11-18 (1975)
- (6) Ghobarkar, H., Schaef, O. and Guth, U. *J. Solid State Chem.*, **142**, 451-454 (1999)
- (7) Ghobarkar, H. and Schaef, O. *J. Phys. D: Appl. Phys.*, **31**, 3172-3176 (1998)



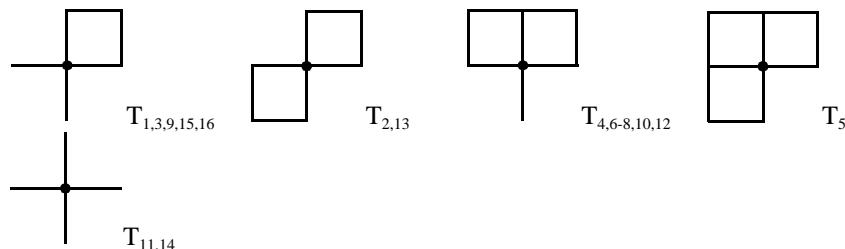
framework viewed normal to [100]

**Idealized cell constants:** monoclinic, P12<sub>1</sub>/n1,  $a = 13.1\text{\AA}$ ,  $b = 21.9\text{\AA}$ ,  $c = 13.6\text{\AA}$ ,  $\beta = 102.9^\circ$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (4, 1)	4 11 22 36 55 84 111 142 179 233	4·5·5·7·6·9
	T <sub>2</sub> (4, 1)	4 10 23 37 57 78 108 146 187 225	4·4·5·9·6·7
	T <sub>3</sub> (4, 1)	4 11 20 35 54 84 108 142 178 225	4·6·5·6·5·7
	T <sub>4</sub> (4, 1)	4 10 19 32 55 81 109 141 174 223	4·5·4·6·5·5
	T <sub>5</sub> (4, 1)	4 9 21 38 57 78 104 144 195 220	4·4·4·9·5·7
	T <sub>6</sub> (4, 1)	4 10 20 40 57 78 105 143 191 226	4·5·4·7·5·9
	T <sub>7</sub> (4, 1)	4 10 20 35 57 77 108 139 184 230	4·6·4·6·5·7
	T <sub>8</sub> (4, 1)	4 10 19 35 54 77 108 146 174 215	4·5·4·9·5·6
	T <sub>9</sub> (4, 1)	4 11 19 34 53 81 115 140 173 219	4·6·5·5·5·5
	T <sub>10</sub> (4, 1)	4 10 21 31 51 83 115 134 172 216	4·6·4·6·5·5
	T <sub>11</sub> (4, 1)	4 12 20 31 50 88 117 137 167 221	5·5·2·5·6·6·6
	T <sub>12</sub> (4, 1)	4 10 18 35 59 77 107 150 179 221	4·5·4·9·5·5
	T <sub>13</sub> (4, 1)	4 10 20 38 55 75 104 149 185 218	4·4·5·6·5·9
	T <sub>14</sub> (4, 1)	4 12 21 33 58 86 112 141 174 226	5·5·5·6·5·7
	T <sub>15</sub> (4, 1)	4 11 21 33 55 78 111 145 175 217	4·5·5·6·5·9
	T <sub>16</sub> (4, 1)	4 11 20 32 54 89 115 138 172 221	4·6·5·5·5·5

**Secondary building units:** 5-1

**Loop configuration of  
T-Atoms:**



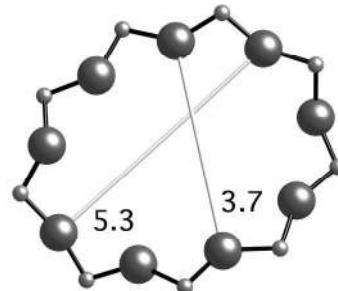
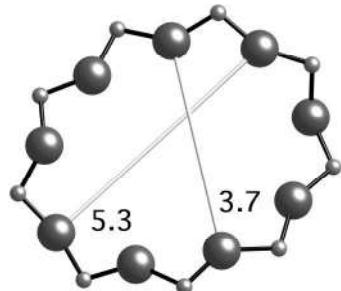
**Isotypic framework  
structures:**

\*SSZ-23<sup>(1)</sup>

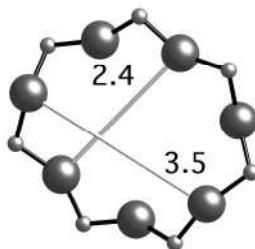
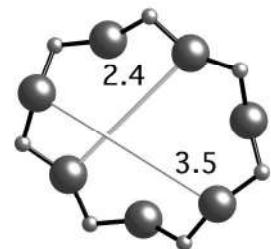
**Crystal chemical data:**  $[(C_{13}H_{24}N^+)^{4.1} F_{3.3}(OH)^{-0.8}[Si_{64}O_{128}]]\text{-STT}$   
 $C_{13}H_{24}N^+$  = N,N,N-trimethyl-1-adamant ammonium  
monoclinic,  $P12_1/n1$   
 $a = 12.959\text{\AA}$ ,  $b = 21.792\text{\AA}$ ,  $c = 13.598\text{\AA}$ ,  $\beta = 101.85^\circ$  <sup>(1)</sup>

**Framework density:**  $17.0 \text{ T}/1000\text{\AA}^3$

**Channels:** [101] **9**  $3.7 \times 5.3^* \leftrightarrow$  [001] **7**  $2.4 \times 3.5^*$



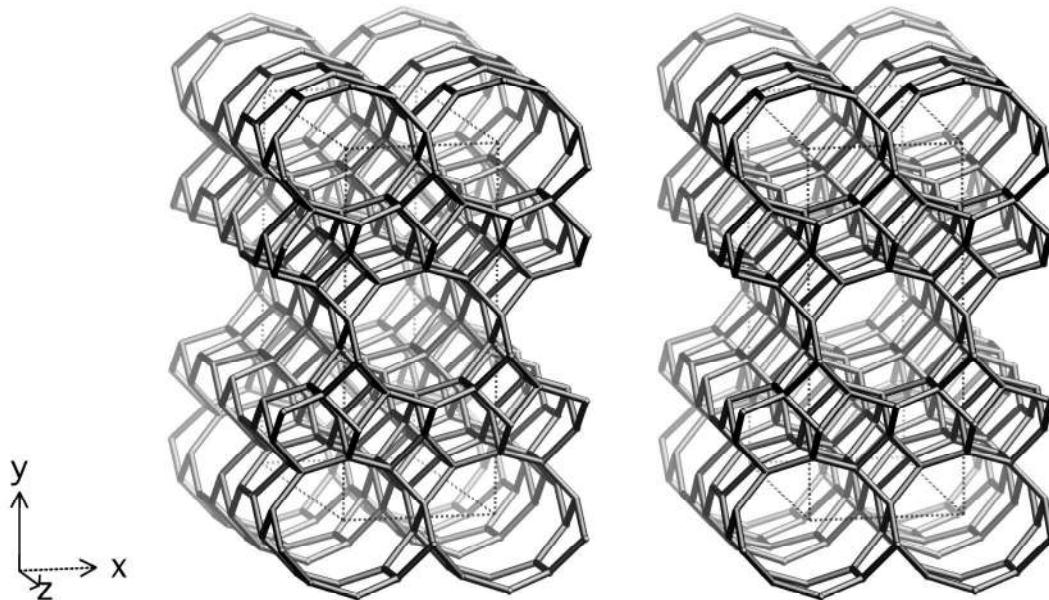
9-ring viewed along [101]



7-ring viewed along [001]

## References:

- (1) Cambor, M.A., Díaz-Cabañas, M.-J., Pérez-Pariente, J., Teat, S.J., Clegg, W., Shannon, I.J., Lightfoot, P., Wright, P.A. and Morris, R.E. *Angew. Chem., Int. Ed.*, **37**, 2122-2126 (1998)

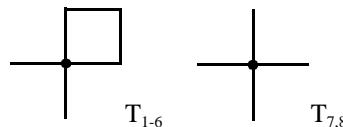
**TER****Framework Type****Cmcm***framework viewed along [001]*

**Idealized cell constants:** orthorhombic, Cmcm,  $a = 9.8\text{\AA}$ ,  $b = 23.6\text{\AA}$ ,  $c = 20.2\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (16, 1)    4    11    19    35    57    83    113    137    184    231	4·5·5·6·5·6 <sub>2</sub>
	T <sub>2</sub> (16, 1)    4    11    21    35    58    87    103    144    188    227	4·5·5·6 <sub>2</sub> ·5·10 <sub>3</sub>
	T <sub>3</sub> (8, m)    4    11    22    39    62    82    104    142    178    225	4·10 <sub>6</sub> ·5·6 <sub>3</sub> ·5·6 <sub>3</sub>
	T <sub>4</sub> (8, m)    4    11    21    41    61    77    107    134    186    232	4·5 <sub>2</sub> ·5·10 <sub>4</sub> ·5·10 <sub>4</sub>
	T <sub>5</sub> (8, m)    4    11    19    34    56    78    116    152    184    208	4·5 <sub>2</sub> ·5·6·5·6
	T <sub>6</sub> (8, m)    4    11    21    33    53    80    112    155    187    214	4·10 <sub>2</sub> ·5·6 <sub>3</sub> ·5·6 <sub>3</sub>
	T <sub>7</sub> (8, m)    4    12    20    35    55    81    119    142    182    216	5·6·5·6·5 <sub>2</sub> ·10 <sub>2</sub>
	T <sub>8</sub> (8, m)    4    12    23    32    49    86    124    147    167    219	5·6 <sub>2</sub> ·5·6 <sub>2</sub> ·10·10 <sub>4</sub>

**Secondary building units:** 2-6-2

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*Terranovaite<sup>(1)</sup>

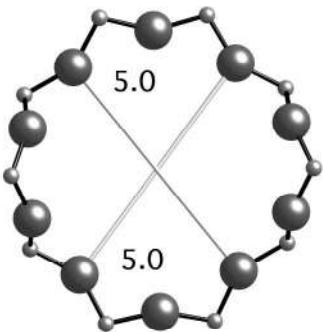
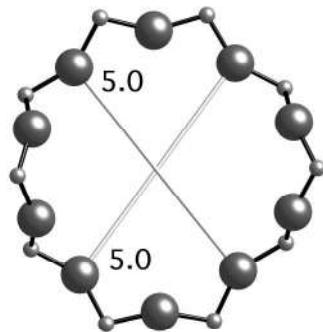
### References:

- (1) Galli, E., Quartieri, S., Vezzalini, G., Alberti, A. and Franzini, M. *Am. Mineral.*, **82**, 423-429 (1997)

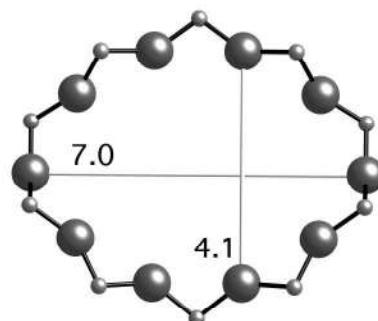
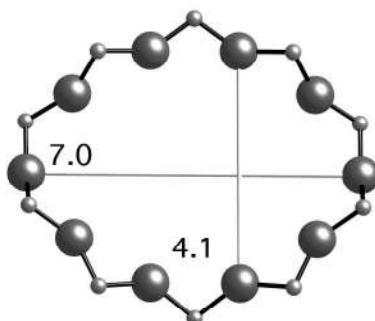
**Crystal chemical data:**  $[\text{Na}^{+}_{4.2}\text{K}^{+}_{0.2}\text{Mg}^{2+}_{0.2}\text{Ca}^{2+}_{3.7}(\text{H}_2\text{O})_{29}] [\text{Al}_{12.3}\text{Si}_{67.7}\text{O}_{160}]$ -TER  
orthorhombic, Cmcm,  $a = 9.747\text{\AA}$ ,  $b = 23.880\text{\AA}$ ,  $c = 20.068\text{\AA}$ <sup>(1)</sup>

**Framework density:** 17.1 T/1000 $\text{\AA}^3$

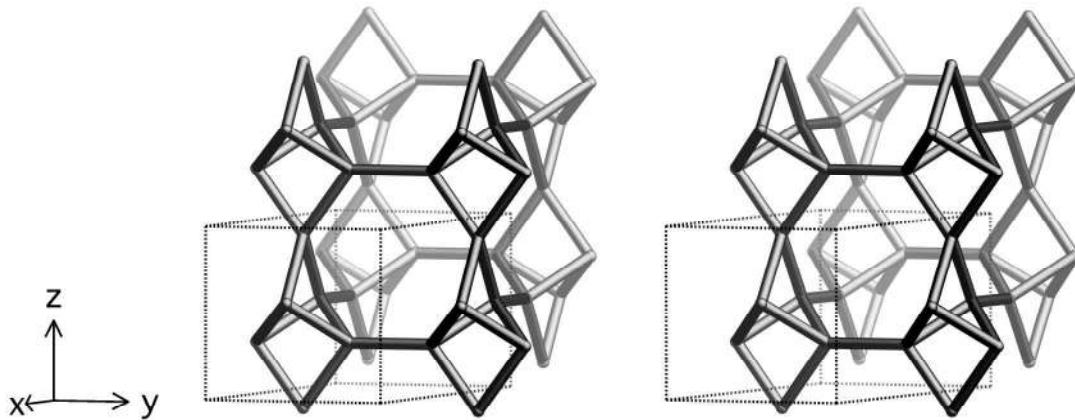
**Channels:** [100] **10** 5.0 x 5.0\*  $\leftrightarrow$  [001] **10** 4.1 x 7.0\*



10-ring viewed along [100]



10-ring viewed along [001]



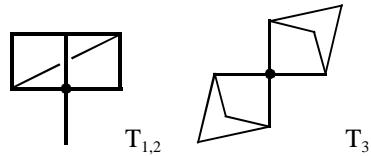
framework viewed normal to [001]

**Idealized cell constants:** orthorhombic, Pmma,  $a = 14.0\text{\AA}$ ,  $b = 7.0\text{\AA}$ ,  $c = 6.5\text{\AA}$

<b>Coordination sequences</b>	T <sub>1</sub> (4, m)	4	9	19	35	52	72	100	131	163	201	4·8 <sub>3</sub> ·4·8 <sub>3</sub> ·4 <sub>2</sub> ·8 <sub>4</sub>
<b>and vertex symbols:</b>	T <sub>2</sub> (4, m)	4	9	19	33	50	74	100	129	165	201	4·8 <sub>3</sub> ·4·8 <sub>3</sub> ·4 <sub>2</sub> ·8 <sub>4</sub>
	T <sub>3</sub> (2, mm2)	4	8	18	34	50	68	100	130	160	204	4 <sub>2</sub> ·4 <sub>2</sub> ·8 <sub>4</sub> ·8 <sub>4</sub> ·8 <sub>4</sub>

