
MINERALS
AND MINERAL ASSEMBLAGES

Ilmenite Group Minerals in the Russia's Oldest Diamondiferous Kimberlites of Kimozero, Karelia

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Abstract—The paper discusses the morphology and compositional variations of ilmenite group minerals from kimberlites of two phases at the Kimozero locality, the oldest in Russia. Phenocrysts of Mn-rich picroilmenite and Fe-rich geikielite in kimberlites of both phases are similar in morphology and composition. Ilmenite from cement in the second-phase kimberlites enriched in Mg and rimming small regularly shaped chrome spinel phenocrysts is not present in the first-phase kimberlites. Ilmenite, manganilmenite, and Fe-bearing pyrophanite (22–24 wt % MnO) abundant in the cement of the second-phase kimberlites are twice as rich in Nb and substantially richer in Mn than ilmenite up to manganilmenite from the cement of the first-phase kimberlites. Ilmenite and manganilmenite of the first-phase kimberlites is enriched in Zn (up to 1.5 wt % ZnO). Ilmenite from the second-phase kimberlites contains up to 3 wt % Cr₂O₃. In Nb concentration, kimberlitic rocks of the Kimozero are similar to those found in other parts of the world (up to 3.5 wt % Nb₂O₅). Significant Mn-enrichment of the ilmenite group minerals is a common feature of Kimozero kimberlitic rocks. It is suggested that kimberlites in which all ilmenite group minerals—from megacrysts and phenocrysts to small segregations in the cement—are enriched in Me, formed with the participation of carbonatite melts with increased alkalinity.

Keywords: kimberlite, ilmenite group minerals, picroilmenite, manganilmenite, manganese-bearing, Kimozero, Karelia

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INTRODUCTION

Kimberlites are the most important source of diamonds and one of the most interesting petrological objects. Kimberlites are produced from alkali ultramafic carbonate–silicate melts with variable proportions of silicate and carbonate constituents. The carbonate constituent is consistent with calcite carbonatite; much less frequently, with dolomite or calcite carbonatites enriched in Mn; and very rarely, with alkali-rich natrocarbonatite (Dowson, 1983; Nikishov, 1984; Gaspar and Wylley, 1984; Jones and Wylley, 1985; Mitchell, 1986; Jaques et al., 1986; Milashev, 2015).

Titanium easily dissolves in alkali melts. Therefore, ilmenite group minerals (ilmenite FeTiO₃, hematite FeFeO₃, geikielite MgTiO₃, pyrophanite MnTiO₃, esclaite CrCrO₃, karelianite VVO₃, ecandrewsite ZnTiO₃, and corundum AlAlO₃) are characteristic constituents of kimberlites. At a high temperature, they form ilmenite–hematite, ilmenite–geikielite, and ilmenite–pyrophanite solid solution series. Ilmenite enriched in Mg or Mn is picroilmenite or manganilmenite, respectively. Ilmenite enriched in

Fe³⁺ is ferri-ilmenite, hemo-ilmenite, or ilmenohe-matite if the hematite endmember predominates.

Several types of the ilmenite group minerals are present in kimberlites: (1) picroilmenite lamellae in xenoliths of deep-seated ultramafic rocks and their disintegration products; (2) subgraphic structures with clinopyroxene or orthopyroxene in deep-seated inclusions; (3) intergrowths with phlogopite, alkali amphibole, rutile, and diopside; (4) large oval crystals (megacrysts, autoliths, and phenocrysts of ilmenite, ferri-geikielite, and rare manganilmenite) commonly enriched in Cr and frequently with chrome spinel exsolution lamellae; and (5) small crystals of ilmenite, manganilmenite, and rare pyrophanite in kimberlite cement (Frantsesson, 1968; Danchin and O'Rey, 1972; Ilupin et al., 1974; Blagulkina et al., 1975; Sobolev et al., 1976; Garanin et al., 1978; Wyatt, 1979; Boctor and Boyd, 1981; Agel et al., 1982; Timofeev et al., 1984; Kostrovitsky, 1986; Genshaft, Ilupin, 1987; Bagdasarov and Ilupin, 1988; Voitkovsky et al., 1991; Edwards et al., 1992; Hood and McCandell, 2004; Kostrovitsky et al., 2004a, 2004b; Wyatt et al., 2004; Malkov and Filippov, 2005). Ilmenite crystals of all types are frequently rimmed by reaction perovskite.

Perovskite pseudomorphs after macro- and micro-crystals of ilmenite are frequent. The most recent manganilmenite and pyrophanite occasionally rim perovskite crystals (Pasteris, 1980; Tompkins and Haggerty, 1985; Chakhmouradian and Mitchell, 1999) or occur as pseudomorphs after perovskite (Malkov and Filippov, 2005). The ilmenite group minerals in kimberlites contain appreciable Nb, usually less than 5 wt % Nb₂O₅, but occasionally up to 12 wt % (Chakhmouradian and Mitchell, 1999).

The Kimozero diamondiferous kimberlites were discovered in 1992 (Ushkov, 2001). Their composition was studied to varying degrees. Ushkov (2001) and Ustinov et al. (2009) reported the first data on the compositions of chrome spinels and ilmenite group minerals. Taking into account the enormous lateral size of the Kimozero deposit (more than 2 km) and the wide variety of kimberlites within it, data supplementing the chemical composition and the evolution of its kimberlite typomorphic minerals, including the ilmenite group minerals, remain relevant.

MATERIALS AND METHODS

A collection of Kimozero kimberlitic rocks from outcrops and drill cores has been studied. Most back-scattered electron images and determinations of the chemical compositions of minerals were obtained with a Jeol JSM-6480 LV scanning electron microscope equipped with an EDS under standard conditions at the Laboratory of High Spatial Resolution Analytical Techniques, Department of Petrology, Moscow State University (analyst N.N. Korotaeva).