**Secondary building units:** 4=1

**Loop configuration of T-Atoms:**



**Isotypic framework structures:**

\*Thomsonite<sup>(1-3)</sup>  
 [Al-Co-P-O]-THO<sup>(4)</sup>  
 [Ga-Co-P-O]-THO<sup>(4)</sup>  
 Na-V<sup>(5)</sup>  
 Synthetic thomsonite<sup>(6)</sup>

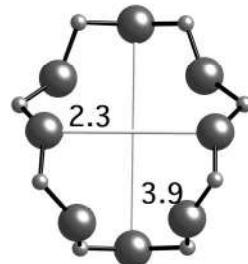
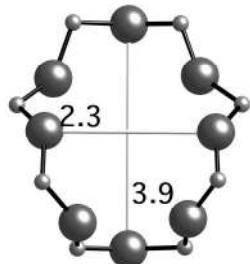
## References:

- (1) Taylor, W.H., Meek, C.A. and Jackson, W.W. *Z. Kristallogr.*, **84**, 373-398 (1933)
- (2) Alberti, A., Vezzalini, G. and Tazzoli, V. *Zeolites*, **1**, 91-97 (1981)
- (3) Pluth, J.J., Smith, J.V. and Kvick, A. *Zeolites*, **5**, 74-80 (1985)
- (4) Feng, P.Y., Bu, X.H. and Stucky, G.D. *Nature*, **388**, 735-741 (1997)
- (5) Barrer, R.M., Baynham, J.W., Bultitude, F.W. and Meier, W.M. *J. Chem. Soc.*, 195-208 (1959)
- (6) Ghobarkar, H. and Schaeff, O. *Cryst. Res. Technol.*, **32**, 653-657 (1997)

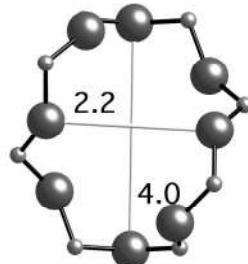
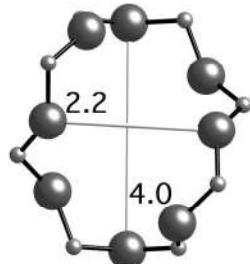
**Crystal chemical data:**  $[\text{Na}^+ \cdot \text{Ca}^{2+}]_8 (\text{H}_2\text{O})_{24} [\text{Al}_{20} \text{Si}_{20} \text{O}_{80}]$ -THO  
orthorhombic, Pncc,  $a = 13.088\text{\AA}$ ,  $b = 13.052\text{\AA}$ ,  $c = 13.229\text{\AA}$ <sup>(3)</sup>  
(Relationship to unit cell of Framework Type:  $a' = a$ ,  $b' = 2b$ ,  $c' = 2c$ )

**Framework density:** 17.7 T/1000 $\text{\AA}^3$

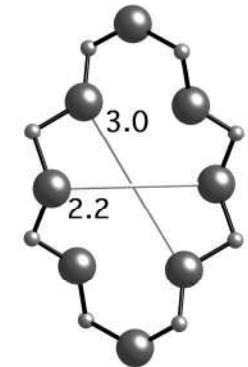
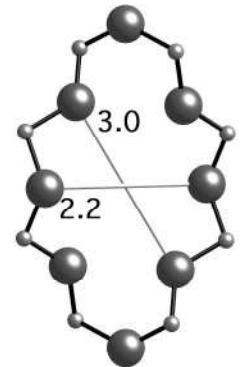
**Channels:** [100] 8 2.3 x 3.9\*  $\leftrightarrow$  [010] 8 2.2 x 4.0\*  $\leftrightarrow$  [001] 8 2.2 x 3.0\* (variable due to considerable flexibility of framework)



8-ring viewed along [100]

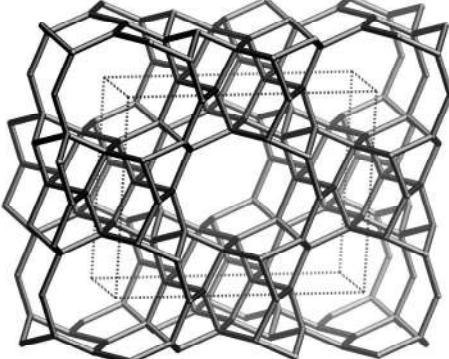


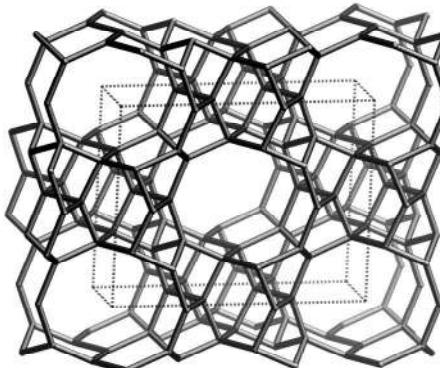
8-ring viewed along [010]



8-ring along [001] (variable)

**TON****Framework Type****Cmcm**



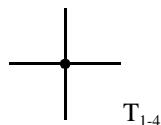
*framework viewed along [001]*

**Idealized cell constants:** orthorhombic, Cmcm,  $a = 14.1\text{\AA}$ ,  $b = 17.8\text{\AA}$ ,  $c = 5.3\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (8, m) 4 12 24 40 61 96 133 163 204 262	5 <sub>2</sub> ·6 <sub>2</sub> ·6·6 <sub>2</sub> ·6·6 <sub>2</sub>
	T <sub>2</sub> (8, m) 4 12 23 43 66 91 128 169 214 258	5·5·5·5·6·10 <sub>2</sub>
	T <sub>3</sub> (4, m2m) 4 12 22 41 68 97 118 166 224 258	5·5·5·5·6 <sub>2</sub> ·10 <sub>2</sub>
	T <sub>4</sub> (4, m2m) 4 12 22 39 66 95 130 158 208 270	5·5·5·5·6 <sub>2</sub> ·*

**Secondary building units:** 6 or 5-1

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Theta-1<sup>(1,2)</sup>  
ISI-1<sup>(3)</sup>  
KZ-2<sup>(4)</sup>  
NU-10<sup>(5)</sup>  
ZSM-22<sup>(6,7)</sup>

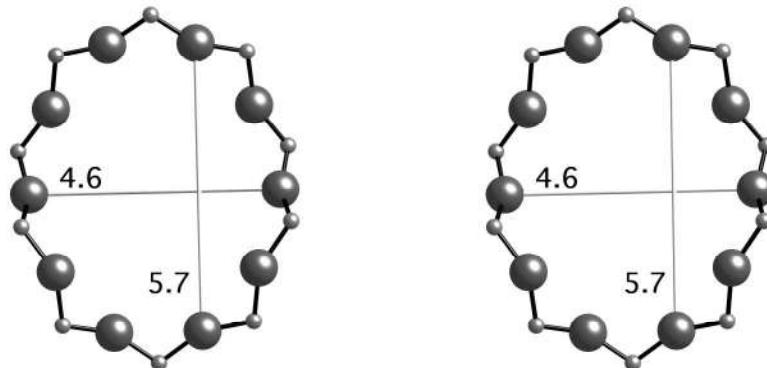
**References:**

- (1) Barri, S.A.I., Smith, G.W., White, D. and Young, D. *Nature*, **312**, 533-534 (1984)
- (2) Highcock, R.M., Smith, G.W. and Wood, D. *Acta Crystallogr.*, **C41**, 1391-1394 (1985)
- (3) Kozo, T. and Noboru, K. *E. Patent A-170,003* (1986)
- (4) Parker, L.M. and Bibby, D.M. *Zeolites*, **3**, 8-11 (1983)
- (5) Araya, A. and Lowe, B.M. *Zeolites*, **4**, 280-286 (1984)
- (6) Kokotailo, G.T., Schlenker, J.L., Dwyer, F.G. and Valyocsik, E.W. *Zeolites*, **5**, 349-351 (1985)
- (7) Marler, B. *Zeolites*, **7**, 393-397 (1987)

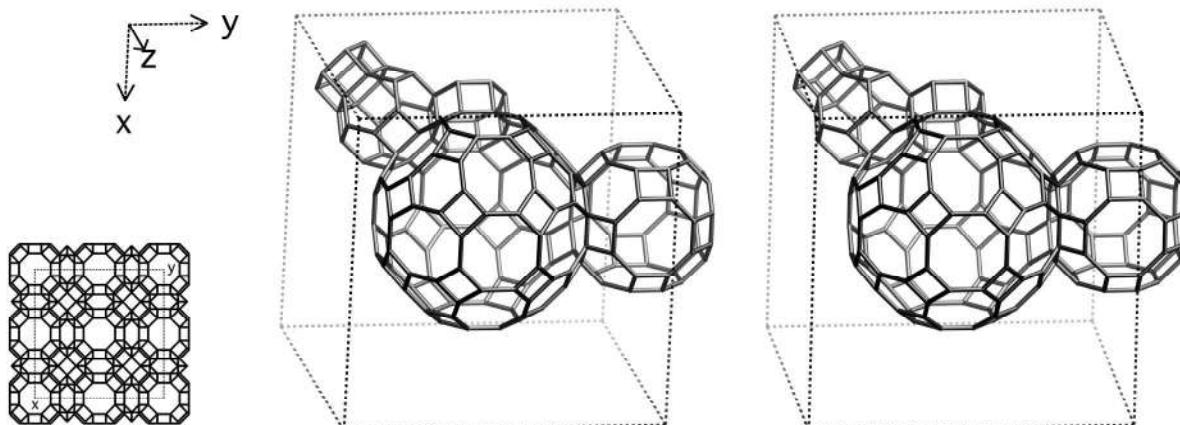
**Crystal chemical data:**  $[\text{Na}^+ \text{n} (\text{H}_2\text{O})_4] [\text{Al}_n \text{Si}_{24-n} \text{O}_{48}]$ -TON,  $n < 2$   
orthorhombic,  $\text{Cmc}2_1$ ,  $a = 13.859\text{\AA}$ ,  $b = 17.420\text{\AA}$ ,  $c = 5.038\text{\AA}$  <sup>(7)</sup>

**Framework density:** 19.7 T/1000 $\text{\AA}^3$

**Channels:** [001] **10** 4.6 x 5.7\*



10-ring viewed along [001]

**TSC****Framework Type****Fm $\bar{3}$ m**

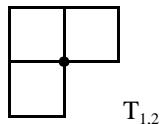
*part of framework viewed along [001]*

**Idealized cell constants:** cubic, Fm $\bar{3}$ m,  $a = 30.7\text{\AA}$

**Coordination sequences  
and vertex symbols:**  $T_1(192, 1)$  4 9 16 25 37 53 74 99 125 151      4·4·4·6·6·8  
 $T_2(192, 1)$  4 9 17 28 41 56 73 93 117 146      4·4·4·8·6·8

**Secondary building units:** 6-6 or 6 or 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Tschörtnerite<sup>(1)</sup>

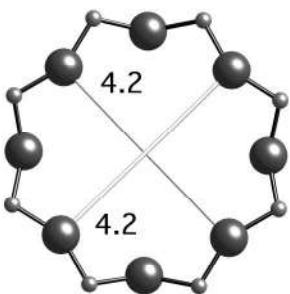
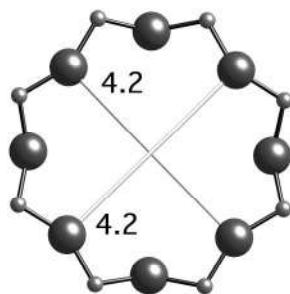
### References:

- (1) Effenberger, H., Giester, G., Krause, W. and Bernhardt, H.J. *Am. Mineral.*, **83**, 607-617 (1998)

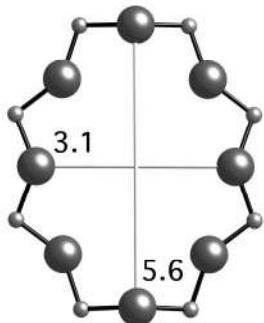
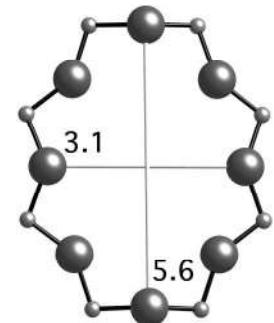
**Crystal chemical data:**  $|\text{Ca}^{2+}_{64}(\text{K}^+)_2, \text{Ca}^{2+}, \text{Sr}^{2+}, \text{Ba}^{2+})_{48}\text{Cu}^{2+}_{48} (\text{OH})_{128} (\text{H}_2\text{O})_x| [\text{Al}_{192}\text{Si}_{192} \text{O}_{768}]$ -TSC  
cubic,  $\text{Fm}\bar{3}\text{m}$ ,  $a = 31.62\text{\AA}$ <sup>(1)</sup>

**Framework density:** 12.1 T/1000 $\text{\AA}^3$

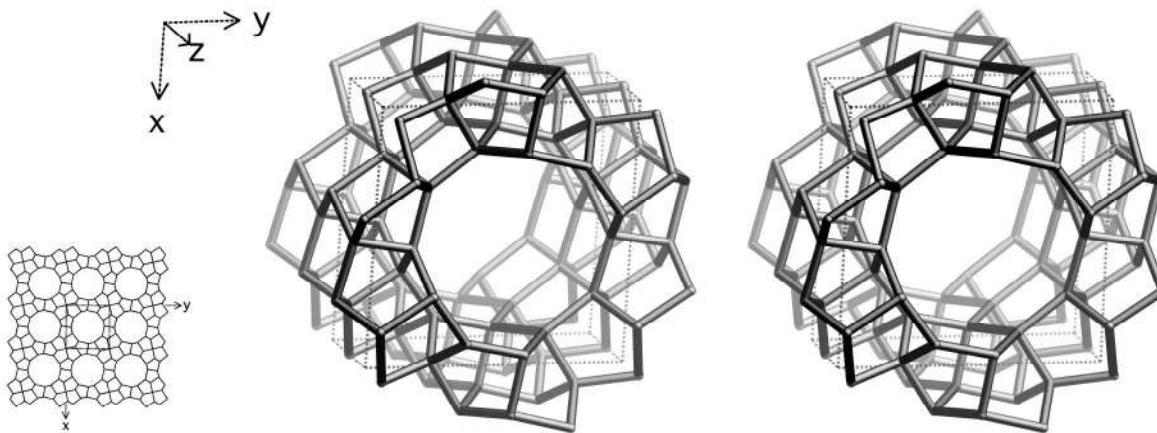
**Channels:**  $<100> \textbf{8} 4.2 \times 4.2^{***} \leftrightarrow <110> \textbf{8} 3.1 \times 5.6^{***}$



8-ring viewed along <100>



8-ring viewed along <110>



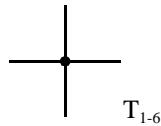
framework viewed along [001] (bottom left: projection down [001])

**Idealized cell constants:** tetragonal, P4,  $a = 13.0\text{\AA}$ ,  $c = 4.9\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (4, 1)	4 12 24 39 61 93 133 179 209 246	5·5·5·6 <sub>2</sub> ·5·7
	T <sub>2</sub> (4, 1)	4 12 26 41 65 94 130 169 218 269	5·5·5·6·7·12 <sub>6</sub>
	T <sub>3</sub> (4, 1)	4 12 24 41 65 95 128 169 218 270	5·6·5·6·5·6 <sub>2</sub>
	T <sub>4</sub> (4, 1)	4 12 23 43 68 94 125 172 226 269	5·6·5·6·5 <sub>2</sub> ·6
	T <sub>5</sub> (1, 4)	4 12 28 38 60 98 152 182 200 246	5·5·5·5·8 <sub>2</sub> ·8 <sub>2</sub>

**Secondary building units:** combinations only

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*VPI-8<sup>(1)</sup>

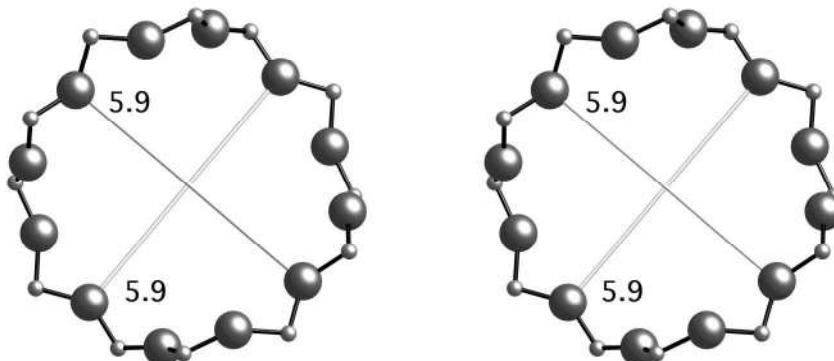
## References:

- (1) Freyhardt, C.C., Lobo, R.F., Khodabandeh, S., Lewis, J.E., Tsapatsis, M., Yoshikawa, M., Camblor, M.A., Pan, M., Helmkamp, M.M., Zones, S.I. and Davis, M.E. *J. Am. Chem. Soc.*, **118**, 7299-7310 (1996)

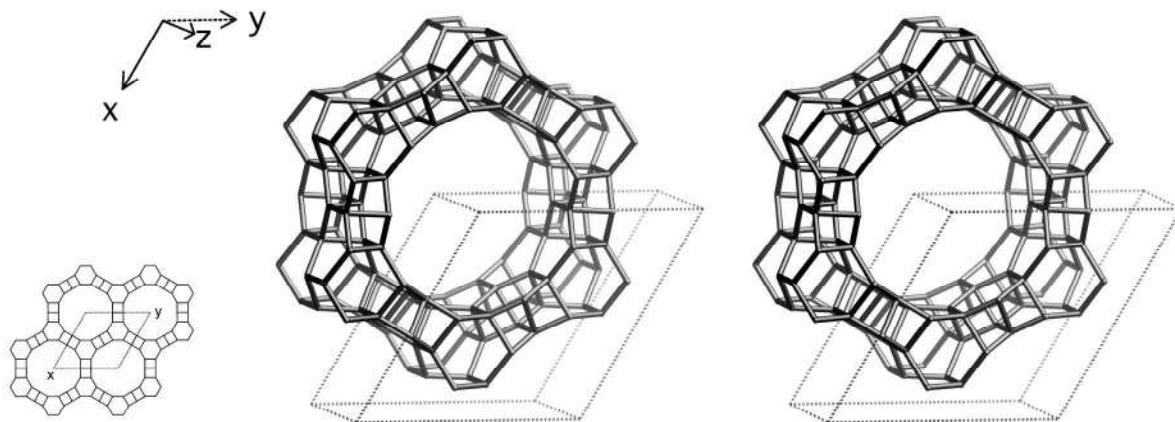
**Crystal chemical data:**  $[\text{Si}_{17} \text{O}_{34}]$ -VET  
tetragonal,  $P\bar{4}$ ,  $a = 13.045\text{\AA}$ ,  $c = 5.034\text{\AA}$ <sup>(1)</sup>

**Framework density:** 19.8 T/1000 $\text{\AA}^3$

**Channels:** [001] **12** 5.9 x 5.9\*



*12-ring viewed along [001]*



framework viewed along [001]

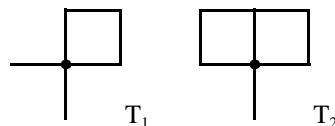
**Idealized cell constants:** hexagonal, P<sub>6</sub><sub>3</sub>/mcm,  $a = 18.3\text{\AA}$ ,  $c = 8.6\text{\AA}$

**Coordination sequences and vertex symbols:**

T <sub>1</sub> (24, 1)	4	11	20	31	44	61	82	108	139	174	4·6 <sub>2</sub> ·6·6 <sub>3</sub> ·6 <sub>2</sub> ·6 <sub>3</sub>
T <sub>2</sub> (12, m)	4	10	18	30	44	60	80	106	135	168	4·6 <sub>3</sub> ·4·6 <sub>3</sub> ·6·6 <sub>4</sub>

**Secondary building units:** 6

**Loop configuration of T-Atoms:**



**Isotypic framework structures:**

- \*VPI-5<sup>(1-3)</sup>
- AlPO-54<sup>(4)</sup>
- H1<sup>(5)</sup>
- MCM-9<sup>(6)</sup>

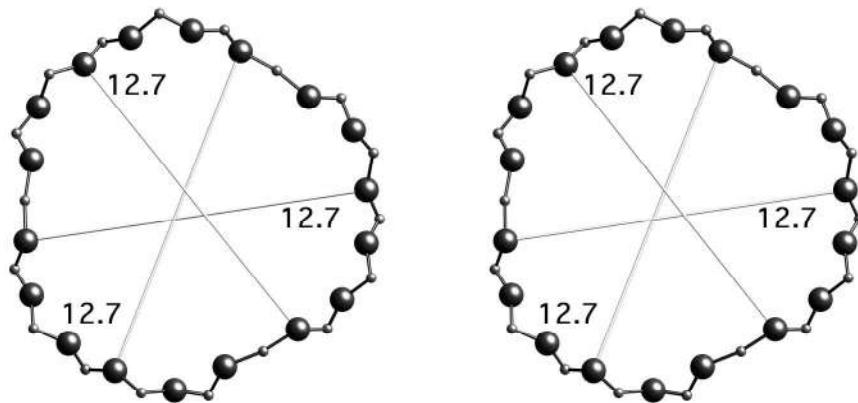
### References:

- (1) Davis, M.E., Saldarriaga, C., Montes, C., Garces, J. and Crowder, C. *Nature*, **331**, 698-699 (1988)
- (2) Richardson Jr., J.W., Smith, J.V. and Pluth, J.J. *J. Phys. Chem.*, **93**, 8212-8219 (1989)
- (3) McCusker, L.B., Baerlocher, Ch., Jahn, E. and Bülow, M. *Zeolites*, **11**, 308-313 (1991)
- (4) Richardson Jr., J.W., Smith, J.V. and Pluth, J.J. *J. Phys. Chem.*, **93**, 8212-8219 (1989)
- (5) d'Yvoire, F. *Bull. Soc. Chim. France*, 1762-1776 (1961)
- (6) Derouane, E.G., Maistreiau, L., Gabelica, Z., Tuel, A., Nagy, J.B. and von Ballmoos, R. *Appl. Catal.*, **51**, L13-L20 (1989)

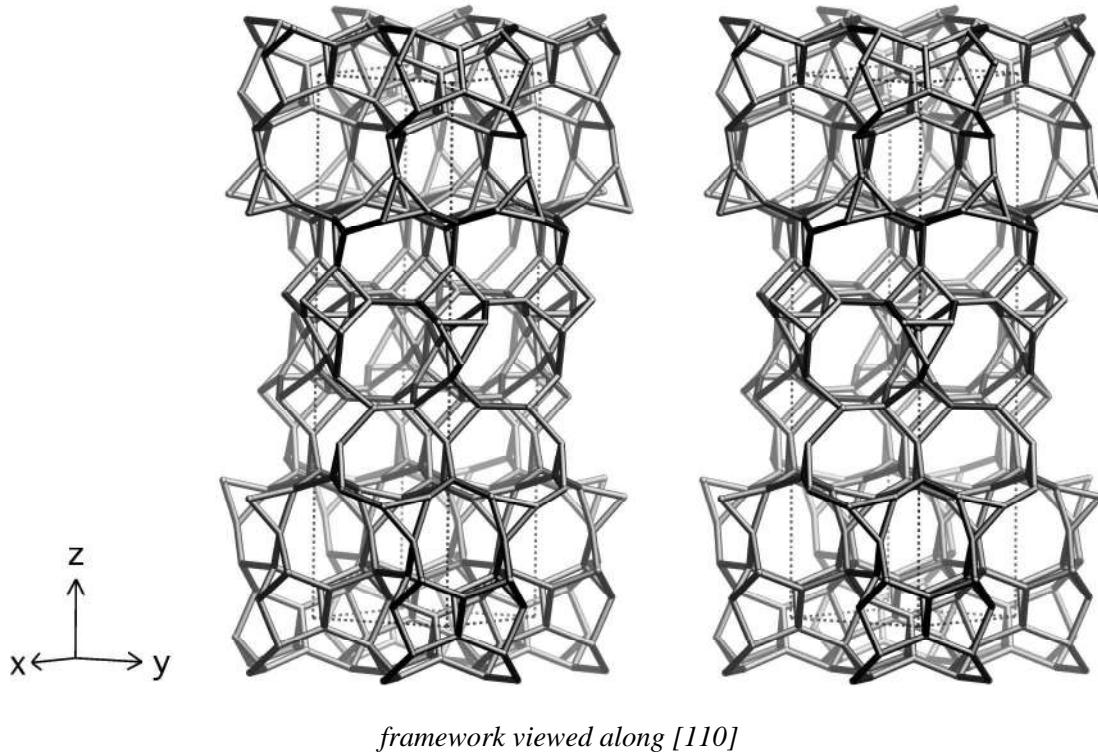
**Crystal chemical data:**  $[(\text{H}_2\text{O})_{42}] [\text{Al}_{18}\text{P}_{18} \text{O}_{72}]$ -VFI  
hexagonal,  $\text{P}6_3$ ,  $a = 18.975\text{\AA}$ ,  $c = 8.104\text{\AA}$ <sup>(3)</sup>

**Framework density:** 14.2 T/1000 $\text{\AA}^3$

**Channels:** [001] **18** 12.7 x 12.7\*



*18-ring viewed along [001]*

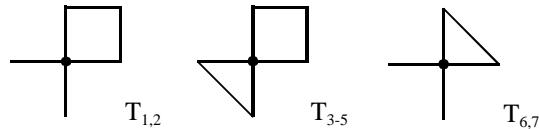


**Idealized cell constants:** tetragonal, P4<sub>2</sub>/ncm (origin choice 2),  $a = 10.0\text{\AA}$ ,  $c = 34.1\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (16, 1) 4 11 23 39 63 93 126 170 210 255	4·8·5·8 <sub>2</sub> ·5 <sub>2</sub> ·8 <sub>2</sub>
	T <sub>2</sub> (8, m) 4 11 19 39 59 89 130 166 207 274	4·5 <sub>2</sub> ·5·8·5·8
	T <sub>3</sub> (8, m) 4 9 20 37 61 92 117 152 201 246	3·4·8·8 <sub>2</sub> ·8·8 <sub>2</sub>
	T <sub>4</sub> (8, m) 4 9 20 37 62 87 119 158 195 248	3·4·8 <sub>2</sub> ·8 <sub>3</sub> ·8 <sub>2</sub> ·8 <sub>3</sub>
	T <sub>5</sub> (8, m) 4 9 21 41 59 85 133 155 195 261	3·4·8·8 <sub>3</sub> ·8·8 <sub>3</sub>
	T <sub>6</sub> (8, m) 4 10 18 39 65 83 119 169 218 236	3·8 <sub>3</sub> ·5·5 <sub>2</sub> ·5·5 <sub>2</sub>
	T <sub>7</sub> (4, 2mm) 4 10 18 36 64 82 118 176 202 264	3·8 <sub>2</sub> ·5·5·5·5

**Secondary building units:** combinations only

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*VPI-9<sup>(1)</sup>

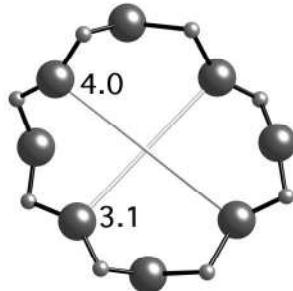
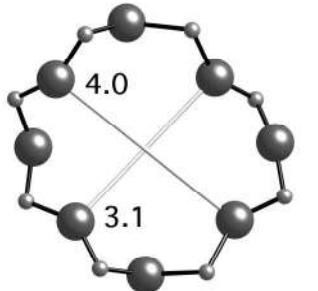
### References:

- (1) McCusker, L.B., Grosse-Kunstleve, R.W., Baerlocher, Ch., Yoshikawa, M. and Davis, M.E. *Microporous Materials*, **6**, 295-309 (1996)

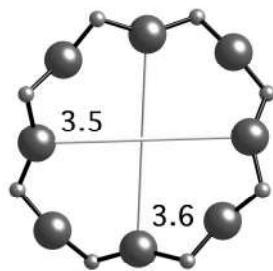
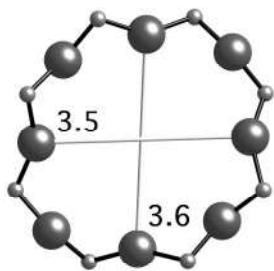
**Crystal chemical data:**  $[\text{Rb}^+_{44}\text{K}^+_4(\text{H}_2\text{O})_{48}] [\text{Zn}_{24}\text{Si}_{96}\text{O}_{240}]$ -VNI  
tetragonal,  $\text{P}4_1\text{2}_1\text{2}$ ,  $a = 9.884\text{\AA}$ ,  $c = 73.650\text{\AA}$ <sup>(1)</sup>  
(Relationship to unit cell of Framework Type:  $a' = a$ ,  $b' = b$ ,  $c' = 2c$ )

**Framework density:** 16.7 T/1000 $\text{\AA}^3$

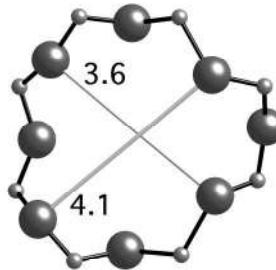
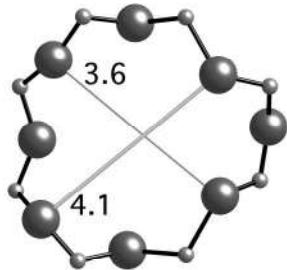
**Channels:**  $\{<110> 8 \text{ } 3.1 \times 4.0 \leftrightarrow [001] 8 \text{ } 3.5 \times .3.6\}^{***}$