BRIEF GEOLOGICAL DESCRIPTION

The Kimozero occurrence of diamondiferous kimberlites is located in the southeastern part of the Baltic Shield of the Russian Platform within the Onega structural formation of the Karelia Craton, the basement of which is the Vedlozero block of the Neoproterozoic stabilization (Ushkov, 2001; Putintseva et al., 2009). Kimozero is one of the oldest kimberlite occurrences; its isotopic age ranges from 1.99 to 1.74 Ga (Makhotkin, 2003; Samsonov et al., 2009). Kimberlitic rocks of at least two intrusion phases make up a large extensive (near 2 km) flattened deposit and some steeply dipping pipelike bodies predominantly composed of kimberlite breccias and tufflike and massive kimberlites.

Kimberlitic rocks of the first phase with numerous olivine and phlogopite phenocrysts are appreciably more magnetic than those of the second phase. Phlogopite (occasionally rimmed by tetraferriphlogopite) is Ti-bearing and enriched in Cr as indicated by abundant titanite and Cr-bearing clinocllore as its alteration products. Magnesiochromite, alomochromite, ilmenite, and picroilmenite are phenocrysts in the first-phase kimberlites. The second-phase kimberlitic

rocks are enriched in bladelike magmatic calcite. Olivine, magnesiochromite, alumomagnesiochromite, ilmenite, and picroilmenite phenocrysts are present in the second-phase kimberlites. Morphology of tremolite–actinolite pseudomorphs riming large grains of serpentinized olivine suggests that olivine from the second-phase kimberlites was initially rimmed by monticellite. Olivine and phlogopite are abundant in the cement of the Kimozero kimberlites; ilmenite, titanomagnetite, chromite, and perovskite (as indicated by the morphology of the replacement products) are frequent; apatite, baddeleyite, and zircon are rare.

Kimberlitic rocks of Kimozero together with Lyudikovian country gabbrodolerite and terrigenous and volcanic rocks, are markedly tectonized and metamorphosed.

ILMENITE GROUP MINERALS OF THE FIRST-PHASE KIMBERLITIC ROCKS

Oval, markedly corroded phenocrysts up to 1.5 mm in size similar in morphology (Fig. 1) and composition to picroilmenite of classic kimberlite megacrysts are rare in the studied rocks. The composition of the Kimozero minerals (analyses 1, 2) is, wt %: 11.95 and 11.19 MgO, 0.25 and 0.29 NiO, 23.27 and 17.53 FeO, 0.75 and 4.31 MnO, 0.12 and 0.30 ZnO, 9.14 and 0.12 CaO, 50.94 and 47.25 TiO₂, 9.83 and 17.26 Fe₂O₃, 0.30 and 0.15 V₂O₅, 2.88 and 0.22 Cr₂O₃, 0.14 and 1.76 Nb₂O₅, total 100.57 and 100.30%; the FeO and Fe₂O₃ contents are calculated from stoichiometry. The end-member contents are, mol %: 41.1 and 39.1 MgTiO₃, 44.9 and 34.4 FeTiO₃, 1.5 and 8.6 MnTiO₃, 0.2 and 0.5 ZnTiO₃, 0.3 CaTiO₃, 0.5 NiTiO₃, 8.6 and 15.3 Fe₂O₃, 2.7 and 0.2 Cr₂O₃, 0.3 and 0.1 V₂O₅, 0.1 and 1.0 Nb₂O₅. Composition 1 is consistent with Cr-bearing picroilmenite; composition 2 corresponds to Mn–Fe-bearing geikielite enriched in Fe³⁺ and Nb.

Isolated unzoned ilmenite crystals up to 150 μm across are common in the cement of the first-phase kimberlites. They are divided into three groups by composition. Group 1 ilmenite crystals depleted in Mg and Mn (Table 1, analyses 3, 4) are the most abundant. This ilmenite contains up to 1 wt % MnO, approximately 6 wt % Fe₂O₃, up to 1 wt % Cr₂O₃, minor Nb, and trace Ca and Zn. Grains of group 2 ilmenite depleted in Mg and enriched in Fe³⁺ and Mn are common. This ilmenite contains, wt %: 2–6 MnO, up to 13.5 Fe₂O₃, up to 1 Cr₂O₃, up to 0.8 ZnO, up to 0.4 V₂O₅, and up to 0.3 Nb₂O₅ (Table 1, analyses 5–9). Group 3 ilmenite is rare. This is manganilmenite depleted in Mg and enriched in Zn. It contains, wt %: 11 MnO, 2 Fe₂O₃, 1.5 ZnO, 0.3 V₂O₅, 0.2 Cr₂O₃ (Table 1, analyses 10). Thus, ilmenite from the first-phase kimberlitic rocks is variable in composition, from Cr-bearing picroilmenite and Mn–Fe-bearing geikielite to common ilmenite and manganilmenite. Most of the ilmenite group minerals, from earliest to latest, have

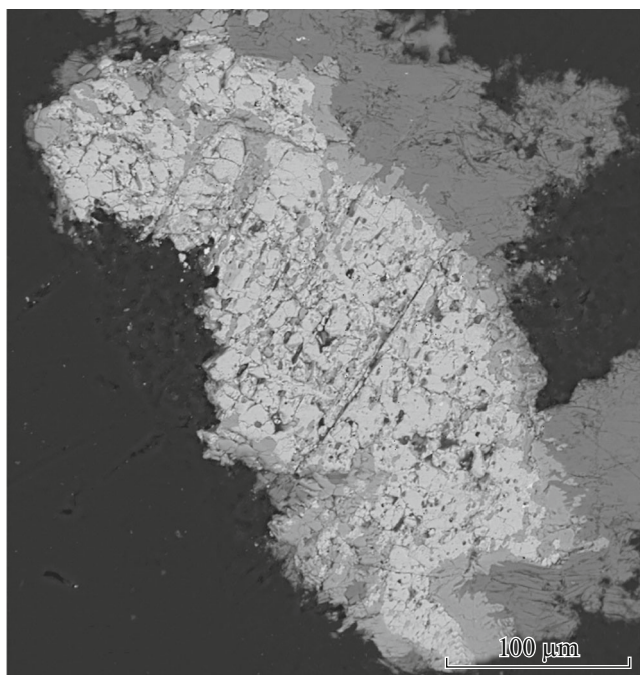


Fig. 1. Backscattered electron image of fragments of Cr-bearing picroilmenite (analysis 1) from first-phase kimberlites of Kimozero. Crystal is markedly corroded and rimmed by replacing titanite (gray).