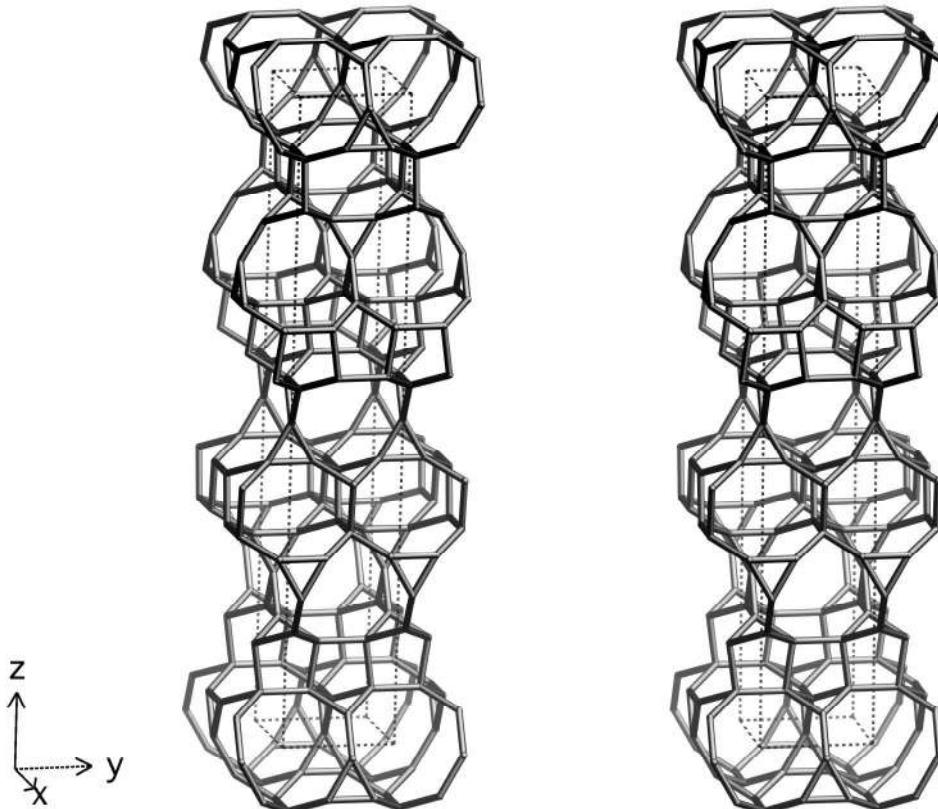
*8-ring along <110>*



*8-ring viewed along [001]*



*2nd 8-ring along <110>*



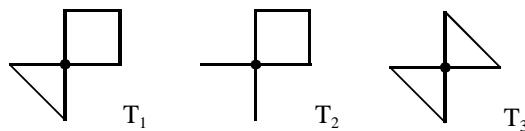
framework viewed along [100]

**Idealized cell constants:** tetragonal, I4<sub>1</sub>/amd,  $a = 7.2\text{\AA}$ ,  $c = 41.8\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	T <sub>1</sub> (16, m) 4 9 21 42 61 81 123 159 198 246	3·4·8 <sub>2</sub> ·9 <sub>4</sub> ·8 <sub>2</sub> ·9 <sub>4</sub>
	T <sub>2</sub> (16, m) 4 11 21 40 61 93 122 151 195 251	4·5 <sub>2</sub> ·5·8·5·8
	T <sub>3</sub> (4, 4m2) 4 8 20 48 56 84 120 160 212 240	3·3·9 <sub>4</sub> ·9 <sub>4</sub> ·9 <sub>4</sub>

**Secondary building units:** combinations only

**Loop configuration of  
T-Atoms:**

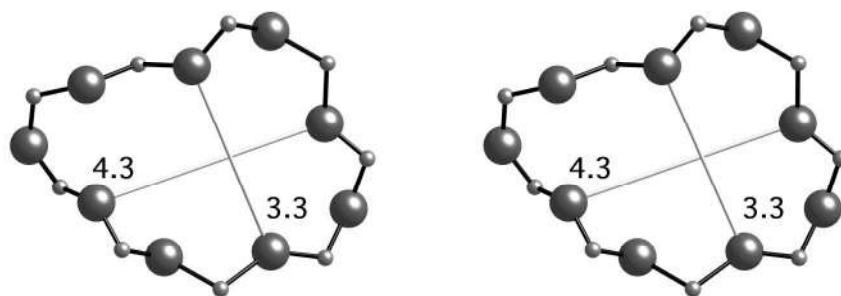


**Isotypic framework  
structures:** \*VPI-7<sup>(1,2)</sup>  
Gaultite<sup>(3)</sup>  
VSV-7#<sup>(4)</sup>

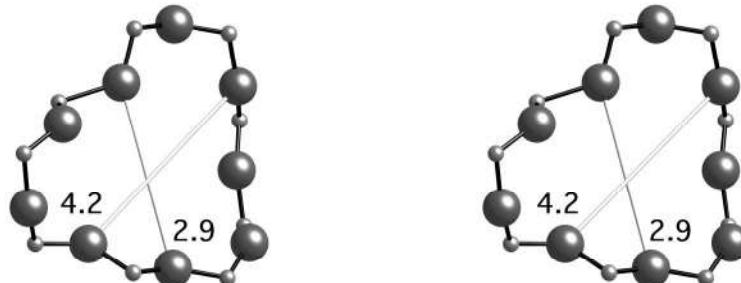
**Crystal chemical data:**  $[\text{Na}^+_{26}\text{H}^+_6(\text{H}_2\text{O})_{44}] [\text{Zn}_{16}\text{Si}_{56}\text{O}_{144}]$ -VSV  
 orthorhombic, Fdd2,  $a = 39.88\text{\AA}$ ,  $b = 10.326\text{\AA}$ ,  $c = 10.219\text{\AA}$ <sup>(2)</sup>  
 (Relationship to unit cell of Framework Type:  
 $\mathbf{a}' = \mathbf{c}$ ,  $\mathbf{b} = \mathbf{a} \sqrt{2}$ ,  $\mathbf{c}' = \mathbf{b} \sqrt{2}$   
 or, as vectors,  $\mathbf{a} = \mathbf{c}$ ,  $\mathbf{b} = \mathbf{b} - \mathbf{a}$ ,  $\mathbf{c} = \mathbf{a} + \mathbf{b}$ )

**Framework density:** 17.1 T/1000 $\text{\AA}^3$

**Channels:**  $[01\bar{1}]$  9 3.3 x 4.3\*  $\leftrightarrow$   $[011]$  9 2.9 x 4.2\*  $\leftrightarrow$   $[011]$  8 2.1 x 2.7\*



9-ring along  $[01\bar{1}]$

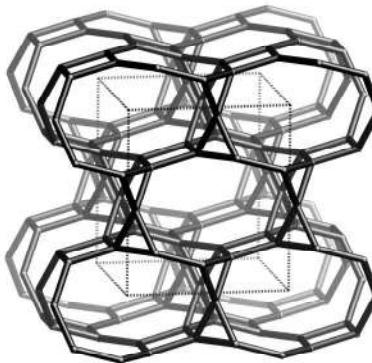
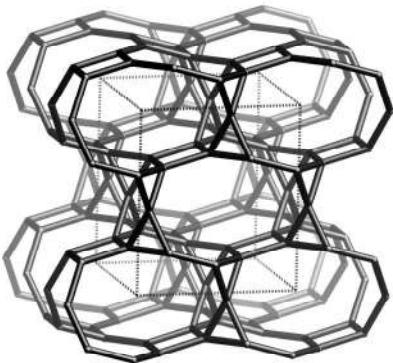
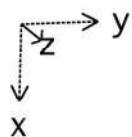


9-ring along  $[011]$

See Appendix A for 8-ring along [011] and 8-ring viewed along [100]

### References:

- (1) Annen, M.J., Davis, M.E., Higgins, J.B. and Schlenker, J.L. *Chem. Commun.*, 1175-1176 (1991)
- (2) Röhrig, C., Gies, H. and Marler, B. *Zeolites*, **14**, 498-503 (1994)
- (3) Ercit, T.S. and van Velthuizen, J. *Can. Mineral.*, **32**, 855-863 (1994)
- (4) Röhrig, C., Dierdorf, I. and Gies, H. *J. Phys. Chem. Solids*, **56**, 1369-1376 (1995)



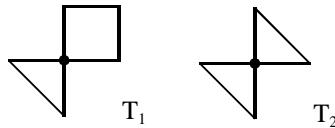
*framework viewed along [001]*

**Idealized cell constants:** orthorhombic, Cccm,  $a = 11.8\text{\AA}$ ,  $b = 10.3\text{\AA}$ ,  $c = 10.0\text{\AA}$

**Coordination sequences  
and vertex symbols:**  $T_1(16, 1) \quad 4 \quad 9 \quad 18 \quad 32 \quad 51 \quad 74 \quad 98 \quad 126 \quad 163 \quad 199 \quad 3 \cdot 4 \cdot 6 \cdot 8 \cdot 8 \cdot 10$   
 $T_2(4, 222) \quad 4 \quad 8 \quad 18 \quad 32 \quad 52 \quad 70 \quad 98 \quad 132 \quad 152 \quad 200 \quad 3 \cdot 3 \cdot 6 \cdot 6 \cdot 10 \cdot 10$

**Secondary building units:** spiro-5

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:**

\*Weinebeneite<sup>(1)</sup>

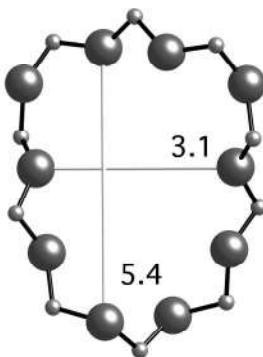
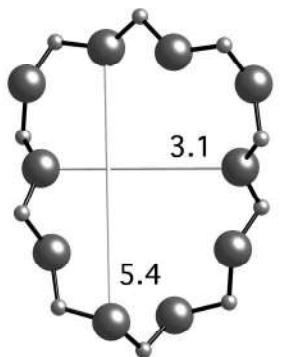
## References:

- (1) Walter, F. *Eur. J. Mineral.*, **4**, 1275-1283 (1992)

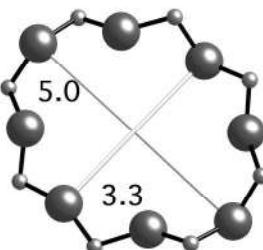
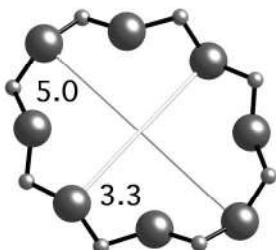
**Crystal chemical data:**  $[\text{Ca}^{2+}_4 (\text{H}_2\text{O})_{16}] [\text{Be}_{12}\text{P}_8 \text{O}_{32}(\text{OH})_8]$ -WEI  
monoclinic, C1c1,  $a = 11.897\text{\AA}$ ,  $b = 9.707\text{\AA}$ ,  $c = 9.633\text{\AA}$ ,  $\beta = 95.76^\circ$  <sup>(1)</sup>

**Framework density:** 18.1 T/1000 $\text{\AA}^3$

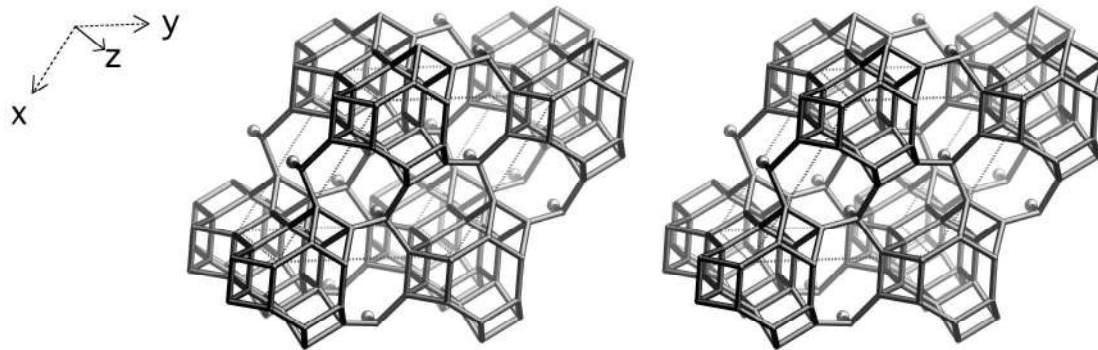
**Channels:** [001] **10** 3.1 x 5.4\*  $\leftrightarrow$  [100] **8** 3.3 x 5.0\*



10-ring viewed along [001]



8-ring viewed along [100]



framework viewed along [001]

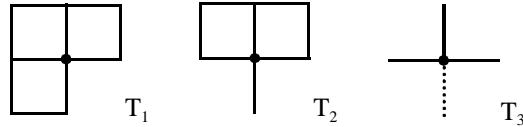
**Idealized cell constants:** hexagonal, P $\bar{6}$ 2m,  $a = 13.6\text{\AA}$ ,  $c = 7.6\text{\AA}$

**Coordination sequences and vertex symbols:**

$T_1(12, 1)$	4	9	16	27	46	73	102	129	157	191
$T_2(6, m)$	4	9	19	34	49	67	94	125	157	195
$T_3(2, 3)$	3	9	21	36	53	69	90	119	156	201

**Secondary building units:** combinations only

**Loop configuration of T-Atoms:**



**Isotypic framework structures:** \*Wenkite<sup>(1,2)</sup>

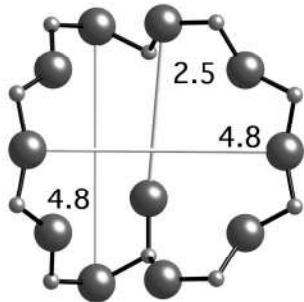
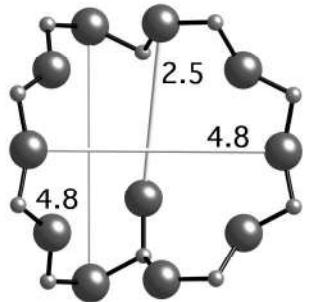
### References:

- (1) Wenk, H.-R. *Z. Kristallogr.*, **137**, 113-126 (1973)
- (2) Merlini, S. *Acta Crystallogr.*, **B30**, 1262-1266 (1974)

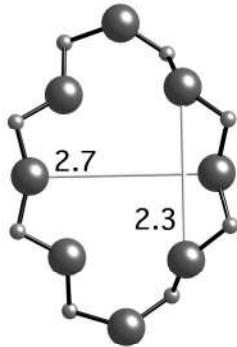
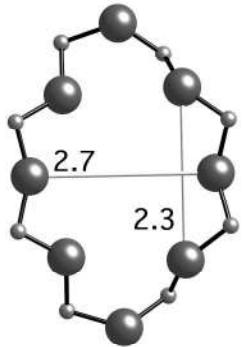
**Crystal chemical data:**  $[\text{Ba}^{2+}_4(\text{Ca}^{2+},\text{Na}^+_2)_3 (\text{SO}_4^{2-})_3 \text{H}_2\text{O}] [\text{Al}_8\text{Si}_{12} \text{O}_{39}(\text{OH})_2]^-$  -WEN  
hexagonal,  $\text{P}\bar{6}2\text{m}$ ,  $a = 13.511\text{\AA}$ ,  $c = 7.462\text{\AA}$ <sup>(2)</sup>

**Framework density:** 17 T/1000 $\text{\AA}^3$

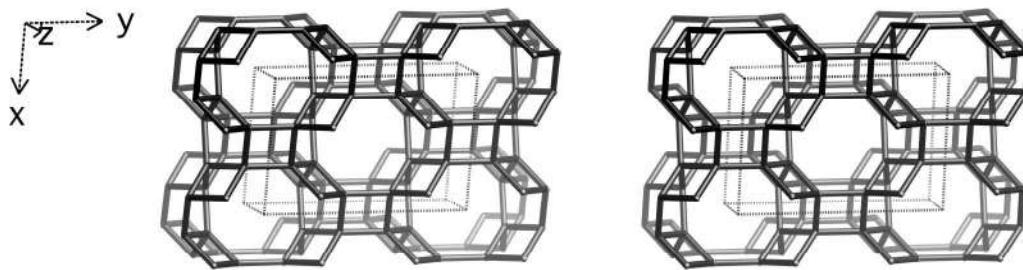
**Channels:**  $<100> \textbf{10} \ 2.5 \times 4.8^{**} \leftrightarrow [001] \textbf{8} \ 2.3 \times 2.7^*$



10-ring viewed along  $<100>$



8-ring viewed along  $[001]$



framework viewed along [001]

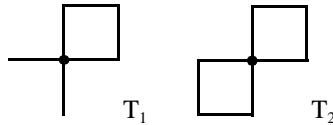
**Idealized cell constants:** monoclinic, C2/m,  $a = 10.2\text{\AA}$ ,  $b = 13.8\text{\AA}$ ,  $c = 6.8\text{\AA}$ ,  $\beta = 111.5^\circ$

**Coordination sequences  
and vertex symbols:**

$T_1(8, 1)$	4 11 22 39 61 88 120 155 192 241	4·5·5·8·8
$T_2(8, 1)$	4 10 22 39 61 89 118 153 198 241	4·4·5·8 <sub>2</sub> ·5·8 <sub>2</sub>

**Secondary building units:** 8 or 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*Yugawaralite<sup>(1-3)</sup>  
Sr-Q<sup>(4,5)</sup>

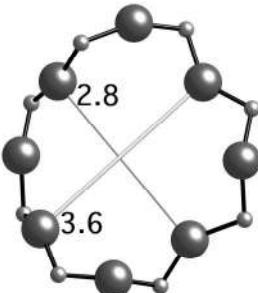
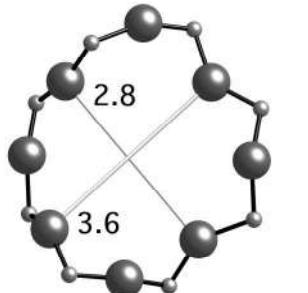
### References:

- (1) Kerr, I.S. and Williams, D.J. *Z. Kristallogr.*, **125**, 220-225 (1967)
- (2) Kerr, I.S. and Williams, D.J. *Acta Crystallogr.*, **B25**, 1183-1190 (1969)
- (3) Leimer, H.W. and Slaughter, M. *Z. Kristallogr.*, **130**, 88-111 (1969)
- (4) Hawkins, D.B. *Mater. Res. Bull.*, **2**, 951-958 (1967)
- (5) Kvick, Å., Artioli, G. and Smith, J.V. *Z. Kristallogr.*, **174**, 265-281 (1986)

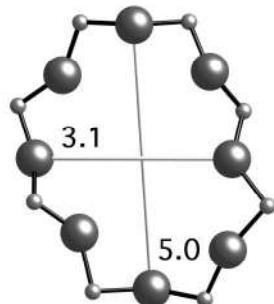
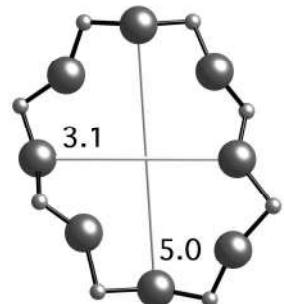
**Crystal chemical data:**  $[\text{Ca}^{2+}_2 (\text{H}_2\text{O})_8] [\text{Al}_4\text{Si}_{12} \text{O}_{32}]$ -YUG  
monoclinic,  $\text{Pc}$ ,  $a = 6.73\text{\AA}$ ,  $b = 13.95\text{\AA}$ ,  $c = 10.03\text{\AA}$ ,  $\beta = 111.5^\circ$  <sup>(2)</sup>  
(Relationship to unit cell of Framework Type:  $a' = c$ ,  $b' = b$ ,  $c' = a$ )

**Framework density:**  $18.3 \text{ T}/1000\text{\AA}^3$

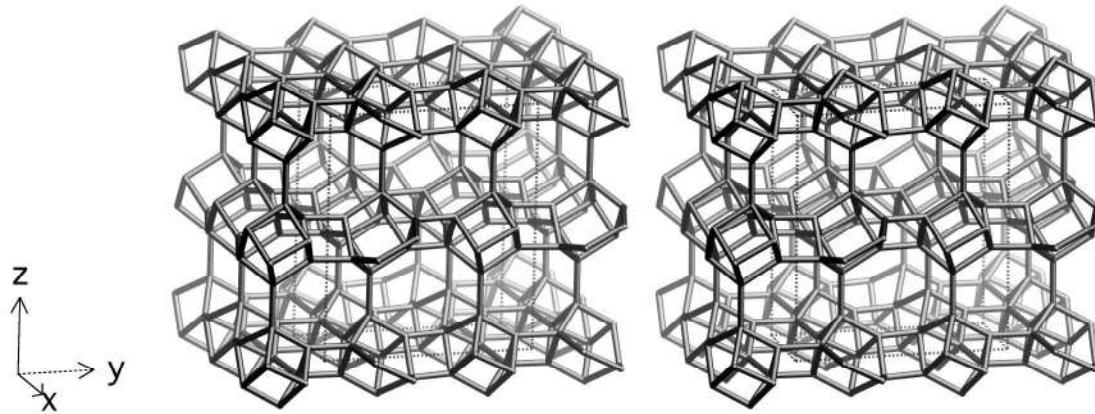
**Channels:**  $[100] \text{ 8 } 2.8 \times 3.6^* \leftrightarrow [001] \text{ 8 } 3.1 \times 5.0^*$



8-ring viewed along [100]



8-ring viewed along [001]



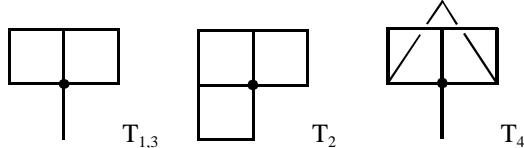
framework viewed along [100]

**Idealized cell constants:** orthorhombic, Pbcm,  $a = 6.9\text{\AA}$ ,  $b = 14.9\text{\AA}$ ,  $c = 17.2\text{\AA}$

<b>Coordination sequences and vertex symbols:</b>	$T_1(8, 1)$	4 10 21 34 47 72 108 136 162 200	4·8·4·8 <sub>2</sub> ·6 <sub>3</sub> ·8
	$T_2(8, 1)$	4 9 19 33 53 78 100 126 166 213	4·4·4·8·6·6 <sub>3</sub>
	$T_3(8, 1)$	4 10 18 33 57 77 95 129 172 209	4·6·4·6·6·8
	$T_4(8, 1)$	4 9 17 32 53 74 98 128 165 208	4·6·4·6 <sub>2</sub> ·4·8

**Secondary building units:** 6-2 or 4

**Loop configuration of  
T-Atoms:**



**Isotypic framework  
structures:** \*ZAPO-M1<sup>(1)</sup>  
GaPO-DAB-2<sup>(2)</sup>  
UiO-7<sup>(3,4)</sup>

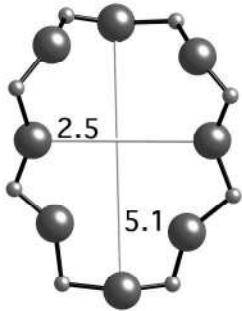
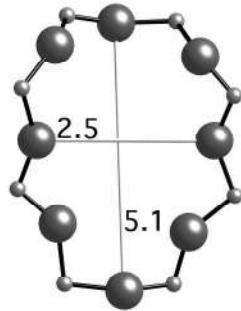
### References:

- (1) Marler, B., Patarin, J. and Sierra, L. *Microporous Materials*, **5**, 151-159 (1995)
- (2) Meden, A., Grosse-Kunstleve, R.W., Baerlocher, Ch. and McCusker, L.B. *Z. Kristallogr.*, **212**, 801-807 (1997)
- (3) Akporiaye, D.E., Fjellvåg, H., Halvorsen, E.N., Hustveit, J., Karlsson, A. and Lillerud, K.P. *Chem. Commun.*, 601-602 (1996)
- (4) Akporiaye, D.E., Fjellvåg, H., Halvorsen, E.N., Hustveit, J., Karlsson, A. and Lillerud, K.P. *J. Phys. Chem.*, **100**, 16641-16646 (1996)

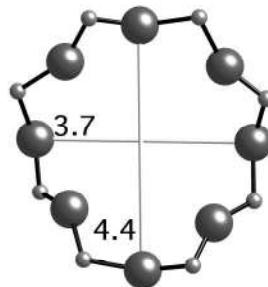
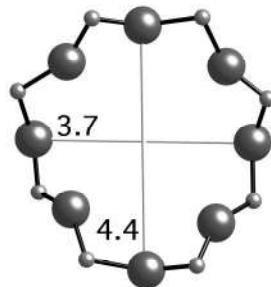
**Crystal chemical data:**  $[(C_4H_{12}N^+)_8][Zn_8Al_{24}P_{32}O_{128}]$ -ZON  
 $C_4H_{12}N^+$  = tetramethylammonium  
orthorhombic, Pbca,  $a = 14.226\text{\AA}$ ,  $b = 15.117\text{\AA}$ ,  $c = 17.557\text{\AA}$ <sup>(1)</sup>  
(Relationship to unit cell of Framework Type:  $a' = 2a$ ,  $b' = b$ ,  $c' = c$ )

**Framework density:** 17 T/1000 $\text{\AA}^3$

**Channels:** [100] 8 2.5 x 5.1\* ↔ [010] 8 3.7 x 4.4\*



8-ring viewed along [100]



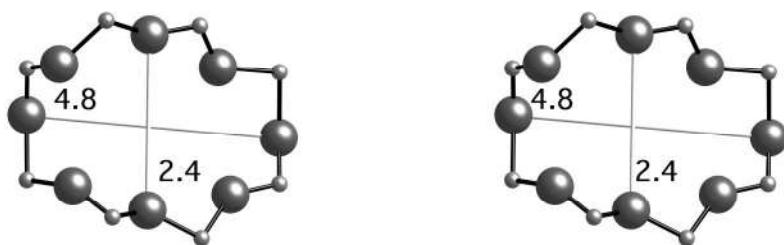
8-ring viewed along [010]



## APPENDIX A

### CGF

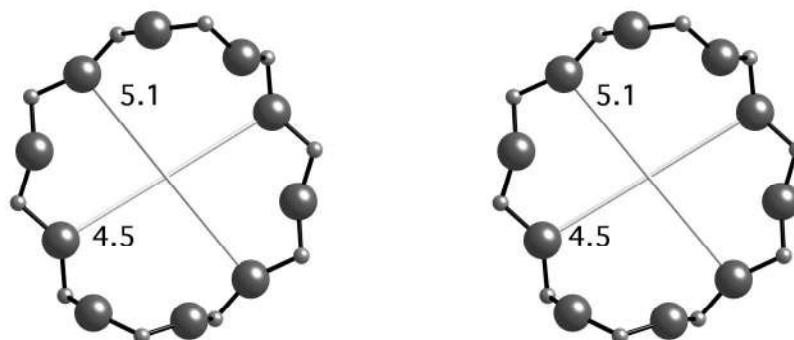
---



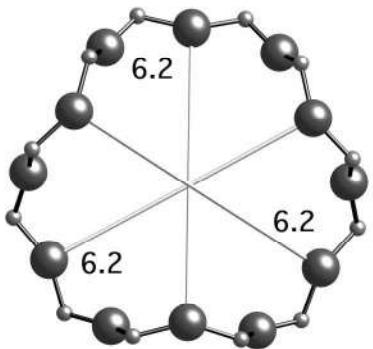
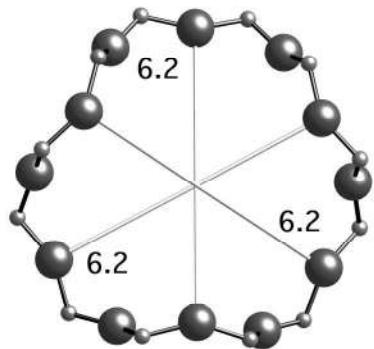
*8-ring viewed along [001]*

### CON

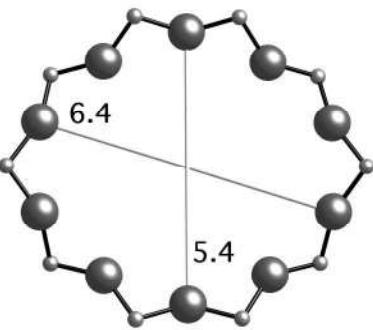
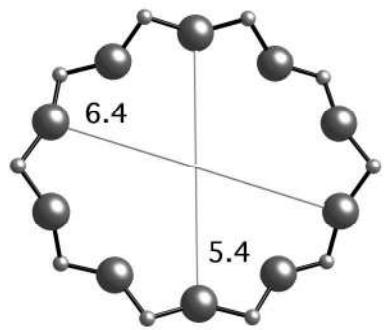
---



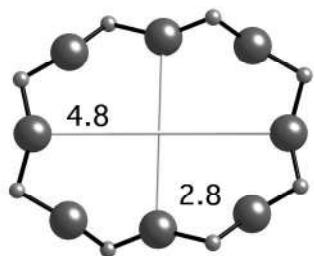
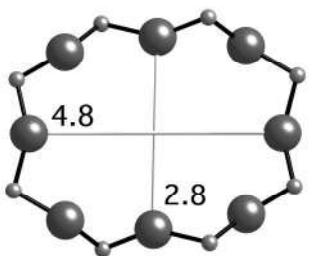
*10-ring viewed along [010]*