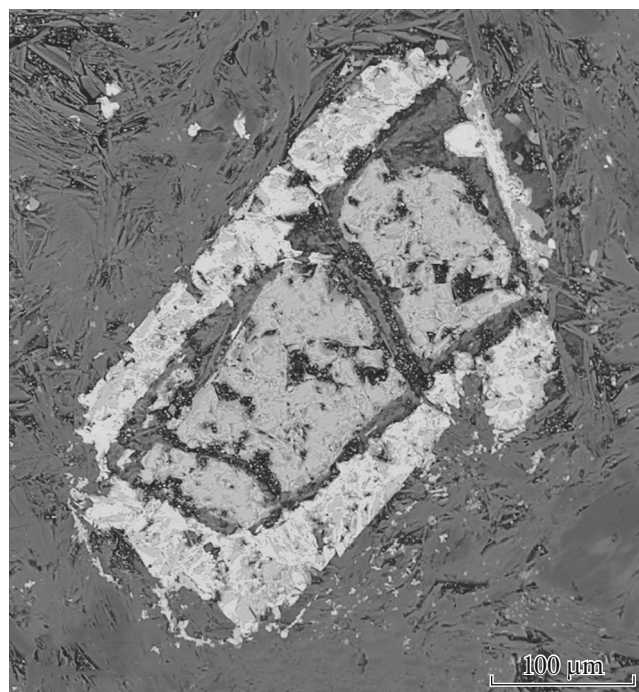


Fig. 2. Backscattered electron image of Mg-bearing bearing ilmenite (analysis 15) rimming small chrome spinel phenocryst (gray) in matrix of second-phase kimberlite at Kimozero. Zircon crystal (white) is on inner side of rim. Ilmenite is markedly corroded.

an elevated Mn concentration. The Fe^{3+} content is highest in the earliest high-temperature Fe-bearing geikielite. The Cr and Nb concentrations are not correlated and are relatively highly variable. Elevated Zn is characteristic of manganilmenite. Taking into account the compositions not given in Table 1, ilmenite from the first-phase kimberlites contains on average, wt %: 2.4 MnO, 0.29 Nb_2O_5 , 0.45 Cr_2O_3 , 0.25 V_2O_5 ($n = 17$).

ILMENITE GROUP MINERALS FROM THE SECOND-PHASE KIMBERLITIC ROCKS

Oval phenocrysts up to 2 mm in size, also close in composition to picroilmenite from classic kimberlite megacrysts, are present in these rocks. The composition of the Kimozero mineral (analysis 11) is, wt %: 11.75 MgO, 0.11 NiO, 12.83 FeO, 2.37 MnO, 0.51 ZnO, 0.13 CaO, 41.45 TiO_2 , 29.69 Fe_2O_3 , 0.16 V_2O_5 , 1.14 Cr_2O_3 , 0.28 Nb_2O_5 , total 100.62; the FeO and Fe_2O_3 contents are calculated from stoichiometry. The endmember contents are, mol %: 41.5 MgTiO_3 , 25.0 FeTiO_3 , 4.7 MnTiO_3 , 0.9 ZnTiO_3 , 0.3 CaTiO_3 , 0.2 NiTiO_3 , 26.1 Fe_2O_3 , 1.1 Cr_2O_3 , 0.1 V_2O_5 , 0.1 Nb_2O_5 . This composition is consistent with Mn–Fe-bearing geikielite in the proportions of the major elements.

Ilmenite enriched in Mg and rimming small regularly shaped chrome spinel phenocrysts (Figs. 2, 3, 4) is common. The thickness of the ilmenite rims ranges

from 20 to 50 μm . Zircon occasionally overgrows their inner side (Fig. 2). This ilmenite contains up to, wt %: 6.5 MgO, 14 Fe_2O_3 , 4 MnO, 3.5 Nb_2O_5 , 1 Cr_2O_3 , and 0.7 V_2O_5 ; Zn was not detected (Table 2). Ilmenite enriched in Fe^{3+} was exsolved and converted to the ilmenite matrix with numerous hematite lamellae (Fig. 2).

Isolated grains of the ilmenite group minerals up to 250 μm across are common in the cement of the second-phase kimberlites (Figs. 5, 6). They are divided into four chemical groups. Unzoned ilmenite crystals of group 1 depleted in Mg and relatively enriched Mn (Fig. 5; Table 3) are the most abundant. This ilmenite contains, wt %: 2–9 MnO, up to 5 Fe_2O_3 , up to 2.5 Cr_2O_3 , up to 1.5 Nb_2O_5 , up to 1 CaO, up to 0.7 ZnO, and up to 0.4 V_2O_5 . Zoned crystals of group 2 ilmenite depleted in Mg and relatively enriched in Mn are observed less frequently. The composition of two such crystals is given in Table 4. Substitution of Fe^{2+} for Mn^{2+} is clearly demonstrated in this ilmenite type: the cores and rims of crystals are enriched in Fe and Mn, respectively. Group 2 ilmenite contains, wt %: 6–15 MnO, up to 5 Fe_2O_3 , up to 2 Nb_2O_5 , up to 1 CaO and ZnO, and up to 0.4 V_2O_5 ; Cr was not detected (Table 4). Unzoned crystals of group 3 ilmenite depleted in Mg and Cr and significantly enriched in Mn are quite common. This manganilmenite contains, wt %: 11.5–

Table 1. Chemical composition (wt %) of isolated ilmenite crystals from cement of first-phase kimberlites