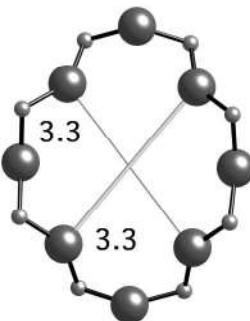
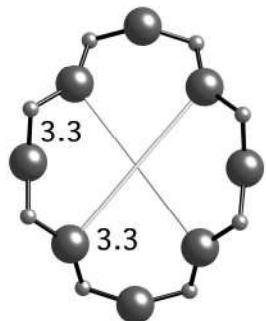
*2nd 12-ring viewed along [001]*



*10-ring viewed normal to [001]*



8-ring viewed along [010]



2nd 8-ring viewed along [001]

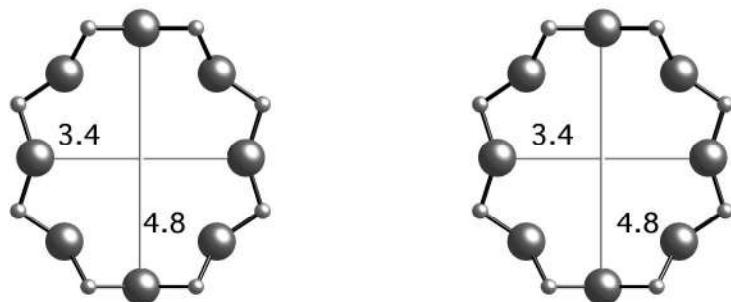
<b>Isotypic framework structures:</b>	[As-Si-O]-MFI <sup>(4)</sup>	NU-4 <sup>(16)</sup>
	[Fe-Si-O]-MFI <sup>(5)</sup>	NU-5 <sup>(17)</sup>
	[Ga-Si-O]-MFI <sup>(6)</sup>	Silicalite <sup>(18)</sup>
	AMS-1B <sup>(7)</sup>	TS-1 <sup>(19)</sup>
	AZ-1 <sup>(8)</sup>	TSZ <sup>(20)</sup>
	Bor-C <sup>(9)</sup>	TSZ-III <sup>(21)</sup>
	Boralite C <sup>(10)</sup>	TZ-01 <sup>(22)</sup>
	Encilite <sup>(11)</sup>	USC-4 <sup>(23)</sup>
	FZ-1 <sup>(12)</sup>	USI-108 <sup>(24)</sup>
	LZ-105 <sup>(13)</sup>	ZBH <sup>(25)</sup>
	Monoclinic H-ZSM-5 <sup>(14)</sup>	ZKQ-1B <sup>(26)</sup>
	Mutinaite <sup>(15)</sup>	ZMQ-TB <sup>(27)</sup>

**References:**

- (4) Bhaumik, A. and Kumar, R. *Chem. Commun.*, 869-870 (1995)
- (5) Patarin, J., Kessler, H. and Guth, J.L. *Zeolites*, **10**, 674-679 (1990)
- (6) Awate, S.V., Joshi, P.N., Shiralkar, V.P. and Kotasthane, A.N. *J. Incl. Phenom.*, **13**, 207-218 (1992)
- (7) Klotz, M.R. *U.S. Patent 4,269,813* (1981)
- (8) Chono, M. and Ishida, H. *E. Patent B-113,116* (1984)
- (9) Taramasso, M., Perego, G. and Notari, B. In *Proc. 5th Int. Zeolite Conf.*, (ed. L.V.C. Rees), pp. 40-48 (1980), Heyden, London
- (10) Taramasso, M., Manara, G., Fattore, V. and Notari, B. *GB Patent 2,024,790* (1980)
- (11) Ratnasamy, P. and Borade, M.B. *E. Patent A-160,136* (1985)
- (12) Suzuki, T., Hashimoto, S. and Nakano, R. *E. Patent B-31,255* (1981)
- (13) Grose, R.W. and Flanigen, E.M. *U.S. Patent 4,257,885* (1981)
- (14) van Koningsveld, H., Jansen, J.C. and van Bekkum, H. *Zeolites*, **10**, 235-242 (1990)
- (15) Vezzalini, G., Quartieri, S., Galli, E., Alberti, A., Cruciani, G. and Kvick, A. *Zeolites*, **19**, 323-325 (1997)
- (16) Whittam, T.V. *E. Patent B-65,401* (1986)
- (17) Whittam, T.V. *E. Patent B-54,386* (1982)
- (18) Flanigen, E.M., Bennett, J.M., Grose, R.W., Cohen, J.P., Patton, R.L., Kirchner, R.M. and Smith, J.V. *Nature*, **271**, 512-516 (1978)
- (19) Taramasso, M., Perego, G. and Notari, B. *U.S. Patent 4,410,501* (1983)
- (20) Ashibe, K., Kobayashi, W., Maejima, T., Sakurada, S. and Tagaya, N. *E. Patent A-101,232* (1984)
- (21) Sakurada, S., Tagaya, N., Miura, T., Maeshima, T. and Hashimoto, T. *E. Patent A-170,751* (1986)
- (22) Iwayama, K., Kamano, T., Tada, K. and Inoue, T. *E. Patent A-57,016* (1982)
- (23) Young, D.A. *U.S. Patent 4,325,929* (1982)
- (24) Hinzenkamp, J.A. and Walatka, V.V. *U.S. Patent 4,423,020* (1983)
- (25) Holderich, W., Mross, W.D. and Schwartzmann, M. *E. Patent B-77,946* (1986)
- (26) Kee Kwee, L.S.L. *E. Patent A-148,038* (1984)
- (27) Kee Kwee, L.S.L. *E. Patent A-104,107* (1983)

## MOR

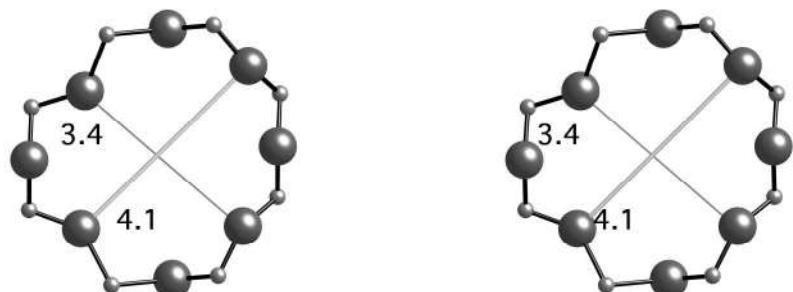
---



8-ring viewed along [010]

## RSN

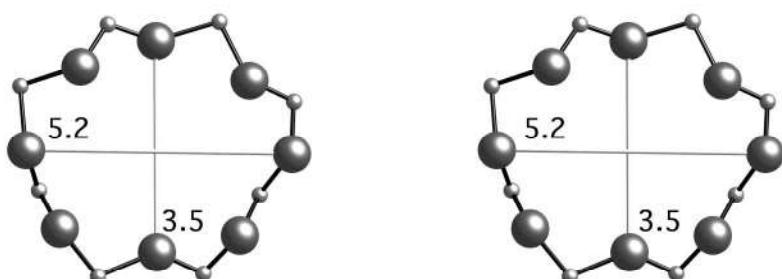
---



8-ring viewed along [010]

## SBS

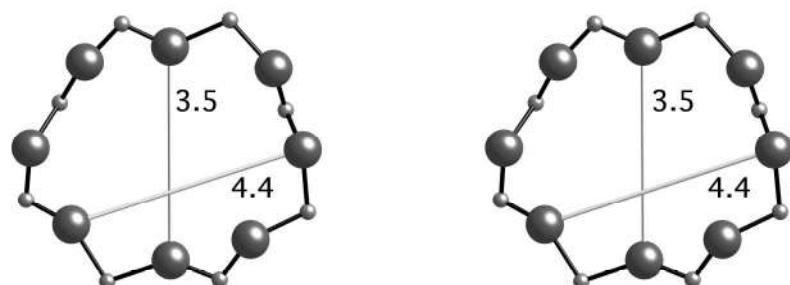
---



8-ring along [001]

## SBT

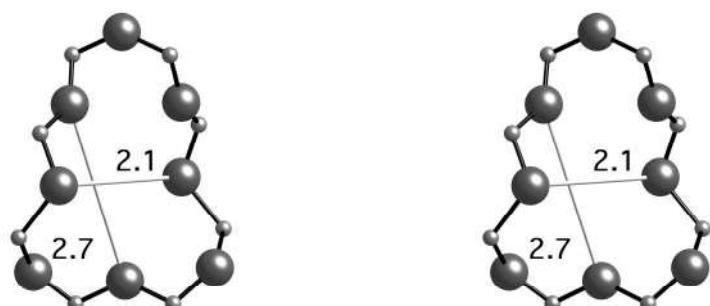
---



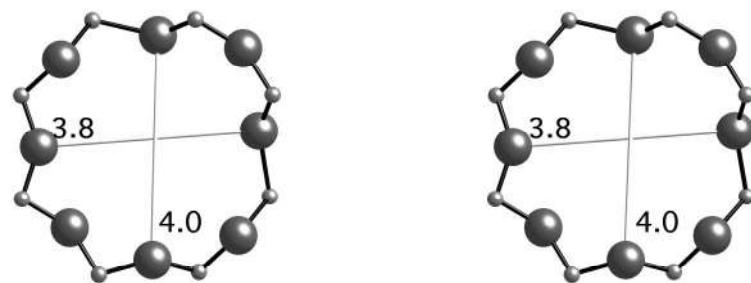
8-ring viewed along [102]

## VSV

---



8-ring along [011]



8-ring viewed along [100]

## APPENDIX B

### **Rules for Framework Type Assignment**

The following is a set of rules to be applied by the Structure Commission of the IZA in assigning a three-letter code to a new framework type. The materials of interest are generally defined as open 4-connected 3D nets which have the general (approximate) composition AB<sub>2</sub>, where A is a tetrahedrally connected atom and B is any 2-connected atom, which may or may not be shared, between two neighboring A atoms. Inclusion of other microporous materials is left to the discretion of the IZA Structure Commission, depending on the interest of the molecular sieve science community at large.

#### **RULES**

- (i) The IZA Structure Commission is the only body that can coordinate the assignment of a code to such frameworks. The framework types are idealized, with no reference to actual materials, symmetry, composition, etc. and, therefore, ONLY refer to the connectivity.
- (ii) The code for a 4-connected 3D framework type shall consist of three capital letters. Other frameworks of interest shall be indicated by a (-) in front of the code.
- (iii) The three letters of the code shall be mnemonic and must refer to an actual material (i.e. the type material). These materials are chosen to be:
  - mineral names (rules of the International Mineralogical Association followed).
  - commonly accepted synthetic material types.
  - in the absence of the above, the workers who first determined the structure have priority in assigning the name.
- (iv) No mnemonic code can be assigned without the structure being determined with the following exception:  
For "polytypic" materials, codes can be assigned as useful. Such codes shall be marked with an asterisk.
- (v) Codes of framework types which turn out to be in error are discredited. Later use of the code is not permitted.
- (vi) For all cases where a decision of the Structure Commission is required, a two thirds majority vote of the full commission shall be required. Such votes are taken verbally at a meeting of the Structure Commission or are done in writing on the initiative of the Chairperson and Co-chairperson. All evidence substantiating a new framework type must accompany the ballot. The members are obliged to respond to this request within one month and the Chairperson and Co-chairperson will make all possible effort to solicit replies from all members.



## APPENDIX C

### Topological Densities

Topological density values  $TD_{10}$  including all values of the coordination sequence from  $N_0$  to  $N_{10}$ .

TD = exact density as defined in the Explanatory Notes.

<b>Code</b>	<b>TD<sub>10</sub></b>	<b>TD</b>	<b>Code</b>	<b>TD<sub>10</sub></b>	<b>TD</b>	<b>Code</b>	<b>TD<sub>10</sub></b>	<b>TD</b>
<b>ABW</b>	833	0.703704	<b>DFO</b>	664	0.576042	<b>MTW</b>	912	0.776004
<b>ACO</b>	787	0.666667	<b>DFT</b>	840	0.711111	<b>MWW</b>	851	0.752127
<b>AEI</b>	689	0.583081	<b>DOH</b>	1002	0.882191	<b>NAT</b>	834	0.740741
<b>AEL</b>	904	0.766975	<b>DON</b>	851	0.728920	<b>NES</b>	922	0.818688
<b>AEN</b>	956	0.857768	<b>EAB</b>	735	0.628571	<b>NON</b>	1038	0.915319
<b>AET</b>	824	0.697917	<b>EDI</b>	786	0.666667	<b>OFF</b>	739	0.628571
<b>AFG</b>	816	0.693333	<b>EMT</b>	584	0.493095	<b>OSI</b>	892	0.777778
<b>AFI</b>	828	0.700000	<b>EPI</b>	979	0.845059	<b>OSO</b>	747	0.645000
<b>AFN</b>	777	0.661111	<b>ERI</b>	738	0.628571	<b>-PAR</b>	773	0.664915
<b>AFO</b>	907	0.769676	<b>ESV</b>	875	0.754994	<b>PAU</b>	728	0.623377
<b>AFR</b>	687	0.579229	<b>EUO</b>	965	0.872973	<b>PHI</b>	751	0.635556
<b>AFS</b>	656	0.568750	<b>FAU</b>	579	0.476190	<b>RHO</b>	641	0.533333
<b>AFT</b>	685	0.585714	<b>FER</b>	1021	0.887635	<b>-RON</b>	771	0.688981
<b>AFX</b>	689	0.585714	<b>FRA</b>	802	0.683983	<b>RSN</b>	914	0.786947
<b>AFY</b>	585	0.488889	<b>GIS</b>	726	0.611111	<b>RTE</b>	844	0.715873
<b>AHT</b>	853	0.729167	<b>GME</b>	694	0.585714	<b>RTH</b>	817	0.695757
<b>ANA</b>	933	0.800000	<b>GON</b>	926	0.787372	<b>RUT</b>	902	0.767499
<b>APC</b>	814	0.696296	<b>GOO</b>	840	0.716755	<b>SAO</b>	632	0.545455
<b>APD</b>	888	0.759019	<b>HEU</b>	909	0.778254	<b>SAS</b>	701	0.586667
<b>AST</b>	742	0.625000	<b>IFR</b>	798	0.678236	<b>SAT</b>	763	0.644444
<b>ASV</b>	787	0.666667	<b>ISV</b>	772	0.694533	<b>SAV</b>	690	0.587302
<b>ATN</b>	833	0.703704	<b>ITE</b>	824	0.711802	<b>SBE</b>	619	0.514323
<b>ATO</b>	894	0.760000	<b>JBW</b>	890	0.753333	<b>SBS</b>	617	0.534092
<b>ATS</b>	752	0.640000	<b>KFI</b>	681	0.571429	<b>SBT</b>	617	0.522344
<b>ATT</b>	768	0.647619	<b>LAU</b>	782	0.658201	<b>SFE</b>	892	0.767358
<b>ATV</b>	960	0.816667	<b>LEV</b>	719	0.605714	<b>SFF</b>	880	0.765073
<b>AWO</b>	828	0.708843	<b>LIO</b>	816	0.693333	<b>SGT</b>	962	0.862901
<b>AWW</b>	772	0.656085	<b>LOS</b>	816	0.693333	<b>SOD</b>	791	0.666667
<b>*BEA</b>	805	0.704545	<b>LOV</b>	879	0.754446	<b>STF</b>	877	0.748986
<b>BIK</b>	1052	0.907407	<b>LTA</b>	641	0.533333	<b>STI</b>	852	0.720342
<b>BOG</b>	781	0.659922	<b>LTL</b>	746	0.619048	<b>STT</b>	859	0.760705
<b>BPH</b>	667	0.568750	<b>LTN</b>	779	0.698452	<b>TER</b>	872	0.739316
<b>BRE</b>	901	0.778031	<b>MAZ</b>	823	0.697032	<b>THO</b>	784	0.666667
<b>CAN</b>	817	0.693333	<b>MEI</b>	728	0.630682	<b>TON</b>	1006	0.867111
<b>CAS</b>	1042	0.895949	<b>MEL</b>	944	0.808638	<b>TSC</b>	590	0.482539
<b>CFI</b>	892	0.765153	<b>MEP</b>	1059	0.955379	<b>VET</b>	1023	0.913737
<b>CGF</b>	819	0.695238	<b>MER</b>	738	0.622222	<b>VFI</b>	669	0.562500
<b>CGS</b>	718	0.613333	<b>MFI</b>	960	0.825819	<b>VNI</b>	971	0.896603
<b>CHA</b>	677	0.566667	<b>MFS</b>	995	0.866320	<b>VSV</b>	948	0.818000
<b>-CHI</b>	913	0.833441	<b>MON</b>	1033	0.885802	<b>WEI</b>	773	0.655864
<b>-CLO</b>	456	0.443290	<b>MOR</b>	938	0.802340	<b>-WEN</b>	755	0.640693
<b>CON</b>	784	0.670229	<b>MSO</b>	822	0.694444	<b>YUG</b>	935	0.797884
<b>CZP</b>	885	0.800000	<b>MTF</b>	1083	0.942112	<b>ZON</b>	798	0.679089
<b>DDR</b>	968	0.850801	<b>MTN</b>	1049	0.927590			
<b>DAC</b>	977	0.841190	<b>MTT</b>	1015	0.883181			



## APPENDIX D

### Origin of 3-Letter Codes and Material Names

<b>Code</b>	<b>Abbreviated Name</b>	<b>Full Name</b>
<b>ABW</b>		Li-A (Barrer and White)
<b>ACO</b>	<b>ACP-1 (one)</b>	Aluminium Cobalt Phosphate - one
<b>AEI</b>	<b>AlPO<sub>4</sub>-18 (eighteen)</b>	Aluminophosphate-eighteen
<b>AEL</b>	<b>AlPO<sub>4</sub>-11 (eleven)</b>	Aluminophosphate-eleven
<b>AEN</b>	<b>AlPO-EN3</b>	Aluminophosphate ethylenediamine ( <b>en</b> ) - three
<b>AET</b>	<b>AlPO<sub>4</sub>-8 (eight)</b>	Aluminophosphate-eight
<b>AFI</b>	<b>AlPO<sub>4</sub>-5 (five)</b>	Aluminophosphate-five
<b>AFN</b>	<b>AlPO<sub>4</sub>-14 (forteen)</b>	Aluminophosphate-forteen
<b>AFO</b>	<b>AlPO<sub>4</sub>-41 (forty-one)</b>	Aluminophosphate-forty-one
<b>AFR</b>	<b>SAPO-40 (forty)</b>	Silico-Aluminophosphate-forty
<b>AFS</b>	<b>MAPSO-46 (forty-six)</b>	<b>MgAl(P, Si)O<sub>4</sub>-46</b>
<b>AFT</b>	<b>AlPO<sub>4</sub>-52 (fifty-two)</b>	Silico-Aluminophosphate-fifty-six
<b>AFX</b>	<b>SAPO-56 (fifty-six)</b>	
<b>AFY</b>	<b>CoAPO-50 (fifty)</b>	
<b>AHT</b>	<b>AlPO<sub>4</sub>-H2 (two)</b>	
<b>APC</b>	<b>AlPO<sub>4</sub>-C</b>	
<b>APD</b>	<b>AlPO<sub>4</sub>-D</b>	
<b>AST</b>	<b>AlPO<sub>4</sub>-16 (sixteen)</b>	
<b>ASV</b>	<b>ASU-7 (seven)</b>	Arizona State University - seven
<b>ATN</b>	<b>MAPO-39 (thirty-nine)</b>	<b>MgAlPO<sub>4</sub>- thirty-nine</b>
<b>ATO</b>	<b>AlPO<sub>4</sub>-31 (thirty-one)</b>	
<b>ATS</b>	<b>MAPO-36 (thirty-six)</b>	
<b>ATT</b>	<b>AlPO<sub>4</sub>-12 (twelve)-TAMU</b>	<b>AlPO<sub>4</sub>-12-Texas A&amp;M University</b>
<b>ATV</b>	<b>AlPO<sub>4</sub>-25 (twenty-five)</b>	
<b>AWO</b>	<b>AlPO<sub>4</sub>-21 (twenty-one)</b>	
<b>AWW</b>	<b>AlPO<sub>4</sub>-22 (twenty-two)</b>	
<b>*BEA</b>	<b>Zeolite Beta</b>	
<b>BPH</b>	<b>Beryllophosphate-H</b>	<b>Beryllophosphate-Harvey (or hexagonal)</b>
<b>CAS</b>	<b>Cesium Aluminsilicate</b>	
<b>CFI</b>	<b>CIT-5 (five)</b>	California Institute of Technology - five
<b>CGF</b>	<b>CoGaPO-5 (five)</b>	Cobalt-Gallium-Phosphate-five
<b>CGS</b>	<b>CoGaPO-6 (six)</b>	Cobalt-Gallium-Phosphate-six
<b>-CLO</b>	<b>Cloverite</b>	Four-leaved <b>clover</b> shaped pore opening
<b>CON</b>	<b>CIT-1 (one)</b>	California Institute of Technology - one
<b>CZP</b>		Chiral Zincophosphate
<b>DDR</b>	<b>Deca-dodecasil 3R</b>	<b>Deca- &amp; dodecahedra, 3 layers, rhombohedral</b>
<b>DFO</b>	<b>DAF-1 (one)</b>	Davy Faraday Research Laboratory - one
<b>DFT</b>	<b>DAF-2 (two)</b>	Davy Faraday Research Laboratory - two
<b>DOH</b>	<b>Dodecasil 1H</b>	<b>Dodecahedra, 1 layer, hexagonally stacked</b>
<b>DON</b>	<b>UTD-1 (one)</b>	University of Texas at Dallas-one
<b>EAB</b>		TMA-E (Aiello and Barrer)
<b>EMT</b>	<b>EMC-2 (two)</b>	Elf (or Ecole Supérieure) Mulhouse Chimie - two
<b>ESV</b>	<b>ERS-7 (seven)</b>	Eniricerche-molecular-sieve-seven
<b>EUO</b>	<b>EU-1 (one)</b>	Edinburgh University - one

<b>Code</b>	<b>Abbreviated Name</b>	<b>Full Name</b>
<b>GON</b>	<b>GUS-1 (one)</b>	Gifu University Molecular Sieve - one
<b>IFR</b>	<b>ITQ-4 (four)</b>	Instituto de Tecnologia Quimica Valencia - four
<b>ISV</b>	<b>ITQ-7 (seven)</b>	Instituto de Tecnologia Quimica Valencia - seven
<b>ITE</b>	<b>ITQ-3 (three)</b>	Instituto de Tecnologia Quimica Valencia - three
<b>JBW</b>		Na-J (Barrer and White)
<b>KFI</b>	<b>ZK-5 (five)</b>	Zeolite Kerr - five
<b>LOS</b>	<b>Losod</b>	Low sodium aluminosilicate
<b>LTA</b>	<b>Linde Type A</b>	Zeolite A (Linde Division, Union Carbide)
<b>LTL</b>	<b>Linde Type L</b>	Zeolite L (Linde Division, Union Carbide)
<b>LTN</b>	<b>Linde Type N</b>	Zeolite N (Linde Division, Union Carbide)
<b>MEI</b>	<b>ZSM-18 (eighteen)</b>	Zeolite Socony Mobil - eighteen
<b>MEL</b>	<b>ZSM-11 (eleven)</b>	Zeolite Socony Mobil - eleven
<b>MFI</b>	<b>ZSM-5 (five)</b>	Zeolite Socony Mobil - five
<b>MFS</b>	<b>ZSM-57 (fifty-seven)</b>	Zeolite Socony Mobil - fifty-seven
<b>MSO</b>	<b>MCM-61 (sixty-one)</b>	Mobil Composition of Matter - sixty-one
<b>MTF</b>	<b>MCM-35 (thirty-five)</b>	Mobil Composition of Matter - thirty-five
<b>MTN</b>	<b>ZSM-39 (thirty-nine)</b>	Zeolite Socony Mobil - thirty-nine
<b>MTT</b>	<b>ZSM-23 (twenty-three)</b>	Zeolite Socony Mobil - twenty-three
<b>MTW</b>	<b>ZSM-12 (twelve)</b>	Zeolite Socony Mobil - twelve
<b>MWW</b>	<b>MCM-22 (twenty-two)</b>	Mobil Composition of Matter-twenty-two
<b>NES</b>	<b>NU-87 (eighty-seven)</b>	New (ICI) - eighty-seven
<b>NON</b>	<b>Nonasil</b>	Nonahedra, all silica composition
<b>OSI</b>	<b>UiO-6 (six)</b>	University of Oslo-six
<b>OSO</b>	<b>OSB-1 (one)</b>	Universities of Oslo and Calif., Santa Barbara – one
<b>-RON</b>	<b>Roggianite</b>	
<b>RSN</b>	<b>RUB-17 (seventeen)</b>	Ruhr University Bochum - seventeen
<b>RTE</b>	<b>RUB-3 (three)</b>	Ruhr University Bochum - three
<b>RTH</b>	<b>RUB-13 (thirteen)</b>	Ruhr University Bochum - thirteen
<b>RUT</b>	<b>RUB-10 (ten)</b>	Ruhr University Bochum - ten
<b>SAO</b>	<b>STA-1 (one)</b>	University of Saint Andrews - one
<b>SAS</b>	<b>STA-6 (six)</b>	University of Saint Andrews - six
<b>SAT</b>	<b>STA-2 (two)</b>	University of Saint Andrews - two
<b>SAV</b>	<b>Mg-STA-7 (seven)</b>	University of Saint Andrews - seven
<b>SBE</b>	<b>UCSB-8 (eight)</b>	University of California, Santa Barbara - eight
<b>SBS</b>	<b>UCSB-6 (six)</b>	University of California, Santa Barbara - six
<b>SBT</b>	<b>UCSB-10 (ten)</b>	University of California, Santa Barbara - ten
<b>SFE</b>	<b>SSZ-48 (forty-eight)</b>	Standard Oil Synthetic Zeolite - fourty-eight
<b>SFF</b>	<b>SSZ-44 (forty-four)</b>	Standard Oil Synthetic Zeolite - fourty-four
<b>SGT</b>	<b>Sigma-2 (two)</b>	
<b>STF</b>	<b>SSZ-35 (thirty-five)</b>	Standard Oil Synthetic Zeolite - thirty-five
<b>STT</b>	<b>SSZ-23 (twenty-three)</b>	Standard Oil Synthetic Zeolite - twenty-three
<b>TON</b>	<b>Theta-1 (one)</b>	
<b>VET</b>	<b>VPI-8 (eight)</b>	Virginia Polytechnic Institute - eight
<b>VFI</b>	<b>VPI-5 (five)</b>	Virginia Polytechnic Institute - five
<b>VNI</b>	<b>VPI-9 (nine)</b>	Virginia Polytechnic Institute - nine
<b>VSV</b>	<b>VPI-7 (seven)</b>	Virginia Polytechnic Institute - seven
<b>ZON</b>	<b>ZAPO-M1 (one)</b>	(Zn,Al)PO <sub>4</sub> -Mulhouse - one