Component	Analysis							
	3	4	5	6	7	8	9	10
MgO	0.19	0.26	1.36	0.23	0.24	0.22	3.05	0.59
NiO	—	—	—	—	—	—	—	—
FeO	42.97	42.30	35.63	40.34	37.55	36.49	29.33	31.92
MnO	0.44	0.82	1.91	2.53	3.36	3.65	5.93	10.79
ZnO	—	—	0.23	0.22	0.71	0.78	—	1.45
CaO	0.05	—	0.10	—	—	—	0.13	—
TiO ₂	48.74	48.49	44.84	48.33	46.68	47.76	45.45	51.08
Fe ₂ O ₃	6.22	5.78	13.26	7.00	10.10	8.05	13.45	1.80
V ₂ O ₃	—	0.33	0.23	0.39	0.33	0.40	0.16	0.34
Cr ₂ O ₃	—	1.07	0.11	0.38	0.38	0.24	—	0.24
Nb ₂ O ₅	0.34	0.24	0.14	0.20	0.17	0.17	0.28	—
Total	98.95	99.29	97.81	100.62	99.52	97.76	98.78	98.21
Atoms per formula unit calculated on basis of two metal atoms								
Mg	0.007	0.010	0.051	0.009	0.009	0.008	0.115	0.022
Fe ²⁺	0.920	0.901	0.766	0.841	0.899	0.810	0.622	0.702
Mn	0.010	0.018	0.042	0.054	0.073	0.079	0.127	0.234
Zn	—	—	0.004	0.004	0.013	0.015	—	0.027
Ca	0.001	—	0.003	—	—	—	0.004	—
Ti	0.938	0.929	0.867	0.918	0.895	0.912	0.868	0.985
Fe ³⁺	0.120	0.113	0.257	0.152	0.104	0.162	0.257	0.018
V	—	0.007	0.005	0.002	0.007	0.008	0.003	0.007
Cr	—	0.022	0.002	0.008	0.008	0.006	—	0.005
Nb	0.004	0.003	0.002	0.002	0.002	0.002	0.003	—

Here and in Tables 2–6, FeO and Fe₂O₃ contents are calculated from stoichiometry; dash denotes element content below detection limit.

21.5 MnO, up to 4 (usually below 1) Fe₂O₃, up to 3 Nb₂O₅, up to 1 CaO and ZnO, and up to 0.5 V₂O₃ (Table 5). Rare Fe-bearing pyrophanite microscopically indistinguishable from above described species (Fig. 6) presents group 4. Pyrophanite contains, wt %: 22–24 MnO, up to 3 Fe₂O₃, up to 0.7 ZnO, and up to 0.5 Nb₂O₅ and V₂O₃ (Table 6).

Thus, ilmenite from the second-phase kimberlitic rocks at Kimozero is variable in morphology and composition, from Mn–Fe-bearing geikielite through ilmenite enriched in Mg and Mn to ilmenite, manganylmenite, and pyrophanite depleted in Mg and enriched Mn. The high Mn concentration is characteristic of the ilmenite group minerals from the earliest to the latest generation. The Fe³⁺ content is the highest in the early highest-temperature Fe-bearing geikielite. The Cr and Nb concentrations in ilmenite are not correlated and are relatively highly variable. Ilmenite with elevated Mg concentration is more frequently enriched in Cr, whereas ilmenite with elevated Mn is

enriched in Nb. Taking into account the compositions not given in Tables 2–6, ilmenite from the second-phase kimberlites contains on average, wt %: 10.2 MnO, 0.54 Nb₂O₅, 0.24 Cr₂O₃, 0.33 V₂O₃ ($n = 39$). Minor Ni and Zn are characteristic of Mg- and Mn-bearing ilmenite, respectively. Constant minor Ca in ilmenites is noteworthy.

COMPARISON OF ILMENITE GROUP MINERALS FROM THE FIRST- AND SECOND-PHASE KIMBERLITIC ROCKS

Early generation microilmenite and Fe-bearing geikielite in kimberlites of the both phases at Kimozero are similar in morphology and composition (Figs. 7, 8). Later ilmenite enriched in Mg and rimming regularly shaped chrome spinel crystals from the second-phase kimberlites was not found in the first-phase kimberlites. Later ilmenite, the most abundant in the cement of the second-phase kimberlites, is two times richer in Nb and substantially richer in Mn (Fig. 8b) than that



Fig. 3. Backscattered electron image of Mg-bearing bearing ilmenite (analysis 12) containing tiny chrome spinel octahedron and rimming small chrome spinel phenocryst (gray) in matrix of second-phase kimberlites. Zircon occurs as fine white crystals in matrix. Ilmenite is strongly corroded.

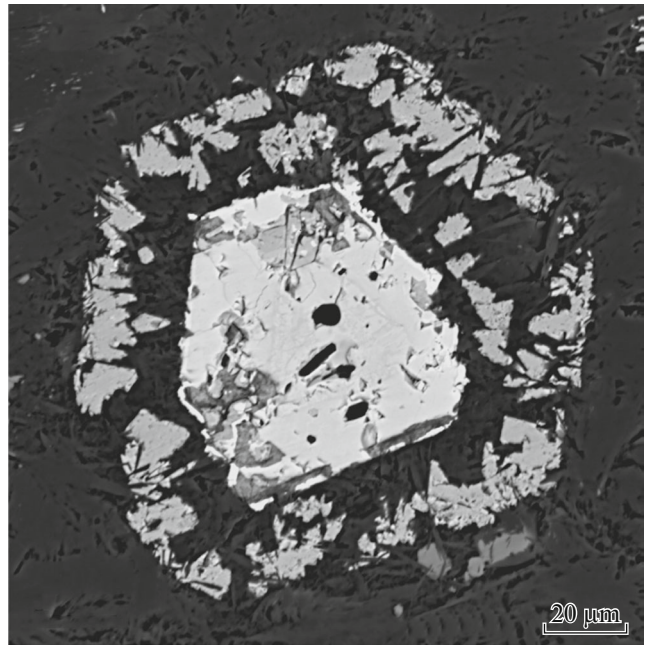


Fig. 4. Backscattered electron image of Mg-bearing bearing ilmenite enriched in Nb (analysis 13) rimming chrome spinel phenocryst (white) in matrix of second-phase kimberlites at Kimozero. Ilmenite is markedly corroded.

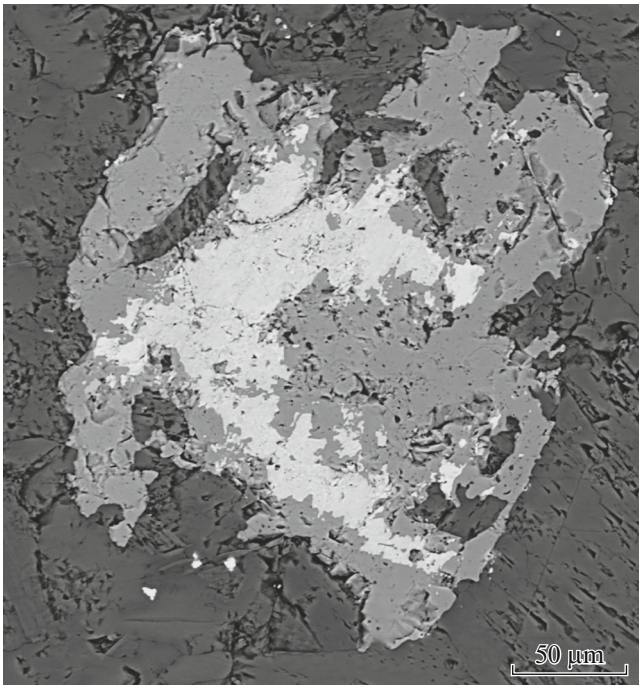


Fig. 5. Backscattered electron image of isolated Mn-bearing ilmenite crystal (analysis 23) in matrix of second-phase kimberlites at Kimozero. Ilmenite is strongly replaced by titanite (gray).

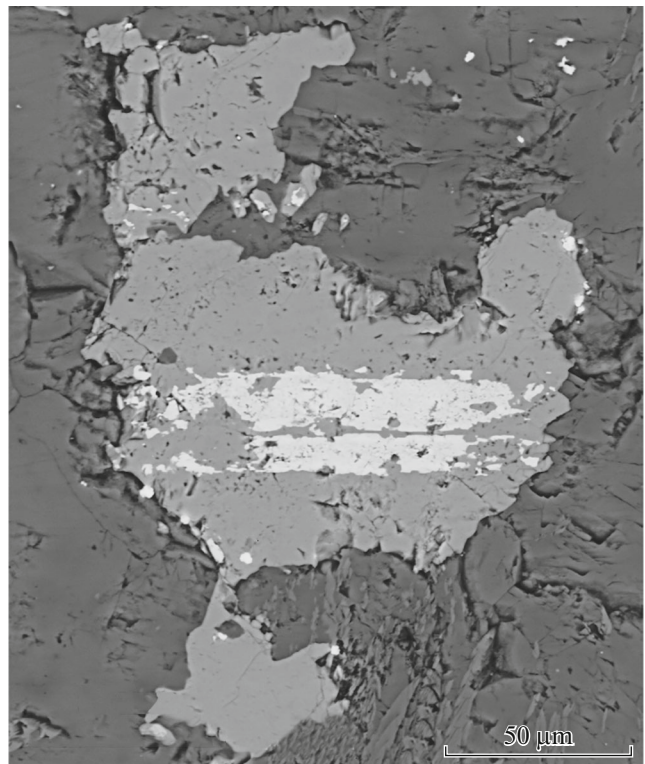


Fig. 6. Backscattered electron image of isolated Fe-bearing pyrophanite crystal (analysis 42) in matrix of second-phase kimberlite at Kimozero. Pyrophanite is massively replaced by titanite (gray).

Table 2. Chemical composition (wt %) of ilmenite enriched in Mg rimming chrome spinel crystals from second-phase kimberlites

Component	Analysis			
	12	13	14	15
MgO	6.30	4.16	4.15	3.67
NiO	0.13	—	—	0.12
FeO	30.36	30.17	34.34	36.38
MnO	3.46	2.61	4.11	0.08
ZnO	—	—	—	—
CaO	0.19	0.32	0.05	0.37
TiO ₂	50.57	50.63	51.07	45.15
Fe ₂ O ₃	7.90	2.55	6.15	13.75
V ₂ O ₃	0.45	0.69	0.31	0.49
Cr ₂ O ₃	0.42	—	0.03	1.13
Nb ₂ O ₅	—	3.25	—	1.03
Total	99.78	99.31	100.21	99.17
Atoms per formula unit calculated on basis of two metal atoms				
Mg	0.227	0.155	0.151	0.137
Ni	0.003	—	—	0.002
Fe ²⁺	0.614	0.734	0.703	0.698
Mn	0.071	0.055	0.085	0.002
Ca	0.005	0.007	0.001	0.010
Ti	0.920	0.951	0.940	0.849
Fe ³⁺	0.144	0.048	0.113	0.259
V	0.009	0.014	0.006	0.010
Cr	0.008	—	0.001	0.022
Nb	—	0.037	—	0.012

Table 3. Chemical composition (wt %) of isolated ilmenite crystals in cement of second-phase kimberlites