# ISOTYPIC MATERIAL INDEX

\* Type materials are marked with an asterisk

*ACP-1	<b>ACO</b>	Bor-C	<b>MFI</b>
ACP-3	<b>DFT</b>	Bor-D ( <b>MFI/MEL</b> intergrowth)	<b>MEL</b>
*Afghanite	<b>AFG</b>	Boralite C	<b>MFI</b>
Alpha	<b>LTA</b>	Boralite D	<b>MEL</b>
AlPO <sub>4</sub> -pollucite	<b>ANA</b>	*Brewsterite	<b>BRE</b>
*AlPO-5	<b>AFI</b>	Bystrite	<b>LOS</b>
*AlPO-8	<b>AET</b>	Ca-D	<b>ANA</b>
*AlPO-11	<b>AEL</b>	Ca-Q	<b>MOR</b>
*AlPO-12-TAMU	<b>ATT</b>	*Cancrinite	<b>CAN</b>
*AlPO-14	<b>AFN</b>	Cancrinite hydrate	<b>CAN</b>
*AlPO-16	<b>AST</b>	*Cesium Aluminosilicate	<b>CAS</b>
AlPO-17	<b>ERI</b>	CF-3	<b>NON</b>
*AlPO-18	<b>AEI</b>	CF-4	<b>MTN</b>
AlPO-20	<b>SOD</b>	CFSAPO-1A	<b>AEN</b>
*AlPO-21	<b>AWO</b>	*Chabazite	<b>CHA</b>
*AlPO-22	<b>AWW</b>	*Chiavennite	<b>-CHI</b>
AlPO-24	<b>ANA</b>	*Chiral Zincophosphate	<b>CZP</b>
*AlPO-25	<b>ATV</b>	*CIT-1	<b>CON</b>
*AlPO-31	<b>ATO</b>	CIT-4	<b>BRE</b>
AlPO-33	<b>ATT</b>	*CIT-5	<b>CFI</b>
AlPO-34	<b>CHA</b>	CIT-6	<b>*BEA</b>
AlPO-35	<b>LEV</b>	Clinoptilolite	<b>HEU</b>
AlPO-40	<b>AFR</b>	*Cloverite	<b>-CLO</b>
*AlPO-41	<b>AFO</b>	*Co-Ga-Phosphate-5	<b>CGF</b>
*AlPO-52	<b>AFT</b>	*Co-Ga-Phosphate-6	<b>CGS</b>
AlPO-53(A)	<b>AEN</b>	Co-STA-7	<b>SAV</b>
AlPO-53(B)	<b>AEN</b>	CoAPO-5	<b>AFI</b>
AlPO-54	<b>VFI</b>	CoAPO-44	<b>CHA</b>
*AlPO-C	<b>APC</b>	CoAPO-47	<b>CHA</b>
*AlPO-D	<b>APD</b>	*CoAPO-50	<b>AFY</b>
*AlPO-EN3	<b>AEN</b>	CoAPSO-40	<b>AFR</b>
*AlPO-H2	<b>AHT</b>	CoDAF-4	<b>LEV</b>
AlPO-H3	<b>APC</b>	CrAPO-5	<b>AFI</b>
Amicite	<b>GIS</b>	Cs berylliosilicate pollucite	<b>ANA</b>
Ammonioleucite	<b>ANA</b>	Cs,Fe silicate pollucite	<b>ANA</b>
AMS-1B	<b>MFI</b>	(Cs,K)-ZK-5	<b>KFI</b>
*Analcime	<b>ANA</b>	CSZ-1	<b>EMT/FAU</b>
*ASU-7	<b>ASV</b>	CZH-5	<b>MTW</b>
AZ-1	<b>MFI</b>	*Dachiardite	<b>DAC</b>
B-NU-1	<b>RUT</b>	*DAF-1	<b>DFO</b>
Ba-dominant brewsterite	<b>BRE</b>	*DAF-2	<b>DFT</b>
Barerite	<b>STI</b>	Danalite	<b>SOD</b>
Basic cancrinite	<b>CAN</b>	Davyne	<b>CAN</b>
Basic sodalite	<b>SOD</b>	*Deca-dodecasil 3R	<b>DDR</b>
Bellbergite	<b>EAB</b>	Dehydrated Ca,NH <sub>4</sub> -Heulandite	<b>HEU</b>
Beryllophosphate X	<b>FAU</b>	Dehydrated Linde Type A	<b>LTA</b>
*Beryllophosphate-H	<b>BPH</b>	Dehydrated Na-Chabazite	<b>CHA</b>
*Beta	<b>*BEA</b>	Dehydrated Na-X	<b>FAU</b>
Bicchulite	<b>SOD</b>	Dehydrated US-Y	<b>FAU</b>
*Bikitaite	<b>BIK</b>	Deuterated Rho	<b>RHO</b>
*Boggsite	<b>BOG</b>	*Dodecasil 1H	<b>DOH</b>

Dodecasil-3C	<b>MTN</b>	K-exchanged Heulandite	<b>HEU</b>
ECR-5	<b>CAN</b>	K-F	<b>EDI</b>
ECR-18	<b>PAU</b>	K-M	<b>MER</b>
ECR-30	<b>EMT/FAU</b>	K-rich gmelinite	<b>GME</b>
*Edingtonite	<b>EDI</b>	Kehoeite	<b>ANA</b>
*EMC-2	<b>EMT</b>	KZ-1	<b>MTT</b>
Encelite	<b>MFI</b>	KZ-2	<b>TON</b>
*Epistilbite	<b>EPI</b>	Large port mordenite	<b>MOR</b>
ERB-1	<b>MWW</b>	*Laumontite	<b>LAU</b>
*Eriomite	<b>ERI</b>	Leonhardite	<b>LAU</b>
*ERS-7	<b>ESV</b>	Leucite	<b>ANA</b>
*EU-1	<b>EUO</b>	*Levyne	<b>LEV</b>
EU-13	<b>MTT</b>	*Li-A (Barrer and White)	<b>ABW</b>
*Faujasite	<b>FAU</b>	Linde D	<b>CHA</b>
Fe-NU-1	<b>RUT</b>	Linde F	<b>EDI</b>
*Ferrierite	<b>FER</b>	Linde Q	<b>BPH</b>
FU-9	<b>FER</b>	Linde R	<b>CHA</b>
FZ-1	<b>MFI</b>	Linde T	<b>ERI/OFF</b>
G	<b>SOD</b>	*Linde Type A	<b>LTA</b>
Ga-NU-1	<b>RUT</b>	*Linde Type L	<b>LTL</b>
Gallosilicate ECR-10	<b>RHO</b>	*Linde Type N	<b>LTN</b>
Gallosilicate L	<b>LTL</b>	Linde W	<b>MER</b>
GaPO-14	<b>AFN</b>	Linde X	<b>FAU</b>
GaPO-34	<b>CHA</b>	Linde Y	<b>FAU</b>
GaPO-DAB-2	<b>ZON</b>	*Liottite	<b>LIO</b>
Garronite	<b>GIS</b>	*Losod	<b>LOS</b>
Gaultite	<b>VSV</b>	*Lovdarite	<b>LOV</b>
Genthelvite	<b>SOD</b>	Low-silica Na-P (MAP)	<b>GIS</b>
*Gismondine	<b>GIS</b>	LZ-105	<b>MFI</b>
*Gmelinite	<b>GME</b>	LZ-132	<b>LEV</b>
Gobbinsite	<b>GIS</b>	LZ-202	<b>MAZ</b>
Gonnardite	<b>NAT</b>	LZ-210	<b>FAU</b>
*Goosecreekite	<b>GOO</b>	LZ-211	<b>MOR</b>
Gottardiite	<b>NES</b>	LZ-212	<b>LTL</b>
*GUS-1	<b>GON</b>	LZ-214	<b>RHO</b>
H1	<b>VFI</b>	LZ-215	<b>LTA</b>
Harmotome	<b>PHI</b>	LZ-217	<b>OFF</b>
Hauyn	<b>SOD</b>	LZ-218	<b>CHA</b>
Helvin	<b>SOD</b>	LZ-219	<b>HEU</b>
*Heulandite	<b>HEU</b>	LZ-220	<b>ERI</b>
High natrolite	<b>NAT</b>	MAP	<b>GIS</b>
High-silica Na-P	<b>GIS</b>	*MAPO-36	<b>ATS</b>
Holdstite	<b>MTN</b>	*MAPO-39	<b>ATN</b>
Hsianghualite	<b>ANA</b>	MAPO-43	<b>GIS</b>
Hydroxo sodalite	<b>SOD</b>	*MAPSO-46	<b>AFS</b>
ISI-1	<b>TON</b>	Maricopaita	<b>MOR</b>
ISI-4	<b>MTT</b>	*Mazzite	<b>MAZ</b>
ISI-6	<b>FER</b>	MCM-9	<b>VFI</b>
ITQ-1	<b>MWW</b>	*MCM-22	<b>MWW</b>
*ITQ-3	<b>ITE</b>	*MCM-35	<b>MTF</b>
*ITQ-4	<b>IFR</b>	MCM-37	<b>AET</b>
*ITQ-7	<b>ISV</b>	MCM-58	<b>IFR</b>
ITQ-9	<b>STF</b>	*MCM-61	<b>MSO</b>
JDF-2	<b>AEN</b>	MeAPO-47	<b>CHA</b>
(K,Ba)-G,L	<b>LTL</b>	MeAPSO-47	<b>CHA</b>
K-Chabazite, Iran	<b>CHA</b>	*Melanophlogite	<b>MEP</b>

*Merlinoite	<b>MER</b>	SAPO-34	<b>CHA</b>
Mesolite	<b>NAT</b>	SAPO-35	<b>LEV</b>
Metanatrolite	<b>NAT</b>	SAPO-37	<b>FAU</b>
*Mg-STA-7	<b>SAV</b>	*SAPO-40	<b>AFR</b>
MgAPO-50	<b>AFY</b>	SAPO-42	<b>LTA</b>
Microsommitte	<b>CAN</b>	SAPO-43	<b>GIS</b>
MnAPO-11	<b>AEL</b>	SAPO-47	<b>CHA</b>
Monoclinic ferrierite	<b>FER</b>	*SAPO-56	<b>AFX</b>
Monoclinic H-ZSM-5	<b>MFI</b>	Scolecite	<b>NAT</b>
*Montesommaite	<b>MON</b>	Si-CHA	<b>CHA</b>
*Mordenite	<b>MOR</b>	Sigma-1	<b>DDR</b>
MSC-1	<b>AEN</b>	*Sigma-2	<b>SGT</b>
Mutinaite	<b>MFI</b>	Silica sodalite	<b>SOD</b>
N-A	<b>LTA</b>	Silicalite	<b>MFI</b>
Na-B	<b>ANA</b>	Silicalite 2	<b>MEL</b>
Na-D	<b>MOR</b>	Siliceous Na-Y	<b>FAU</b>
Na-P1	<b>GIS</b>	*Sodalite	<b>SOD</b>
Na-P2	<b>GIS</b>	Sr-D	<b>FER</b>
Na-V	<b>THO</b>	Sr-Q	<b>YUG</b>
*Na-J (Barrer and White)	<b>JBW</b>	SSZ-16	<b>AFX</b>
*Natrolite	<b>NAT</b>	*SSZ-23	<b>STT</b>
NaZ-21	<b>LTN</b>	SSZ-24	<b>AFI</b>
Nepheline hydrate	<b>JBW</b>	SSZ-25	<b>MWW</b>
*Nonasil	<b>NON</b>	SSZ-26	<b>CON</b>
Nosean	<b>SOD</b>	SSZ-33	<b>CON</b>
NU-1	<b>RUT</b>	*SSZ-35	<b>STF</b>
NU-3	<b>LEV</b>	SSZ-42	<b>IFR</b>
NU-4	<b>MFI</b>	*SSZ-44	<b>SFF</b>
NU-5	<b>MFI</b>	SSZ-46	<b>MEL</b>
NU-10	<b>TON</b>	*SSZ-48	<b>SFE</b>
NU-13	<b>MTW</b>	*STA-1	<b>SAO</b>
NU-23	<b>FER</b>	*STA-2	<b>SAT</b>
*NU-87	<b>NES</b>	STA-5	<b>BPH</b>
Octadecasil	<b>AST</b>	*STA-6	<b>SAS</b>
*Offretite	<b>OFF</b>	Stellerite	<b>STI</b>
Omega	<b>MAZ</b>	*Stilbite	<b>STI</b>
*OSB-1	<b>OSO</b>	Svetlozarite (disordered variant)	<b>DAC</b>
P	<b>KFI</b>	Synthetic amicite	<b>GIS</b>
Pahasapaita	<b>RHO</b>	Synthetic analcime	<b>ANA</b>
*Partheite	<b>-PAR</b>	Synthetic  Na-[Al-Si-O]-JBW	<b>JBW</b>
*Paulingite	<b>PAU</b>	Synthetic barrerite	<b>STI</b>
Periallite	<b>LTL</b>	Synthetic brewsterite	<b>BRE</b>
Phi	<b>CHA</b>	Synthetic Ca-garrisonite	<b>GIS</b>
*Phillipsite	<b>PHI</b>	Synthetic cancrinitite	<b>CAN</b>
Pollucite	<b>ANA</b>	Synthetic edingtonite	<b>EDI</b>
PSH-3	<b>MWW</b>	Synthetic epistilbite	<b>EPI</b>
Q	<b>KFI</b>	Synthetic fault-free gmelinite	<b>GME</b>
*Rho	<b>RHO</b>	Synthetic garrisonite	<b>GIS</b>
*Roggianite	<b>-RON</b>	Synthetic gobbinosite	<b>GIS</b>
*RUB-3	<b>RTE</b>	Synthetic gonnardite	<b>NAT</b>
*RUB-10	<b>RUT</b>	Synthetic hsinghualite	<b>ANA</b>
*RUB-13	<b>RTH</b>	Synthetic laumontite	<b>LAU</b>
*RUB-17	<b>RSN</b>	Synthetic lovdarite	<b>LOV</b>
SAPO-5	<b>AFI</b>	Synthetic melanophlogite	<b>MEP</b>
SAPO-11	<b>AEL</b>	Synthetic merlinoite	<b>MER</b>
SAPO-31	<b>ATO</b>	Synthetic mesolite	<b>NAT</b>

Synthetic natrolite	NAT	UiO-20	DFT
Synthetic offretite	OFF	USC-4	MFI
Synthetic scolecite	NAT	USI-108	MFI
Synthetic stellerite	STI	UTD-1	DON
Synthetic stilbite	STI	*UTD-1F	DON
Synthetic thomsonite	THO	UTM-1	MTF
Synthetic wairakite	ANA	Vishnevite	CAN
*Terranovaite	TER	*VPI-5	VFI
Tetragonal edingtonite	EDI	*VPI-7	VSV
*Theta-1	TON	*VPI-8	VET
Theta-3	MTW	*VPI-9	VNI
*Thomsonite	THO	VS-12	MTW
Tiptopite	CAN	VSV-7#	VSV
TMA sodalite	SOD	Wairakite	ANA
*TMA-E	EAB	*Weinebeneite	WEI
TMA-gismondine	GIS	Wellsite	PHI
TMA-O	OFF	*Wenkite	-WEN
TNU-1	CGS	Willhendersonite	CHA
TPAF AlPO-5	AFI	*Yugawaralite	YUG
TPZ-3	EUO	*ZAPO-M1	ZON
TPZ-12	MTW	ZBH	MFI
Triclinic bikitaite	BIK	Zeolite N	EDI
TS-1	MFI	Zeolite W	MER
TS-2	MEL	Zincophosphate X	FAU
*Tschörtnerite	TSC	ZK-4	LTA
Tschernichite	*BEA	*ZK-5	KFI
TsG-1	CGS	ZK-14	CHA
TSZ	MFI	ZK-19	PHI
TSZ-III	MFI	ZK-20	LEV
Tugtupite	SOD	ZK-21	LTA
TZ-01	MFI	ZK-22	LTA
UCSB-3GaGe	DFT	ZKQ-1B	MFI
UCSB-3ZnAs	DFT	ZMQ-TB	MFI
UCSB-6Co	SBS	Zn-STA-7	SAV
*UCSB-6GaCo	SBS	ZnAPSO-40	AFR
UCSB-6GaMg	SBS	ZSM-3	EMT/FAU
UCSB-6GaZn	SBS	ZSM-4	MAZ
UCSB-6Mg	SBS	*ZSM-5	MFI
UCSB-6Mn	SBS	*ZSM-11	MEL
UCSB-6Zn	SBS	*ZSM-12	MTW
*UCSB-8Co	SBE	*ZSM-18	MEI
UCSB-8Mg	SBE	ZSM-20	EMT/FAU
UCSB-8Mn	SBE	ZSM-22	TON
UCSB-8Zn	SBE	*ZSM-23	MTT
UCSB-10Co	SBT	ZSM-35	FER
*UCSB-10GaZn	SBT	*ZSM-39	MTN
UCSB-10Mg	SBT	ZSM-50	EUO
UCSB-10Zn	SBT	ZSM-51	NON
*UiO-6	OSI	*ZSM-57	MFS
UiO-7	ZON	ZSM-58	DDR
UiO-12-as	AEN	ZYT-6	CHA
UiO-12-500	AEN		