Component	Analysis							
	16	17	18	19	20	21	22	23
MgO	0.17	0.23	0.13	1.51	0.37	0.09	0.07	0.17
NiO	—	—	—	—	0.05	—	—	—
FeO	41.98	42.46	40.45	37.53	38.27	38.50	36.43	35.27
MnO	2.29	2.64	3.44	3.45	4.26	6.07	8.53	9.04
ZnO	—	—	—	0.36	0.67	—	0.18	—
CaO	0.91	0.04	0.93	0.15	0.10	0.31	0.44	1.01
TiO ₂	50.96	51.27	50.36	49.22	48.97	50.18	51.02	51.15
Fe ₂ O ₃	0.67	0.93	1.98	5.09	3.17	4.25	2.94	0.31
V ₂ O ₃	0.34	0.19	0.31	0.20	0.40	0.33	0.23	0.31
Cr ₂ O ₃	0.44	—	0.92	1.29	2.64	—	—	—
Nb ₂ O ₅	0.41	0.17	0.26	0.12	0.16	0.14	—	1.47
Total	98.07	98.73	98.78	98.92	98.96	99.87	99.84	98.63
Atoms per formula unit calculated on basis of two metal atoms								
Mg	0.006	0.009	0.005	0.057	0.014	0.003	0.003	0.006
Ni	—	—	—	—	0.001	—	—	—
Fe ²⁺	0.902	0.921	0.863	0.794	0.815	0.814	0.769	0.756
Mn	0.050	0.057	0.074	0.074	0.092	0.130	0.182	0.196
Zn	—	—	—	0.007	0.013	—	0.003	—
Ca	0.025	0.001	0.025	0.004	0.003	0.008	0.012	0.028
Ti	0.983	0.988	0.967	0.936	0.938	0.955	0.969	0.986
Fe ³⁺	0.013	0.018	0.038	0.097	0.061	0.081	0.056	0.006
V	0.007	0.004	0.006	0.004	0.008	0.007	0.005	0.004
Cr	0.009	—	0.019	0.025	0.053	—	—	—
Nb	0.005	0.002	0.003	0.002	0.002	0.002	—	0.017

Table 4. Chemical composition (wt %) of isolated zoned ilmenite crystals from cement of second-phase kimberlites

Component	Analysis							
	24 core	25 intermediate	26 intermediate	27 rim	28 core	29 intermediate	30 intermediate	31 rim
MgO	0.09	0.07	0.13	0.29	0.12	0.17	0.08	0.17
NiO	—	—	—	—	—	—	—	—
FeO	38.15	36.82	31.33	26.20	36.75	384.02	28.41	29.10
MnO	6.07	8.53	8.89	15.33	6.62	9.04	15.01	15.28
ZnO	0.31	0.18	0.48	0.98	0.47	0.39	0.62	0.79
CaO	0.31	0.44	0.89	0.90	0.93	1.01	0.65	0.73
TiO ₂	50.18	51.02	49.80	49.20	50.40	50.15	50.18	51.69
Fe ₂ O ₃	2.42	1.40	1.98	5.93	2.54	2.81	4.02	0.27
V ₂ O ₃	0.33	0.23	0.36	0.31	0.29	0.21	—	0.27
Cr ₂ O ₃	—	—	—	—	—	—	—	—
Nb ₂ O ₅	0.14	0.11	0.12	2.18	0.21	1.47	0.19	1.05
Total	98.08	98.80	98.70	100.43	99.92	99.27	99.16	99.32
Atoms per formula unit calculated on basis of two metal atoms								
Mg	0.003	0.003	0.005	0.011	0.005	0.006	0.003	0.006
Fe ²⁺	0.822	0.786	0.728	0.554	0.787	0.726	0.604	0.619
Mn	0.132	0.184	0.192	0.329	0.143	0.195	0.323	0.329
Zn	0.006	0.003	0.009	0.018	0.009	0.007	0.012	0.015
Ca	0.009	0.003	0.024	0.024	0.026	0.028	0.018	0.020
Ti	0.972	0.979	0.958	0.936	0.970	0.962	0.960	0.989
Fe ³⁺	0.047	0.027	0.075	0.096	0.049	0.054	0.077	0.005
V	0.007	0.005	0.007	0.006	0.006	0.004	—	0.005
Nb	0.002	0.001	0.001	0.025	0.002	0.017	0.002	0.012

Table 5. Chemical composition (wt %) of isolated ilmenite crystals enriched in Mn from cement of second-phase kimberlites

Component	Analysis							
	32	33	34	35	36	37	38	39
MgO	0.18	0.09	0.16	0.17	0.29	—	0.60	0.22
NiO	—	—	—	—	—	—	—	—
FeO	34.38	32.01	30.83	29.77	27.23	29.59	24.37	22.66
MnO	11.41	13.63	14.52	15.28	15.33	16.10	18.62	21.48
ZnO	—	—	—	—	—	0.29	0.32	0.51
CaO	0.19	0.36	0.25	0.73	0.89	0.28	0.56	0.07
TiO ₂	51.70	51.61	51.26	51.69	49.20	51.70	50.51	50.40
Fe ₂ O ₃	0.58	0.73	1.20	1.74	3.90	0.94	0.21	3.03
V ₂ O ₃	0.30	0.49	0.54	0.24	0.31	0.28	0.34	0.55
Cr ₂ O ₃	—	—	—	—	—	0.35	0.13	—
Nb ₂ O ₅	—	—	—	1.74	2.18	0.30	2.88	0.47
Total	98.84	98.92	98.76	100.67	99.33	99.83	98.54	99.09
Atoms per formula unit calculated on basis of two metal atoms								
Mg	0.007	0.003	0.006	0.006	0.011	—	0.026	0.008
Fe ²⁺	0.733	0.681	0.657	0.624	0.581	0.625	0.524	0.481
Mn	0.246	0.294	0.313	0.325	0.332	0.345	0.405	0.462
Zn	—	—	—	—	—	0.005	0.006	0.010
Ca	0.005	0.010	0.007	0.020	0.024	0.008	0.015	0.002
Ti	0.991	0.988	0.983	0.975	0.945	0.983	0.976	0.963
Fe ³⁺	0.011	0.014	0.023	0.033	0.075	0.018	0.004	0.058
V	0.006	0.010	0.011	0.005	0.006	0.006	0.007	0.011
Cr	—	—	—	—	—	0.007	0.003	—
Nb	—	—	—	0.012	0.025	0.003	0.033	0.005

Table 6. Chemical composition (wt %) of isolated Fe-bearing pyrophanite from cement of second-phase kimberlites

Component, wt %				Atoms per formula unit			
Analysis	40	41	42	Analysis	40	41	42
MgO	0.14	0.26	0.18	Mg	0.005	0.010	0.007
NiO	—	—	—	Ni	—	—	—
FeO	22.26	21.00	21.42	Fe ²⁺	0.480	0.447	0.452
MnO	21.99	22.57	23.93	Mn	0.461	0.487	0.511
ZnO	0.55	0.71	0.71	Zn	0.011	0.013	0.013
CaO	0.50	0.18	0.12	Ca	0.014	0.005	0.003
TiO ₂	51.04	50.25	51.97	Ti	0.991	0.962	0.986
Fe ₂ O ₃	0.31	3.07	1.05	Fe ³⁺	0.006	0.059	0.020
V ₂ O ₃	0.49	0.61	0.737	V	0.010	0.012	0.007
Cr ₂ O ₃	—	—	—	Cr	—	—	—
Nb ₂ O ₅	0.22	0.38	0.14	Nb	0.003	0.004	0.002
Total	97.50	99.03	99.85			2	

from the cement of the first-phase kimberlites (Fig. 7b). Ilmenite from the first-phase kimberlites is markedly richer in Zn. The positive correlation between Mg and Fe³⁺ concentrations and between contents of geikielite and hematite endmembers is characteristic of ilmenite group minerals from kimberlites of the both phases. This correlation is more pronounced in the second-phase kimberlites (Figs. 7a, 8a). Significant enrichment in Mn is a common feature of ilmenite group minerals from the Kimozero kimberlites (Figs. 7b, 8b).

COMPARISON OF ILMENITE GROUP MINERALS FROM KIMBERLITIC ROCKS AT KIMOZERO AND OTHER DIAMONDIFEROUS KIMBERLITES

Diamondiferous kimberlites contain picroilmenite and ferri-geikielite, which are enriched in Cr to vary-

ing degrees (Holmes, 1936; Nixon et al., 1963; Boctor and Boyd, 1980; Frantsesson et al., 1982; Haggerty and Tompkins, 1985; Mitchell, 1986; Genshaft and Ilupin, 1987; Bagdasarov and Ilupin, 1988; Edwards et al., 1992; Hood and McCandell, 2004; Kjarsgaard et al., 2004; Kostrovitsky et al., 2004; Masum et al., 2004; Wyatt et al., 2004; Kostrovitsky et al., 2006; Milashev, 2015). The positive correlation between Mg and Fe³⁺ concentrations and between contents of geikielite and hematite endmembers is characteristic of ilmenite group minerals from many diamondiferous kimberlites. Megacrysts of ilmenite group minerals frequently have appreciable Nb. The content of these ilmenite group minerals is low in kimberlites of both phases at Kimozero, which is consistent with the low diamond grade in these kimberlites (Ushkov, 2001; Lukyanova et al., 2006; Ustinov et al., 2009).

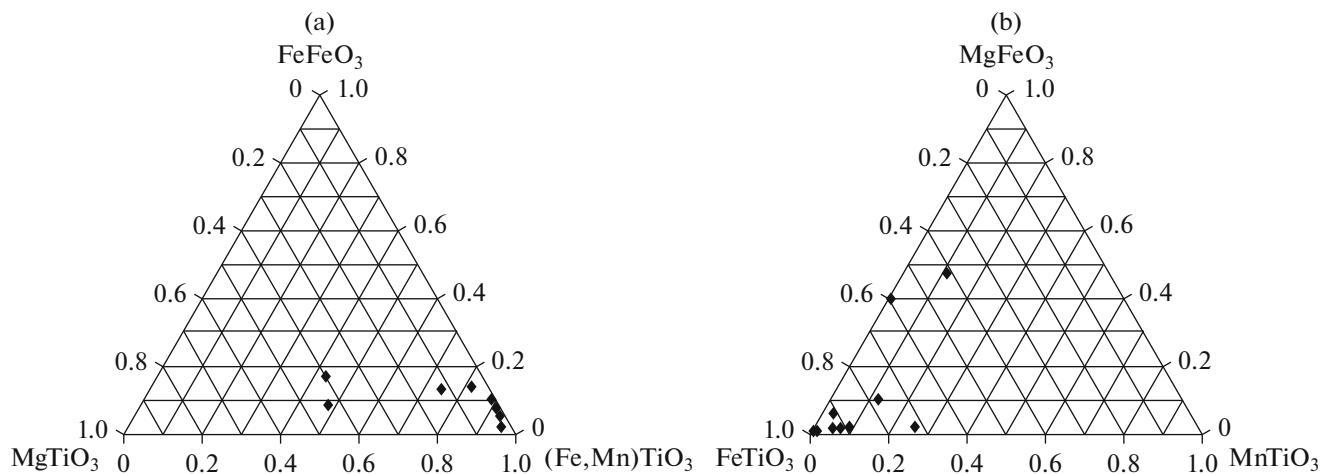


Fig. 7. Triangular plots in terms of (a) FeFeO₃–MgTiO₃–(Fe,Mn)TiO₃ and (b) MgTiO₃–FeTiO₃–MnTiO₃ illustrating composition of ilmenite group minerals from first-phase kimberlites at Kimozero.

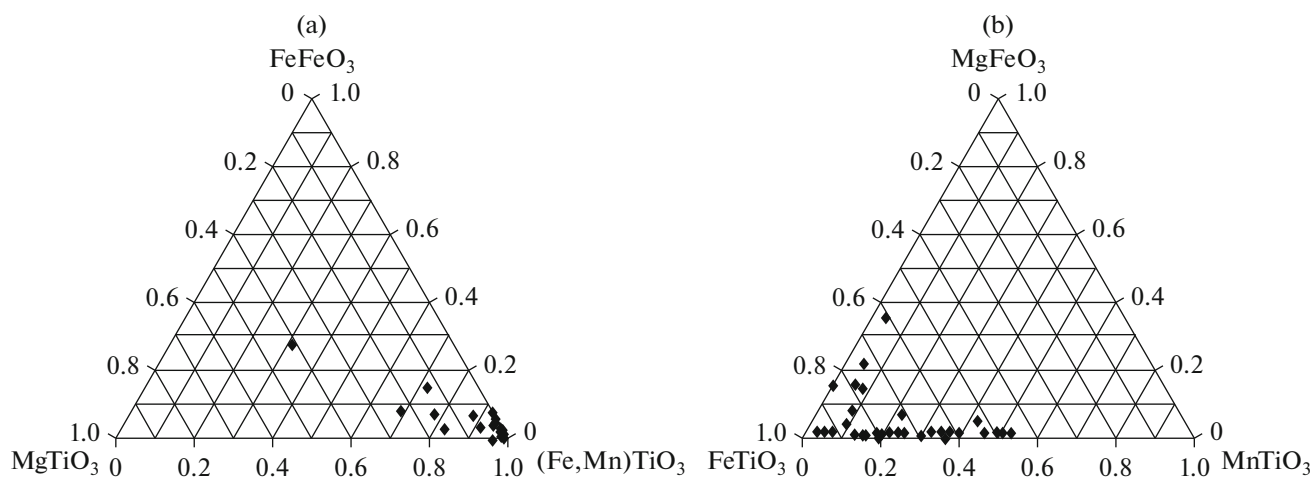


Fig. 8. Triangular plots in terms of (a) FeFeO_3 – MgTiO_3 – $(\text{Fe},\text{Mn})\text{TiO}_3$ and (b) MgTiO_3 – FeTiO_3 – MnTiO_3 illustrating composition of ilmenite group minerals from second-phase kimberlites.

The vast majority of picroilmenite and ferri-geikielite megacrysts from diamondiferous kimberlites is Mn-poor, 0.3–0.5 wt % MnO. The Kimozero picroilmenite and Fe-bearing geikielite are enriched in this element, 0.8–4.3 wt % MnO. Rare types of diamondiferous kimberlites containing megacrysts of ilmenite enriched in Mn and manganilmenite are known in Brazil (Kaminsky and Belousova, 2009). In this respect, the Kimozero kimberlites are no exception.

Microcrystalline ilmenite from the cement of the vast majority of kimberlites is Mn-poor: only rims of crystals are enriched in this element (Blagulkina et al., 1975; Garanin et al., 1978; Boctor and Boyd, 1981; Agel et al., 1982; Dowson, 1983; Nikoshov, 1984; Timofeev et al., 1984; Tompkins and Haggerty, 1985; Mitchell, 1986; Genshaft and Ilupin, 1988; Voitkovsky et al., 1991; Ilupin, 1997; Edwards et al., 1992; Hood and McCandell, 2004; Wyatt et al., 2004; Malkov and Filippov, 2005; Kostrovitsky et al., 2006; Kaminsky and Belousova, 2009). Rare types of kimberlites contain Mn-rich ilmenite, manganilmenite, and pyrophanite in the cement (Wyatt, 1979; Pasteris, 1980; Tompkins, Haggerty, 1985; Chakhmouradian and Mitchell, 1999; Malkov and Filippov, 2005; Kaminsky and Belousova, 2009). Such are the Kimozero kimberlites. These kimberlites contain a significant proportion of carbonatites. The same Mn-rich ilmenite group minerals are present in high-alkali carbonatites (Jakupiranga, Brazil; Western Australia) (Mitchell, 1978; Gaspar and Wylley, 1983; Jaques et al., 1986).

It can be suggested that kimberlites in which all types of ilmenite group minerals from megacrysts and phenocrysts to fine grains in the cement are enriched in Mn originated with the participation of high-alkali carbonatite melts, whereas standard kimberlites with ilmenite group minerals depleted in Mn were formed with the involvement of low-alkali carbonatite melts. High-alkali carbonatites enriched in Mn are known in

the eastern part of the Baltic Shield (Dudkin et al., 1984).

The Kimozero kimberlitic rocks are similar in Nb concentration to standard kimberlites.

ALTERATION

The Kimozero kimberlites, country Lyudikovian gabbrodolerite, and sedimentary and volcanic rocks are tectonized and predominantly metamorphosed under prehnite–pumpellyite facies conditions. The Kimozero metakimberlites are composed of antigorite and less frequent lizardite, actinolite–tremolite, carbonates, chlorites, and titanite; their proportions are highly variable (Lukyanova et al., 1986; Ustinov et al., 2009; Putintseva, 2015). Many of the ilmenite group minerals in metakimberlites are corroded (Figs. 1, 2, 3), partly replaced by titanite (Figs. 4, 5, 7), and less frequent ferro-pseudobrookite FeTi_2O_5 and fine-grained aggregates of rutile and hematite. Complete pseudomorphs of titanite after ilmenite are rather common.

CONCLUSIONS

Oval Mn-bearing picroilmenite and Fe-bearing geikielite phenocrysts in the Kimozero kimberlites of both phases are close in morphology and composition to megacrysts from standard diamondiferous kimberlites. The low content of these ilmenite group minerals in the Kimozero kimberlites is consistent with the low diamond grade in these rocks. Ilmenite enriched in Mg and rimming small regularly shaped chrome spinel phenocrysts in the cement of the second-phase kimberlites was not found in the first-phase kimberlites. Ilmenite, manganilmenite, and Fe-bearing pyrophanite (22–24 wt % MnO) common in the cement of the second-phase kimberlites are two times richer in Nb and substantially richer in Mn than ilmenite and

manganilmenite from the cement of the first-phase kimberlites. Ilmenite and manganilmenite from the first-phase kimberlites are richer in Zn (up to 1.5 wt % ZnO). Ilmenite from the second-phase kimberlites contains up to 3 wt % Cr₂O₃. The Nb content (up to 3.5 wt % Nb₂O₅) in the Kimozero kimberlitic rocks is similar in to that in standard kimberlites. Significant enrichment in Mn is a common feature of ilmenite group minerals of the Kimozero kimberlitic rocks. It can be suggested that kimberlites in which all types of ilmenite group minerals—from megacrysts and phenocrysts to fine grains in the cement—are enriched in Mn formed with the participation high-alkali carbonatite melts.

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